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 (54) Title: CONTAINER-PACKED COFFEE BEVERAGE AND PROCESS FOR PRODUCTION THEREOF

(57) Abrégé/Abstract:

A container-packed coffee beverage is provided, containing chlorogenic acids to a suitable extent, and furthermore having excellent taste, and with deterioration at heating has been suppressed. A container-packed coffee beverage containing the following constituents (A), (B) and (C): (A) Constituent  $\alpha$  (B) Constituent  $\beta$  (C) chlorogenic acids wherein the peak area ratio of constituent (A) appearing in a retention time of 2.0 to 2.3 minutes and constituent (B) appearing in a retention time of 2.7 to 3.2 minutes on a chromatogram obtained by constituent analysis of a coffee extract at a detection wavelength of 260nm using a high performance liquid chromatography apparatus is  $\{(A)/(B)\} \times 100 = 30$  to 100, and the content ratio of 5-caffeoylquinic acid within the constituent (C) is 28.0 to 35.0% by weight, and a preparation method thereof is provided.

### Abstract

A container-packed coffee beverage is provided, containing chlorogenic acids to a suitable extent, and furthermore having excellent taste, and with deterioration at heating has been suppressed. A container-packed coffee beverage containing the following constituents (A), (B) and (C):

(A) Constituent  $\alpha$

(B) Constituent  $\beta$

(C) chlorogenic acids

wherein the peak area ratio of constituent (A) appearing in a retention time of 2.0 to 2.3 minutes and constituent (B) appearing in a retention time of 2.7 to 3.2 minutes on a chromatogram obtained by constituent analysis of a coffee extract at a detection wavelength of 260nm using a high performance liquid chromatography apparatus is  $\{(A)/(B)\} \times 100 = 30$  to 100, and the content ratio of 5-caffeoylquinic acid within the constituent (C) is 28.0 to 35.0% by weight, and a preparation method thereof is provided.

## SPECIFICATION

CONTAINER-PACKED COFFEE BEVERAGE AND PROCESS FOR  
PRODUCTION THEREOF

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## Technical Field

The present invention relates to a container-packed coffee beverage containing chlorogenic acids to a suitable extent, and furthermore, having an excellent taste and whose deterioration at heating has been suppressed.

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## Background Art

Coffee is a tasty beverage people enjoy drinking world-wide. In addition to conventional extraction from roasted beans and instant coffee, readily ingestible container-packed coffee beverages, are widely popular. In addition, useful constituents contained in coffee have been drawing attention in recent years. Chlorogenic acids are one among the useful constituents contained in coffee beverage. SOD-like activity is known as an effect of chlorogenic acids, reported to have the activity of suppressing the generation of reactive oxygen species and lipid peroxide involved in aging and a variety of sicknesses, and to have antioxidant activity (refer for instance to Japanese Patent Application

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Laid-open No. 2001-136910 and Japanese Patent Application Laid-open No. 2002-080351). The approximate amount of natural polyphenols such as chlorogenic acids to be ingested per day is 1g per adult (refer for instance to Japanese Patent Application Laid-open No. 2005-312301), and simpler  
5 ingestion methods have been desired. As one such method, there is a method of increasing antioxidant activity by electrically processing the food source containing chlorogenic acids (refer for instance to Japanese Patent Application Laid-open No. 2003-102403).

In addition, among the chlorogenic acids, 3-caffeoylquinic acid,  
10 generally called neochlorogenic acid (refer for instance to Japanese Patent Application Laid-open No. 2004-033023), has been reported in particular to have an excellent antioxidant activity, and as simple ingestion methods thereof, there exist a method of separating chlorogenic acids by dextran gel to modify the mixing ratio (refer for instance to Japanese Patent Application Laid-open No.  
15 H09-143465) and a method of alkaline processing an aqueous solution containing chlorogenic acids (refer for instance to Japanese Patent Application Laid-open No. 2000-063827) .

Elsewhere, 3-caffeoylquinic acid is known to be generated by heating 5-caffeoylquinic acid in alkali (refer for instance to "Quantification of chlorogenic  
20 acids and caffeine in coffee-containing beverage by high performance liquid

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chromatography" Takashi Ueki, Hiroshi Honda and Shiro Sakurai, Agricultural and Forestry Products Inspection Institute Issue No. 10, March 1986.

5           In addition, chlorogenic acids are known to be in an unstable state when coffee beans are roasted, and that if heating history accumulates such as carrying out sterilization treatments on coffee extract, they decompose into small molecule organic acids, such as, acetic acid, formic acid, malic acid and citric acid, and as a result, sourness increases over time triggering deterioration  
10 (for instance Japanese Patent Application Laid-open No. H07-050993).

          Meanwhile, there is a problem that when a container-packed beverage is heated at a store or in an automatic vending machine, the quality thereof remarkably deteriorates due to heating, and in order to solve this, for instance, a method of masking by adding flavor, or a method of adding cyclodextrin and  
15 ascorbic acid or ascorbic acid salt have been proposed (refer for instance to Japanese Patent Application Laid-open No. 2004-073057).

          However, even if any of the above methods is selected, there are problems such as, caused by the addition of another constituent or carrying out special treatment subsequently to coffee beverage extraction; the original  
20 coffee taste is lost and the production cost increases due to the additional steps



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of production. In addition, since chlorogenic acids cause distracting flavors when contained in large quantities (refer for instance to Japanese Patent Application Laid-open No. 2005-333927, Japanese Patent Application Laid-open No. 2003-204755, Japanese Patent Application Laid-open No. 2003-204756), preserving the taste while leaving high concentrations of chlorogenic acids contained is difficult.

#### Disclosure of the Invention

An object of the present invention is to provide a container-packed coffee beverage containing chlorogenic acids to a suitable extent, and furthermore, having an excellent taste and whose deterioration at heating has been suppressed.

After earnest studies in order to resolve the above issues, the present inventors reached the ideas for the present invention. Specifically, the present inventors discovered that deterioration during heating of a container-packed coffee beverage could be suppressed by setting the content ratio of 5-caffeoylquinic acid among the chlorogenic acids within a specific range, and

that taste could be improved by setting the peak area ratio of constituents  $\alpha$  and  $\beta$  within a specific range.

More specifically, the present invention is as follows:

1. A container-packed coffee beverage containing the following constituents (A),

5 (B) and (C):

(A) Peak  $\alpha$  constituent

(B) Peak  $\beta$  constituent

(C) chlorogenic acids

10 wherein the peak area ratio of constituent (A) appearing in a retention time of 2.0 to 2.3 minutes and constituent (B) appearing in a retention time of 2.7 to 3.2 minutes on a chromatogram obtained by constituent analysis of a coffee extract at a detection wavelength of 260nm using a high performance liquid chromatography apparatus is  $\{(A)/(B)\} \times 100 = 30$  to 100, and the content ratio of 5-caffeoylquinic acid within the constituent (C) is 28.0 to 35.0% by weight.

15 2. In the above 1, a container-packed coffee beverage wherein the content in constituent (C) is 360ppm or greater and 840ppm or less in value converted into per 1.0% by weight of amount of solid coffee.

3. A method for preparing a container-packed coffee beverage, wherein adjustments are made in such a way that, in a coffee extract obtained by

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extracting coffee beans, the following constituents (A), (B) and (C) are contained:

(A) Peak  $\alpha$  constituent

(B) Peak  $\beta$  constituent

5 (C) chlorogenic acids

and the peak area ratio of constituent (A) appearing in a retention time of 2.0 to 2.3 minutes and constituent (B) appearing in a retention time of 2.7 to 3.2 minutes on a chromatogram obtained by constituent analysis of a coffee extract at a detection wavelength of 260nm using high performance liquid  
10 chromatography apparatus is  $\{(A)/(B)\} \times 100 = 30$  to 100, and the content ratio of 5-caffeoylquinic acid within the constituent (C) is 28.0 to 35.0% by weight.

4. A method for preparing a container-packed coffee beverage, wherein for the adjustment in the above 3, roasted coffee beans having an L value of 16 to 21.5 and roasted coffee beans having an L value of 22 to 33 are blended at a ratio of  
15 100:0 to 60:40.

5. A method for preparing a container-packed coffee beverage, wherein in the above 3 or 4, chlorogenic acids are added to the obtained coffee extract.



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6. A method for preparing a container-packed coffee beverage, wherein roasted coffee beans having an L value of 16 to 21.5 and roasted coffee beans having an L value of 22 to 33 are blended at a ratio of 95:5 to 70:30 to obtain a coffee extract, and;

5 in a coffee extract obtained by extracting coffee beans, the following constituents (A), (B) and (C) are contained;

(A) Peak  $\alpha$  constituent

(B) Peak  $\beta$  constituent

(C) chlorogenic acids

10 and the peak area ratio of constituent (A) having a retention time of 2.0 to 2.3 minutes and maximum absorption wavelength of 257.9nm and constituent (B) having a retention time of 2.7 to 3.2 minutes and maximum absorption wavelength of 263.8nm on a chromatogram obtained by constituent analysis of a coffee extract using high performance liquid chromatography apparatus is  $\{(A)/(B)\} \times 100 = 30$  to 100, and the content ratio of 5-caffeoylquinic acid  
15 within the constituent (C) is 28.0 to 35.0% by weight;

and the conditions for performing the high performance liquid chromatography being as follows:

Column: Inertsil ODS-2  $\varnothing$  3.0 x 250mm;

20 Mobile phase: Gradient elution method by

A Solution 3% acetonitrile solution containing 0.05M acetic acid;

B Solution 100% acetonitrile solution containing 0.05M acetic acid;

Volume injected: 10  $\mu$ L;

Flow rate: 0.43 mL/min;

25 Column temperature: 40 degree C;

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Detector: PDA UV 210nm to 400nm.

7. A container-packed coffee beverage, comprising a coffee extract blended from roasted coffee beans having an L value of 16 to 21.5 and roasted coffee beans having an L value of 22 to 33 at a ratio of 95:5 to 70:30 and  
5 containing the following constituents (A), (B) and (C);

(A) Peak  $\alpha$  constituent

(B) Peak  $\beta$  constituent

(C) chlorogenic acids

- and the peak area ratio of constituent (A) has a retention time of 2.0  
10 to 2.3 minutes and maximum absorption wavelength of 257.9nm and constituent (B) has a retention time of 2.7 to 3.2 minutes and maximum absorption wavelength of 263.8nm on a chromatogram obtained by constituent analysis of a coffee extract using high performance liquid chromatography apparatus is  
 $\{(A)/(B)\} \times 100 = 30$  to 100, and the content ratio of 5-caffeoylquinic acid within the  
15 constituent (C) is 28.0 to 35.0% by weight; and

the conditions for performing the high performance liquid chromatography are as follows:

Column: Inertsil ODS-2  $\varnothing$  3.0 x 250mm;

Mobile phase: Gradient elution method by

- 20 A Solution 3% acetonitrile solution containing 0.05M acetic acid;

B Solution 100% acetonitrile solution containing 0.05M acetic acid;

Volume injected: 10  $\mu$ L;

Flow rate: 0.43 mL/min;

Column temperature: 40 degree C;

- 25 Detector: PDA UV 210nm to 400nm.

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### Brief Description of the Drawings

Fig. 1 shows chromatograms of the peak areas of constituents  $\alpha$  and  $\beta$  test-produced coffee beverages at a detection wavelength of 260 nm.

Fig. 2 is a graph showing the results (generation of PBN-OH)  
5 obtained in Example 1.

Fig. 3 is a graph showing the results (generation of PBN-OH)  
obtained in Example 2.

Fig. 4 is a graph showing the results (generation of PBN-OH)  
obtained in Example 3.

10 Fig. 5 is a graph showing the results (amount of radical generated)  
obtained in Example 3.

### Description of Preferred Embodiments

Hereinafter, embodiments of the present invention will be described.  
The terms used in the specification are defined as follows.

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“(A) Peak  $\alpha$  constituent” is an unidentified constituent, whose amount increases as the coffee beans are the more roasted; concretely, a constituent corresponding to peak obtained during constituent analysis of coffee extract using high performance liquid chromatography (HPLC), appearing at a detection wavelength of 260nm and a retention time of 2.0 to 2.3 minute. The constituent can be measured by an HPLC apparatus. As HPLC apparatus, commercially available apparatus, for instance, “LaChrom<sup>\*</sup>” manufactured by Hitachi Ltd. and “LC-10A<sup>\*</sup>” manufactured by Shimadzu Corporation can be used; however, there are no particular limitations. The measurement conditions are set suitably so that the constituent can be detected. The column is not limited in particular as long as it can detect the constituent; however, the use of “Inertsil<sup>\*</sup> ODS-2  $\Phi$  3.0  $\times$  250mm” (manufactured by GL Sciences Inc.) is preferred.

“(B) Peak  $\beta$  constituent” is, in contrast to the  $\alpha$  constituent, an unidentified constituent decreasing as coffee beans are more roasted, and can be measured by an HPLC apparatus under the same conditions as the  $\alpha$  constituent. Concretely, it is a constituent corresponding to a peak obtained during constituent analysis of a coffee extract using an HPLC apparatus, appearing at a detection wavelength of 260nm and a retention time of 2.7 to 3.2 minute.

\*Trade-mark



In the specification, "(C) chlorogenic acids" is a collective term grouping 3-caffeoylquinic acid, 5-caffeoylquinic acid, 4-caffeoylquinic acid, 3-feruloylquinic acid, 5-feruloylquinic acid and 4-feruloylquinic acid. In addition, there are no particular limitations with respect to derivation thereof, and addition  
5 or elimination treatment may be carried out artificially.

"5-caffeoylquinic acid" is one of the constituents contained in constituent (C) chlorogenic acids. Weakest against heat among the chlorogenic acids, it has the property of either decomposing into small molecule organic acid, or isomerizing into 3-caffeoylquinic acid, as coffee beans are roasted. Therefore,  
10 the content in 5-caffeoylquinic acid varies during roasting of coffee beans, and the total amount of chlorogenic acids decreases as a result. In addition, although the antioxidant activity of the chlorogenic acids becomes higher when the content ratio of 3-caffeoylquinic acid rises due to the isomerization, if the content increases excessively the contribution of 3-caffeoylquinic acid to the  
15 antioxidant activity decreases. The small molecule organic acids generated by the decomposition of 5-caffeoylquinic acid have strong sourness and may become a cause of taste deterioration over time. Note that the deterioration is known to be attributable to generation of reactive oxygen species (radical). The higher the amount of radicals generated the more advanced the deterioration,



and conversely, if the amount of radicals generated can be held down, the deterioration can be suppressed.

According to the present invention, by adjusting the above-mentioned four constituents to appropriate ranges, a container-packed coffee beverage  
5 can be obtained, containing chlorogenic acids to a suitable extent, and furthermore, having excellent taste and which deterioration at heating has been suppressed. From the viewpoints of chlorogenic acids amount and taste, the peak area ratio  $\{(A)/(B)\} \times 100$  of (A) Constituent  $\alpha$  and (B) Constituent  $\beta$  is adjusted to 30 to 100. Furthermore, in order to maintain an excellent taste, the  
10 adjustment is such that the ratio becomes 40 to 75, and preferably 50 to 65. The reasons are, at less than 30, a decrease in taste is observed, such as chlorogenic acids-derived distracting flavor becomes noticeable, and at 100 or greater, the amount of chlorogenic acids becomes excessively small.

From the viewpoints of quality and antioxidant activity, 360ppm to  
15 840ppm, preferably 440ppm to 710ppm and more preferably 490ppm to 600ppm of chlorogenic acids (C) in values converted to per 1.0% by weight of amount of solid coffee may be contained. If 840ppm is exceeded, care must be taken, as a large amount of chlorogenic acids decompose into small molecule organic acid due to heat, readily generating sourness, or the like, over time. From the  
20 amount of chlorogenic acids and the viewpoints of quality and antioxidant

activity, the content ratio of 5-caffeoylquinic acid is adjusted to 28.0% to 35.0%, more preferably 28.8% to 31.5% and even more preferably 29.5% to 31.0%. At less than 28.0%, on top of the amount of chlorogenic acids becoming excessively small, the content ratio of 3-caffeoylquinic acid becomes

5 excessively high, and contribution to antioxidant activity decreases. If 35.0% is exceeded, on top of the content ratio of 3-caffeoylquinic acid becomes small, and the antioxidant activity becoming low, sourness increases over time due to the content in 5-caffeoylquinic acid being high.

While the container-packed coffee beverage of the present invention

10 can also be obtained by extracting roasted coffee beans in water or hot water by a conventional method, as the composition of the present invention is difficult to fulfill only with a single extract, it is preferably obtained by adjusting the constituents (A), (B) and (C) to give the previously mentioned composition, by blending of a plurality of extracts and/or addition of each constituent. For

15 instance, deeply roasted coffee beans with an L value of 16 or greater and slightly roasted coffee beans with an L value of 33 or less may be blended at a ratio of 90:10 and extracted. In more detail, the L value of deeply roasted coffee beans is adjusted to 16 to 21.5, more preferably to 16.5 to 20, even more preferably to 17 to 19.5, and particularly preferably to 17.5 to 18.0. The L value

20 of slightly roasted coffee beans is adjusted to 22 to 33, more preferably to 22.5

to 27, even more preferably to 23 to 26, and particularly preferably to 23.5 to 25.5. An object of the present invention is achieved by bringing the blending ratio to 100:0 to 60:40, more preferably to 95:5 to 70:30, and even more preferably to 90:10 to 80:20.

5           As places of production of coffee beans to be used, Brazil, Columbia, Tanzania, Mocha and the like can be cited, without any in particular restrictions. In addition, as bean product species, arabica species, robusta species, and the like may be cited. The coffee beans may be used as one species alone or by blending a plurality of species. Roasting is carried out using a general method,  
10 and the extent of roasting is adjusted suitably to obtain the extract necessary for the adjustment of each constituent. Concretely, bitterness becomes strong if roasting is deep and sourness becomes strong if roasting is slight.

The L value of the coffee beverage is preferably 28 to 62, more preferably 33 to 57, and even more preferably 38 to 52. Herein, an L value is a  
15 value that indicates luminosity. The L value of a coffee beverage can be measured according to common procedure using a colorimeter CT-310 (manufactured by Minolta) in a 1cm cell.

The container-packed coffee beverage of the present invention can also be prepared by adding each constituent specified in the present invention alone.  
20 When adjusting the composition by adding chlorogenic acids alone, the addition

of an extract of a plant containing chlorogenic acids is preferred. As plant raw materials containing chlorogenic acids that can be used herein, coffee (bean), hawthorn, grape, *Ligusticum wallichii*, *angelica acutiloba*, *coptis japonica*, turmeric, *asafoetida*, sweet potato, *molokheiya*, sunflower (seed), apple (fruit),  
5 tobacco (leaf), pear (leaf), peach, and the like, may be cited. From the viewpoints of the taste of the coffee beverage, coffee extract is preferred.

Furthermore, the coffee extract is preferably obtained from green coffee beans. Green coffee bean (i.e., raw coffee bean) extract can be obtained by grinding green coffee beans as necessary and extracting using ethanol, hydrous ethanol,  
10 methanol or the like, at a temperature of room temperature to 100°C. In addition a commercially available product, such as, flavor holder RC-30R, FH-1242, or the like, (Manufactured by T. Hasegawa Co.,Ltd.) can also be used. In addition, when adding 5-caffeoylquinic acid alone, after grinding with sunflower seed as a material and extracting with water or hydrous alcohol, the obtained sunflower  
15 seed extract is adsorbed onto a styrene -divinyl benzene series porous polymerization resin, then eluted with an organic solvent such as methanol, the obtained chlorogenic acids-containing extract is further passed over a dextran gel-packed column and the eluate obtained by eluting with water can be used.

In addition, as desired, sugars, such as, sucrose, glucose, fructose,  
20 xylose, fructose-glucose solution and sugar alcohol, milk constituent,

antioxidant, pH adjuster, emulsifier, flavor, and the like, can be added to the container-packed coffee beverage of the present invention. As milk constituents, raw milk, cow milk, whole dry milk, nonfat dry milk, raw cream, concentrated milk, nonfat milk, partially nonfat milk, condensed milk, and the like, may be  
5 cited. As the pH of the coffee beverage of the present invention, 3 to 7, furthermore 4 to 7, and in particular 5 to 7 are desirable from the aspect stability of the beverage.

As antioxidants, ascorbic acid or a salt thereof, erythorbic acid or a salt thereof, and the like, may be cited, among which, ascorbic acid or a salt thereof  
10 is particularly desirable.

As emulsifiers, sucrose fatty acid ester, glycerin fatty acid ester, microcrystal cellulose, lecithins, sorbitan fatty acid ester, polyglycerol fatty acid ester, and the like are preferred.

As containers used in the present invention, PET bottle, can (aluminum,  
15 steel), paper, retort pouch, bottle (glass), and the like, may be cited.

If heat sterilization is possible after filling into a container, as in the case of a metal can, sterilization treatment of coffee extract in the present invention is carried out under sterilization conditions defined in food sanitation law.

Regarding those containers for which retort sterilization is not possible, such as  
20 PET bottle and paper container, methods such as sterilizing beforehand under



equivalent sterilization conditions to conditions defined in the Food Sanitation Law, for instance, with a plate heat exchanger at high temperature and for a short-time, then cooling to a given temperature and filling a container, are adopted.

5 Example

Hereinafter, the present invention will be described in further detail with examples; however, the scope of the present invention is not limited by these examples and the like.

[1. Analysis of chlorogenic acids and constituents ( $\alpha$  and  $\beta$ )]

10 The gradient elution method by HPLC was used as the method for analyzing chlorogenic acids and constituent ( $\alpha$  and  $\beta$ ). As the analyzer, Waters\* 2695 separation module (manufactured by Nihon Waters K.K.) was used as the HPLC main unit and Waters\* 2996 PDA (manufactured by Nihon Waters K.K.) was used as the detector. The analytical conditions were as follows:

15 (Table 1)

Column: Inertsil ODS-2  $\phi$  3.0 x 250mm

Mobile phase:

A Solution 3% acetonitrile solution containing 0.05M acetic acid

B Solution 100% acetonitrile solution containing 0.05M acetic acid

20 Gradient elution method by (the program is shown in Table 1)

\*Trade-mark

Volume injected: 10  $\mu$  L

Flow rate: 0.43 mL/min

Column temperature: 40 degree C

- 5 Detector: PDA UV 210nm to 400nm (retention time and detection wavelength of each constituent are shown in Table 2)

Gradient Program

(min)	Flow rate (mL)	A Solution (%)	B Solution (%)
INITIAL	1.0	100	0
5.0	1.0	100	0
20.0	1.0	80	20
35.0	1.0	80	20
45.0	1.0	0	100
60.0	1.0	0	100
70.0	1.0	100	0
100.0	1.0	100	0

Chlorogenic acids	Retention time (min)	Detection wavelength
3-caffeoylquinic acid (3-CQA)	16.396	UV 325 nm
5-caffeoylquinic acid (5-CQA)	18.398	UV 325 nm
4-caffeoylquinic acid (4-CQA)	19.490	UV 325 nm
3-feruloylquinic acid (3-FQA)	20.204	UV 325 nm
5-feruloylquinic acid (5-FQA)	22.563	UV 325 nm
4-feruloylquinic acid (4-FQA)	23.267	UV 325 nm

Constituent	Retention time (min)	Detection wavelength
$\alpha$ constituent	2.217	UV 260 nm
$\beta$ constituent	2.915	UV 260 nm

[(1)-1 Chlorogenic acids quantification method]

Calibration curve construction method:

- 5 · Standards for 5-caffeoylquinic acid (5-CQA) (Manufactured by Wako Pure Chemical Industries Ltd.) were prepared with distilled water so as to yield three steps of concentrations in the order range of 20ppm to 100ppm to serve as standard solutions. A standard solution of each concentration was injected into the HPLC, and a three-point calibration curve was constructed from the peak
- 10 area values and concentrations.

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· Sample adjustment and quantification method:

Accurately, 5mL of each sample was measured, placed in a 25mL measuring flask, brought to a defined volume with distilled water, filtered with a 0.45 $\mu$ m filter, and then injected into HPLC to quantify 5-CQA by the three-point calibration curve method mentioned previously. In addition, since no standard was available for the quantification of 3-caffeoylquinic acid (3-CQA), 4-caffeoylquinic acid (4-CQA), 3-feruloylquinic acid (3-FQA), 5-feruloylquinic acid (5-FQA) and 4-feruloylquinic acid (4-FQA), only qualification was carried out, the area value of each peak was calculated as an estimated value using the 5-CQA calibration curve, the total value of the 6 types chlorogenic acids was used as the amount of chlorogenic acids.

[(1)-2 Method for the analysis of constituents ( $\alpha$  and  $\beta$ )]

Although constituents ( $\alpha$  and  $\beta$ ) are unidentified, the peak area value of each constituent was analyzed using the chromatogram at the detection wavelength of 260nm obtained during (1)-1 Analysis of chlorogenic acids, and the area ratio ( $\alpha/\beta$  value) determined by {area of  $\alpha$  constituent/area of  $\beta$  constituent} $\times 100$  was calculated.

The results are shown in Fig. 1.

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Regarding the  $\alpha$  constituent, it became clear that it was a constituent detected in a retention time of 2.217min under the HPLC conditions mentioned previously, with a maximum absorption wavelength of 257.9nm, and increased by roasting. Regarding the  $\beta$  constituent, it became clear that it was a

5 constituent detected in a retention time of 2.915min under the HPLC conditions mentioned previously, with a maximum absorption wavelength of 263.8nm and decreased by roasting. In Fig. 1, chromatograms at a detection wavelength of 260nm are shown for coffee beverages test-produced with each raw material bean. At the top of Fig. 1, it is shown that  $\alpha$  constituent and  $\beta$  constituent of a

10 coffee beverage test-produced with slightly roasted beans, In the middle of Fig. 1, it is shown that  $\alpha$  constituent and  $\beta$  constituent of a coffee beverage test-produced with medium roasted beans, and at the bottom of Fig. 1, it is



shown that  $\alpha$  constituent and  $\beta$  constituent of a coffee beverage test-produced with deeply roasted beans.

[(1)-3 Method for measurement of hydroxy radical]

A radical (reactive oxygen species) is an index of quality deterioration, indicating the higher the quantity generated the more progressed the deterioration. For the method for measurement of hydroxy radical, each sample was heated at 60°C, and hydroxy radical generated at each time point was measured by the ESR method. (according to T. Am. Soc. Brew. Chem. 54 (4): 198-205, 205-211, 1996) For the reagent, spin trapping agent: PBN (N-tert-Butyl- $\alpha$ -phenylnitron) (manufactured by Wako Pure Chemical Industries) was used, and the spin trapping agent reagent was prepared to 2.55M with 50% ethanol, 0.16mL of which was added to 8mL of sample and then stirred thoroughly. Heating was carried out at 60°C with this solution in a thermostatic bath, hydroxy radical generated at each time was measured by the ESR, and the peak height of the hydroxy radical (PBN-OH) was calculated as a relative ratio against Mn (internal standard). ESR measurement conditions were as shown in Table 3. When the amount of radicals generated (relative ratio against Mn) is taken as the vertical axis and the heating time (min) is taken as the horizontal axis, the hydroxy radical generated when each sample was heated can be represented by an approximate linear equation having a certain

slope, and the slope of this linear equation (radical generation velocity) was evaluated as the amount of radicals generated.

(Table 3)

Power	4 mW
ModWid	0.1 mT
C. Field	335.500 mT
Amp	1000
SwWid	(±) 5 mT
TimeC	1 sec
SwTime	4 min
Mn	550 nm

## 5 [Example 1]

(Comparison of coffee beverages test-produced with beans having different degrees of roasting)

Chlorogenic acids content, area ratio of Constituents  $\alpha$  and  $\beta$ , and amount of radicals generated at heating, of coffee beverages test-produced separately with coffee beans having different degrees of roasting were investigated.

A semi-hot air drum sample roaster having a 5kg pan was used for roasting of coffee beans. After ignition to the burner, adjustment was made to medium heat, at the point where the internal temperature of the drum became 200°C, 3kg of Brazil-produced beans were fed. While maintaining the heating power of the burner at medium heat, roasting was carried out until the target degree of roasting was reached, and after roasting, discharging and cooling

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were performed. The roasting time and drum internal temperature at discharge time for each roast bean used in the present test were as shown in Table 5. The roasted coffee beans were ground with a coffee cutter and the surface color was measured with the spectrophotometric colorimeter SE2000 from Nippon

5 Denshoku Industries Co.,Ltd.

(Table 4)

	L value	Roasting time	Discharge temperature
Slight roast roasted coffee bean	32	10 minutes 00 seconds	190 degree C
Medium roast roasted coffee bean	21	11 minutes 50 seconds	200 degree C
Deep roast roasted coffee bean	17	12 minute 10 seconds	218 degree C

Extraction of coffee beverage was done by feeding 300g of ground roasted coffee beans into a coffee extractor. After the bean surface was leveled,  
 10 it was showered from the upper part of the extractor with hot water at 95°C, and after verifying that hot water penetrated the entirety of the beans, extract was recovered while continuing the showering.





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	32	21	17	(ppm) <sup>a</sup>			(slope of linear equation)
Sample 1	100%			1540	32.9	2.9	0.0482
Sample 2		100%		830	31.4	19.0	0.0173
Sample 3			100%	360	28.0	91.4	0.0050

Amount of radicals generated during heating of each coffee beverage

From the results of Table 5 and Fig. 2, in the coffee beverages test-produced with raw material beans having different degrees of roasting, for sample 1 with slightly roasted beans as raw materials, the content in chlorogenic acids was



high, and since there was little  $\alpha$  constituent and abundant  $\beta$  constituent, the area ratio  $\alpha/\beta$  value was a low value. In addition, for sample 1, the amount of radicals generated at heating was also found to be large. On the other hand, for sample 3 with deeply roasted beans as raw material, the content in chlorogenic acids decreased, and as  $\alpha$  constituent increased and  $\beta$  constituent decreased, the area ratio  $\alpha/\beta$  value was a higher value. In addition, for sample 3, the amount of radicals generated at heating was also found to be small.

[Example 2]

(Comparison of coffee beverages blending coffee beans having different degrees of roasting)

Chlorogenic acids content, area ratio of Constituents  $\alpha$  and  $\beta$ , and the amount of radicals generated at heating of coffee beverages test-produced by blending coffee beans having different degrees of roasting were investigated.

Green beans roasting method and coffee beverage preparation method were carried out by methods similar to Example 1. Coffee beans having different degrees of roasting were blended as described below, a coffee beverage was prepared by the preparation method of Example 1 such that the amount of solid coffee was 1.0% by weight, and the amount of chlorogenic acids, constituents ( $\alpha$  and  $\beta$ ), and the amount of radicals generated were investigated. The results are shown in Table 6.

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(Table 6)

	Blending ratio of material coffee beans		Amount of chlorogenic acids (ppm) <sup>a</sup>	5-CQA/a* 100 (%)	$\alpha/\beta \times 100$ (area ratio)	Amount of radicals generated (slope of linear equation)
	L value	L value				
	32	17				
Sample 1	100%	0%	32.9	1540	2.9	0.0482
Sample 2	75%	25%	32.6	1230	12.2	0.0290
Sample 3	50%	50%	32.0	930	25.9	0.0190
Sample 4	25%	75%	30.9	630	47.5	0.0090
Sample 5	0%	100%	28.0	360	91.4	0.0050

Amount of radicals generated during heating of each coffee beverage

From the results of Table 6 and Fig. 3, also with coffee beverages blending raw  
5 material beans having different degrees of roasting, for samples 1 and 2 with  
high blending ratio of slightly roasted beans, the chlorogenic acid content was  
high and the area ratio of  $\alpha$  constituent and  $\beta$  constituent had a low value. In  
addition, samples 1 and 2 were also found to have large amounts of radical  
generated at heating. On the other hand, for samples 4 and 5 with high blending  
10 ratio of deeply roasted beans, chlorogenic acids content was low, and the area  
ratio of  $\alpha$  constituent and  $\beta$  constituent also had a high value. In addition, for  
samples 4 and 5, the amount of radicals generated at heating was also found to  
be lower. From these results, it was found that, in coffee beverages  
test-produced with coffee beans blending slightly roasted beans in abundance,  
15 the chlorogenic acids content was high, furthermore, the area ratio of  $\alpha$   
constituent and  $\beta$  constituent was low and the amount of radicals generated at  
heating was large.

[Example 3]

(Relationship between chlorogenic acids content,  $\alpha$  constituent/ $\beta$  constituent area ratio and amount of radicals generated at heating, and sensory evaluation after heat storage)

The influence on the deterioration of quality after heat storage (at 70°C for two weeks) exerted by the chlorogenic acids content, the area ratio of  $\alpha$  constituent and  $\beta$  constituent, and the amount of radicals generated at heating of coffee beverages test-produced by blending different degrees of roasting raw material beans was investigated.

Green beans roasting method and coffee beverage preparation method were carried out by methods similar to Example 1. After storage at 70°C for two weeks, sensory evaluation and a five-stage sourness intensity evaluation were carried out for each sample. Note that the sensory evaluation was carried out by five panelists who have excellent identification capabilities regarding coffee flavor.

Similarly to Example 2, coffee beans having different degrees of roasting were blended as shown in Table 8, a coffee beverage was prepared by the preparation method of Example 1 such that the amount of solid coffee was 1.0% by weight, and the amount of chlorogenic acids, constituents ( $\alpha$  and  $\beta$ ), and the amount of radicals generated were investigated.

(Table 7)



	Blending ratio of material coffee beans				Amount of chlorogenic acids (ppm) <sup>a</sup>	5-CQA/ a*100 (%)	$\alpha/\beta*100$ (area ratio)	Amount of radicals generated (slope of linear equation)
	L value 32	L value 17	L value 16	L value 15				
Sample 1	75%	25%			1230	32.6	12.2	0.0290
Sample 2	50%	50%			970	32.0	23.7	0.0190
Sample 3	40%	60%			840	31.6	30.8	0.0130
Sample 4	30%	70%			710	31.4	40.5	0.0078
Sample 5	20%	80%			600	30.7	49.4	0.0084
Sample 6	10%	90%			490	29.5	62.2	0.0081
Sample 7	5%	95%			440	28.8	72.2	0.0070
Sample 8	0%	5%			360	28.0	91.4	0.0050
Sample 9			100%		230	25.2	150.2	0.0043
Sample 10				100%	140	24.8	247.7	0.0044

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(Table 8)

	Sensory evaluation	Sourness intensity	Overall evaluation
Sample 1	Strong fermentation odor, abundant distracting flavors	0.25	X

Sample 2	Fermentation-like deterioration odor	0.25	X
Sample 3	Somewhat fermentation-like deterioration odor	1.38	△
Sample 4	Somewhat fermentation-like deterioration odor	1.25	△
Sample 5	Weakest sourness	1.75	○
Sample 6	Weakest sourness	2.00	◎
Sample 7	Somewhat strong sourness	2.13	○
Sample 8	Strong sourness	2.50	△
Sample 9	Strong sourness and abundant distracting flavors	3.00	X
Sample 10	Abundant distracting flavors and burnt odor sensed	2.75	X

From the results of Table 7, Table 8, Fig. 4 and Fig. 5, similarly as before, with coffee beverages blending coffee beans having different degrees of roasting, when the blending ratio of slightly roasted beans became higher, the chlorogenic acids content was high and the area ratio of  $\alpha$  constituent and  $\beta$  constituent had a low value. In addition, amount of radicals generated at heating also became large

when the blending ratio of slightly roasted beans became higher. On the other hand, when the blending ratio of deeply roasted beans became higher, chlorogenic acids content was low and the area ratio of  $\alpha$  constituent and  $\beta$  constituent had also a high value. In addition, the amount of radicals generated at heating also became low when the blending ratio of deeply roasted beans became higher.

Resulting from the sensory evaluation after conservation at 70°C for two weeks, for samples 1 and 2 with high blending ratio of slightly roasted beans, fermentation-like deterioration odor could be strongly sensed, and for sample 1 distracting flavor could also be sensed abundantly. On the other hand, for samples 9 and 10, in which deeply roasted beans with a strong degree of roasting were used, distracting flavor, burnt odor, sourness and the like, were strongly sensed, and for sample 8, sourness was also sensed strongly. From the above results, it was thought that, when the blending ratio of slightly roasted beans became higher, chlorogenic acids content became large and the amount of radicals generated, which are considered to be one of the causes of deterioration, also became large, deterioration due to heat storage occurred, and fermentation odor and the like could be sensed strongly. In addition, with deeply roasted beans, if the degree of roasting became strong, the amount of chlorogenic acids became low and the amount of radicals generated also



became low; however, after heat storage, distracting flavor, burnt taste, sourness and the like, tended to be stronger. From the above results, sample 3 to sample 8 had satisfactory sensory evaluation and good overall evaluation.

[Example 4]

5 (Comparison between coffee beverage test-produced with medium roasted beans and coffee beverage blending slightly roasted beans and deeply roasted beans)

The amount of radicals generated at heating and the situation of quality deterioration after heat storage (at 70°C for two week) of coffee beverage from 10 medium roasted beans and coffee beverage blending slightly roasted beans and deeply roasted beans with the same extent in chlorogenic acids content were investigated.

Green beans roasting method and coffee beverage preparation method were carried out by methods similar to Example 1. Sensory evaluation method 15 was carried out by a similar method to Example 3.

Similarly to Example 2, coffee beans having different degrees of roasting were blended as shown in Table 9, a coffee beverage was prepared by the preparation method of Example 1 such that the amount of solid coffee was 1.0% by weight, and the amount of chlorogenic acids, constituents ( $\alpha$  and 20  $\beta$ ), and the amount of radicals generated were investigated.

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(Table 9)

	Blending ratio of material coffee beans				Amount of chlorogenic acids (ppm) <sup>a</sup>	5-CQA/ a*100 (%)	$\alpha/\beta*100$ (area ratio)	Amount of radicals generated (slope of linear equation)
	L value	L value	L value	L value				
	32	25	20	17				
Sample 1			100%		760	30.0	28.4	0.0099
Sample 2	20%			80%	730	30.2	37.9	0.0105
Sample 3		100%			1330	33.1	5.8	0.0416
Sample 4	50%			50%	1070	31.5	20.5	0.0183

(Table 10)

5 Sensory evaluation results and sourness intensity after storage at 70°C for two weeks

	Sensory evaluation	Sourness intensity	Overall evaluation
Sample 1	Somewhat sourness taste sensed,	3.00	X

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	abundant distracting flavors, strong sourness		
Sample 2	Somewhat sourness taste sensed but satisfactory flavor and also has bitterness	2.00	○
Sample 3	Grainy taste sensed, abundant distracting flavors, strong sourness	2.25	X
Sample 4	Distracting flavors somewhat present, fermentation-like deterioration odor	1.25	X

From the results of Table 9 and Table 10, samples 1 and 3 test-produced with medium roasted beans alone had graininess, and distracting flavor, sourness and the like could be sensed strongly; however, by blending

5 deeply roasted beans and slightly roasted beans, for sample 2, deeply roasted beans-derived bitterness could be sensed and the flavor was satisfactory. Consequently, even with the same extent of chlorogenic acids concentration, blending deeply roasted beans and slightly roasted beans gave satisfactory flavor and stability after heat storage. However, for sample 4, with a high

blending ratio of slightly roasted beans and large amount of chlorogenic acids, fermentation odor could be sensed.

#### Industrial applicability

- 5 According to the present invention, by giving the above-mentioned composition, a container-packed coffee beverage, can be provided, having an excellent taste while containing chlorogenic acids to a suitable extent, which are useful constituents found in coffee, and furthermore, whose deterioration at heating has been suppressed.



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CLAIMS:

1. A method for preparing a container-packed coffee beverage,  
wherein roasted coffee beans having an L value of 16 to 21.5 and roasted coffee  
beans having an L value of 22 to 33 are blended at a ratio of 95:5 to 70:30 to  
5 obtain a coffee extract, and;

in a coffee extract obtained by extracting coffee beans, the following  
constituents (A), (B) and (C) are contained;

(A) Peak  $\alpha$  constituent

(B) Peak  $\beta$  constituent

10 (C) chlorogenic acids

and the peak area ratio of constituent (A) having a retention time of  
2.0 to 2.3 minutes and maximum absorption wavelength of 257.9nm and  
constituent (B) having a retention time of 2.7 to 3.2 minutes and maximum  
absorption wavelength of 263.8nm on a chromatogram obtained by constituent  
15 analysis of a coffee extract using high performance liquid chromatography  
apparatus is  $\{(A)/(B)\} \times 100 = 30$  to 100, and the content ratio of 5-caffeoylquinic acid  
within the constituent (C) is 28.0 to 35.0% by weight;

and the conditions for performing the high performance liquid  
chromatography being as follows:

20 Column: Inertsil ODS-2  $\varnothing$  3.0 x 250mm;

Mobile phase: Gradient elution method by

A Solution 3% acetonitrile solution containing 0.05M acetic acid;

B Solution 100% acetonitrile solution containing 0.05M acetic acid;

Volume injected: 10  $\mu$ L;

25 Flow rate: 0.43 mL/min;

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Column temperature: 40 degree C;

Detector: PDA UV 210nm to 400nm.

2. The method for preparing a container-packed coffee beverage as recited in claim 1, wherein chlorogenic acids are added to said obtained coffee  
5 extract.

3. A container-packed coffee beverage, comprising a coffee extract blended from roasted coffee beans having an L value of 16 to 21.5 and roasted coffee beans having an L value of 22 to 33 at a ratio of 95:5 to 70:30 and containing the following constituents (A), (B) and (C);

10 (A) Peak  $\alpha$  constituent

(B) Peak  $\beta$  constituent

(C) chlorogenic acids

and the peak area ratio of constituent (A) has a retention time of 2.0 to 2.3 minutes and maximum absorption wavelength of 257.9nm and constituent  
15 (B) has a retention time of 2.7 to 3.2 minutes and maximum absorption wavelength of 263.8nm on a chromatogram obtained by constituent analysis of a coffee extract using high performance liquid chromatography apparatus is  $\{(A)/(B)\} \times 100 = 30$  to 100, and the content ratio of 5-caffeoylquinic acid within the constituent (C) is 28.0 to 35.0% by weight; and

20 the conditions for performing the high performance liquid chromatography are as follows:

Column: Inertsil ODS-2  $\varnothing$  3.0 x 250mm;

Mobile phase: Gradient elution method by

A Solution 3% acetonitrile solution containing 0.05M acetic acid;

25 B Solution 100% acetonitrile solution containing 0.05M acetic acid;

Volume injected: 10  $\mu$ L;

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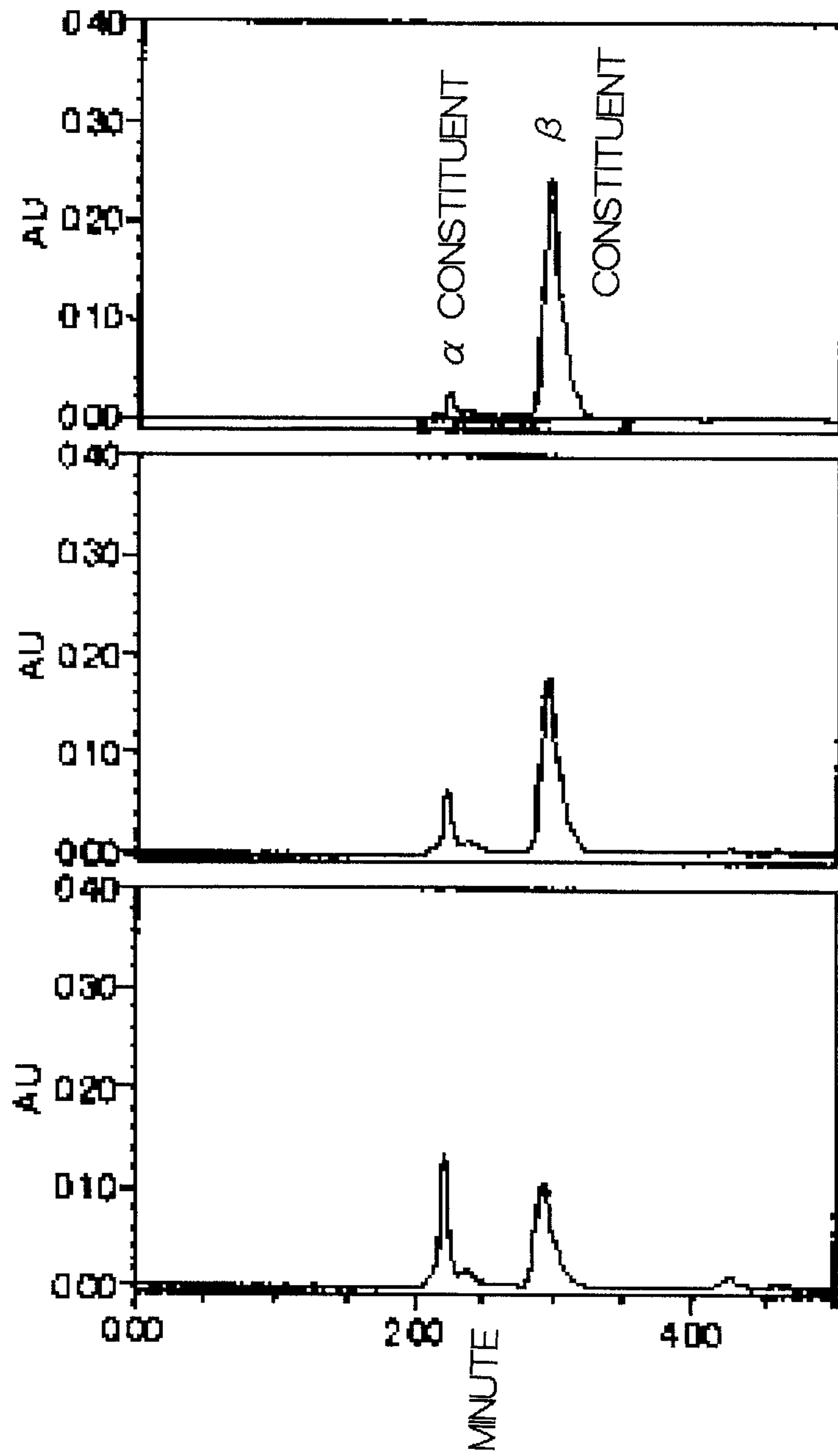
38

Flow rate: 0.43 mL/min;

Column temperature: 40 degree C;

Detector: PDA UV 210nm to 400nm.

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**FIG. 1**



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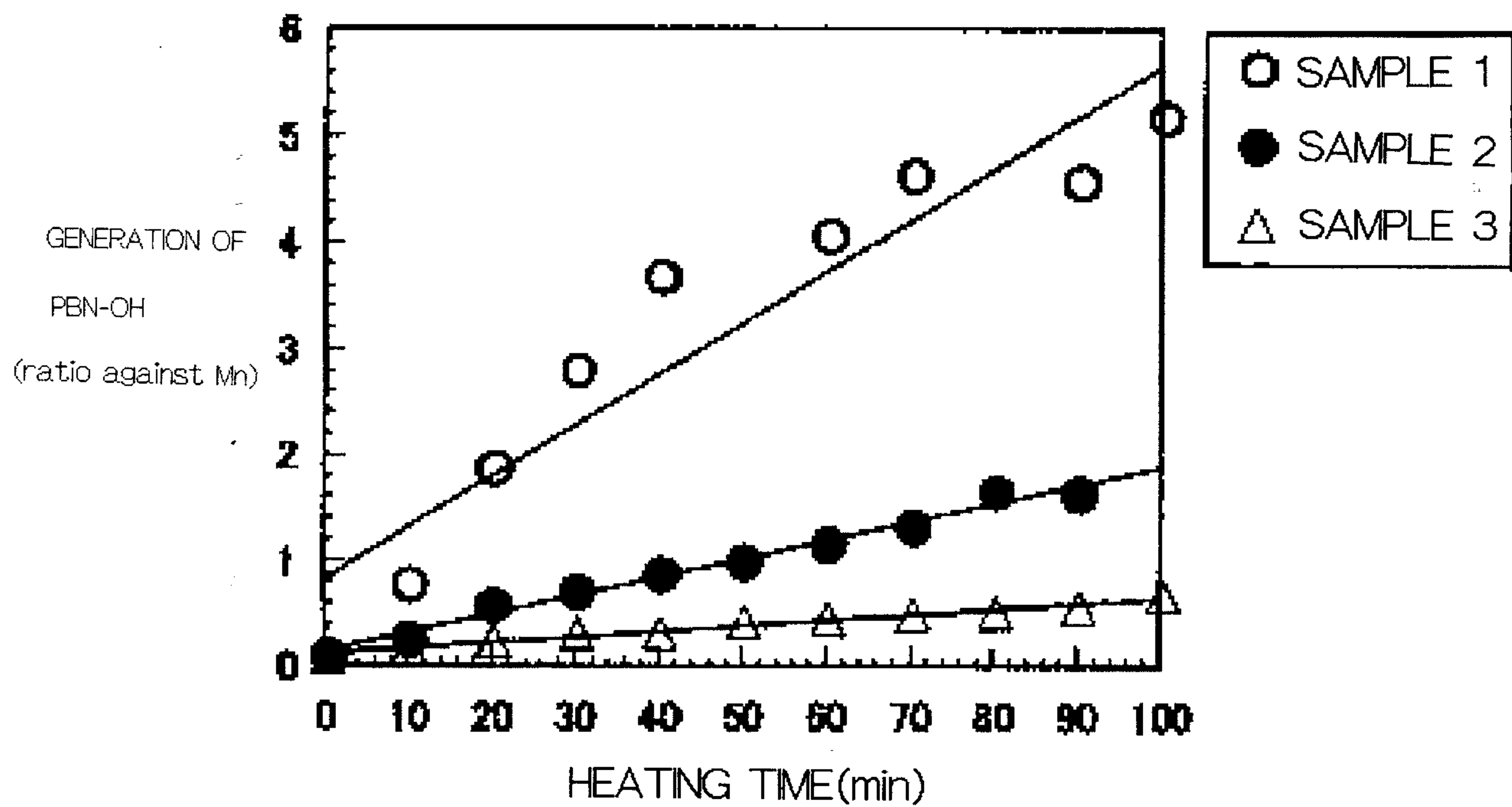


FIG. 2

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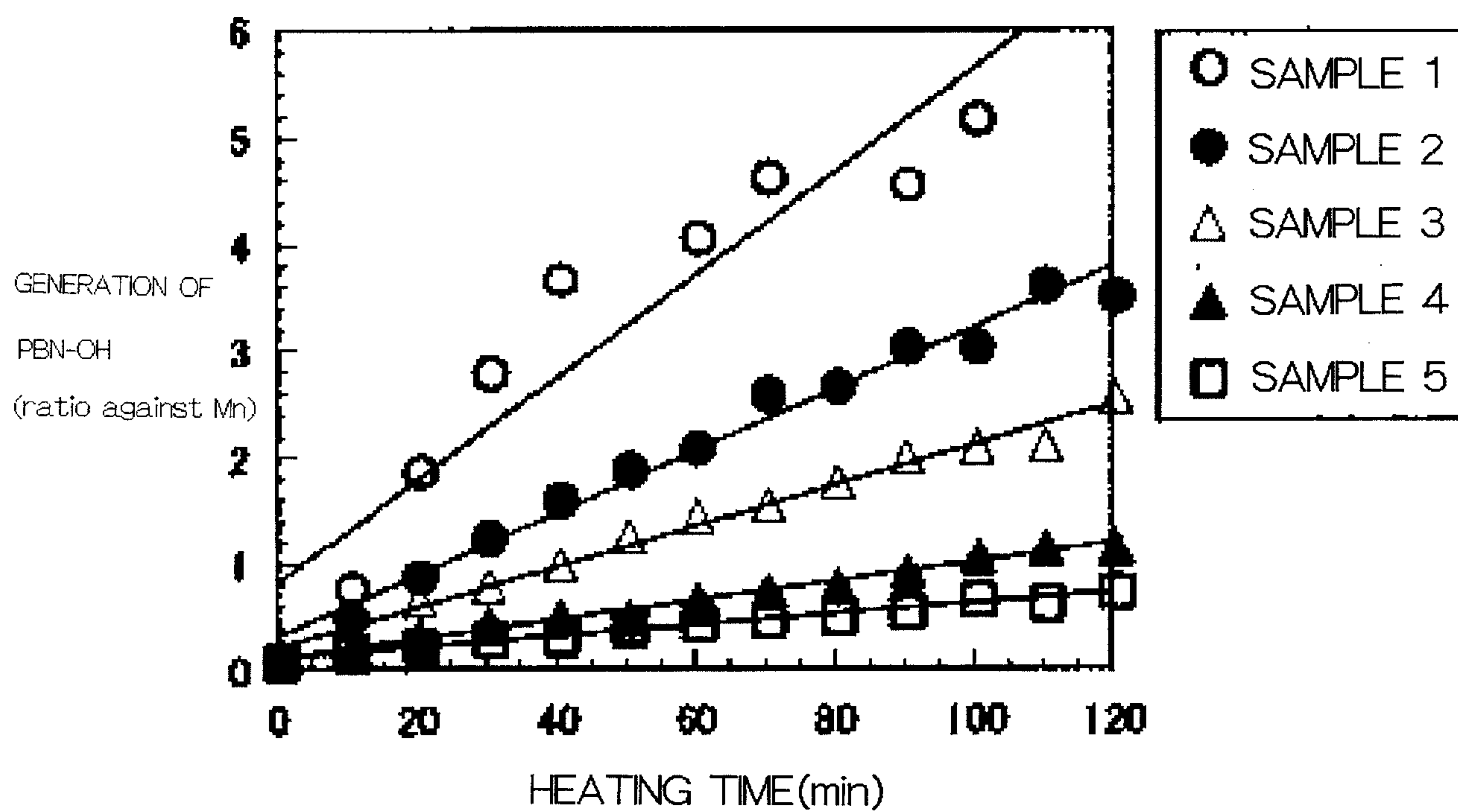


FIG. 3

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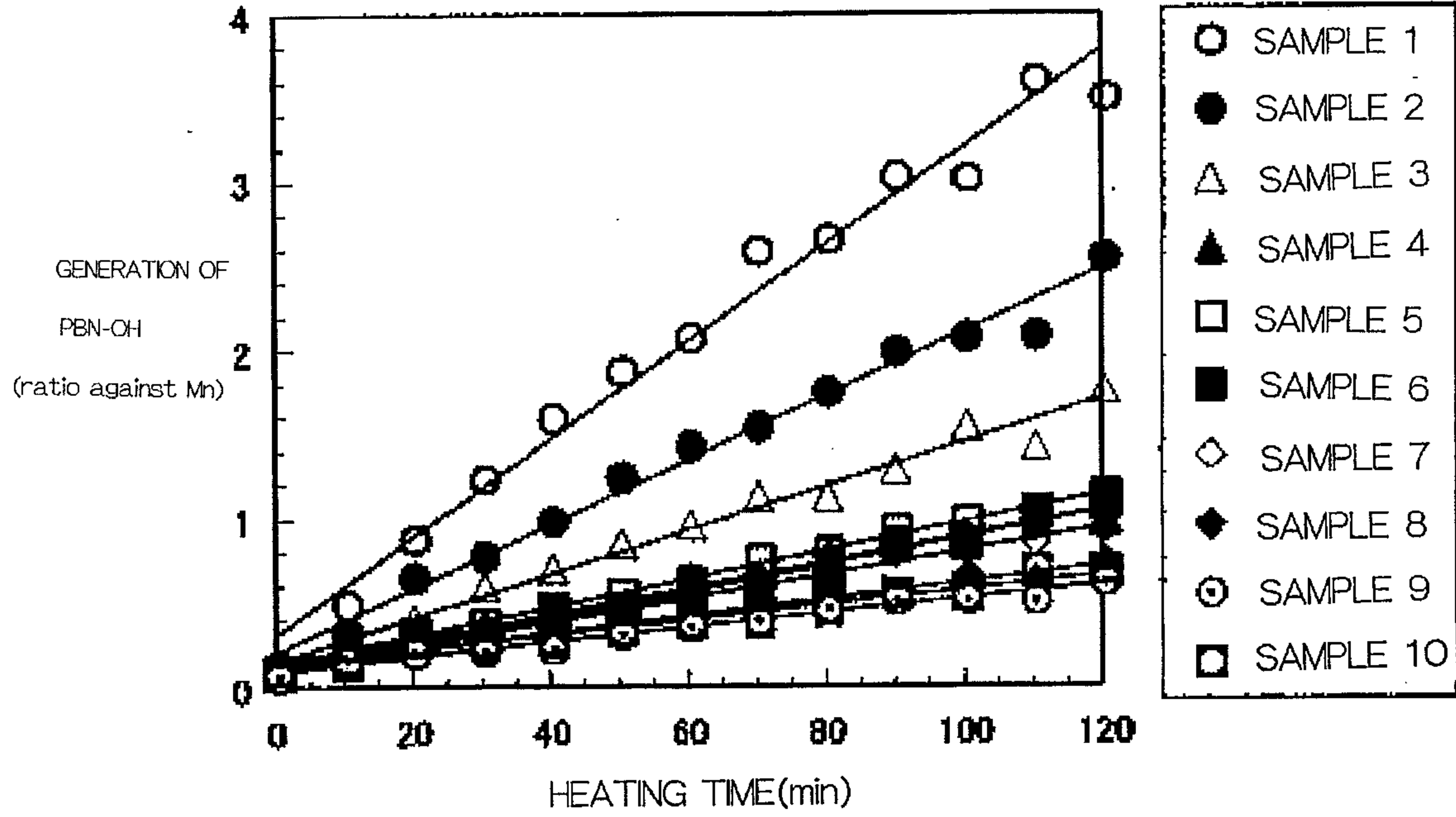


FIG. 4

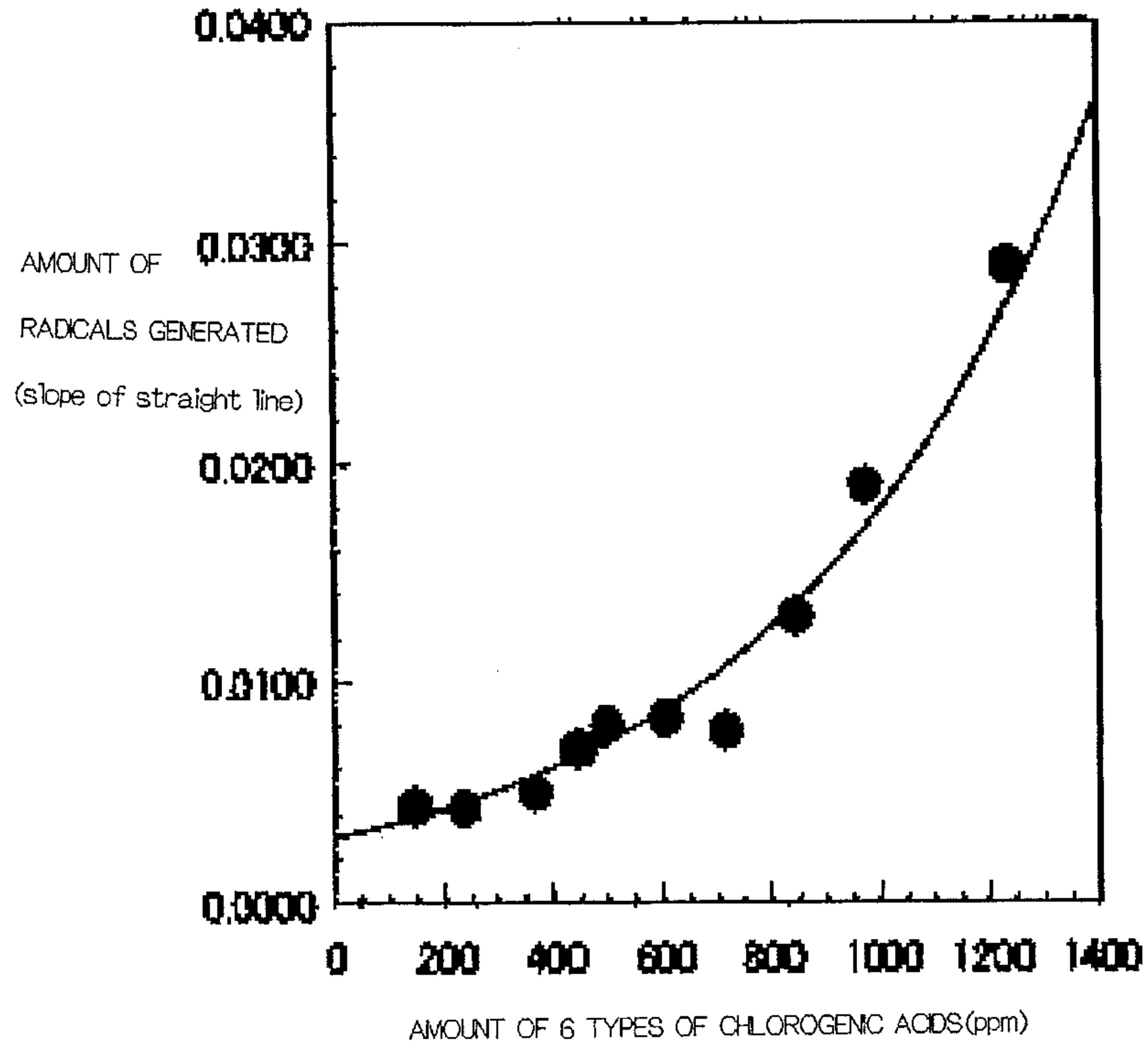


FIG. 5