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# (12) United States Patent Kim et al.

# (54) POLYCYCLIC COMPOUND AND ORGANIC LIGHT EMITTING DIODE COMPRISING SAME

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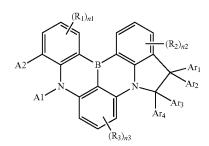
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#### (57) ABSTRACT

Provided is a compound of Formula 1:



 $Ar_1$  to  $Ar_4$  are each independently hydrogen, deuterium, a halogen group, a nitrile group, or a substituted or unsubstituted: silyl, boron, alkyl, alkenyl, alkynyl, alkoxy, aryloxy, cycloalkyl, aryl, or heterocyclic group, or adjacent groups are bonded together to form a substituted or unsubstituted aliphatic hydrocarbon ring; A1 and A2 are each independently hydrogen, deuterium, a halogen group, a nitrile group, or a substituted or (Continued)

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unsubstituted: silyl, boron, alkyl, alkenyl, alkynyl, alkoxy, aryloxy, cycloalkyl, aryl, or heterocyclic group, or are bonded together to form a substituted or unsubstituted ring;

R<sub>1</sub> to R<sub>3</sub> are each independently hydrogen, deuterium, a halogen group, a nitrile group, or a substituted or unsubstituted; silyl, boron, alkyl, alkenyl, alkynyl, alkoxy, aryloxy, cycloalkyl, aryl, amine, or heterocyclic group; and

n1 to n3 are each 0 to 3, and 2 or more, the substituents are the same as or different from each other,

and an organic light emitting device including the same.

#### 19 Claims, 1 Drawing Sheet

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	H10K 101/40	(2023.01)

#### (52) U.S. Cl.

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FIG. 1

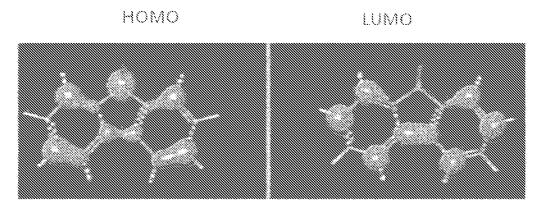


FIG. 2

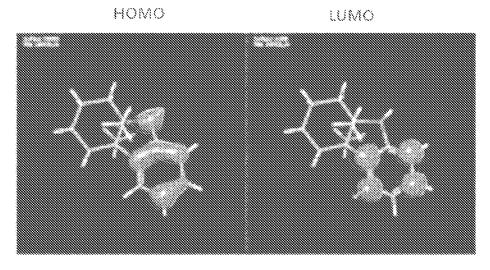


FIG. 3

#### POLYCYCLIC COMPOUND AND ORGANIC LIGHT EMITTING DIODE COMPRISING **SAME**

#### CROSS-REFERENCE TO RELATED APPLICATION(S)

This application is a National Stage Application of International Application No. PCT/KR2019/008121 filed on Jul. 3, 2019, which claims priority to and the benefit of Korean Patent Application No. 10-2019-0006153 filed in the Korean Intellectual Property Office on Jan. 17, 2019, and to Korean Patent Application No. 10-2018-0077111 filed in the Korean Intellectual Property Office on Jul. 3, 2018, the entire 15 contents of which are incorporated herein by reference.

#### TECHNICAL FIELD

The present specification relates to a polycyclic com- 20 pound and an organic light emitting device including the same.

#### BACKGROUND

In the present specification, an organic light emitting device is a light emitting device using an organic semiconductor material, and requires an exchange of holes and/or electrons between electrodes and organic semiconductor materials. The organic light emitting device can be roughly 30 divided into the following two light emitting devices depending on the operation principle. The first organic light emitting device is a light emitting device in which an exciton is formed in an organic material layer by a photon that flows from an external light source to the device, the exciton is 35 separated into electrons and holes, and the electrons and the holes are each transferred to different electrodes and used as a current source (voltage source). The second organic light emitting device is a light emitting device in which holes and/or electrons are injected into organic semiconductor 40 organic light emitting device including the same. material layers forming an interface with an electrode by applying a voltage or current to two or more electrodes, and the device is operated by the injected electrons and holes.

In general, an organic light emitting phenomenon refers to a phenomenon in which electric energy is converted into 45 provides a compound of Formula 1: light energy by using an organic material. An organic light emitting device using the organic light emitting phenomenon usually has a structure including a positive electrode, a negative electrode, and an organic material layer interposed therebetween. Here, the organic material layer has in 50 many cases a multi-layered structure composed of different materials in order to improve the efficiency and stability of the organic light emitting device, and for example, can be composed of a hole injection layer, a hole transport layer, a light emitting layer, an electron blocking layer, an electron 55 transport layer, an electron injection layer, and the like. In such a structure of the organic light emitting device, if a voltage is applied between the two electrodes, holes are injected from the positive electrode into the organic material layer and electrons are injected from the negative electrode 60 into the organic material layer, and when the injected holes and electrons meet each other, an exciton is formed, and light is emitted when the exciton falls down again to a ground state. Such an organic light emitting device has been known to have characteristics such as self-emission, high 65 brightness, high efficiency, a low driving voltage, a wide viewing angle, and high contrast.

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In an organic light emitting device, materials used as an organic material layer can be classified into a light emitting material and a charge transport material, for example, a hole injection material, a hole transport material, an electron blocking material, an electron transport material, an electron injection material, and the like depending on the function. The light emitting materials include blue, green, and red light emitting materials according to the light emitting color, and yellow and orange light emitting materials required for implementing a much better natural color.

Furthermore, a host/dopant system can be used as a light emitting material for the purpose of enhancing color purity and light emitting efficiency through energy transfer. The principle is that when a small amount of dopant which has a smaller energy band and better light emitting efficiency than those of a host mainly constituting a light emitting layer is mixed with the light emitting layer, the excitons generated by the host are transported to the dopant to emit light with high efficiency. In this case, it is possible to obtain light with a desired wavelength according to the type of dopant used because the wavelength of the host moves to the wavelength range of the dopant.

In order to fully exhibit the above-described excellent characteristics of the organic light emitting device, a material constituting an organic material layer in a device, for example, a hole injection material, a hole transport material, a light emitting material, an electron blocking material, an electron transport material, an electron injection material, and the like need to be supported by stable and efficient materials, so that there is a continuous need for developing a new material.

Prior Art Document—Japanese Patent Application Laid-Open No. 2018-043984

#### BRIEF DESCRIPTION OF THE INVENTION

#### Technical Problem

The present specification describes a compound and an

#### Technical Solution

An exemplary embodiment of the present specification

Formula 1

In Formula 1:

 $Ar_1$  to  $Ar_2$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted

alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted cycloalkyl group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heterocyclic group, or adjacent groups are bonded to each other to form a substituted or unsubstituted aliphatic hydrocarbon ring;

Al and A2 are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted or unsubstituted or unsubstituted aryloxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heterocyclic group, or are bonded to each other to form a substituted or unsubstituted ring;

R<sub>1</sub> to R<sub>3</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group; a substituted or unsubstituted alkyl group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted or unsubsti

n1 to n3 are each an integer from 0 to 3, and when n1 to n3 are each 2 or more, the substituents in a plurality of parentheses are the same as or different from each other

Further, an exemplary embodiment of the present specification provides an organic light emitting device including: a first electrode; a second electrode provided to face the first electrode; and an organic material layer having one or more layers provided between the first electrode and the second electrode, in which one or more layers of the organic 40 material layer include the above-described compound.

#### Advantageous Effects

The compound of the present invention can be used as a material for an organic material layer of an organic light emitting device. The compound of the present invention has high stability of the compound by heat during the deposition process by including a non-aromatic pentagonal ring (cycloalkene ring) including N in the molecule and having a distorted structure instead of a planar structure to lower the sublimation temperature, so that it is possible to obtain an organic light emitting device having high efficiency, low voltage, and long service life characteristics when the compound is applied to the organic light emitting device.

In addition, the compound of the present invention includes an aliphatic hydrocarbon ring, so that it is possible to obtain an organic light emitting device having a narrow full width at half maximum and excellent color purity.

Furthermore, the compound of the present invention has 60 high solubility, and thus can also be used in a solution process.

#### BRIEF DESCRIPTION OF DRAWINGS

FIG. 1 illustrates an organic light emitting device composed of a substrate 1, a positive electrode 2, a hole injection

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layer 3, a hole transport layer 4, an electron blocking layer 5, a light emitting layer 6, a first electron transport layer 7, a second electron transport layer 8, and a negative electrode 9.

FIG. 2 illustrates the HOMO and LUMO electron distribution diagram of 9H-carbazole.

FIG. 3 illustrates the HOMO and LUMO electron distribution diagram of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole.

# EXPLANATION OF REFERENCE NUMERALS AND SYMBOLS

1: Substrate

**2**: Positive electrode

3: Hole injection layer

4: Hole transport layer

5: Electron blocking layer

6: Light emitting layer

7: First electron transport layer

8: Second electron transport layer

9: Negative electrode

#### DETAILED DESCRIPTION

Hereinafter, the present specification will be described in more detail.

The present specification provides a compound of the following Formula 1. The compound of the following Formula 1 has a low sublimation temperature, and thus is stable, and the efficiency and service life characteristics of the organic light emitting device are improved when the compound is applied to an organic material layer of an organic light emitting device:

Formula 1  $A2 \xrightarrow{(R_1)_{n1}} B \xrightarrow{(R_2)_{n2}} Ar_1.$   $Ar_2 \xrightarrow{(R_3)_{n3}} Ar_3$ 

In Formula 1:

Ar<sub>1</sub> to Ar<sub>4</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted or unsubstituted or unsubstituted alkyl group, a substituted or unsubstituted alkenyl group, a substituted alkenyl group, a substituted alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heterocyclic group, or adjacent groups are bonded to each other to form a substituted or unsubstituted aliphatic hydrocarbon ring;

A1 and A2 are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group; a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted alkyl group, a substituted or unsubstituted or unsubstituted or unsubstituted alkyl group, a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted alkyl group, a substituted or unsubstituted or unsubstit

stituted alkenyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted cycloalkyl group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heterocyclic group, or are bonded to each other to form a substituted or unsubstituted ring;

 $\rm R_1$  to  $\rm R_3$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted or unsubstituted or unsubstituted alkyl group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkenyl group, a substituted alkoxy group, a substituted or unsubstituted aryl group, a substituted or unsubstituted aryl group, a substituted or unsubstituted or unsubstituted aryl group, a substituted or unsubstituted aryl group, a substituted or unsubstituted amine group, or a substituted or unsubstituted heterocyclic group; and

n1 to n3 are each an integer from 0 to 3, and when n1 to n3 are each 2 or more, the substituents in a plurality of parentheses are the same as or different from each other.

According to an exemplary embodiment of the present specification, the compound of Formula 1 includes a hexahydrocarbazole ring at a central fused ring core, or includes a hexahydrocarbazole group at a R<sub>3</sub> position. A hexahydrocarbazole has reduced conjugation compared to a carbazole, and thus exhibits different properties from the carbazole:

In the following table, the HOMO, LUMO, T1, and S1 means the highest occupied energy, lowest unoccupied energy, triplet energy, and singlet energy, respectively. 9H-carbazole is a derivative including an aromatic ring and 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1-carbazole is a derivative including an aliphatic ring, and can correspond to the derivative of hexahydrocarbazole of Formula 1.

	НОМО	LUMO	T1	S1
H	5.44	0.64	3.19	4.15
9H-CARBAZOLE	5.19	0.43	3.56	4.74
4a,9a-dimethyl-2,3,4,4a,9,9a- hexahydro-1H-carbazole				

When the HOMO and LUMO values are compared to 65 each other in the table, 9H-carbazole has a deeper HOMO value than that of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-

1H-carbazole. This shows that 9H-carbazole has a greater influence on amines having an electron donor characteristic. As the HOMO energy of a compound becomes deeper, the electron donor characteristic of the compound to another compound in a device deteriorates, so that 9H-carbazole has a lower electron donor characteristic in a device than 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole.

Further, in the table, 9H-carbazole has a lower triplet energy value than 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole when the triplet energy values (T1) are compared. The more extended the conjugation is, the lower the triplet energy is, so that 9H-carbazole having an extended conjugation has a low triplet energy value. When the triplet energy values are compared, it can be seen that 9H-carbazole and 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole are different materials having quite different characteristics.

FIG. 2 illustrates the HOMO and LUMO electron distribution diagram of 9H-carbazole, and FIG. 3 illustrates the HOMO and LUMO electron distribution diagram of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole. When the HOMO and LUMO electron distribution diagrams in FIGS. 2 and 3 are compared, 9H-carbazole has widely spread HOMO orbital and LUMO orbital unlike 4a,9a-dimethyl-2, 3,4,4a,9,9a-hexahydro-1H-carbazole. These widely spread orbitals are caused by the extension of the conjugation, and it can be confirmed that the amine of carbazole affects two benzene rings. Therefore, when 9H-carbazole is used as a dopant of a light emitting layer of an organic light emitting device, 9H-carbazole affects the full width at half maximum and the wavelength, thereby having a wide full width at half maximum.

In contrast, in the case of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, amine affects only one benzene ring, so that when 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole is used as a dopant material, the color purity is high because a device has the narrow full width at half maximum.

Hereinafter, the substituents and terms will be described. When one part "includes" one constituent element in the present specification, unless otherwise specifically described, this does not mean that another constituent element is excluded, but means that another constituent element can be further included.

When one member is disposed "on" another member in the present specification, this includes not only a case where the one member is brought into contact with another member, but also a case where still another member is present 50 between the two members.

In the present specification,

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means a moiety bonded to another substituent or a bonding portion.

Examples of the substituents in the present specification will be described below, but are not limited thereto.

The term "substitution" means that a hydrogen atom bonded to a carbon atom of a compound is changed into another substituent, and a position to be substituted is not limited as long as the position is a position at which the hydrogen atom is substituted, that is, a position at which the

substituent can be substituted, and when two or more are substituted, the two or more substituents can be the same as or different from each other.

In the present specification, the term "substituted or unsubstituted" means being substituted with one or two or 5 more substituents selected from the group consisting of deuterium (—D), a halogen group, a nitrile group (—CN), a silyl group, a boron group, an alkyl group, an alkenyl group, an alkynyl group, a cycloalkyl group, an alkoxy group, an aryloxy group, an amine group, an aryl group, and a heterocyclic group, being substituted with a substituent to which two or more substituents among the substituents are linked, or having no substituent. For example, "the substituent to which two or more substituents are linked" can be a biphenyl group. That is, the biphenyl group can also be an aryl group, and can be interpreted as a substituent to which two phenyl groups are linked.

In the present specification, the term "substituted with A or B" includes i) the case of being substituted with only A,  $_{20}$  ii) the case of being substituted with only B, and iii) the case of being substituted with A and B.

Examples of the substituents will be described below, but are not limited thereto.

In the present specification, examples of a halogen group 25 include fluorine (F), chlorine (Cl), bromine (Br) or iodine (I).

In the present specification, a silyl group can be —Si (Y101) (Y102) (Y103), and Y101, Y102, and Y103 can be each hydrogen, a substituted or unsubstituted alkyl group, or 30 a substituted or unsubstituted aryl group. Examples of the silyl group include a trialkylsilyl group and a triarylsilyl group, and specific examples thereof include a trimethylsilyl group, a triethylsilyl group, a t-butyldimethylsilyl group, a vinyldimethylsilyl group, a propyldimethylsilyl group, a 35 triphenylsilyl group, a diphenylsilyl group, a phenylsilyl group, and the like, but the examples are not limited thereto.

In the present specification, a boron group can be —B(Y104) (Y105), and Y104 and Y105 can be each hydrogen, a substituted or unsubstituted alkyl group, or a substituted or unsubstituted aryl group. Specific examples of the boron group include a trimethylboron group, a triethylboron group, a t-butyldimethylboron group, a triphenylboron group, a phenylboron group, and the like, but are not limited thereto.

In the present specification, the alkyl group can be straight-chained or branched, and the number of carbon atoms thereof is not particularly limited, but is preferably 1 to 60. According to an exemplary embodiment, the number of carbon atoms of the alkyl group is 1 to 30. According to 50 another exemplary embodiment, the number of carbon atoms of the alkyl group is 1 to 20. According to still another exemplary embodiment, the number of carbon atoms of the alkyl group is 1 to 10. Specific examples of the alkyl group include a methyl group, an ethyl group, a propyl group, an isopropyl group, a butyl group, an isobutyl group, a tertbutyl group, a pentyl group, a hexyl group, a heptyl group, an octyl group, and the like, but are not limited thereto.

In the present specification, a cycloalkyl group is not particularly limited, but has preferably 3 to 60 carbon atoms, 60 and according to an exemplary embodiment, the number of carbon atoms of the cycloalkyl group is 3 to 30. According to another exemplary embodiment, the number of carbon atoms of the cycloalkyl group is 3 to 20. According to still another exemplary embodiment, the number of carbon 65 atoms of the cycloalkyl group is 3 to 6. Specific examples thereof include a cyclopropyl group, a cyclobutyl group, a

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cyclopentyl group, a cyclohexyl group, a cycloheptyl group, a cyclooctyl group, and the like, but are not limited thereto.

In the present specification, an amine group can be selected from the group consisting of —NH<sub>2</sub>; an alkylamine group; an arylalkylamine group; an arylamine group; an arylamine group; an alkylheteroarylamine group; and a heteroarylamine group, and is not limited thereto. The number of carbon atoms of the amine group is not particularly limited, but is preferably 1 to 60.

In the present specification, the number of carbon atoms of an alkylamine group is not particularly limited, but can be 1 to 40, and can be 1 to 20 according to an exemplary embodiment. Specific examples of the alkylamine group include a methylamine group, a dimethylamine group, an ethylamine group, and the like, but are not limited thereto.

In the present specification, examples of the arylamine group include a substituted or unsubstituted monoarylamine group, a substituted or unsubstituted diarylamine group, or a substituted or unsubstituted triarylamine group. The aryl group in the arylamine group can be a monocyclic aryl group or a polycyclic aryl group. The arylamine group including the two or more aryl groups can include a monocyclic aryl group, a polycyclic aryl group, or both a monocyclic aryl group and a polycyclic aryl group.

Specific examples of the arylamine group include a phenylamine group, a naphthylamine group, a biphenylamine group, an anthracenylamine group, a diphenylamine group, a phenylnaphthylamine group, a biphenylamine group, a dibiphenylamine group, a fluorenylphenylamine group, and the like, but are not limited thereto.

In the present specification, examples of the heteroarylamine group include a substituted or unsubstituted monoheteroarylamine group, a substituted or unsubstituted diheteroarylamine group, or a substituted or unsubstituted triheteroarylamine group.

The heteroaryl group in the heteroarylamine group can be a monocyclic heteroaryl group or a polycyclic heteroaryl group. The heteroarylamine group including two or more heteroaryl groups can include a monocyclic heteroaryl group, a polycyclic heteroaryl group, or both a monocyclic heteroaryl group and a polycyclic heteroaryl group.

In the present specification, an arylheteroarylamine group means an amine group substituted with an aryl group and a 45 heteroaryl group.

In the present specification, an arylalkylamine group means an amine group substituted with an aryl group and an alkyl group.

In the present specification, an alkylheteroarylamine group means an amine group substituted with an alkyl group and a heteroaryl group.

In the present specification, an aryl group is not particularly limited, but has preferably 6 to 60 carbon atoms, and can be a monocyclic aryl group or a polycyclic aryl group. According to an exemplary embodiment, the number of carbon atoms of the aryl group is 6 to 30. According to an exemplary embodiment, the number of carbon atoms of the aryl group is 6 to 20. Examples of the monocyclic aryl group include a phenyl group, a biphenyl group, a terphenyl group, and the like, but are not limited thereto. Examples of the polycyclic aryl group include a naphthyl group, an anthracenyl group, a phenanthrenyl group, a pyrenyl group, a perylenyl group, a triphenyl group, a chrysenyl group, a fluorenyl group, and the like, but are not limited thereto.

In the present specification, a fluorenyl group can be substituted, and two substituents can be bonded to each other to form a spiro structure.

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When the fluorenyl group is substituted, the fluorenyl group can be a spiro fluorenyl group such as

and a substituted fluorenyl group such as

(a 9,9-dimethylfluorenyl group) and

(a 9,9-diphenylfluorenyl group). However, the substituent is not limited thereto.

In the present specification, a heterocyclic group is a cyclic group including one or more of N, O, P, S, Si, and Se <sup>40</sup> as a heteroatom, and the number of carbon atoms thereof is not particularly limited, but is preferably 2 to 60. According to an exemplary embodiment, the number of carbon atoms of the heterocyclic group is 2 to 30. Examples of the heterocyclic group include a pyridine group, a pyrrole group, a pyrimidine group, a pyridazinyl group, a furan group, a thiophene group, an imidazole group, a pyrazole group, a dibenzofuran group, a dibenzothiophene group, a carbazole group, and the like, but are not limited thereto.

In the present specification, the alkenyl group can be straight-chained or branched as a substituent including a double bond between a carbon atom and a carbon atom, and the number of carbon atoms thereof is not particularly 55 limited, but is preferably 2 to 40. According to an exemplary embodiment, the number of carbon atoms of the alkenyl group is 2 to 20. According to another exemplary embodiment, the number of carbon atoms of the alkenyl group is 2 to 10. Specific examples thereof include vinyl, 1-propenyl, isopropenyl, 1-butenyl, 2-butenyl, 3-butenyl, 1-pentenyl, and the like, but are not limited thereto.

In the present specification, the alkynyl group can be straight-chained or branched as a substituent including a 65 triple bond between a carbon atom and a carbon atom, and the number of carbon atoms thereof is not particularly

limited, but is preferably 2 to 40. According to an exemplary embodiment, the number of carbon atoms of the alkenyl group is 2 to 20. According to another exemplary embodiment, the number of carbon atoms of the alkenyl group is 2 to 10.

In the present specification, the alkoxy group can be straight-chained, branched, or cyclic. The number of carbon atoms of the alkoxy group is not particularly limited, but is preferably 1 to 40. Specific examples thereof include methoxy, ethoxy, n-propoxy, isopropoxy, and the like, but are not limited thereto.

A substituent including an alkyl group, an alkoxy group, and other alkyl group moieties described in the present specification includes both a straight-chained form and a branch-chained form.

In the present specification, the above-described description on the aryl group can be applied to an aryl of an aryloxy group.

In the present specification, in a substituted or unsubstituted ring formed by bonding substituents, the "ring" means a hydrocarbon ring; or a hetero ring.

The hydrocarbon ring can be an aromatic ring, an aliphatic ring, or a fused ring of the aromatic ring and the aliphatic ring, and can be selected from the examples of the cycloalkyl group or the aryl group, except for a divalent hydrocarbon ring.

In the present specification, the description on the aryl group can be applied to an aromatic hydrocarbon ring except 35 for a divalent aromatic hydrocarbon ring.

The description on the heterocyclic group can be applied to the hetero ring except for a divalent hetero ring.

In the present specification, the aromatic hydrocarbon ring means a planar ring in which pi electrons are completely conjugated.

In the present specification, an aliphatic hydrocarbon ring means all hydrocarbon rings except for aromatic hydrocarbon rings. A substituted aliphatic hydrocarbon ring also includes an aliphatic hydrocarbon ring in which aromatic rings are fused.

In the present specification, "substituents are bonded to 50 each other to form an aliphatic hydrocarbon ring" means that one hydrocarbon ring formed by linking the two corresponding substituents is an aliphatic ring. "One hydrocarbon ring formed by linking the two corresponding substituents" refers to a ring including all the two corresponding substituents. Not only an aliphatic hydrocarbon ring, but also an aliphatic hetero ring, an aromatic hydrocarbon ring, or an aromatic hetero ring can be fused to the aliphatic hydrocarbon ring formed by linking the two corresponding substituents. For example, the case where the following  $Y_{106}$  and  $Y_{107}$  are bonded to each other to form an aliphatic hydrocarbon ring also includes the case of including not only the following 1 (forming a cyclohexane ring) or 2 (forming a cyclohexene ring), but also the following 3 (a cyclohexane ring is fused to a cyclohexane ring) or 4 (a benzene ring is fused to a cyclohexane ring):

Hereinafter, the compound of the present invention will be described.

In an exemplary embodiment of the present specification, A1 and A2 are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, a substituted or unsub- 35 stituted alkenyl group having 2 to 20 carbon atoms, a substituted or unsubstituted alkynyl group having 2 to 20 carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 20 carbon atoms, a substituted or unsubstituted aryloxy group having 6 to 30 carbon atoms, a substituted or 40 unsubstituted cycloalkyl group having 3 to 30 carbon atoms, a substituted or unsubstituted aryl group having 6 to 30 carbon atoms, or a substituted or unsubstituted heterocyclic group having 2 to 30 carbon atoms, or are bonded to each other to form a substituted or unsubstituted ring having 3 to 45 60 carbon atoms.

According to an exemplary embodiment of the present specification, A1 and A2 are the same as or different from each other, and are each independently hydrogen; deuterium; a halogen group; a nitrile group; a substituted or 50 unsubstituted alkyl group having 1 to 20 carbon atoms; a substituted or unsubstituted aryl group having 6 to 30 carbon atoms; or a substituted or unsubstituted heterocyclic group having 2 to 30 carbon atoms, or are bonded to each other to form a substituted or unsubstituted ring having 3 to 60 55 carbon atoms.

According to another exemplary embodiment, A1 and A2 are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group 60 having 1 to 20 carbon atoms, or a substituted or unsubstituted aryl group having 6 to 30 carbon atoms, or are bonded to each other to form a substituted or unsubstituted ring having 3 to 60 carbon atoms.

In still another exemplary embodiment, A1 and A2 are the 65 same as or different from each other, and are each independently hydrogen, deuterium, an alkyl group having 1 to 20

carbon atoms, or an aryl group having 3 to 30 carbon atoms, which is unsubstituted or substituted with an alkyl group having 1 to 20 carbon atoms, or are bonded to each other to form a ring having 3 to 60 carbon atoms, which is unsubstituted or substituted with one or more substituents selected from the group consisting of deuterium, a halogen group, a trialkylsilyl group having 1 to 20 carbon atoms, a triarylsilyl group having 6 to 30 carbon atoms, an alkyl group having 1 to 20 carbon atoms, and an aryl group having 6 to 30 carbon atoms.

In yet another exemplary embodiment, A1 and A2 are the same as or different from each other, and are each independently hydrogen, deuterium, an alkyl group having 1 to 4 carbon atoms, or an aryl group having 6 to 12 carbon atoms, which is unsubstituted or substituted with an alkyl group having 1 to 4 carbon atoms or an aryl group having 6 to 12 carbon atoms, or are bonded to each other to form a ring having 3 to 60 carbon atoms, which is unsubstituted or substituted with one or more substituents selected from the group consisting of deuterium, a halogen group, a trimethylsilyl group, a triphenylsilyl group, an alkyl group having 1 to 4 carbon atoms, and an aryl group having 6 to 12 carbon atoms.

According to still yet another exemplary embodiment, A1 and A2 are the same as or different from each other, and are each independently hydrogen; deuterium; a methyl group; a phenyl group which is unsubstituted or substituted with a methyl group, a tert-butyl group, a phenyl group, or a naphthyl group; or a biphenyl group which is unsubstituted or substituted with a tert-butyl group, or are bonded to each other to form a ring having 3 to 60 carbon atoms, which is unsubstituted or substituted with one or more substituents selected from the group consisting of deuterium, a halogen group, a trimethylsilyl group, a triphenylsilyl group, a methyl group, an ethyl group, a tert-butyl group, and a phenyl group.

In a further exemplary embodiment, A1 and A2 are the same as or different from each other, and are each independently hydrogen; deuterium; an alkyl group having 1 to 20 carbon atoms; or an aryl group having 3 to 30 carbon atoms, which is unsubstituted or substituted with an alkyl group having 1 to 20 carbon atoms, or are bonded to each other to form a five-membered ring, which is unsubstituted or substituted with one or more substituents selected from the group consisting of deuterium, a halogen group, a trialkylsilyl group having 1 to 20 carbon atoms, a triarylsilyl group having 6 to 30 carbon atoms, an alkyl group having 1 to 20 carbon atoms, and an aryl group having 6 to 30 carbon atoms and in which an aliphatic hydrocarbon ring or an aromatic hydrocarbon ring is fused or unfused.

According to another further exemplary embodiment, A1 and A2 are the same as or different from each other, and are each independently hydrogen; deuterium; a methyl group; a phenyl group which is unsubstituted or substituted with a methyl group, a tert-butyl group, a phenyl group, or a naphthyl group; or a biphenyl group which is unsubstituted or substituted with a tert-butyl group, or are bonded to each other to form a five-membered ring, which is unsubstituted or substituted with one or more substituents selected from the group consisting of deuterium, a halogen group, a trimethylsilyl group, a triphenylsilyl group, a methyl group, an ethyl group, a tert-butyl group, and a phenyl group and in which a monocyclic to tricyclic aliphatic hydrocarbon ring or a monocyclic to tricyclic aromatic hydrocarbon ring is fused or unfused.

According to an exemplary embodiment of the present specification, Formula 1 is any one of the following Formula 1-1 or 1-2:

$$\begin{array}{c|c} & & & & & \\ & & & & \\ Ar_5 & & & & \\ Ar_7 & Ar_8 & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ &$$

In Formulae 1-1 and 1-2:

the definitions of  $R_1$  to  $R_3$ ,  $Ar_1$  to  $Ar_4$ , and n1 to n3 are the 30 same as those defined in Formula 1;

A11 and A12 are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted alkenyl group, a substituted alkynyl group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted or unsubstituted aryloxy group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted aryl group, or a substituted or unsubstituted heterocyclic group; and

 ${
m Ar}_5$  to  ${
m Ar}_8$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted alkynyl group, a substituted alkynyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted aryl group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heterocyclic group, or adjacent groups are bonded to each other to form a substituted or unsubstituted ring.

In an exemplary embodiment of the present specification, A11 and A12 are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms; a substituted or unsubstituted alkenyl group having 2 to 20 carbon atoms, a substituted or unsubstituted alkynyl group having 2 to 20 carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 20 carbon atoms, a substituted or unsubstituted aryloxy group having 6 to 30 carbon atoms; a substituted or unsubstituted aryl group having 6 to 30

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carbon atoms, or a substituted or unsubstituted heterocyclic group having 2 to 30 carbon atoms.

According to another exemplary embodiment, A11 and A12 are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, or a substituted or unsubstituted aryl group having 6 to 30 carbon atoms.

According to still another exemplary embodiment, A11 and A12 are the same as or different from each other, and are each independently hydrogen, deuterium, an alkyl group having 1 to 4 carbon atoms; or an aryl group having 6 to 12 carbon atoms, which is unsubstituted or substituted with an alkyl group having 1 to 4 carbon atoms or an aryl group having 6 to 12 carbon atoms.

In yet another exemplary embodiment, A11 and A12 are the same as or different from each other, and are each independently hydrogen, deuterium, a substituted or unsubstituted methyl group, a substituted or unsubstituted phenyl group, or a substituted or unsubstituted biphenyl group.

In still yet another exemplary embodiment, A11 and A12 are the same as or different from each other, and are each independently hydrogen, deuterium, a methyl group, a phe25 nyl group which is unsubstituted or substituted with a methyl group, a tert-butyl group, a phenyl group, or a naphthyl group, or a biphenyl group which is unsubstituted or substituted with a tert-butyl group.

According to an exemplary embodiment of the present specification, A11 is an aryl group having 6 to 12 carbon atoms, which is unsubstituted or substituted with an alkyl group having 1 to 4 carbon atoms or an aryl group having 6 to 12 carbon atoms.

In another exemplary embodiment, A11 is a phenyl group which is unsubstituted or substituted with a methyl group, a tert-butyl group, a phenyl group, or a naphthyl group; or a biphenyl group which is unsubstituted or substituted with a tert-butyl group.

According to an exemplary embodiment of the present specification, A12 is hydrogen, deuterium, or an alkyl group having 1 to 4 carbon atoms.

In another exemplary embodiment, A12 is hydrogen, deuterium, or a methyl group.

According to an exemplary embodiment of the present specification, Ar<sub>1</sub> to Ar<sub>4</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, a substituted or unsubstituted alkenyl group having 2 to 20 carbon atoms, a substituted or unsubstituted alkynyl group having 2 to 20 carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 20 carbon atoms, a substituted or unsubstituted aryloxy group having 6 to 30 carbon atoms, a substituted or unsubstituted cycloalkyl group having 3 to 30 carbon atoms; 55 a substituted or unsubstituted aryl group having 6 to 30 carbon atoms, or a substituted or unsubstituted heterocyclic group having 2 to 30 carbon atoms, or adjacent groups are bonded to each other to form a substituted or unsubstituted aliphatic hydrocarbon ring having 3 to 60 carbon atoms.

According to another exemplary embodiment,  $Ar_1$  to  $Ar_4$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, or a substituted or unsubstituted aryl group having 6 to 30 carbon atoms, or adjacent groups are bonded to each other to form a substituted or unsubstituted monocyclic to tricyclic aliphatic hydrocarbon

ring having 3 to 60 carbon atoms, in which an aliphatic hydrocarbon ring or an aromatic hydrocarbon ring is fused or unfused.

In still another exemplary embodiment,  ${\rm Ar_1}$  to  ${\rm Ar_4}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted methyl group, a substituted or unsubstituted ethyl group; or a substituted or unsubstituted phenyl group, or adjacent groups are bonded to each other to form a substituted or unsubstituted monocyclic to tricyclic aliphatic hydrocarbon ring having 3 to 60 carbon atoms, in which an aliphatic hydrocarbon ring or an aromatic hydrocarbon ring is fused or unfused.

In yet another exemplary embodiment, Ar<sub>1</sub> to Ar<sub>4</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, a substituted or unsubstituted aryl group having 6 to 30 carbon atoms, or adjacent groups are bonded to each other to form a six-membered aliphatic hydrocarbon ring in which an aliphatic hydrocarbon ring or an aromatic hydrocarbon ring is fused or unfused.

In still yet another exemplary embodiment,  ${\rm Ar_1}$  to  ${\rm Ar_4}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, a substituted or unsubstituted aryl group having 6 to 30 carbon atoms, or adjacent groups are bonded to each other to form a six-membered aliphatic hydrocarbon ring in which monocyclic to bicyclic aliphatic hydrocarbon rings or monocyclic to bicyclic aromatic hydrocarbon rings are fused or unfused.

In a further exemplary embodiment,  $Ar_1$  to  $Ar_4$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, a substituted or unsubstituted aryl group having 6 to 30 carbon atoms, or adjacent groups are bonded to each other to form a six-membered aliphatic hydrocarbon ring in which one or two cyclohexane(s) or benzene(s) is or are fused or unfused. 45

According to another further exemplary embodiment,  ${\rm Ar_1}$  to  ${\rm Ar_4}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a methyl group, an ethyl group, or a phenyl group, or adjacent groups are bonded to each other to form a substituted or unsubstituted cyclohexane, a substituted or unsubstituted tetradecahydrophenanthrene, a substituted or unsubstituted tetrahydronaphthalene, or a substituted or unsubstituted decahydronaphthalene.

In another further exemplary embodiment, two of  $\mathrm{Ar_1}$  to  $\mathrm{Ar_4}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a methyl group, an ethyl group, or a phenyl group, and the other two are bonded to each other to form cyclohexane, tetradecahydrophenanthrene, tetrahydronaphthalene, or decahydronaphthalene.

According to an exemplary embodiment of the present specification, when two of  $Ar_1$  to  $Ar_4$  are bonded to each  $_{65}$  other to form an aliphatic hydrocarbon ring, any one ring selected from the following rings is formed:

In the ring, Ar<sub>11</sub> and Ar<sub>12</sub> are substituents which do not form an aliphatic hydrocarbon ring in Ar<sub>1</sub> to Ar<sub>4</sub>, and are the same as or different from each other; and

the ring is unsubstituted or substituted with deuterium, an alkyl group having 1 to 10 carbon atoms, or an aryl group having 6 to 30 carbon atoms.

According to another exemplary embodiment, the ring is unsubstituted or substituted with deuterium.

According to still another exemplary embodiment, the ring is unsubstituted. That is, the ring does not have another substituent except for  $Ar_{11}$  and  $Ar_{12}$ .

According to yet another exemplary embodiment,  $Ar_{11}$  and  $Ar_{12}$  are each independently a straight-chained or branched alkyl group having 1 to 4 carbon atoms; or an aryl group having 6 to 20 carbon atoms.

In still yet another exemplary embodiment, Ar<sub>11</sub> and Ar<sub>12</sub>
55 are the same as or different from each other, and are each independently a methyl group or a phenyl group.

In a further exemplary embodiment,  $Ar_{11}$  and  $Ar_{12}$  are a methyl group.

According to an exemplary embodiment of the present specification, two of  ${\rm Ar_1}$  to  ${\rm Ar_4}$  are the same as or different from each other, and are each independently a substituted or unsubstituted straight-chained or branched alkyl group having 1 to 20 carbon atoms, or an aryl group having 6 to 30 carbon atoms. In this case, the present invention has an effect in which the efficiency of a device is increased by preventing the phenomenon of an aggregation among compounds to suppress the quenching.

According to another exemplary embodiment, two of  ${\rm Ar_1}$  to  ${\rm Ar_4}$  are the same as or different from each other, and are each independently a straight-chained or branched alkyl group having 1 to 4 carbon atoms, or an aryl group having 6 to 20 carbon atoms.

According to still another exemplary embodiment, two of  $Ar_1$  to  $Ar_4$  are the same as or different from each other, and are each independently a methyl group or a phenyl group.

In yet another exemplary embodiment, two of  $Ar_1$  to  $Ar_4$  are a methyl group.

In an exemplary embodiment of the present specification, Formula 1 is any one of the following Formula 2-1 or 2-2:

Formula 2-1 15

$$R_{31}$$
 $R_{32}$ 
 $R_{31}$ 
 $R_{33}$ 
 $R_{33}$ 
 $R_{33}$ 

 $(R_3)_{n3}$ 

Formula 2-2

25

35

In Formulae 2-1 and 2-2:

the definitions of A1, A2,  $R_1$  to  $R_3$ , and n1 to n3 are the same as those defined in Formula 1;

 $\rm R_{31}$  to  $\rm R_{35}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl 45 group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted alkenyl group, a substituted alkenyl group, a substituted alkenyl group, a substituted alkoxy group, a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted aryloxy group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heterocyclic group;

adjacent R<sub>35</sub>'s can be bonded to each other to form a substituted or unsubstituted aliphatic hydrocarbon ring;

r33 is an integer from 0 to 8;

r34 and r35 are each an integer from 0 to 4; and

when r33 to r35 are each 2 or more, the substituents in a plurality of parentheses are the same as or different from each other.

According to an exemplary embodiment of the present specification,  $R_{31}$  to  $R_{35}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, an alkyl group having 1 to 4 carbon atoms, or an aryl group having 6 to 20 carbon atoms.

According to an exemplary embodiment of the present specification,  $R_{31}$  and  $R_{32}$  are the same as or different from

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each other, and are each independently hydrogen, deuterium, a halogen group, an alkyl group having 1 to 4 carbon atoms, or an aryl group having 6 to 20 carbon atoms.

In another exemplary embodiment,  $R_{31}$  and  $R_{32}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a methyl group, an ethyl group, or a phenyl group.

According to an exemplary embodiment of the present specification,  $R_{33}$  to  $R_{35}$  are the same as or different from each other, and are each independently hydrogen or deuterium

In another exemplary embodiment,  $R_{33}$  to  $R_{35}$  are each hydrogen.

According to an exemplary embodiment of the present specification, adjacent R<sub>35</sub>'s are bonded to each other to form a substituted or unsubstituted aliphatic hydrocarbon ring.

In another exemplary embodiment, adjacent  $R_{35}$ 's are bonded to each other to form a substituted or unsubstituted cyclohexane.

In still another exemplary embodiment, adjacent  $R_{35}$ 's are bonded to each other to form one or two cyclohexane(s).

In an exemplary embodiment of the present specification, r33 to r35 are each 0.

According to an exemplary embodiment of the present specification, Formula 1-1 is any one of the following Formulae 2 to 7:

Formula 2
$$Ar_{5}$$

$$Ar_{6}$$

$$Ar_{7}$$

$$Ar_{8}$$

$$R_{11}$$

$$R_{12}$$

$$R_{13}$$

$$R_{14}$$

$$R_{13}$$

Formula 3

$$Ar_5$$
 $Ar_7$ 
 $Ar_8$ 
 $R_{23}$ 
 $R_{22}$ 
 $R_{21}$ 
 $R_{21}$ 

Formula 4

$$Ar_5$$
 $Ar_7$ 
 $Ar_8$ 
 $R_{23}$ 
 $R_{23}$ 
 $R_{24}$ 
 $R_{24}$ 
 $R_{24}$ 

Formula 5

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-continued

$$Ar_{5}$$
 $Ar_{7}$ 
 $Ar_{8}$ 
 $R_{23}$ 
 $R_{23}$ 
 $R_{23}$ 
 $R_{25}$ 
 $R_{25}$ 

Formula 6
$$Ar_{5}$$

$$Ar_{6}$$

$$Ar_{7}$$

$$Ar_{8}$$

$$R_{23}$$

$$R_{23}$$

$$R_{23}$$

$$R_{24}$$

$$R_{25}$$

$$R_{25}$$

$$R_{26}$$

$$R_{26}$$

$$R_{26}$$

In Formulae 2 to 7:

definitions of  $R_1$  to  $R_3$ ,  $Ar_5$  to  $Ar_8$ , and n1 to n3 are the same as those defined in Formula 1-1;

 $R_{11}$  to  $R_{14}$  and  $R_{21}$  to  $R_{27}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkenyl group, a substituted alkoxy group, a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted aryl group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heterocyclic group:

p1 is an integer from 0 to 8;

p2 to p4 are each an integer from 0 to 14;

p5 is an integer from 0 to 20; and

when p1 to p5 are each 2 or more, the substituents in a plurality of parentheses are the same as or different from 60 each other.

According to an exemplary embodiment of the present specification,  $R_{11}$  to  $R_{14}$  and  $R_{21}$  to  $R_{27}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, a substituted or unsubstituted alkenyl group having 2

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to 20 carbon atoms, a substituted or unsubstituted alkynyl group having 2 to 20 carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 20 carbon atoms, a substituted or unsubstituted aryloxy group having 6 to 30 carbon atoms, a substituted or unsubstituted cycloalkyl group having 3 to 30 carbon atoms, a substituted or unsubstituted aryl group having 6 to 30 carbon atoms, or a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted heterocyclic group having 2 to 30 carbon atoms.

According to another exemplary embodiment,  $R_{11}$  to  $R_{14}$  and  $R_{21}$  to  $R_{27}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, or a substituted or unsubstituted aryl group having 6 to 30 carbon atoms.

In still another exemplary embodiment,  $R_{11}$  to  $R_{14}$  and  $R_{21}$  to  $R_{27}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted methyl group, a substituted or unsubstituted ethyl group, or a substituted or unsubstituted phenyl group.

In yet another exemplary embodiment, R<sub>11</sub> to R<sub>14</sub> and R<sub>21</sub> to R<sub>27</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, 25 a methyl group, an ethyl group, or a phenyl group.

According to an exemplary embodiment of the present specification, R<sub>11</sub> to R<sub>14</sub>, R<sub>22</sub>, and R<sub>23</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, an alkyl group having 1 to 4 carbon atoms, or an aryl group having 6 to 20 carbon atoms.

In another exemplary embodiment, R<sub>11</sub> to R<sub>14</sub>, R<sub>22</sub>, and R<sub>23</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a methyl group, an ethyl group, or a phenyl group.

According to an exemplary embodiment of the present specification,  $R_{21}$  and  $R_{24}$  to  $R_{27}$  are the same as or different from each other, and are each independently hydrogen or deuterium.

In another exemplary embodiment,  $R_{21}$  and  $R_{24}$  to  $R_{27}$  are each hydrogen.

According to an exemplary embodiment of the present specification, p1 to p5 are each an integer from 0 to 2, and when p1 to p5 are each 2 or more, substituents in a plurality of parentheses are the same as or different from each other.

According to another exemplary embodiment, p1 to p5 are each 0 or 1.

According to an exemplary embodiment of the present specification, Ar<sub>5</sub> to Ar<sub>8</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, a substituted or unsubstituted alkenyl group having 2 to 20 carbon atoms, a substituted or unsubstituted alkynyl group having 2 to 20 carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 20 carbon atoms, a substituted or unsubstituted aryloxy group having 6 to 30 carbon atoms, a substituted or unsubstituted cycloalkyl group having 3 to 30 carbon atoms, a substituted or unsubstituted aryl group having 6 to 60 carbon atoms, or a substituted or unsubstituted heterocyclic group having 2 to 60 carbon atoms, or adjacent groups are bonded to each other to form a substituted or unsubstituted ring having 3 to 60 carbon atoms.

According to an exemplary embodiment of the present specification, Ar<sub>5</sub> to Ar<sub>8</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubsti-

tuted alkyl group having 1 to 20 carbon atoms, or a substituted or unsubstituted aryl group having 6 to 30 carbon atoms, or adjacent groups are bonded to each other to form a substituted or unsubstituted ring having 3 to 60 carbon atoms.

In an exemplary embodiment of the present specification,  $Ar_5$  to  $Ar_8$  are bonded to each other to form a substituted or unsubstituted aromatic hydrocarbon ring, or any one of  $Ar_5$  and  $Ar_6$  and any one of  $Ar_7$  and  $Ar_8$  are bonded to each other to form a substituted or unsubstituted aliphatic hydrocarbon 10 ring, and groups which do not form a ring among  $Ar_5$  to  $Ar_8$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, or a substituted or unsubstituted aryl group having 6 to 30 carbon atoms.

In another exemplary embodiment,  $Ar_5$  to  $Ar_8$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted methyl group, a substituted or unsubstituted ethyl group, or a substituted or unsubstituted phenyl group, or adjacent groups are bonded to each other to form a substituted or unsubstituted ring having 3 to 60 carbon atoms.

In still another exemplary embodiment,  ${\rm Ar_5}$  to  ${\rm Ar_8}$  are the 25 same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted methyl group, a substituted or unsubstituted or unsubstituted or unsubstituted phenyl group, or adjacent groups are bonded 30 to each other to form a ring having 3 to 60 carbon atoms, which is unsubstituted or substituted with one or more substituents selected from the group consisting of an alkyl group having 1 to 20 carbon atoms, a trialkylsilyl group having 1 to 20 carbon atoms, a trialylsilyl group having 6 to 30 carbon atoms, and an aryl group having 6 to 30 carbon atoms.

According to yet another exemplary embodiment,  $Ar_5$  to  $Ar_8$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a 40 nitrile group, a substituted or unsubstituted methyl group, a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted phenyl group, or adjacent groups are bonded to each other to form a substituted or unsubstituted aromatic hydrocarbon ring having 6 to 60 carbon atoms, or an 45 aliphatic hydrocarbon ring having 3 to 60 carbon atoms.

According to still yet another exemplary embodiment,  ${\rm Ar_5}$  to  ${\rm Ar_8}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted methyl group, 50 a substituted or unsubstituted ethyl group, or a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted or each other to form an aromatic hydrocarbon ring having 6 to 60 carbon atoms, which is unsubstituted or substituted with one or more substituents selected from the group 55 consisting of an alkyl group having 1 to 20 carbon atoms, a triarlylsilyl group having 6 to 30 carbon atoms, and an aryl group having 6 to 30 carbon atoms, or an aliphatic hydrocarbon ring having 3 to 60 carbon atoms.

In a further exemplary embodiment,  $Ar_5$  to  $Ar_8$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted methyl group, a substituted or unsubstituted ethyl group, or a substituted or unsubstituted phenyl group, or adjacent groups are bonded to each other to form a substituted or unsubstituted cyclo-

hexane, a substituted or unsubstituted tetradecaphenanthrene, a substituted or unsubstituted decahydronaphthalene, a substituted or unsubstituted benzene, or a substituted or unsubstituted naphthalene.

In another further exemplary embodiment,  ${\rm Ar}_5$  to  ${\rm Ar}_8$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a methyl group, an ethyl group, or a phenyl group, or adjacent groups are bonded to each other to form cyclohexane, tetradecaphenanthrene, decahydronaphthalene, a benzene which is unsubstituted or substituted with one or more substituents selected from the group consisting of a methyl group, a tert-butyl group, a trimethylsilyl group, a triphenylsilyl group, and a phenyl group, or a naphthalene, which is unsubstituted or substituted with one or more substituents selected from the group consisting of a methyl group, a tert-butyl group, a trimethylsilyl group, a triphenylsilyl group, and a phenyl group.

In the present specification, the case where adjacent groups among  $Ar_5$  to  $Ar_8$  are bonded to each other to form a ring means that i) two of  $Ar_5$  to  $Ar_8$  are bonded to each other to form an aliphatic hydrocarbon ring, or ii) all of  $Ar_5$  to  $Ar_8$  participate in the formation of a ring to form an aromatic hydrocarbon ring.

According to an exemplary embodiment of the present specification, when adjacent groups of  $Ar_5$  to  $Ar_8$  are bonded to each other to form a ring, any one ring selected from the rings of the following Group A or B is formed:

Group A

In the rings of Groups A and B,  $Ar_{13}$  and  $Ar_{14}$  are substituents which do not form a ring among  $Ar_5$  to  $Ar_8$ , and are the same as or different from each other.

The rings of Groups A and B are unsubstituted or substituted with one or more substituents selected from the group consisting of deuterium, an alkyl group having 1 to 20 carbon atoms, a trialkylsilyl group having 1 to 20 carbon atoms, a triarylsilyl group having 6 to 30 carbon atoms, and an aryl group having 6 to 30 carbon atoms.

According to an exemplary embodiment of the present specification, the rings of Group A are unsubstituted. That is, the rings do not have another substituent except for  $Ar_{13}$  and  $^{25}$   $Ar_{14}$ .

According to an exemplary embodiment of the present specification, the rings of Group B are unsubstituted or substituted with one or more substituents selected from the group consisting of deuterium, an alkyl group having 1 to 20 carbon atoms, a trialkylsilyl group having 1 to 20 carbon atoms, a triarylsilyl group having 6 to 30 carbon atoms, and an aryl group having 6 to 30 carbon atoms.

In another exemplary embodiment, the rings of Group B <sup>35</sup> are unsubstituted or substituted with one or more substituents selected from the group consisting of a methyl group, a tert-butyl group, a trimethylsilyl group, and a phenyl group.

According to still another exemplary embodiment,  $Ar_{13}$  and  $Ar_{14}$  are each independently a straight-chained or branched alkyl group having 1 to 4 carbon atoms, or an aryl group having 6 to 20 carbon atoms.

In yet another exemplary embodiment,  $Ar_{13}$  and  $Ar_{14}$  are  $^{45}$  the same as or different from each other, and are each independently a methyl group, or a phenyl group.

In still yet another exemplary embodiment,  $Ar_{13}$  and  $Ar_{14}$  are a methyl group.

According to an exemplary embodiment of the present invention, Formula 1-1 is any one of the following Formulae 8 to 10:

Formula 8

$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $(R_3)_{n3}$ 
 $(R_4)_{n4}$ 
 $(R_5)_{n4}$ 
 $(R_7)_{n4}$ 
 $(R_7)_{n4}$ 

-continued

Formula 9
$$(R_1)_{n1}$$

$$(R_2)_{n2}$$

$$Ar_1$$

$$Ar_2$$

$$(R_3)_{n3}$$

Formula 10

In Formulae 8 to 10:

the definitions of  $R_1$  to  $R_3$ , n1 to n3, and  $Ar_1$  to  $Ar_4$  are the same as those defined in Formula 1-1;

 $R_4$  to  $R_9$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted or unsubstituted or unsubstituted alkyl group, a substituted alkenyl group, a substituted or unsubstituted alkenyl group, a substituted alkoxy group, a substituted or unsubstituted aryl group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heterocyclic group;

 $Y_1$  and  $Y_2$  are bonded to each other to form a substituted or unsubstituted aliphatic hydrocarbon ring; and

Cy1 is a substituted or unsubstituted aromatic hydrocarbon ring.

According to an exemplary embodiment of the present specification,  $R_4$  to  $R_7$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, a substituted or unsubstituted alkenyl group having 2 to 20 carbon atoms, a substituted or unsubstituted alkynyl group having 2 to 20 carbon atoms, a substituted aryloxy group having 1 to 20 carbon atoms, a substituted or unsubstituted aryloxy group having 6 to 30 carbon atoms, a substituted or unsubstituted aryl group having 6 to 30 carbon atoms, or a substituted or unsubstituted heterocyclic group having 2 to 30 carbon atoms.

According to another exemplary embodiment,  $R_4$  to  $R_7$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, or a substituted or unsubstituted aryl group having 6 to 30 carbon atoms.

In still another exemplary embodiment,  $R_4$  to  $R_7$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile

group, a substituted or unsubstituted methyl group, a substituted or unsubstituted ethyl group, or a substituted or unsubstituted phenyl group.

According to yet another exemplary embodiment,  $R_4$  to  $R_7$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a methyl group, an ethyl group, or a phenyl group.

According to an exemplary embodiment,  $R_8$  and  $R_9$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, or a substituted or unsubstituted aryl group having 6 to 30 carbon atoms.

According to another exemplary embodiment,  $R_8$  and  $R_9$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted methyl group, a substituted or unsubstituted ethyl group, or a substituted or unsubstituted phenyl group.

According to still another exemplary embodiment,  $R_8$  and  $R_9$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a methyl group, an ethyl group, or a phenyl group.

According to an exemplary embodiment,  $Y_1$  and  $Y_2$  are bonded to each other to form a substituted or unsubstituted aliphatic hydrocarbon ring having 3 to 60 carbon atoms.

According to another exemplary embodiment,  $Y_1$  and  $Y_2$  are bonded to each other to form a substituted or unsubstituted monocyclic to tricyclic aliphatic hydrocarbon ring having 3 to 60 carbon atoms.

In still another exemplary embodiment,  $Y_1$  and  $Y_2$  are bonded to each other to form a substituted or unsubstituted  $^{35}$  cyclohexane, a substituted or unsubstituted tetradecahydrophenanthrene, or a substituted or unsubstituted decahydronaphthalene.

In yet another exemplary embodiment,  $Y_1$  and  $Y_2$  are bonded to each other to form cyclohexane, tetradecahydrophenanthrene, or decahydronaphthalene.

According to an exemplary embodiment of the present specification, Cy1 is a substituted or unsubstituted aromatic hydrocarbon ring having 6 to 60 carbon atoms.

According to another exemplary embodiment, Cy1 is an aromatic hydrocarbon ring having 6 to 30 carbon atoms, which is unsubstituted or substituted with one or more substituents selected from the group consisting of an alkyl group having 1 to 20 carbon atoms, a trialkylsilyl group having 1 to 20 carbon atoms, a triarylsilyl group having 6 to 30 carbon atoms, and an aryl group having 6 to 30 carbon atoms.

In still another exemplary embodiment, Cy1 is a substituted or unsubstituted benzene, or a substituted or unsubstituted naphthalene.  $^{55}$ 

According to yet another exemplary embodiment, Cy1 is a benzene which is unsubstituted or substituted with one or more substituents selected from the group consisting of a methyl group, a tert-butyl group, a trimethylsilyl group, a triphenylsilyl group, and a phenyl group, or a naphthalene which is unsubstituted or substituted with one or more substituents selected from the group consisting of a methyl group, a tert-butyl group, a trimethylsilyl group, a triphenylsilyl group, and a phenyl group.

In an exemplary embodiment of the present specification,  $R_1$  to  $R_3$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted or unsubstituted alkyl group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted cycloalkyl group, a substituted or unsubstituted aryloxy group, a substituted aryloxy group, a substituted or unsubstituted heterocyclic group.

In another exemplary embodiment, R<sub>1</sub> to R<sub>3</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted triarlylsilyl group, a substituted or unsubstituted triarylsilyl group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group, a substituted arylamine group, or a substituted or unsubstituted heterocyclic group.

In still another exemplary embodiment,  $R_1$  to  $R_3$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a trialkylsilyl group having 1 to 20 carbon atoms, a triarylsilyl group having 6 to 30 carbon atoms, an alkyl group having 1 to 10 carbon atoms, an aryl group having 6 to 30 carbon atoms, an arylamine group having 6 to 50 carbon atoms, which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms, a trialkylsilyl group having 1 to 20 carbon atoms, or a triarylsilyl group having 6 to 30 carbon atoms, a dihydroacridine group which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms or an aryl group having 6 to 30 carbon atoms, a dihydrodibenzoazasiline group which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms or an aryl group having 6 to 30 carbon atoms, a spiro(dibenzosilole-dibenzoazasiline) group, a spiro(acridine-fluorene) group, or a hexahydrocarbazole group which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms or an aryl group having 6 to 30 carbon atoms and in which a benzene ring is fused or unfused.

In yet another exemplary embodiment,  $R_1$  to  $R_3$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a trialkylsilyl group having 1 to 20 carbon atoms, a triarylsilyl group having 6 to 30 carbon atoms, an alkyl group having 1 to 10 carbon atoms, an aryl group having 6 to 30 carbon atoms, an arylamine group having 6 to 50 carbon atoms, which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms, a trialkylsilyl group having 1 to 20 carbon atoms, or a triarylsilyl group having 6 to 30 carbon atoms, a dihydroacridine group

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which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms or an aryl group having 6 to 30 carbon atoms, a dihydrodibenzoazasiline group

which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms or an aryl group having 6 to 30 carbon atoms, or a tetrahydrobenzocarbazole group

10

which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms or an aryl group having 6 to 30 carbon atoms, a spiro(dibenzosilole-dibenzoazasiline) group 20

a spiro(acridine-fluorene) group

a hexahydrocarbazole group

15 which unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms or an aryl group having 6 to 30 carbon atoms.

In still yet another exemplary embodiment, R<sub>1</sub> to R<sub>3</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a trimethylsilyl group, a trimethylsilyl group, a methyl group, a tert-butyl group, a phenyl group, a biphenyl group, a diphenylamine group which is unsubstituted or substituted with a methyl group, a tert-butyl group, a trimethylsilyl group, or a triphenylsilyl group, a dihydroacridine group

which is unsubstituted or substituted with a methyl group, a tert-butyl group, or a phenyl group, a dihydrodibenzoazasiline group

which is unsubstituted or substituted with a methyl group, a tert-butyl group, or a phenyl group, a spiro(dibenzosiloledibenzoazasiline) group

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a spiro(acridine-fluorene) group

a hexahydrocarbazole group

which is unsubstituted or substituted with a methyl group, a tert-butyl group, or a phenyl group, or a tetrahydrobenzo-carbazole group

which is unsubstituted or substituted with a methyl group, a tert-butyl group, or a phenyl group.

In an exemplary embodiment of the present invention,  $R_1$  and  $R_2$  are the same as or different from each other, and are each independently hydrogen, deuterium, a trialkylsilyl group having 1 to 20 carbon atoms, a triarylsilyl group having 6 to 30 carbon atoms, an alkyl group having 1 to 10 carbon atoms, an aryl group having 6 to 30 carbon atoms, or an arylamine group having 6 to 50 carbon atoms, which is unsubstituted or substituted with an alkyl group having 1 to 55 10 carbon atoms, a trialkylsilyl group having 1 to 20 carbon atoms, or a triarylsilyl group having 6 to 30 carbon atoms.

According to another exemplary embodiment,  $R_1$  and  $R_2$  are the same as or different from each other, and are each independently hydrogen, deuterium, a trimethylsilyl group, 60 a trimethylsilyl group, a methyl group, a tert-butyl group, a phenyl group, a biphenyl group, a diphenylamine group which is unsubstituted or substituted with a methyl group, a tert-butyl group, a trimethylsilyl group, or a triphenylsilyl group.

According to an exemplary embodiment of the present invention, R<sub>3</sub> is hydrogen, deuterium, an alkyl group having

1 to 10 carbon atoms, an arylamine group having 6 to 50 carbon atoms, which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms, a trialkylsilyl group having 1 to 20 carbon atoms, or a triarylsilyl group having 6 to 30 carbon atoms, a dihydroacridine group

which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms or an aryl group having 6 to 30 carbon atoms, a dihydrodibenzoazasiline group

which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms or an aryl group having 6 to 30 carbon atoms, a spiro(dibenzosilole-dibenzoazasiline) group

a spiro(acridine-fluorene) group

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a hexahydrocarbazole group

which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms or an aryl group having 6 to 30 carbon atoms, or a tetrahydrobenzocarbazole group

which is unsubstituted or substituted with an alkyl group having 1 to 10 carbon atoms or an aryl group having 6 to 30 carbon atoms.

According to another exemplary embodiment,  $R_3$  is  $^{35}$ hydrogen, deuterium, a methyl group, a tert-butyl group, a diphenylamine group which is unsubstituted or substituted with a methyl group, a tert-butyl group, a trimethylsilyl group, or a triphenylsilyl group, a dihydroacridine group

which is unsubstituted or substituted with a methyl group, a tert-butyl group, or a phenyl group, a dihydrobenzoazasiline group

which is unsubstituted or substituted with a methyl group, a tert-butyl group, or a phenyl group, a spiro(dibenzosiloledibenzoazasiline) group

20 a spiro(acridine-fluorene) group

a hexahydrocarbazole group

<sup>50</sup> which is unsubstituted or substituted with a methyl group, a tert-butyl group, or a phenyl group, or a tetrahydrobenzocarbazole group

which is unsubstituted or substituted with a methyl group, a tert-butyl group, or a phenyl group.

The tetrahydrobenzocarbazole group of  $\boldsymbol{R}_1$  to  $\boldsymbol{R}_3$  is preferably

According to an exemplary embodiment of the present specification, Formula 1 is any one of the following For- 15 mulae 101 to 108:

Formula 101
$$A2$$

$$A1$$

$$N$$

$$R_{1})_{n1}$$

$$R_{2})_{n2}$$

$$Ar_{1}$$

$$Ar_{2}$$

$$Ar_{3}$$

$$R_{3}')_{n3'}$$

$$Ar_{4}$$

Formula 102 30

$$A2$$
 $A1$ 
 $A1$ 

Ar<sub>103</sub> 
$$Ar_{104}$$
 Formula 103

Ar<sub>103</sub>  $Ar_{104}$   $Ar_{103}$   $Ar_{104}$   $Ar_{105}$   $Ar$ 

Formula 104 
$$_{55}$$

A2

 $(R_{1})_{n1}$ 
 $(R_{13})_{n13}$ 
 $(R_{13})_{n13}$ 
 $(R_{105})$ 
 $(R_{$ 

Formula 105

$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $Ar_1$ 
 $Ar_2$ 
 $Ar_3$ 
 $Ar_4$ 
 $(R_{10})_{n16}$ 
 $(R_{19})_{n17}$ 

$$(R_{1})_{n1}$$
 $(R_{2})_{n2}$ 
 $(R_{41})_{n41}$ 
 $(R_{42})_{n42}$ 
 $(R_{43})_{n43}$ 

$$(R_{1})_{n1}$$
 $(R_{2})_{n2}$ 
 $Ar_{1}$ 
 $Ar_{2}$ 
 $Ar_{3}$ 
 $(R_{44})_{n44}$ 
 $(R_{45})_{n45}$ 
 $(R_{46})_{n46}$ 
 $(R_{47})_{n47}$ 

In Formulae 101 to 108:

the definitions of A1, A2,  $R_1$  to  $R_3$ , n1 to n3, and  $Ar_1$  to  $Ar_4$  are the same as those defined in Formula 1;

 $Q_1$  is  $C(R_{48})$   $(R_{49})$  or  $Si(R_{48})$   $(R_{49})$ ;

 $Q_2$  is C or Si;

 $R_3^{\prime}$ ,  $R_{11}$  to  $R_{20}$ , and  $R_{41}$  to  $R_{49}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted alkenyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted aryloxy group, a substituted aryloxy

n3' is an integer from 0 to 3;

n11 to n14 and n41 are each an integer form 0 to 2;

n15 is an integer from 0 to 8;

n16 to n18 and n42 to n47 are each an integer from 0 to  $\frac{4}{}$ 

when n3', n15 to n18, and n42 to n47 are each 2 or more, substituents in a plurality of parentheses are the same as or different from each other;

when n11 to n14 and n41 are each 2, substituents in a plurality of parentheses are the same as or different from each other; and

Ar<sub>101</sub> to Ar<sub>106</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a 30 substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heteroaryl group.

According to an exemplary embodiment of the present invention,  $R_{48}$  and  $R_{49}$  are the same as or different from each 35 other, and are each independently an alkyl group having 1 to 10 carbon atoms, or an aryl group having 6 to 30 carbon atoms.

According to another exemplary embodiment,  $R_{48}$  and  $R_{49}$  are the same as or different from each other, and are each 40 independently a methyl group, or a phenyl group.

According to an exemplary embodiment of the present invention,  $R_{48}$  and  $R_{49}$  are the same as each other.

According to an exemplary embodiment of the present invention,  $Ar_{101}$  to  $Ar_{106}$  are the same as or different from 45 each other, and are each independently a substituted or unsubstituted aryl group having 6 to 60 carbon atoms.

In another exemplary embodiment,  $Ar_{101}$  to  $Ar_{106}$  are the same as or different from each other, and are each independently a substituted or unsubstituted phenyl group.

In still another exemplary embodiment,  $Ar_{101}$  to  $Ar_{106}$  are the same as or different from each other, and are each independently an aryl group having 6 to 60 carbon atoms, which is unsubstituted or substituted with an alkyl group having 1 to 20 carbon atoms, a trialkylsilyl group having 1 to 20 carbon atoms, or a triarylsilyl group having 6 to 30 carbon atoms.

According to yet another exemplary embodiment,  $Ar_{101}$  to  $Ar_{106}$  are the same as or different from each other, and are each independently a phenyl group which is unsubstituted or 60 substituted with a methyl group, a tert-butyl group, a trimethylsilyl group, or a triphenylsilyl group.

According to an exemplary embodiment of the present invention,  $Ar_{101}$  and  $Ar_{102}$  are the same as or different from each other, and are each independently an aryl group having 65 to 60 carbon atoms, which is unsubstituted or substituted with an alkyl group having 1 to 4 carbon atoms, a trialkyl-

silyl group having 1 to 20 carbon atoms, or a triarylsilyl group having 6 to 30 carbon atoms.

In another exemplary embodiment,  $Ar_{101}$  and  $Ar_{102}$  are the same as or different from each other, and are each independently a phenyl group which is unsubstituted or substituted with a methyl group, a tert-butyl group, a trimethylsilyl group, or a triphenylsilyl group.

According to an exemplary embodiment of the present invention,  $Ar_{103}$  to  $Ar_{106}$  are the same as or different from each other, and are each independently a phenyl group which is unsubstituted or substituted with an alkyl group having 1 to 4 carbon atoms.

In another exemplary embodiment,  $Ar_{103}$  to  $Ar_{106}$  are the same as or different from each other, and are each independently a phenyl group which is unsubstituted or substituted with a tert-butyl group.

According to an exemplary embodiment of the present invention,  $R_3$ ',  $R_{11}$  to  $R_{20}$ , and  $R_{41}$  to  $R_{49}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, or a substituted or unsubstituted aryl group having 6 to 30 carbon atoms.

According to another exemplary embodiment,  $R_3$ ',  $R_{11}$  to  $R_{20}$ , and  $R_{41}$  to  $R_{41}$  to  $R_{41}$  to  $R_{42}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a substituted or unsubstituted methyl group, a substituted or unsubstituted butyl group, or a substituted or unsubstituted phenyl group.

In still another exemplary embodiment,  $R_3$ ',  $R_{11}$  to  $R_{20}$ , and  $R_{41}$  to  $R_{49}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a methyl group, a tert-butyl group, or a phenyl group.

According to an exemplary embodiment of the present specification, n1 is an integer from 0 to 3, and when n1 is 2 or more, a plurality of  $R_1$ 's is the same as or different from each other.

According to another exemplary embodiment, n1 is 0 or

According to an exemplary embodiment of the present invention, n2 is an integer from 0 to 3, and when n2 is 2 or more, a plurality of R<sub>2</sub>'s is the same as or different from each other

According to another exemplary embodiment, n2 is 0 or 1.

According to an exemplary embodiment of the present specification, n3 is an integer from 0 to 3, and when n3 is 2 or more, a plurality of  $R_3$ 's is the same as or different from each other.

According to another exemplary embodiment, n3 is 0 or

According to an exemplary embodiment of the present invention, Formula 1 is any one of the following Formulae 11 to 39:

Formula 11 
$$R_{101}$$
  $R_{102}$   $R_{103}$   $R_{104}$   $R_{105}$   $R_{105}$ 

#### -continued

-continued

Formula 12

$$(R_{1})_{n1}$$
 $(R_{2})_{n2}$ 
 $(R_{100})_{m1}$ 
 $(R_{3})_{n3}$ 

$$(R_{119})_{m11} \\ R_{2)n2} \\ R_{120})_{m12}$$

## Formula 13 15

$$(R_{1})_{n1}$$
 $(R_{2})_{n2}$ 
 $(R_{111})_{m3}$ 
 $(R_{3})_{n3}$ 
 $(R_{112})_{m4}$ 
 $(R_{112})_{m4}$ 

#### Formula 14

30

35

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$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $(R_{113})_{m5}$ 
 $(R_{114})_{m6}$ 

Formula 17

$$(R_{127})_{m15} \\ R_{130} \\ R_{128} \\ R_{129}$$

$$(R_{1})_{n1}$$

$$(R_{2})_{n2}$$

$$(R_{116})_{m7}$$

$$(R_{3})_{n3}$$

$$(R_{3})_{n3}$$

$$(R_{116})_{m8}$$

$$(R_{132})_{m16} \\ (R_{133})_{m16} \\ (R_{3})_{n3}$$

### Formula 21

Formula 16

$$(R_{117})_{m9}$$
 $(R_{1})_{n1}$ 
 $(R_{1})_{n1}$ 
 $(R_{1})_{n2}$ 
 $(R_{1})_{n2}$ 
 $(R_{1})_{n3}$ 

60

$$(R_{1})_{n1}$$

$$(R_{2})_{n2}$$

$$(R_{135})_{m19}$$

$$(R_{134})_{m18}$$

20

25

Formula 25

 $(R_{146})_{m26}$ 

Formula 22

-continued

-continued

Formula 27]

$$(R_{13})_{n2}$$
 $(R_{137})_{m21}$ 
 $(R_{137})_{m21}$ 
 $(R_{137})_{m21}$ 

 $(R_3)_{n3}$ 

$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $(R_{150})_{m30}$ 
 $(R_{149})_{m29}$ 
 $(R_{3})_{n3}$ 

Formula 23  $(R_1)_{n1}$  $(R_2)_{n2}$  $(\dot{R}_{138})_{m22}$  $\dot{R}_{139}$ 

 $(R_3)_{n3}$ 

$$(R_1)_{n1} \qquad \qquad (R_2)_{n2}$$
 Formula 28 
$$(R_{153})_{n3} \qquad \qquad (R_{154})_{m31}$$

$$(R_1)_{n1}$$
 30
 $(R_2)_{n2}$  35
 $(R_{143})_{m23}$   $(R_3)_{n3}$ 

$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $(R_{155})_{n32}$ 
 $(R_{3})_{n3}$ 

$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $(R_{145})_{n25}$ 
 $(R_{145})_{n25}$ 
 $(R_{10})_{n1}$ 
 $(R_{10})_{n2}$ 
 $(R_{10})_{n2}$ 
 $(R_{10})_{n2}$ 
 $(R_{10})_{n2}$ 

Formula 30

Formula 29

Formula 26 55 
$$(R_{1})_{n1} \qquad (R_{148})_{m28}$$
  $(R_{147})_{m27} \qquad (R_{3})_{n3} \qquad 65$ 

Formula 
$$(R_1)_{n1}$$
  $(R_2)_{n2}$   $(R_{158})_{m33}$   $(R_{157})_{n4}$   $(R_{158})_{m33}$   $(R_{160})_{m35}$ 

Formula 31

Formula 33

Formula 35

-continued

-continued

$$(R_{161})_{n1} \qquad (R_{2})_{n2} \qquad \qquad 5$$

$$R_{161} \qquad (R_{162})_{m36} \qquad \qquad 10$$

$$(R_{163})_{m37} \qquad \qquad \qquad 15$$

Formula 32

$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $(R_3)_{n4}$ 
 $(R_{166})_{m39}$ 
 $(R_{167})_{m40}$ 
 $(R_{168})_{m41}$ 

$$(R_{171})_{m43}$$
 $(R_{171})_{m43}$ 
 $(R_{172})_{m44}$ 
 $(R_{172})_{m44}$ 
 $(R_{172})_{m44}$ 
 $(R_{172})_{m44}$ 
 $(R_{172})_{m44}$ 

Formula 34

$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $(R_{173})_{m45}$ 
 $(R_{174})_{m46}$ 
 $(R_{174})_{m46}$ 
 $(R_{174})_{m46}$ 
 $(R_{174})_{m46}$ 
 $(R_{174})_{m46}$ 

$$(R_{177})_{m49}$$
 $(R_{177})_{m49}$ 
 $(R_{178})_{m50}$ 
 $(R_{178})_{m51}$ 

Formula 36
$$(R_{181})_{m53}$$

$$(R_{181})_{m53}$$

$$(R_{182})_{m54}$$

$$(R_{183})_{m55}$$

$$(R_{184})_{m56}$$

Formula 37 
$$(R_{185})_{m57}$$

$$(R_{185})_{m59}$$

$$(R_{187})_{m59}$$

$$(R_{188})_{m60}$$

$$(R_{188})_{m60}$$

[Formula 38]
$$(R_{189})_{m61}$$

$$(R_{191})_{m63}$$

$$(R_{192})_{m64}$$

Formula 39

$$(R_{193})_{m65}$$
 $(R_{194})_{m66}$ 
 $(R_{195})_{m67}$ 
 $(R_{196})_{m68}$ 
 $(R_{197})_{m69}$ 
 $(R_{198})_{m70}$ 

In Formulae 11 to 39:

the definitions of  $R_1$  to  $R_3$ , and n1 to n3 are the same as those defined in Formula 1;

$$Q_1$$
 is  $C(R_{199})$   $(R_{200})$  or  $Si(R_{199})$   $(R_{200})$ ;  $Q_2$  is C or Si;

R<sub>101</sub> to R<sub>200</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted alkynyl group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted aryl group, a substituted or unsubstituted aryl group, a substituted or unsubstituted or unsu

n4 is an integer from 0 to 2; and

m1 to m70 are each an integer from 0 to 3, and when m1 40 to m70 and n4 are each 2 or more, substituents in two or more parentheses are the same as or different from each other.

According to an exemplary embodiment of the present invention,  $R_{101}$  to  $R_{200}$  are the same as or different from each 45 other, and are each independently hydrogen, deuterium, a halogen group, a substituted or unsubstituted trialkylsilyl group having 1 to 20 carbon atoms, a substituted or unsubstituted triarylsilyl group having 6 to 30 carbon atoms, an alkyl group having 1 to 20 carbon group, a substituted or unsubstituted aryl group having 6 to 30 carbon atoms, or a substituted or unsubstituted arylamine group having 6 to 50 carbon atoms

In another exemplary embodiment, R<sub>101</sub> to R<sub>200</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a substituted or unsubstituted trimethylsilyl group, a substituted or unsubstituted a triphenylsilyl group, a substituted or unsubstituted methyl group, a substituted or unsubstituted biphenyl group, or a substituted or unsubstituted biphenyl group, or a substituted or unsubstituted diphenylamine group.

According to still another exemplary embodiment,  $R_{101}$  to  $R_{200}$  are the same as or different from each other, and are 65 each independently hydrogen, deuterium, a halogen group, a trimethylsilyl group, a triphenylsilyl group, a methyl

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group, an ethyl group, a tert-butyl group, a phenyl group which is unsubstituted or substituted with a methyl group, a tert-butyl group, a phenyl group, or a naphthyl group, a biphenyl group which is unsubstituted or substituted with a tert-butyl group, or a diphenylamine group which is unsubstituted or substituted with a methyl group, a tert-butyl group, a trimethylsilyl group, or a triphenylsilyl group.

According to an exemplary embodiment of the present invention,  $R_{199}$  and  $R_{200}$  are the same as or different from each other, and are each independently an alkyl group having 1 to 10 carbon atoms, or an aryl group having 6 to 30 carbon atoms.

According to another exemplary embodiment, R<sub>199</sub> and <sup>15</sup> R<sub>200</sub> are the same as or different from each other, and are each independently a methyl group, or a phenyl group.

According to an exemplary embodiment of the present invention,  $R_{199}$  and  $R_{200}$  are the same as each other.

In an exemplary embodiment of the present invention, Formula 1 can be any one of the following structures:

$$\bigcup_{N} \bigcup_{N} \bigcup_{N$$

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & \\ & & \\ & \\ & & \\ & \\ & & \\ & & \\ & \\ & & \\ & & \\ & \\ & & \\ & &$$

-continued

$$\bigcup_{D} \bigvee_{N} \bigvee_{N}$$

$$\begin{array}{c|c} D & D & D \\ \hline D & N & D \\ \hline \end{array}$$

-continued

5

N

10

N N N 

N N 

-continued

$$\begin{array}{c} \\ \\ \\ \\ \\ \\ \end{array}$$

$$\begin{array}{c} \\ \\ \\ \\ \\ \\ \\ \end{array}$$

-continued

-continued

-continued

-continued

-continued

-continued

A core structure can be prepared using the following reaction scheme from the compound of Formula 1 according to an exemplary embodiment of the present specification.

25 The substituent can be bonded by a method known in the art, and the kind and position of the substituent or the number of substituents can be changed according to the technology known in the art.

Starting from a bromochloride compound, an aryl intermediate substituted with various types of amines is synthesized by an amination reaction using a palladium catalyst. 20 Next, the final product can be obtained by using boron triiodide to introduce boron. Reaction Scheme 1 exemplifies a process of synthesizing a compound in which a specific substituent is bonded to a specific position, but compounds corresponding to the range of Formula 1 can be synthesized 25 by any synthesis method known in the art using a starting material, an intermediate material, and the like known in the

In the present invention, various substituents can be introduced into the core structure as described above to 30 synthesize compounds having various energy bandgaps. Further, in the present invention, various substituents can be introduced into the core structure described above to adjust the HOMO and LUMO energy levels of compounds.

In addition, various substituents can be introduced into 35 the core structure having the structure described above to synthesize compounds having inherent characteristics of the introduced substituents. For example, a substituent usually used for a hole injection layer material, a material for transporting holes, a light emitting layer material, and an 40 electron transporting layer material, which are used for manufacturing an organic light emitting device, can be introduced into the core structure to synthesize a material which satisfies conditions required for each organic material layer.

Furthermore, the organic light emitting device according to the present invention is an organic light emitting device including: a first electrode; a second electrode provided to face the first electrode; and an organic material layer having one or more layers provided between the first electrode and 50 the second electrode, in which one or more layers of the organic material layer include the above-described compound.

The organic light emitting device of the present invention can be manufactured using typical preparation methods and 55 materials of an organic light emitting device, except that the above-described compound is used to form an organic material layer having one or more layers.

The compound can be formed as an organic material layer by not only a vacuum deposition method, but also a solution 60 application method when an organic light emitting device is manufactured. Here, the solution application method means spin coating, dip coating, inkjet printing, screen printing, a spray method, roll coating, and the like, but is not limited thereto.

The organic material layer of the organic light emitting device of the present invention can also be composed of a single-layered structure, but can be composed of a multilayered structure in which organic material layers having two or more layer are stacked. For example, the organic light emitting device of the present invention can have a structure including a hole injection layer, a hole transport layer, a layer which injects and transports holes simultaneously, a light emitting layer, an electron transport layer, an electron injection layer, and the like as organic material layers. However, the structure of the organic light emitting device is not limited thereto, and can include a fewer or greater number of organic material layers.

In the organic light emitting device of the present invention, the organic material layer can include one or more layers of an electron transport layer, an electron injection layer, and a layer which injects and transports electrons simultaneously, and one or more layers of the layers can include the compound of Formula 1.

In another organic light emitting device, the organic material layer can include an electron transport layer or an electron injection layer, and the electron transport layer or the electron injection layer can include the compound of Formula 1.

In the organic light emitting device of the present invention, the organic material layer can include one or more layers of a hole injection layer, a hole transport layer, and a layer which injects and transports holes simultaneously, and one or more layers of the layers can include the compound of Formula 1.

In still another organic light emitting device, the organic material layer can include a hole injection layer or a hole transport layer, and the hole transport layer or the hole injection layer can include the compound of Formula 1.

In still yet another exemplary embodiment, the organic material layer includes a light emitting layer, and the light emitting layer includes the compound of Formula 1. As an example, the compound of Formula 1 can be included as a dopant of the light emitting layer.

In an exemplary embodiment of the present specification, the organic light emitting device is a green organic light emitting device in which the light emitting layer includes the compound of Formula 1 as a dopant.

According to an exemplary embodiment of the present specification, the organic light emitting device is a red organic light emitting device in which the light emitting layer includes the compound of Formula 1 as a dopant.

In another exemplary embodiment, the organic light emitting device is a blue organic light emitting device in which the light emitting layer includes the compound of Formula 1 as a dopant.

As another example, the organic material layer including the compound of Formula 1 can include the compound of Formula 1 as a dopant, and can include an organic compound such as an anthracene-based compound as a host.

As still another example, the organic material layer including the compound of Formula 1 can include the compound of Formula 1 as a dopant, and can further include a fluorescent host or a phosphorescent host.

In still another exemplary embodiment, the organic material layer including the compound of Formula 1 can include the compound of Formula 1 as a dopant, include a fluorescent host or a phosphorescent host, and include another organic compound, a metal or a metal compound as a dopant.

As yet another example, the organic material layer including the compound of Formula 1 can include the compound

of Formula 1 as a dopant and include a fluorescent host or a phosphorescent host, and can be used with an iridium (Ir)-based dopant.

When the light emitting layer includes a dopant and a host, the dopant can be included in an amount of 0.01 to 10 parts by weight based on 100 parts by weight of the host.

In an exemplary embodiment of the present specification, the first electrode is a positive electrode, and the second electrode is a negative electrode.

According to another exemplary embodiment, the first electrode is a negative electrode, and the second electrode is a positive electrode.

The organic light emitting device can have, for example, the stacking structure described below, but the stacking  $_{15}$  structure is not limited thereto:

- (1) Positive electrode/Hole transport layer/Light emitting layer/Negative electrode
- (2) Positive electrode/Hole injection layer/Hole transport layer/Light emitting layer/Negative electrode
- (3) Positive electrode/Hole injection layer/Hole buffer layer/Hole transport layer/Light emitting layer/Negative electrode
- (4) Positive electrode/Hole transport layer/Light emitting layer/Electron transport layer/Negative electrode
- (5) Positive electrode/Hole transport layer/Light emitting layer/Electron transport layer/Electron injection layer/Negative electrode
- (6) Positive electrode/Hole injection layer/Hole transport layer/Light emitting layer/Electron transport layer/Negative electrode
- (7) Positive electrode/Hole injection layer/Hole transport layer/Light emitting layer/Electron transport layer/Electron injection layer/Negative electrode
- (8) Positive electrode/Hole injection layer/Hole buffer layer/Hole transport layer/Light emitting layer/Electron transport layer/Negative electrode
- (9) Positive electrode/Hole injection layer/Hole buffer layer/Hole transport layer/Light emitting layer/Electron 40 transport layer/Electron injection layer/Negative electrode
- (10) Positive electrode/Hole transport layer/Electron blocking layer/Light emitting layer/Electron transport layer/Negative electrode
- (11) Positive electrode/Hole transport layer/Electron <sup>45</sup> blocking layer/Light emitting layer/Electron transport layer/ Electron injection layer/Negative electrode
- (12) Positive electrode/Hole injection layer/Hole transport layer/Electron blocking layer/Light emitting layer/Electron transport layer/Negative electrode
- (13) Positive electrode/Hole injection layer/Hole transport layer/Electron blocking layer/Light emitting layer/Electron transport layer/Electron injection layer/Negative electrode
- (14) Positive electrode/Hole transport layer/Light emitting layer/Hole blocking layer/Electron transport layer/Negative electrode
- (15) Positive electrode/Hole transport layer/Light emitting layer/Hole blocking layer/Electron transport layer/Electron injection layer/Negative electrode
- (16) Positive electrode/Hole injection layer/Hole transport layer/Light emitting layer/Hole blocking layer/Electron transport layer/Negative electrode
- (17) Positive electrode/Hole injection layer/Hole trans- 65 port layer/Light emitting layer/Hole blocking layer/Electron transport layer/Electron injection layer/Negative electrode

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- (18) Positive electrode/Hole injection layer/Hole transport layer/Electron blocking layer/Light emitting layer/Hole blocking layer/Electron injection and transport layer/Negative electrode
- (19) Positive electrode/Hole injection layer/Hole transport layer/Electron blocking layer/Light emitting layer/First electron transport layer/Second electron transport layer/Negative electrode.

The structure of the organic light emitting device of the present invention can have a structure illustrated in FIG. 1, but is not limited thereto.

FIG. 1 exemplifies a structure of an organic light emitting device in which a positive electrode 2, a hole injection layer 3, a hole transport layer 4, an electron blocking layer 5, a light emitting layer 6, a first electron transport layer 7, a second electron transport layer 8, and a negative electrode 9 are sequentially stacked on a substrate 1. In the structure described above, the compound of Formula 1 can be included in the light emitting layer 6.

For example, the organic light emitting device according to the present invention can be manufactured by depositing a metal or a metal oxide having conductivity, or an alloy thereof on a substrate to form a positive electrode, forming an organic material layer having one or more layers selected from the group consisting of a hole injection layer, a hole transport layer, an electron blocking layer, a layer which transports and injects holes simultaneously, a light emitting layer, an electron transport layer, an electron injection layer, a hole blocking layer, and a layer which transports and injects electrons simultaneously, thereon, and then depositing a material, which can be used as a negative electrode, thereon, by using a physical vapor deposition (PVD) method such as sputtering or e-beam evaporation. In addition to the method described above, an organic light emitting device can also be made by sequentially depositing a negative electrode material, an organic material layer, and a positive electrode material on a substrate.

The organic material layer can have a multi-layered structure including a hole injection layer, a hole transport layer, a light emitting layer, and an electron transport layer, and the like, but is not limited thereto and can have a single-layered structure. Further, the organic material layer can be manufactured with a fewer number of layers by a method, such as a solvent process, for example, spin coating, dip coating, doctor blading, a screen printing, inkjet printing, or a thermal transfer method, using various polymers, instead of a deposition method.

The positive electrode is an electrode which injects holes, and as the positive electrode material, materials having a high work function are usually preferred so as to facilitate the injection of holes into an organic material layer. Specific examples of a positive electrode material which can be used in the present invention include: a metal, such as vanadium, chromium, copper, zinc, and gold, or an alloy thereof; a metal oxide, such as zinc oxide, indium oxide, indium tin oxide (ITO), and indium zinc oxide (IZO); a combination of metal and oxide, such as ZnO:Al or SnO<sub>2</sub>:Sb; a conductive polymer, such as poly(3-methylthiophene), poly[3,4-(ethylene-1,2-dioxy)thiophene] (PEDOT), polypyrrole, and polyaniline; and the like, but are not limited thereto.

The negative electrode is an electrode which injects electrons, and as the negative electrode material, materials having a low work function are usually preferred so as to facilitate the injection of electrons into an organic material layer. Specific examples of a negative electrode material include: a metal such as magnesium, calcium, sodium, potassium, titanium, indium, yttrium, lithium, gadolinium,

aluminum, silver, tin, and lead, or an alloy thereof; a multi-layer structured material, such as LiF/Al or  ${\rm LiO_2/Al}$ , and the like, but are not limited thereto.

The hole injection layer is a layer which serves to facilitate the injection of holes from a positive electrode to a light 5 emitting layer, and a hole injection material is a material which can proficiently receive holes injected from a positive electrode at low voltage, and it is preferred that the highest occupied molecular orbital (HOMO) of the hole injection material is between the work function of the positive electrode material and the HOMO of the peripheral organic material layer. Specific examples of the hole injection material include metal porphyrin, oligothiophene, arylamine-based organic materials, hexanitrile hexaazatriphenylene-based organic materials, quinacridone-based organic 15 materials, perylene-based organic materials, anthraquinone, polyaniline-based and polythiophene-based conductive polymers, and the like, but are not limited thereto.

The hole transport layer can serve to smoothly transport holes. A hole transport material is suitably a material having 20 high hole mobility which can receive holes transported from a positive electrode or a hole injection layer and transfer the holes to a light emitting layer. Specific examples thereof include arylamine-based organic material, conductive polymers, block copolymers having both conjugated portions 25 and non-conjugated portions, and the like, but are not limited thereto.

A hole buffer layer can be additionally provided between the hole injection layer and the hole transport layer, and include hole injection or transport materials known in the 30 art.

An electron blocking layer can be provided between the hole transport layer and the light emitting layer. As the electron blocking layer, a spiroindoloacridine-based compound or a material known in the art can be used.

The light emitting layer can emit red, green, or blue light, and can be composed of a phosphorescent material or a fluorescent material. The light emitting material is a material which can receive holes and electrons from a hole transport layer and an electron transport layer, respectively, and 40 combine the holes and the electrons to emit light in a visible ray region, and is preferably a material having good quantum efficiency to fluorescence or phosphorescence. Specific examples thereof include: 8-hydroxy-quinoline aluminum complexes (Alq<sub>3</sub>); carbazole-based compounds; dimerized 45 styryl compounds; BAlq<sub>3</sub>; 10-hydroxybenzoquinoline-metal compounds; benzoxazole-based, benzothiazole-based and benzimidazole-based compounds; poly(p-phenylenevinylene) (PPV)-based polymers; spiro compounds; polyfluorene, rubrene, and the like, but are not limited thereto.

Examples of a host material for the light emitting layer include fused aromatic ring derivatives, or hetero ring-containing compounds, and the like. Specifically, examples of the fused aromatic ring derivative include anthracene derivatives, pyrene derivatives, naphthalene derivatives, 55 pentacene derivatives, phenanthrene compounds, fluoranthene compounds, and the like, and examples of the hetero ring-containing compound include carbazole derivatives, dibenzofuran derivatives, ladder-type furan compounds, pyrimidine derivatives, and the like, but the 60 examples thereof are not limited thereto.

When the light emitting layer emits red light, it is possible to use a phosphorescent material such as bis(1-phenyliso-quinoline) acetylacetonate iridium (PIQIr(acac)), bis(1-phenylquinoline)acetylacetonate iridium (PQIr(acac)), tris(1-65 phenylquinoline)iridium (PQIr), or octaethylporphyrin platinum (PtOEP), or a fluorescent material such as tris(8-

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hydroxyquinolino)aluminum (Alq<sub>3</sub>), as a light emitting dopant, but the light emitting dopant is not limited thereto. When the light emitting layer emits green light, it is possible to use a phosphorescent material such as fac-tris(2-phenylpyridine)iridium (Ir(ppy)<sub>3</sub>), or a fluorescent material such as tris(8-hydroxyquinolino)aluminum (Alq<sub>3</sub>), as the light emitting dopant, but the light emitting dopant is not limited thereto. When the light emitting layer emits blue light, it is possible to use a phosphorescent material such as (4,6- $F_2$ PPY)<sub>2</sub>Irpic, or a fluorescent material such as spiro-DPVBi, spiro-6P, distyrylbenzene (DSB), distyrylarylene (DSA), PFO-based polymers or PPV-based polymer, as the light emitting dopant, but the light emitting dopant is not limited thereto.

The electron transport layer can serve to smoothly transport electrons. An electron transport material is suitably a material having high electron mobility which can proficiently receive electrons injected from a negative electrode and transfer the electrons to a light emitting layer. Specific examples thereof include: Al complexes of 8-hydroxyquinoline, complexes including Alq<sub>3</sub>, organic radical compounds, hydroxyflavone-metal complexes, 8-quinolinolato lithium (LiQ), benzoimidazole-based compounds, or a combination thereof, and the like, but are not limited thereto. Further, the electron transport layer can be formed of one layer, but can be formed of two or more layers.

The electron injection layer can serve to smoothly inject electrons. An electron injection material is preferably a compound which has a capability of transporting electrons, an effect of injecting electrons from a negative electrode, and an excellent effect of injecting electrons into a light emitting layer or a light emitting material, prevents excitons produced from a light emitting layer from moving to a hole injection layer, and is also excellent in the ability to form a thin film. Specific examples thereof include fluorenone, anthraquinodimethane, diphenoquinone, thiopyran dioxide, oxazole, oxadiazole, triazole, imidazole, perylenetetracarboxylic acid, fluorenylidene methane, anthrone, and the like, and derivatives thereof, metal complex compounds, nitrogen-containing 5-membered ring derivatives, and the like, but are not limited thereto.

Examples of the metal complex compounds include 8-hydroxyquinolinato lithium, bis(8-hydroxyquinolinato) zinc, bis(8-hydroxyquinolinato) copper, bis(8-hydroxy-quinolinato) manganese, tris(8-hydroxyquinolinato) aluminum, tris (2-methyl-8-hydroxyquinolinato) aluminum, tris(8-hydroxyquinolinato) gallium, bis(10-hydroxy-benzo[h] quinolinato) beryllium, bis(10-hydroxybenzo[h]quinolinato) zinc. bis(2-methyl-8-quinolinato) chlorogallium, bis(2-methyl-8-quinolinato) (o-cresolato) gallium, bis(2-methyl-8-quinolinato) (1-naphtholato) aluminum, bis(2-methyl-8-quinolinato) (2-naphtholato) gallium, and the like, but are not limited thereto.

The hole blocking layer is a layer which blocks holes from reaching a negative electrode, and can be generally formed under the same conditions as those of the hole injection layer. Specific examples thereof include oxadiazole derivatives or triazole derivatives, phenanthroline derivatives, BCP, aluminum complexes, and the like, but are not limited thereto.

tives, dibenzofuran derivatives, ladder-type furan compounds, pyrimidine derivatives, and the like, but the examples thereof are not limited thereto.

When the light emitting layer emits red light, it is possible

The organic light emitting device according to the present invention can be a top emission type, a bottom emission type, or a dual emission type according to the material to be used.

# Examples

Hereinafter, the present specification will be described in detail with reference to Examples in order to specifically

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3) 4a,9a-dimethyl-6-(trimethylsilyl)-2,3,4,4a,9,9ahexahydro-1H-carbazole

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explain the present specification. However, the Examples according to the present specification can be modified in various forms, and it is not interpreted that the scope of the present application is limited to the Examples described in detail below. The Examples of the present application are provided for more completely explaining the present specification to the person with ordinary skill in the art.

## Synthesis Examples

Reactant Synthesis

1) 6-bromo-4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & &$$

After 50 g of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole was dissolved in 1,000 ml of tetrahydrofuran under nitrogen atmosphere, the temperature was lowered to 0° C., and then 44.2 g of N-bromosuccinimide was slowly added thereto, and 1 hour later, the resulting product was extracted after the completion of the reaction, and then purified with an ethyl acetate:hexane column to obtain 62 g of 6-bromo-4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole (yield 89%).

MS[M+H]+=281

2) 4a,9a-dimethyl-6-phenyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole

After 50 g of 6-bromo-4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 23.9 g of phenyl boronic acid, 49.32 g of potassium carbonate, and 2.73 g of bis(tri-tert-butylphosphine)palladium(0) were dissolved in 600 m1 of 60 tetrahydrofuran under nitrogen atmosphere, and 3 hours later, the resulting product was extracted under reflux conditions using 300 ml of water after the completion of the reaction, and then purified with an ethyl acetate:hexane column to obtain 40 g of 4a,9a-dimethyl-6-phenyl-2,3,4,4a, 65 9,9a-hexahydro-1H-carbazole (yield 81%).

MS[M+H]+=278

$$+ Cl - Si$$

$$Br$$

After 50 g of 6-bromo-4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole was dissolved in 1,000 ml of tetrahydrofuran under nitrogen atmosphere, the temperature was lowered to  $-78^{\circ}$  C., and then 22.86 g of N-butyllithium (2.5 M) was slowly added dropwise thereto, 1 hour later, 25.2 g of chlorotrimethylsilane was slowly added dropwise thereto, and 3 hours later, the resulting product was extracted after the completion of the reaction, and then purified with an ethyl acetate:hexane column to obtain 34 g of 4a,9a-dimethyl-6-(trimethylsilyl)-2,3,4,4a,9,9a-hexahydro-1H-carbazole (yield 70%).

MS[M+H]+=274

4) 4a,9a-dimethyl-6-(trimphenylsilyl)-2,3,4,4a,9,9a-hexahydro-1H-carbazole

After 50 g of 6-bromo-4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole was dissolved in 1,000 ml of tetrahydrofuran under nitrogen atmosphere, the temperature

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was lowered to -78° C., and then 22.86 g of N-butyllithium (2.5 M) was slowly added dropwise thereto, 1 hour later, 68.39 g of chlorotriphenylsilane dissolved in tetrahydrofuran was slowly added dropwise thereto, and 6 hours later, the resulting product was extracted after the completion of 5 the reaction, and then purified with an ethyl acetate:hexane column to obtain 61 g of 4a,9a-dimethyl-6-(triphenylsilyl)-2,3,4,4a,9,9a-hexahydro-1H-carbazole (yield 74%).

MS[M+H]+=460

#### 5) 3,6-dibromo-9H-carbazole

After 50 g of 9H-carbazole was dissolved in 1,000 ml of N,N-dimethylformamide under nitrogen atmosphere, 95.6 g 25 of bromine was slowly added thereto at room temperature, and 6 hours later, a solid was obtained by putting the resulting product into water after the completion of the reaction, and then purified with an ethyl acetate:hexane column to obtain 84 g of 3,6-dibromo-9H-carbazole (yield  $^{30}$ 

MS[M+H]+=326

# 6) 3,6-diphenyl-9H-carbazole

After 50 g of 3,6-dibromo-9H-carbazole, 39.3 g of phenyl boronic acid, 42.42 g of potassium carbonate, and 2.35 g of bis(tri-tert-butylphosphine)palladium(0) were dissolved in 700 ml of tetrahydrofuran under nitrogen atmosphere, and 3 hours later, the resulting product was extracted under reflux conditions using 300 ml of water after the completion of the reaction, and then purified with an ethyl acetate:hexane column and then recrystallized to obtain 41 g of 3,6diphenyl-9H-carbazole (yield 84%).

MS[M+H]+=320

After 50 g of 3,6-dibromo-9H-carbazole was dissolved in 1,200 ml of tetrahydrofuran under nitrogen atmosphere, the temperature was lowered to -78° C., and then 30.6 g of N-butyllithium (2.5 M) was slowly added dropwise thereto, and then 1 hour later, 35.1 g of chlorotrimethylsilane was slowly added dropwise thereto, and 3 hours later, the resulting product was extracted after the completion of the reaction, and then purified with an ethyl acetate:hexane column to obtain 37 g of 3,6-bis(trimethylsilyl)-9H-carbazole (yield 77%).

MS[M+H]+=312

### 8) 3,6-bis(triphenylsilyl)-9H-carbazole

After 50 g of 3,6-dibromo-9H-carbazole was dissolved in 1,000 ml of tetrahydrofuran under nitrogen atmosphere, the temperature was lowered to -78° C., and then 22.86 g of N-butyllithium (2.5 M) was slowly added dropwise thereto, and then 1 hour later, 95.3 g of chlorotriphenyl-silane dissolved in tetrahydrofuran was slowly added dropwise thereto, and 6 hours later, the resulting product was extracted

after the completion of the reaction, and then purified with an ethyl acetate:hexane column and then recrystallized to obtain 79 g of 3,6-bis(triphenylsilyl)-9H-carbazole (yield

MS[M+H]+=685

Synthesis Example 1: Synthesis of Compound 1

### 1) Synthesis of Intermediate I1

After 15.1 g of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 9.4 g of 1,3-dibromo-5-methylbenzene, 24 g of sodium-tert-butoxide, and 0.96 g of bis(tri-tert-butylphosphine)palladium(0) were put into 150 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then recrystallized with an ethyl acetate:hexane, and then 16 g of Intermediate I1 was obtained. (Yield 87%).

MS[M+H]+=491

### 2) Synthesis of Compound 1

Under nitrogen atmosphere, 5 g of Intermediate I1, 10 g of boron triiodide, and 2.7 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 4 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then recrystallized with ethyl acetate: hexane, and then 3 g of Compound 1 was obtained (yield 59%).

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MS[M+H]+=499

Synthesis Example 2. Synthesis of Compound 2

### 1) Synthesis of Intermediate I2

After 30 g of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 29.4 g of 1-bromo-3-chloro-5-methylbenzene, 62 g of sodium-tert-butoxide, and 2.23 g of bis(tritert-butylphosphine)palladium(0) were put into 300 ml of toluene under nitrogen atmosphere, the resulting mixture 45 was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 40 g of Intermediate I2 which was in a liquid state (yield 84%).

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MS[M+H]+=326

Synthesis Example 3. Synthesis of Compound 3

5 N

I3 15

After 6.9 g of Intermediate I2, 8 g of bis(3-(tert-butyl) phenyl)amine, 15.6 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 150 20 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 10 g of Intermediate I3 (yield 71%).

MS[M+H]+=571

### 3) Synthesis of Compound 2

Under nitrogen atmosphere, 2 g of Intermediate I3, 3.4 g of boron triiodide, and 0.9 g of triphenylborane were stirred using dichlorobenzene at  $160^{\circ}$  C. for 4 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then recrystallized with ethyl acetate: hexane, and then 1.5 g of Compound 2 was obtained (yield 6574%)

1) Synthesis of Intermediate I4

$$+ \frac{\operatorname{Br}}{\operatorname{Cl}}$$

After 26.75 g of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 30 g of 1-bromo-3,5-dichlorobenzene, 56 g of sodium-tert-butoxide, and 2.03 g of bis(tri-tert-butylphosphine)palladium(0) were put into 300 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 35 g of Intermediate I4 (yield 76%).

MS[M+H]+=346

2) Synthesis of Intermediate I5

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After 41.3 g of Intermediate I4, 33.6 g of bis(4-(tert-butyl) phenyl)amine, 76.1 g of sodium-tert-butoxide, and 1.8 g of bis(tri-tert-butylphosphine)palladium(0) were put into 500 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 50 g of Intermediate I5 (yield 71%).

MS[M+H]+=591

### 3) Synthesis of Intermediate I6

Under nitrogen atmosphere, 2 g of Intermediate I5, 3.3 g of boron triiodide, and 0.9 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 4 hours. The reaction was completed, the resulting product was extracted at room temperature, and then recrystallized with ethyl acetate: hexane to obtain 1.3 g of Intermediate I6 (yield 66%).

MS[M+H]+=599

4) Synthesis of Compound 3

After 5 g of Intermediate I6, 1.9 g of diphenylamine, 3.6 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 4 g of Compound 3 (yield 71%).

MS[M+H]+=731

Synthesis Example 4. Synthesis of Compound 4

### 1) Synthesis of Intermediate I7

After 14.1 g of 2,2,3,3-tetramethylindolin, 10 g of 1,3-dibromo-5-methylbenzene, 25 g of sodium-tert-butoxide,

and 1.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 100 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then column-purified with ethyl acetate:hexane, and then 12 g of Intermediate I7 was obtained. (Yield 68%).

MS[M+H]+=438

2) Synthesis of Compound 4

Under nitrogen atmosphere, 4 g of Intermediate 17, 8.9 g of boron triiodide, and 2.4 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 4 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 3.1 g of Compound 4 (yield 76%).

MS[M+H]+=447

Synthesis Example 5. Synthesis of Compound 5

#### 1) Synthesis of Intermediate I8

After 10.0 g of 2,2,3,3-tetramethylindolin, 11.72 g of 65 1-bromo-3-chloro-5-methylbenzene, 24.2 g of sodium-tert-butoxide, and 0.87 g of bis(tri-tert-butylphosphine)palla-

dium(0) were put into 100 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 5 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 12 g of Intermediate I8 (yield 70%).

MS[M+H]+=300

2) Synthesis of Intermediate 19

After 10.65 g of Intermediate 18, 10.00 g of bis(3-(tert-butyl)phenyl)amine, 22.63 g of sodium-tert-butoxide, and 0.54 g of bis(tri-tert-butylphosphine)-palladium(0) were put into 150 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain

MS[M+H]+=545

3) Synthesis of Compound 5

14 g of Intermediate I9 (yield 72%).

Under nitrogen atmosphere, 4.00 g of Intermediate 19, 7.18 g of boron triiodide, and 1.95 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 4 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 3.0 g of Compound 5 (yield 74%).

MS[M+H]+=553

Synthesis Example 6. Synthesis of Compound 6

### 1) Synthesis of Intermediate I10

After 17.93 g of Intermediate I8, 10.00 g of 9H-carbazole, 38.08 g of sodium-tert-butoxide, and 0.92 g of bis(tri-tert-60 butylphosphine)palladium(0) were put into 200 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 8 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 17 g of Intermediate 65 I10 (yield 66%).

MS[M+H]+=431

2) Synthesis of Compound 6

Under nitrogen atmosphere, 3.00 g of Intermediate I10, 6.82 g of boron triiodide, and 1.86 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 5 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 2.3 g of Compound 6 (yield 75%).

MS[M+H] + = 553

Synthesis Example 7. Synthesis of Compound 7

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After 14.89 g of Intermediate 18, 10 g of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 31.63 g of sodium-tert-butoxide, and 0.76 g of bis(tri-tert-butylphosphine)palladium(0) were put into 150 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. 5 After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 16 g of Intermediate I11 (yield 69%).

MS[M+H]+=465

2) Synthesis of Compound 7

Under nitrogen atmosphere, 3.00 g of Intermediate II1, 6.31 g of boron triiodide, and 1.72 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was 40 extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 2.5 g of Compound 7 (yield 82%).

MS[M+H]+=472

Synthesis Example 8. Synthesis of Compound 8

1) Synthesis of Intermediate I12

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I12

After 10 g of Intermediate I4, 5.06 g of 2,2,3,3-tetramethylindolin, 12.26 g of sodium-tert-butoxide, and 0.44 g of bis(tri-tert-butylphosphine)palladium(0) were put into 110 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 4 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 11 g of Intermediate I12 (yield 79%).

MS[M+H]+=486

2) Synthesis of Intermediate I13

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ \end{array}$$

Under nitrogen atmosphere, 5.00 g of Intermediate I12, 10.07 g of boron triiodide, and 2.74 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 3 g of Intermediate I13 (yield 59%).

# 3) Synthesis of Compound 8

After 3 g of Intermediate I13, 1.02 g of diphenylamine, 35 3.87 g of sodium-tert-butoxide, and 0.09 g of bis(tri-tertbutylphosphine)palladium(0) were put into 40 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 5 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an  $\,^{40}$ ethyl acetate:hexane column, and then recrystallized to obtain 4 g of Compound 8 (yield 71%).

MS[M+H]+=626

Synthesis Example 9. Synthesis of Compound 9

#### 1) Synthesis of Intermediate I14

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After 10 g of Intermediate I4, 4.82 g of 9H-carbazole, 12.25 g of sodium-tert-butoxide, and 0.44 g of bis(tri-tertbutylphosphine)palladium(0) were put into 110 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 8 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 9 g of Intermediate I14 (yield 65%).

MS[M+H]+=478

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## 2) Synthesis of Intermediate I15

Under nitrogen atmosphere, 5.00 g of Intermediate I14, 10.23 g of boron triiodide, and 2.78 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 5 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 3.4 g of Intermediate I15 (yield 67%).

# 3) Synthesis of Compound 9

After 3 g of Intermediate I15, 1.03 g of diphenylamine,  $^{30}$  3.87 g of sodium-tert-butoxide, and 0.09 g of bis(tri-tert-butylphosphine)palladium(0) were put into 40 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 7 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.3 g of Compound 9 (yield 61%).

Synthesis Example 10. Synthesis of Compound 10

### 1) Synthesis of Intermediate I16

After 6.9 g of Intermediate I2, 8 g of bis(4-(tert-butyl) phenyl)amine, 15.6 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 150 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 11 g of Intermediate I16 (yield 78.1%).

MS[M+H]+=571

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2) Synthesis of Compound 10

Under nitrogen atmosphere, 3.00 g of Intermediate I16, 5.1 g of boron triiodide, and 1.35 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified

65 with ethyl acetate:hexane, and then recrystallized to obtain 2 g of Compound 10 (yield 66.0%).

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Synthesis Example 11. Synthesis of Compound 11

#### 1) Synthesis of Intermediate I17

After 6 g of Intermediate 12, 4 g of di-o-tolylamine, 5.31 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 150 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 6 g of Intermediate 117 (yield 67%).

MS[M+H]+=487

### 2) Synthesis of Compound 11

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Under nitrogen atmosphere, 3.00 g of Intermediate 117, 6.02 g of boron triiodide, and 1.64 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.8 g of Compound 11 (yield 59.0%).

MS[M+H]+=495

Synthesis Example 12. Synthesis of Compound 12

### 1) Synthesis of Intermediate I18

After 6 g of Intermediate I2, 4 g of di-m-tolylamine, 5.31 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 150 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 5.5 g of Intermediate I18 (yield 61%).

Under nitrogen atmosphere, 3.00 g of Intermediate I18, 6.02 g of boron triiodide, and 1.64 g of triphenylborane were 30 stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.5 g of Compound 12 (yield 49.0%).

MS[M+H]+=495

Synthesis Example 13. Synthesis of Compound 13

### 1) Synthesis of Intermediate I19

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After 6 g of Intermediate I2, 4 g of di-p-tolylamine, 5.31 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tertbutylphosphine)palladium(0) were put into 150 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 5.8 g of Intermediate I19 (yield 65%).

MS[M+H]+=487

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2) Synthesis of Compound 13

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Under nitrogen atmosphere, 3.00 g of Intermediate I19, 6.02 g of boron triiodide, and 1.64 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.8 g of Compound 13 (yield 59.0%).

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Synthesis Example 14. Synthesis of Compound 14

#### 1) Synthesis of Intermediate I20

After 6 g of Intermediate I2, 3.38 g of 9H-carbazole, 5.31 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 150 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 5.1 g of Intermediate I20 (yield 61%).

 $M\tilde{S}[M+H] + = 457$ 

# 2) Synthesis of Compound 14

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Under nitrogen atmosphere, 3.00 g of Intermediate I20, 6.42 g of boron triiodide, and 1.74 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.9 g of Compound 14 (yield 62.0%).

MS[M+H]+=465

Synthesis Example 15. Synthesis of Compound 15

### 1) Synthesis of Intermediate I21

$$\longrightarrow^{\mathbb{N}}$$

After 6 g of Intermediate I2, 3.95 g of 3,6-dimethyl-9H-carbazole, 5.31 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 150 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 5.5 g of Intermediate I21 (yield 62%).

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I21

Under nitrogen atmosphere, 3.00 g of Intermediate I21, 6.42 g of boron triiodide, and 1.74 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.4 g of Compound 15 (yield 46.0%).

MS[M+H]+=465

Synthesis Example 16. Synthesis of Compound 16

### 1) Synthesis of Intermediate I22

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After 6 g of Intermediate I2, 3.95 g of 2.7-dimethyl-9H-carbazole, 5.31 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 150 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 5.4 g of Intermediate I22 (yield 61%).

MS[M+H]+=485

# 2) Synthesis of Compound 16

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & &$$

Under nitrogen atmosphere, 3.00 g of Intermediate I22, 6.42 g of boron triiodide, and 1.74 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.5 g of Compound 16 (yield 49.0%).

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Synthesis Example 17. Synthesis of Compound 17

#### 1) Synthesis of Intermediate I23

After 6 g of Intermediate I2, 5.66 g of 3,6-di-tert-butyl-9H-carbazole, 5.31 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 150 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then <sup>45</sup> purified with an ethyl acetate:hexane column to obtain 5.5 g of Intermediate I23 (yield 53%).

MS[M+H]+=569

# 2) Synthesis of Compound 17

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Under nitrogen atmosphere, 3.00 g of Intermediate 123, 6.42 g of boron triiodide, and 1.74 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.5 g of Compound 17 (yield 49.0%).

MS[M+H]+=465

Synthesis Example 18. Synthesis of Compound 18

### 1) Synthesis of Intermediate I24

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After 50 g of 4a,6,9a-trimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 52.48 g of 1-bromo-3-chloro-5-methyl-benzene, 66.94 g of sodium-tert-butoxide, and 1.19 g of bis(tritert-butylphosphine)palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 60 g of Intermediate I24 (yield 76%).

2) Synthesis of Intermediate I25

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & &$$

Under nitrogen atmosphere, 3.00 g of Intermediate I25, 6.02 g of boron triiodide, and 1.74 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.9 g of Compound 18 (yield 63.0%).

MS[M+H]+=593

Synthesis Example 19. Synthesis of Compound 19

1) Synthesis of Intermediate I26

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After 6 g of Intermediate I24, 5.46 g of bis(3-(tert-butyl) <sup>40</sup> Cl. phenyl)amine, 5.08 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 100 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then <sup>45</sup> purified with an ethyl acetate:hexane column to obtain 6.0 g of Intermediate I25 (yield 58%).

MS[M+H]+=585

3) Synthesis of Compound 18

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After 6 g of Intermediate I24, 5.46 g of bis(4-(tert-butyl) phenyl)amine, 5.08 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 100 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 6.1 g of Intermediate I26 (yield 59%).

MS[M+H]+=585

2) Synthesis of Compound 19

Under nitrogen atmosphere, 3.00 g of Intermediate I26, 6.02 g of boron triiodide, and 1.74 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The 65 reaction was completed, and the resulting product was extracted at room temperature, and then column-purified

with ethyl acetate:hexane, and then recrystallized to obtain 1.8 g of Compound 19 (yield 59.0%).

MS[M+H]+=593

Synthesis Example 20. Synthesis of Compound 20

1) Synthesis of Intermediate I27

After 6 g of Intermediate I24, 3.24 g of 9H-carbazole, 5.08 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 5.9 g of Intermediate I27 (yield 71%).

MS[M+H]+=471

2) Synthesis of Compound 20

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Under nitrogen atmosphere, 3.00 g of Intermediate I27, 7.48 g of boron triiodide, and 1.74 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified <sup>20</sup> with ethyl acetate:hexane, and then recrystallized to obtain 1.4 g of Compound 20 (yield 46.0%).

MS[M+H]+=479

Synthesis Example 21. Synthesis of Compound 21

#### 1) Synthesis of Intermediate I28

After 6 g of Intermediate 124, 5.42 g of 3,6-di-tert-butyl-9H-carbazole, 5.08 g of sodium-tert-butoxide, and 0.1 g of 65 bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting

mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 5.6 g of Intermediate I28 (yield 54%).

MS[M+H]+=583

### 2) Synthesis of Compound 21

Under nitrogen atmosphere, 3.00 g of Intermediate 128, 6.04 g of boron triiodide, and 1.37 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.3 g of Compound 21 (yield 43.0%).

MS[M+H]+=591

Synthesis Example 22. Synthesis of Compound 22

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  $Cl$   $Br$   $-$ 

After 50 g of 6-(tert-butyl)-4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 59.70 g of 1-bromo-3-chloro-5-methylbenzene, 55.99 g of sodium-tert-butoxide, and 1.00 g of bis(tri-tert-butylphosphine)palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 58 g of Intermediate I29 (yield 78%).

MS[M+H]+=382

### 2) Synthesis of Intermediate I30

After 6 g of Intermediate I29, 4.86 g of bis(3-(tert-butyl) phenyl)amine, 4.53 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 100 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate: hexane column to obtain 6.2 g of Intermediate I30 (yield 63%).

MS[M+H]+=627

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# 3) Synthesis of Compound 22

Under nitrogen atmosphere, 3.00 g of Intermediate I30, 5.6 g of boron triiodide, and 1.27 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was

extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.8 g of Compound 22 (yield 59.0%).

MS[M+H]+=635

Synthesis Example 23. Synthesis of Compound 23

### 1) Synthesis of Intermediate I31

After 6 g of Intermediate I29, 4.86 g of bis(4-(tert-butyl) phenyl)amine, 4.52 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 100 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 6.3 g of Intermediate I31 (yield 64%).

MS[M+H]+=627

2) Synthesis of Compound 23

Under nitrogen atmosphere, 3.00 g of Intermediate I31, 5.62 g of boron triiodide, and 1.27 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.9 g of Compound 23 (yield 63.0%).

MS[M+H]+=635

Synthesis Example 24. Synthesis of Compound 24

-continued

After 6 g of Intermediate I29, 2.89 g of 9H-carbazole, 4.53 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tertbutylphosphine)palladium(0) were put into 120 ml of tolu- 20 ene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 5.5 g of Intermediate I32 (yield 68%).

MS[M+H]+=513

2) Synthesis of Compound 24

Under nitrogen atmosphere, 3.00 g of Intermediate I32, 60 6.86 g of boron triiodide, and 1.55 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 65 1.4 g of Compound 24 (yield 46.0%).

MS[M+H]+=521

Synthesis Example 25. Synthesis of Compound 25

1) Synthesis of Intermediate I33

After 6 g of Intermediate I29, 4.82 g of 3,6-di-methyl-9H-carbazole, 4.53 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting 45 mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 6.6 g of Intermediate I33 (yield 67%).

MS[M+H]+=625

2) Synthesis of Compound 25

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Under nitrogen atmosphere, 3.00 g of Intermediate I33, 5.63 g of boron triiodide, and 1.27 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.5 g of Compound 25 (yield 49.0%).

MS[M+H]+=633

Synthesis Example 26. Synthesis of Compound 26

1) Synthesis of Intermediate I34

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After 50 g of 4a,5,7,9a-tetramethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 49.27 g of 1-bromo-3-chloro-5-methylbenzene, 62.85 g of sodium-tert-butoxide, and 1.11 g of bis(tri-tert-butylphosphine)palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 58 g of Intermediate I34 (yield 75%).

2) Synthesis of Intermediate I35

After 6 g of Intermediate I34, 5.24 g of bis(3-(tert-butyl) phenyl)amine, 4.89 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 100 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 6.1 g of Intermediate I35 (yield 60%).

MS[M+H]+=599

3) Synthesis of Compound 26

Under nitrogen atmosphere, 3.00 g of Intermediate I35, 5.88 g of boron triiodide, and 1.33 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.7 g of Compound 26 (yield 56.0%).

MS[M+H]+=607

Synthesis Example 27. Synthesis of Compound 27

# 1) Synthesis of Intermediate I36

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After 6 g of Intermediate I34, 5.24 g of bis(4-(tert-butyl) phenyl)amine, 4.89 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 100 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 6.0 g of Intermediate I36 (yield 59%).

MS[M+H]+=599

2) Synthesis of Compound 27

Under nitrogen atmosphere, 3.00 g of Intermediate I36, 5.88 g of boron triiodide, and 1.33 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain

1.8 g of Compound 27 (yield 59.0%). MS[M+H]+=607

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Synthesis Example 28. Synthesis of Compound 28

#### 1) Synthesis of Intermediate I37

After 6 g of Intermediate I34, 3.12 g of 9H-carbazole, 4.88 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 5.8 g of Intermediate I37 (yield 71%).

MS[M+H]+=485

### 2) Synthesis of Compound 28

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Under nitrogen atmosphere, 3.00 g of Intermediate 137, 7.27 g of boron triiodide, and 1.64 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.8 g of Compound 28 (yield 59.0%).

MS[M+H]+=493

Synthesis Example 29. Synthesis of Compound 29

### 1) Synthesis of Intermediate I38

After 6 g of Intermediate I34, 5.21 g of 3,6-di-methyl-9H-carbazole, 4.88 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then

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688 Synthesis Example 30. Synthesis of Compound 30

purified with an ethyl acetate:hexane column to obtain 6.1 g of Intermediate I38 (yield 60%).

MS[M+H]+=597

2) Synthesis of Compound 29

1) Synthesis of Intermediate I39

After 50 g of 4a, 9a-dimethyl-6-phenyl-2,3,4,4a,9,9a-

hexahydro-1H-carbazole, 40.74 g of 1-bromo-3-chloro-5-40 methylbenzene, 51.96 g of sodium-tert-butoxide, and 0.92 g of bis(tri-tert-butylphosphine)-palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then

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purified with an ethyl acetate: hexane column to obtain 61 g of Intermediate I39 (yield 84%). MS[M+H]+=4022) Synthesis of Intermediate I40

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After 6 g of Intermediate 139, 4.62 g of bis(3-(tert-butyl) phenyl)amine, 4.30 g of sodium-tert-butoxide, and 0.08 g of bis(tri-tert-butylphosphine)palladium(0) were put into 100 ml of toluene under nitrogen atmosphere, the resulting 40 mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 6.3 g of Intermediate I40 (yield 65%).

MS[M+H]+=647

3) Synthesis of Compound 30

-continued

Under nitrogen atmosphere, 3.00 g of Intermediate I40, 5.88 g of boron triiodide, and 1.33 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.9 g of Compound 30 (yield 63.0%).

MS[M+H]+=655

Synthesis Example 31. Synthesis of Compound 31

After 6 g of Intermediate 139, 4.62 g of bis(4-(tert-butyl) phenyl)amine, 4.30 g of sodium-tert-butoxide, and 0.08 g of 20 bis(tri-tert-butylphosphine)palladium(0) were put into 100 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 6.1 g <sup>25</sup> of Intermediate I41 (yield 63%).

MS[M+H]+=647

2) Synthesis of Compound 31

Under nitrogen atmosphere, 3.00 g of Intermediate I41, 5.88 g of boron triiodide, and 1.33 g of triphenylborane were 65 stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was

extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.7 g of Compound 31 (yield 58.0%).

MS[M+H]+=655

Synthesis Example 32. Synthesis of Compound 32

1) Synthesis of Intermediate I42

I42

After 6 g of Intermediate I39, 3.12 g of 9H-carbazole, 4.88 g of sodium-tert-butoxide, and 0.1 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 5.8 g of Intermediate I42 (yield 68%).

2) Synthesis of Compound 32

Under nitrogen atmosphere, 3.00 g of Intermediate I42, 6.61 g of boron triiodide, and 1.50 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.8 g of Compound 32 (yield 59.0%).

MS[M+H]+=541

Synthesis Example 33. Synthesis of Compound 33

1) Synthesis of Intermediate I43

-continued

After 6 g of Intermediate I39, 4.58 g of 3,6-di-(tert-butyl)9H-carbazole, 4.30 g of sodium-tert-butoxide, and 0.08 g of
bis(tri-tert-butylphosphine)palladium(0) were put into 120
ml of toluene under nitrogen atmosphere, the resulting
mixture was stirred for 10 hours. After the completion of the
reaction, the resulting product was extracted, and then
purified with an ethyl acetate:hexane column to obtain 6.4 g
of Intermediate I43 (yield 66%).

MS[M+H]+=645

2) Synthesis of Compound 33

-continued

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Under nitrogen atmosphere, 3.00 g of Intermediate I43, <sup>20</sup> 5.90 g of boron triiodide, and 1.34 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain <sup>25</sup> 1.8 g of Compound 33 (yield 59.0%).

MS[M+H]+=653

Synthesis Example 34. Synthesis of Compound 34

1) Synthesis of Intermediate I44

After 6 g of Intermediate I2, 6.51 g of di([1,1'-biphenyl]-3-yl)amine, 5.30 g of sodium-tert-butoxide, and 0.09 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 7.2 g of Intermediate I44 (yield 64%).

MS[M+H]+=611

30 2) Synthesis of Compound 34

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Under nitrogen atmosphere, 3.00 g of Intermediate I44, 5.77 g of boron triiodide, and 1.31 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.9 g of Compound 34 (yield 63.0%).

MS[M+H]+=619

Synthesis Example 35. Synthesis of Compound 35

1) Synthesis of Intermediate I45

After 6 g of Intermediate I24, 6.24 g of di([1,1'-biphenyl]-3-yl)amine, 5.09 g of sodium-tert-butoxide, and 0.09 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 7.3 g of Intermediate I45 (yield 66%).

I45

MS[M+H]+=625

2) Synthesis of Compound 35

Under nitrogen atmosphere, 3.00 g of Intermediate I45, 5.64 g of boron triiodide, and 1.27 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.8 g of Compound 35 (yield 59.0%).

MS[M+H]+=633

Synthesis Example 36. Synthesis of Compound 36

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After 6 g of Intermediate I29, 5.55 g of di([1,1'-biphenyl]-3-yl)amine, 4.52 g of sodium-tert-butoxide, and 0.08 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 7.3 g of Intermediate I46 (yield 70%).

MS[M+H]+=667

2) Synthesis of Compound 36

I46

Under nitrogen atmosphere, 3.00 g of Intermediate I46, 5.28 g of boron triiodide, and 1.20 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.7 g of Compound 36 (yield 56.0%).

MS[M+H]+=675

Synthesis Example 37. Synthesis of Compound 37

After 6 g of Intermediate I34, 5.99 g of di([1,1'-biphenyl]-3-yl)amine, 4.89 g of sodium-tert-butoxide, and 0.08 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 7.1 g of Intermediate I47 (yield 67%).

MS[M+H]+=639

### 2) Synthesis of Compound 37

Under nitrogen atmosphere, 3.00 g of Intermediate I47, 5.51 g of boron triiodide, and 1.25 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.8 g of Compound 37 (yield 59.0%).

MS[M+H] + = 647

Synthesis Example 38. Synthesis of Compound 38

### 1) Synthesis of Intermediate I48

After 6 g of Intermediate I39, 5.28 g of di([1,1'-biphenyl]-60 3-yl)amine, 4.30 g of sodium-tert-butoxide, and 0.08 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 6.4 g of Intermediate I48 (yield 62%).

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# 2) Synthesis of Compound 38

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Synthesis Example 39. Synthesis of Compound 39

1) Synthesis of Intermediate I49

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Under nitrogen atmosphere, 3.00 g of Intermediate I47, 60 5.51 g of boron triiodide, and 1.25 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 65 1.9 g of Compound 38 (yield 58.0%).

MS[M+H]+=695

After 6 g of Intermediate I2, 6.51 g of di([1,1'-biphenyl]-4-yl)amine, 5.30 g of sodium-tert-butoxide, and 0.09 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 7.1 g of Intermediate I49 (yield 62%).

2) Synthesis of Compound 39

Under nitrogen atmosphere, 3.00 g of Intermediate I49,  $_{40}$  5.77 g of boron triiodide, and 1.31 g of triphenylborane were stirred using dichlorobenzene at  $_{160^{\circ}}$  C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain  $_{15}$  g of Compound 39 (yield 49.0%).

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MS[M+H]+=619

Synthesis Example 40. Synthesis of Compound 40

# 1) Synthesis of Intermediate I50

-continued

After 6 g of Intermediate I24, 6.24 g of di([1,1'-biphenyl]-4-yl)amine, 5.09 g of sodium-tert-butoxide, and 0.09 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 7.1 g of Intermediate I50 (yield 64%).

I50

MS[M+H]+=625

## 2) Synthesis of Compound 40

Under nitrogen atmosphere, 3.00 g of Intermediate I50, 5.64 g of boron triiodide, and 1.27 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.6 g of Compound 40 (yield 53.0%).

MS[M+H]+=633

Synthesis Example 41. Synthesis of Compound 41

# 1) Synthesis of Intermediate I51

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After 6 g of Intermediate I29, 5.55 g of di([1,1'-biphenyl]-4-yl)amine, 4.52 g of sodium-tert-butoxide, and 0.08 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 7.0 g of Intermediate I51 (yield 67%).

MS[M+H]+=667

30 2) Synthesis of Compound 41

Under nitrogen atmosphere, 3.00 g of Intermediate I51, 5.28 g of boron triiodide, and 1.20 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.6 g of Compound 41 (yield 53.0%).

MS[M+H]+=675

Synthesis Example 42. Synthesis of Compound 42

### 1) Synthesis of Intermediate I52

After 6 g of Intermediate I34, 5.99 g of di([1,1'-biphenyl]-4-yl)amine, 4.89 g of sodium-tert-butoxide, and 0.08 g of 60 bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 7.0 g 65 of Intermediate I52 (yield 65%).

MS[M+H]+=639

2) Synthesis of Compound 42

Under nitrogen atmosphere, 3.00 g of Intermediate I52, 5.51 g of boron triiodide, and 1.25 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.7 g of Compound 42 (yield 56.0%).

MS[M+H]+=647

Synthesis Example 43. Synthesis of Compound 43

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After 6 g of Intermediate I39, 5.28 g of di([1,1'-biphenyl]-4-yl)amine, 4.30 g of sodium-tert-butoxide, and 0.08 g of bis(tri-tert-butylphosphine)palladium(0) were put into 120 ml of toluene under nitrogen atmosphere, the resulting 40 mixture was stirred for 10 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 6.5 g of Intermediate I53 (yield 63%).

MS[M+H]+=687

2) Synthesis of Compound 43

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Under nitrogen atmosphere, 3.00 g of Intermediate I53, 5.13 g of boron triiodide, and 1.16 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was <sup>25</sup> extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.5 g of Compound 43 (yield 49.0%).

MS[M+H]+=695

Synthesis Example 44. Synthesis of Compound 44

1) Synthesis of Intermediate I54

45 g of Intermediate I54 was obtained using the same conditions as in the synthesis method of Intermediate I2 under nitrogen atmosphere (yield 82%).

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6.2 g of Intermediate I55 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 76%).

MS[M+H]+=501

3) Synthesis of Compound 44

Under nitrogen atmosphere, 3.00 g of Intermediate 155, 7.03 g of boron triiodide, and 1.60 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The 65 reaction was completed, and the resulting product was extracted at room temperature, and then column-purified

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with ethyl acetate:hexane, and then recrystallized to obtain 1.6 g of Compound 44 (yield 53.0%).

MS[M+H]+=509

Synthesis Example 45. Synthesis of Compound 45

1) Synthesis of Intermediate I56

6.5 g of Intermediate I56 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 65%).

MS[M+H]+=613

2) Synthesis of Compound 45

Under nitrogen atmosphere, 3.00~g of Intermediate I56, 5.67~g of boron triiodide, and 1.28~g of triphenylborane were stirred using dichlorobenzene at  $160^{\circ}$  C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.7~g of Compound 45 (yield 56.0%).

MS[M+H]+=621

Synthesis Example 46. Synthesis of Compound 46

## 1) Synthesis of Intermediate I57

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6.6 g of Intermediate I57 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 66%).

MS[M+H]+=613

2) Synthesis of Compound 46

Under nitrogen atmosphere, 3.00 g of Intermediate 157, 5.74 g of boron triiodide, and 1.30 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain

MS [M+H]+=621

1.6 g of Compound 46 (yield 53.0%).

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Synthesis Example 47. Synthesis of Compound 47

# 1) Synthesis of Intermediate I58

6.4 g of Intermediate I58 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 79%).

$$MS[M+H]+=499$$

## 2) Synthesis of Compound 47

Under nitrogen atmosphere, 3.00 g of Intermediate I58,
 7.06 g of boron triiodide, and 1.60 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours.

The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.3 g of Compound 47 (yield 52.0%).

MS[M+H]+=507

Synthesis Example 48. Synthesis of Compound 48

## 1) Synthesis of Intermediate I59

6.1 g of Intermediate I59 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 61%).

MS[M+H]+=611

2) Synthesis of Compound 48

Under nitrogen atmosphere, 3.00 g of Intermediate I59, 40 5.76 g of boron triiodide, and 1.31 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 45 1.5 g of Compound 48 (yield 50.0%).

MS[M+H]+=619

Synthesis Example 49. Synthesis of Compound 49

1) Synthesis of Intermediate I60

-continued

7.5 g of Intermediate I60 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 70%).

MS[M+H]+=653

2) Synthesis of Compound 49

Under nitrogen atmosphere, 3.00 g of Intermediate I60, 5.40 g of boron triiodide, and 1.22 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.6 g of Compound 49 (yield 53.0%).

MS[M+H]+=661

Synthesis Example 50. Synthesis of Compound 50

# 1) Synthesis of Intermediate I61

7.7 g of Intermediate I61 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 72%).

MS[M+H]+=653

2) Synthesis of Compound 50

Under nitrogen atmosphere, 3.00 g of Intermediate I61, 5.40 g of boron triiodide, and 1.22 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The

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reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.7 g of Compound 50 (yield 56.0%).

MS[M+H]+=661

Synthesis Example 51. Synthesis of Compound 51

1) Synthesis of Intermediate I62

40 g of Intermediate I62 was obtained using the same conditions as in the synthesis method of Intermediate I2 under nitrogen atmosphere (yield 81%).

MS[M+H]+=425

2) Synthesis of Intermediate I63

· – ·

6.6 g of Intermediate I63 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 84%).

MS[M+H]+=557

3) Synthesis of Compound 51

Under nitrogen atmosphere, 3.00 g of Intermediate I63, 6.32 g of boron triiodide, and 1.43 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.7 g of Compound 51 (yield 56.0%).

Synthesis Example 52. Synthesis of Compound 52

# 1) Synthesis of Intermediate I64

2) Synthesis of Compound 52

Under nitrogen atmosphere, 3.00 g of Intermediate I64, 5.26 g of boron triiodide, and 1.19 g of triphenylborane were stirred using dichlorobenzene at  $160^{\circ}$  C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.4 g of Compound 52 (yield 46.0%). MS[M+H]+=677

Synthesis Example 53. Synthesis of Compound 53

# 1) Synthesis of Intermediate I65

6.9 g of Intermediate I64 was obtained using the same conditions as in the synthesis method of Intermediate I3  $_{65}$ under nitrogen atmosphere (yield 73%).

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-continued

 $5.9~{\rm g}$  of Intermediate I65 was obtained using the same conditions as in the synthesis method of Intermediate I3  $^{20}$  under nitrogen atmosphere (yield 75%).

MS[M+H]+=555

2) Synthesis of Compound 53

Under nitrogen atmosphere, 3.00 g of Intermediate I65, 60 6.35 g of boron triiodide, and 1.44 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 65 1.3 g of Compound 53 (yield 43.0%).

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MS[M+H]+=563

Synthesis Example 54. Synthesis of Compound 54

1) Synthesis of Intermediate I66

6.6 g of Intermediate I66 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 70%).

I66

MS[M+H]+=668

2) Synthesis of Compound 54

Under nitrogen atmosphere, 3.00 g of Intermediate I66, 20 6.35 g of boron triiodide, and 1.44 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 25 1.4 g of Compound 54 (yield 46.0%).

MS[M+H]+=675

Synthesis Example 55. Synthesis of Compound 55

# 1) Synthesis of Intermediate I67

7.6 g of Intermediate I67 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 76%).

MS[M+H]+=710

2) Synthesis of Compound 55

Under nitrogen atmosphere, 3.00 g of Intermediate 167, 4.97 g of boron triiodide, and 1.13 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was

extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.6 g of Compound 55 (yield 53.0%).

MS[M+H]+=717

Synthesis Example 56. Synthesis of Compound 56

1) Synthesis of Intermediate I68

38 g of Intermediate I68 was obtained using the same conditions as in the synthesis method of Intermediate I2 under nitrogen atmosphere (yield 82%).

MS[M+H]+=312

2) Synthesis of Intermediate I69

7.1 g of Intermediate I69 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 83%).  $_{65}$ 

MS [M+H]+=445

3) Synthesis of Compound 56

Under nitrogen atmosphere, 3.00 g of Intermediate I69, 6.32 g of boron triiodide, and 1.43 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.4 g of Compound 56 (yield 46.0%).

MS[M+H]+=453

Synthesis Example 57. Synthesis of Compound 57

1) Synthesis of Intermediate I70

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7.2 g of Intermediate I70 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 67%).

MS[M+H]+=557

2) Synthesis of Compound 57

Under nitrogen atmosphere, 3.00 g of Intermediate I70, 6.32 g of boron triiodide, and 1.43 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.5 g of Compound 57 (yield 49.0%).

Synthesis Example 58. Synthesis of Compound 58

1) Synthesis of Intermediate I71

7.1 g of Intermediate I71 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 66%).

I71

MS[M+H]+=557

2) Synthesis of Compound 58

Under nitrogen atmosphere, 3.00 g of Intermediate I71, 6.32 g of boron triiodide, and 1.43 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The <sup>20</sup> reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.4 g of Compound 58 (yield 46.0%).

MS [M+H] = 565

Synthesis Example 59. Synthesis of Compound 59

## 1) Synthesis of Intermediate I72

7.1 g of Intermediate I72 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 83%).  $^{65}$ 

2) Synthesis of Compound 59

Under nitrogen atmosphere, 3.00 g of Intermediate 172, 7.96 g of boron triiodide, and 1.80 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 30 1.3 g of Compound 59 (yield 43.0%).

MS [M+H]+=451

Synthesis Example 60. Synthesis of Compound 60

## 1) Synthesis of Intermediate I73

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7.7 g of Intermediate I73 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 72%).

MS[M+H]+=554

2) Synthesis of Compound 60

Under nitrogen atmosphere, 3.00 g of Intermediate I73, 6.35 g of boron triiodide, and 1.44 g of triphenylborane were stirred using dichlorobenzene at  $160^{\circ}$  C. for 6 hours.

The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.5 g of Compound 60 (yield 49.0%).

MS[M+H]+=563

Synthesis Example 61. Synthesis of Compound 61

# 1) Synthesis of Intermediate I74

8.3 g of Intermediate I74 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 72%).

MS[M+H]+=597

2) Synthesis of Compound 61

Under nitrogen atmosphere, 3.00 g of Intermediate 174, 5.90 g of boron triiodide, and 1.34 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified

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with ethyl acetate:hexane, and then recrystallized to obtain 1.4 g of Compound 61 (yield 46.0%).

MS[M+H]+=605

Synthesis Example 62. Synthesis of Compound 62

1) Synthesis of Intermediate I75

8.4 g of Intermediate I75 was obtained using the same conditions as in the synthesis method of Intermediate I3 65 under nitrogen atmosphere (yield 73%).

2) Synthesis of Compound 62

Under nitrogen atmosphere, 3.00 g of Intermediate I75, 5.90 g of boron triiodide, and 1.34 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.4 g of Compound 62 (yield 46.0%).

MS[M+H]+=605

Synthesis Example 63. Synthesis of Compound 63

1) Synthesis of Intermediate I76

$$+$$
  $CI$   $\rightarrow$   $Br$ 

39 g of Intermediate I76 was obtained using the same conditions as in the synthesis method of Intermediate I2 under nitrogen atmosphere (yield 88%).

MS[M+H]+=340

2) Synthesis of Intermediate I77

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ \end{array}$$

 $6.6~{\rm g}$  of Intermediate I77 was obtained using the same  $_{50}$  conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 79%).

MS[M+H]+=473

3) Synthesis of Compound 63

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-continued

Under nitrogen atmosphere, 3.00 g of Intermediate 177, 7.45 g of boron triiodide, and 1.69 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.3 g of Compound 63 (yield 43.0%).

MS[M+H]+=481

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Synthesis Example 64. Synthesis of Compound 64

1) Synthesis of Intermediate I78

8.1 g of Intermediate I78 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 78%).

I78

I79

Under nitrogen atmosphere, 3.00 g of Intermediate I78, 6.02 g of boron triiodide, and 1.37 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.4 g of Compound 64 (yield 46.0%).

MS[M+H]+=593

8.0 g of Intermediate I79 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 77%).

MS[M+H]+=585

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2) Synthesis of Compound 65

Synthesis Example 65. Synthesis of Compound 65

1) Synthesis of Intermediate I79

Under nitrogen atmosphere, 3.00~g of Intermediate I79, 6.02~g of boron triiodide, and 1.37~g of triphenylborane were stirred using dichlorobenzene at  $160^{\circ}$  C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.3~g of Compound 65 (yield 43.0%).

MS [M+H]+=593

Synthesis Example 66. Synthesis of Compound 66

## 1) Synthesis of Intermediate I80

7.1 g of Intermediate I80 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 85%).

2) Synthesis of Compound 66

Under nitrogen atmosphere, 3.00 g of Intermediate 180, 7.49 g of boron triiodide, and 1.70 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.2 g of Compound 66 (yield 39.0%).

MS [M+H]+=479

Synthesis Example 67. Synthesis of Compound 67

# 1) Synthesis of Intermediate I81

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7.7 g of Intermediate I81 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 75%).

MS[M+H]+=583

2) Synthesis of Compound 67

Under nitrogen atmosphere, 3.00 g of Intermediate I81, 6.05 g of boron triiodide, and 1.37 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.3 g of Compound 67 (yield 43.0%).

MS[M+H]+=591

Synthesis Example 68. Synthesis of Compound 68

1) Synthesis of Intermediate I82

-continued

8.1 g of Intermediate I82 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 73%).

MS [M+H]+=625

2) Synthesis of Compound 68

Under nitrogen atmosphere, 3.00 g of Intermediate I82, 5.64 g of boron triiodide, and 1.28 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 1.3 g of Compound 68 (yield 43.0%).

$$MS[M+H]+=633$$

Synthesis Example 69. Synthesis of Compound 69

## 1) Synthesis of Intermediate I83

**750** 

8.0 g of Intermediate I83 was obtained using the same conditions as in the synthesis method of Intermediate I3 under nitrogen atmosphere (yield 73%).

MS[M+H]+=625

## 2) Synthesis of Compound 69

Under nitrogen atmosphere, 3.00 g of Intermediate 183, 5.64 g of boron triiodide, and 1.28 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified

with ethyl acetate:hexane, and then recrystallized to obtain 1.4 g of Compound 69 (yield 46.0%).

MS[M+H]+=633

Synthesis Example 70. Synthesis of Compound 70

## 1) Synthesis of Intermediate I84

After 40 g of Intermediate I4, 19.56 g of diphenylamine, 33.28 g of sodium-tert-butoxide, and 0.56 g of bis(tri-tertbutylphosphine)palladium(0) were put into 800 ml of tolu- 45 ene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 48 g of Intermediate I84 (yield 87%).

MS[M+H]+=480

### 2) Synthesis of Intermediate I85

Under nitrogen atmosphere, 30 g of Intermediate I84, 73.55 g of boron triiodide, and 16.65 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified <sup>20</sup> with ethyl acetate:hexane, and then recrystallized to obtain 14.5 g of Intermediate I85 (yield 48.0%).

MS[M+H]+=487

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## 3) Synthesis of Compound 70

After 2 g of Intermediate I85, 0.69 g of diphenylamine, 1.18 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tertbutylphosphine)palladium(0) were put into 30 ml of toluene 65 under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an

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ethyl acetate:hexane column, and then recrystallized to obtain  $2.0~{\rm g}$  of Compound 70 (yield 79%).

MS[M+H]+=620

Synthesis Example 71. Synthesis of Compound 71

## 1) Synthesis of Compound 71

After 2 g of Intermediate 185, 1.16 g of bis(4-(tert-butyl) phenyl)amine, 1.18 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.3 g of Compound 71 (yield 77%).

MS[M+H]+=732

1) Synthesis of Compound 72

After 2 g of Intermediate I85, 1.29 g of bis(4-(trimethylsilyl)phenyl)amine, 1.18 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.4 g of Compound 72 (yield 76%). MS [M+H]+=764

Synthesis Example 73. Synthesis of Compound 73

Synthesis Example 74. Synthesis of Compound 74

## 1) Synthesis of Compound 73

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1) Synthesis of Compound 74

After 2 g of Intermediate I85, 2.82 g of bis(4-(triphenyl-silyl)phenyl)amine, 1.18 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.5 g of Compound 73 (yield 54%).

After 2 g of Intermediate I85, 1.32 g of di([1,1'-biphenyl]-3-yl)amine, 1.18 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.2 g of Compound 74 (yield 69%).

MS[M+H]+=772

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Synthesis Example 75. Synthesis of Compound 75

2) Synthesis of Intermediate I87

## 1) Synthesis of Intermediate I86

After 40 g of Intermediate I4, 32.5 g of bis(3-(tert-butyl) phenyl)amine, 33.3 g of sodium-tert-butoxide, and 0.59 g of bis(tri-tert-butylphosphine)palladium(0) were put into 800 60 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 50 g 65 of Intermediate I86 (yield 73%).

Under nitrogen atmosphere, 30 g of Intermediate I86, 59.59 g of boron triiodide, and 13.51 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 14.1 g of Intermediate I87 (yield 46.0%).

MS[M+H]+=6003) Synthesis of Compound 75

After 2 g of Intermediate I87, 0.57 g of diphenylamine, <sup>20</sup> 0.96 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an <sup>25</sup> ethyl acetate:hexane column, and then recrystallized to obtain 1.9 g of Compound 75 (yield 78%).

MS [M+H]+=732

Synthesis Example 76. Synthesis of Compound 76

1) Synthesis of Compound 76

-continued

After 2 g of Intermediate I87, 0.94 g of bis(4-(tert-butyl) phenyl)amine, 0.96 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.1 g of Compound 76 (yield 75%).

 $^{0}$  MS[M+H]+=845

Synthesis Example 77. Synthesis of Compound 77

1) Synthesis of Compound 77

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After 2 g of Intermediate I87, 1.05 g of bis(4-(trimethylsilyl)phenyl)amine, 0.96 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.0 g of Compound 77 (yield 68%). 35

MS[M+H]+=877

Synthesis Example 78. Synthesis of Compound 78

1) Synthesis of Compound 78

After 2 g of Intermediate I87, 2.29 g of bis(4-(triphenylsilyl)phenyl)amine, 0.96 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then 65 recrystallized to obtain 2.3 g of Compound 78 (yield 55%).

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# 1) Synthesis of Intermediate I88

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Synthesis Example 80. Synthesis of Compound 80

## 1) Synthesis of Compound 79

After 2 g of Intermediate I87, 1.07 g of di([1,1'-biphenyl]-3-yl)amine, 0.96 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.4 g of Compound 79 (yield 81%).

After 40 g of Intermediate I4, 19.3 g of 9H-carbazole, 33.3 g of sodium-tert-butoxide, and 0.59 g of bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 43 g of Intermediate I88 (yield 78%).

MS[M+H]+=478

2) Synthesis of Intermediate I89

Under nitrogen atmosphere, 30 g of Intermediate I88, 15 73.86 g of boron triiodide, and 16.75 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 20 14.5 g of Intermediate I89 (yield 48.0%).

MS[M+H]+=485

## 3) Synthesis of Compound 80

After 2 g of Intermediate I89, 0.69 g of diphenylamine, 1.19 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred 65 for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an

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ethyl acetate:hexane column, and then recrystallized to obtain 1.9 g of Compound 80 (yield 78%).

MS[M+H]+=732

Synthesis Example 81. Synthesis of Compound 81

## 1) Synthesis of Compound 81

After 2 g of Intermediate I89, 1.16 g of bis(4-(tert-butyl) phenyl)amine, 1.19 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.3 g of Compound 81 (yield 76%).

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Synthesis Example 82. Synthesis of Compound 82

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Synthesis Example 83. Synthesis of Compound 83

1) Synthesis of Compound 82

B N N

1) Synthesis of Compound 83

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After 2 g of Intermediate 189, 1.29 g of bis(4-(trimethylsilyl)phenyl)amine, 1.19 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.6 g of Compound 82 (yield 83%).

After 2 g of Intermediate I89, 2.82 g of bis(4-(triphenyl-silyl)phenyl)amine, 1.19 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.8 g of Compound 83 (yield 60%).

MS[M+H]+=1135

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Synthesis Example 85. Synthesis of Compound 85

1) Synthesis of Compound 84

After 2 g of Intermediate I89, 1.32 g of di([1,1'-biphenyl]-3-yl)amine, 1.19 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml  $\,^{60}$ of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with obtain 2.1 g of Compound 84 (yield 66%).

MS[M+H]+=770

After 40 g of Intermediate I4, 32.28 g of 3,6-di-tert-butyl-9H-carbazole, 33.3 g of sodium-tert-butoxide, and 0.59 g of bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then an ethyl acetate:hexane column, and then recrystallized to  $_{65}$  purified with an ethyl acetate:hexane column to obtain 47 g of Intermediate I90 (yield 69%).

# 2) Synthesis of Intermediate I91

Under nitrogen atmosphere, 30 g of Intermediate 190, 59.80 g of boron triiodide, and 13.56 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 14.6 g of Intermediate 191 (yield 48.0%).

MS[M+H]+=598

## 3) Synthesis of Compound 85

After 2 g of Intermediate I91, 0.57 g of diphenylamine, 0.97 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.2 g of Compound 85 (yield 49%).

MS[M+H]+=730

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Synthesis Example 86. Synthesis of Compound 86

# 1) Synthesis of Compound 86

After 2 g of Intermediate I91, 0.94 g of bis(4-(tert-butyl) phenyl)amine, 1.19 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.1 g of Compound 86 (yield 74%).

MS[M+H]+=843

Synthesis Example 87. Synthesis of Compound 87

1) Synthesis of Compound 87

After 2 g of Intermediate 191, 1.05 g of bis(4-(trimethyl-silyl)phenyl)amine, 0.97 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.0 g of Compound 87 (yield 68%).

MS[M+H]+=875

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Synthesis Example 88. Synthesis of Compound 88

1) Synthesis of Compound 88

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Synthesis Example 89. Synthesis of Compound 89

1) Synthesis of Compound 89

After 2 g of Intermediate I91, 2.30 g of bis(4-(triphenyl-silyl)phenyl)amine, 0.97 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 3.1 g of Compound 88 (yield 74%).

MS[M+H]+=1247

88

After 2 g of Intermediate I91, 1.08 g of di([1,1'-biphenyl]-3-yl)amine, 0.97 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.0 g of Compound 89 (yield 78%).

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Synthesis Example 90. Synthesis of Compound 90

2) Synthesis of Intermediate I93

## 1) Synthesis of Intermediate I92

After 40 g of Intermediate 14, 32.28 g of 3,6-di-tert-butyl-9H-carbazole, 33.3 g of sodium-tert-butoxide, and 0.59 g of 60 bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 50 g 65 of Intermediate I92 (yield 69%).

Under nitrogen atmosphere, 30 g of Intermediate 192, 56.0 g of boron triiodide, and 12.7 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 13.9 g of Intermediate 193 (yield 46.0%).

MS[M+H]+=638

## 3) Synthesis of Compound 90

After 2 g of Intermediate I93, 0.53 g of diphenylamine, 0.90 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tertbutylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the  $_{40}$ resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.5 g of Compound 90 (yield 62%).

MS [M+H] = 770

Synthesis Example 91. Synthesis of Compound 91

## 1) Synthesis of Compound 91

**780** 

After 2 g of Intermediate I93, 0.88 g of bis(4-(tert-butyl) phenyl)amine, 0.91 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.9 g of Compound 91 (yield 69%).

MS[M+H]+=883

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Synthesis Example 92. Synthesis of Compound 92

## 1) Synthesis of Compound 92

After 2 g of Intermediate I93, 0.98 g of bis(4-(trimethylsilyl)phenyl)amine, 0.91 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 40 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.8 g of Compound 92 (yield 63%). 45 MS[M+H]+=915

Synthesis Example 93. Synthesis of Compound 93

## 1) Synthesis of Compound 93

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After 2 g of Intermediate I93, 2.15 g of bis(4-(trimethylsilyl)phenyl)amine, 0.91 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.3 g of Compound 93 (yield 57%).

MS[M+H]+=1287

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Synthesis Example 94. Synthesis of Compound 94

Synthesis Example 95. Synthesis of Compound 95

## 1) Synthesis of Compound 94

1) Synthesis of Intermediate I94

After 2 g of Intermediate I93, 1.01 g of di([1,1'-biphenyl]-3-yl)amine, 0.91 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml 60 of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with obtain 1.7 g of Compound 94 (yield 59%).

After 20 g of Intermediate I4, 39.5 g of 3,6-bis(triphenylsilyl)-9H-carbazole, 16.7 g of sodium-tert-butoxide, and 0.3 g of bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then an ethyl acetate:hexane column, and then recrystallized to  $_{65}$  purified with an ethyl acetate:hexane column to obtain 40 g of Intermediate I94 (yield 70%).

I94

MS[M+H]+=923

# 2) Synthesis of Intermediate 195

Under nitrogen atmosphere, 30 g of Intermediate I94, 40 35.5 g of boron triiodide, and 8.1 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then recrystallized to obtain 12.0 g of Intermediate I95 (yield 40.0%).

## MS[M+H]+=1002

#### 3) Synthesis of Compound 95

I95

After 2 g of Intermediate 195, 0.33 g of diphenylamine, 0.58 g of sodium-tert-butoxide, and 0.01 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then recrystallized to obtain 1.4 g of Compound 95 (yield 62%).

# MS[M+H]+=1135

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Synthesis Example 96. Synthesis of Compound 96

## 1) Synthesis of Intermediate I96

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After 30 g of Intermediate I4, 27.8 g of di([1,1'-biphenyl]-3-yl)amine, 24.9 g of sodium-tert-butoxide, and 0.4 g of bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 41 g of Intermediate I96 (yield 75%).

MS[M+H]+=632

2) Synthesis of Intermediate 197

Under nitrogen atmosphere, 30 g of Intermediate 196, 55.82 g of boron triiodide, and 12.6 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then recrystallized to obtain 14.0 g of Intermediate 197 (yield 46.0%).

MS [M+H]+=640

3) Synthesis of Compound 96

After 2 g of Intermediate 197, 0.53 g of diphenylamine, 0.9 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then recrystallized to obtain 1.3 g of Compound 96 (yield 54%).

Synthesis Example 97. Synthesis of Compound 97

1) Synthesis of Compound 97

After 2 g of Intermediate I97, 0.88 g of bis(4-(tert-butyl) phenyl)amine, 0.91 g of sodium-tert-butoxide, and 0.02 g of  $_{60}$ bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to 65 reaction, the resulting product was extracted, and then obtain 1.8 g of Compound 97 (yield 65%).

MS [M+H]+=885

1) Synthesis of Compound 98

After 2 g of Intermediate I97, 0.98 g of bis(4-(trimethylsilyl)phenyl)amine, 0.91 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the recrystallized to obtain 1.9 g of Compound 98 (yield 66%).

# **792**Synthesis Example 100. Synthesis of Compound 100

#### 1) Synthesis of Compound 99

After 2 g of Intermediate I97, 2.14 g of bis(4-(triphenyl-silyl)phenyl)amine, 0.90 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then recrystallized to obtain 2.4 g of Compound 99 (yield 60%).

MS[M+H]+=1289

1) Synthesis of Compound 100

After 2 g of Intermediate 197, 1.01 g of di([1,1'-biphenyl]-3-yl)amine, 0.90 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then recrystallized to obtain 2.1 g of Compound 100 (yield 73%).

MS[M+H]+=923

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Synthesis Example 101. Synthesis of Compound

2) Synthesis of Intermediate I99

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & &$$

After 10 g of 4a,9a-dimethyl-6-(triphenylsilyl)-2,3,4,4a, 9,9a-hexahydro-1H-carbazole, 4.91 g of 1-bromo-3,5-dichlorobenzene, 6.28 g of sodium-tert-butoxide, and 0.11 g of bis(tri-tert-butylphosphine)palladium(0) were put into 150 60 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 11 g  $_{65}$  ethyl acetate:hexane column to obtain 10 g of Intermediate of Intermediate I98 (yield 84%).

After 11 g of Intermediate I98, 3.04 g of 9H-carbazole, 0.09~g~of~bis(tri-tert-butylphosphine)palladium(0), and 5.24g of sodium-tert-butoxide were put into 180 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an 199 (yield 75%).

MS[M+H]+=605

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# 3) Synthesis of Intermediate I100

4) Synthesis of Compound 101

ČI I100

Under nitrogen atmosphere, 10~g of Intermediate I99, 15.97~g of boron triiodide, and 3.62~g of triphenylborane were stirred using dichlorobenzene at  $160^{\circ}$  C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then recrystallized to 65~obtain 4.0~g of Intermediate I100 (yield 40.0%).

After 2 g of Intermediate I100, 0.53 g of di-o-tolylamine, 0.78 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then recrystallized to obtain 1.8 g of Compound 101 (yield 74%).

MS[M+H]+=744

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#### 1) Synthesis of Compound 102

After 2 g of Intermediate I100, 0.75 g of bis(4-(tert-butyl) phenyl)amine, 0.77 g of sodium-tert-butoxide, and 0.01 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture 65 was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with

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an ethyl acetate:hexane column, and then recrystallized to obtain 1.8 g of Compound 102 (yield 68%).

$$MS[M+H]+=989$$

Synthesis Example 103. Synthesis of Compound 103

## 1) Synthesis of Intermediate I100

After 15 g of 4a,9a-dimethyl-6-(trimethylsilyl)-2,3,4,4a, 9,9a-hexahydro-1H-carbazole, 12.39 g of 1-bromo-3,5-dichlorobenzene, 15.81 g of sodium-tert-butoxide, and 0.28 g of bis(tri-tert-butylphosphine)palladium(0) were put into 300 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 18 g of Intermediate 1100 (yield 78%).

## MS[M+H]+=419

## 2) Synthesis of Intermediate I101

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Under nitrogen atmosphere, 15 g of Intermediate I101, 25.4 g of boron triiodide, and 5.76 g of triphenylborane were stirred using dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 5.5 g of Intermediate I102 (yield 36.0%).

MS[M+H]+=702

4) Synthesis of Compound 103

After 15 g of Intermediate I100, 11.16 g of 3,6-bis 40 (trimethylsilyl)-9H-carbazole, 0.18 g of bis(tri-tert-butylphosphine)palladium(0), and 10.33 g of sodium-tert-butoxide were put into 180 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product 45 was extracted, and then purified with an ethyl acetate:hexane column to obtain 16 g of Intermediate I101 (yield 64%).

MS[M+H]+=694

3) Synthesis of Intermediate I102

103

0.82 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-

butylphosphine)palladium(0) were put into 30 ml of toluene

under nitrogen atmosphere, the resulting mixture was stirred

resulting product was extracted, and then recrystallized to

for 3 hours. After the completion of the reaction, the 5

of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.6 g of Compound 104 (yield 59%).

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MS[M+H]+=947

Synthesis Example 105. Synthesis of Compound 105

#### 1) Synthesis of Compound 105

After 2 g of Intermediate I85, 0.86 g of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 1.18 g of sodium-tert-butoxide, and 0.06 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.1 g of Compound 105 (yield 78%).

MS[M+H]+=652

Synthesis Example 104. Synthesis of Compound

#### 1) Synthesis of Compound 104

MS[M+H]+=835

obtain 1.5 g of Compound 103 (yield 63%).

After 2 g of Intermediate I102, 0.80 g of bis(4-(tert-butyl) 65 phenyl)amine, 0.82 g of sodium-tert-butoxide, and 0.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml

104

Synthesis Example 106. Synthesis of Compound 106

## 1) Synthesis of Intermediate I103

After 40 g of 4a,5,7,9a-tetramethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 39.4 g of 1-bromo-3,5-dichlorobenzene, 50.3 g of sodium-tert-butoxide, and 2.67 g of bis(tritert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 45 g of Intermediate I103 (yield 69%).

#### MS[M+H]+=375

## 2) Synthesis of Intermediate I104

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After 30 g of Intermediate I103, 32.7 g of 4-(tert-butyl)-N-(4-(tert-butyl)phenyl)-2-(naphthalen-2-yl)aniline, 23.1 g of sodium-tert-butoxide, and 1.23 g of bis(tri-tert-butylphosphine)palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 4 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 31 g of Intermediate I104 (yield 52%).

MS[M+H]+=746

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3) Synthesis of Intermediate I105

Under nitrogen atmosphere, 15 g of Intermediate I104, 23.6 g of boron triiodide, and 5.4 g of triphenylborane were stirred using 150 ml of dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to

obtain 6.9 g of Intermediate 1105 (yield 47%).

MS [M+H] = 736

4) Synthesis of Compound 106

**806**Synthesis Example 107. Synthesis of Compound

1) Synthesis of Intermediate I106

After 40 g of 9a-methyl-4a-phenyl-2,3,4,4a,9,9a-hexa30 hydro-1H-carbazole, 34.3 g of 1-bromo-3,5-dichlorobenzene, 43.8 g of sodium-tert-butoxide, and 2.32 g of bis(tritert-butylphosphine)palladium(0) were put into 800 ml of
toluene under nitrogen atmosphere, the resulting mixture
was stirred for 6 hours. After the completion of the reaction,
35 the resulting product was extracted, and then purified with
an ethyl acetate:hexane column to obtain 41 g of Intermediate I106 (yield 66%).

MS[M+H]+=409

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2) Synthesis of Intermediate I107

+ H

After 2 g of Intermediate I105, 0.83 g of 8-(tert-butyl)-6,6a,11,11a-tetrahydro-5H-benzo[a]carbazole, 0.78 g of sodium-tert-butoxide, and 0.04 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.8 g of Compound 106 (yield 65%).

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After 30 g of Intermediate I106, 29.9 g of 4-(tert-butyl)-N-(4-(tert-butyl)phenyl)-2-(naphthalen-2-yl)aniline, 21.2 g of sodium-tert-butoxide, and 1.13 g of bis(tri-tert-butylphos-phine)palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 4 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 29 g of Intermediate I107 (yield 51%).

MS[M+H]+=780

3) Synthesis of Intermediate I108

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Under nitrogen atmosphere, 15 g of Intermediate I107, 22.6 g of boron triiodide, and 5.1 g of triphenylborane were stirred using 150 ml of dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 6.9 g of Intermediate I108 (yield 46%).

MS[M+H]+=788

4) Synthesis of Compound 107

After 2 g of Intermediate I108, 0.78 g of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 0.73 g of sodium-tert-butoxide, and 0.04 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.6 g of Compound 107 (yield 66%).

Synthesis Example 108. Synthesis of Compound 108

1) Synthesis of Intermediate I109

After 40 g of 4a,9a-dimethyl-6-phenyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 32.6 g of 1-bromo-3,5-dichlorobenzene, 41.6 g of sodium-tert-butoxide, and 2.21 g of bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate: hexane column to obtain 39 g of Intermediate I109 (yield 64%).

MS[M+H]+=423

2) Synthesis of Intermediate I110

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After 30 g of Intermediate I109, 28.9 g of 5-(tert-butyl)-N-(3-(tert-butyl)phenyl)-[1,1'-biphenyl]-2-amine, 20.5 g of sodium-tert-butoxide, and 1.01 g of bis(tri-tert-butylphos-phine)palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 4 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 25 g of Intermediate I110 (yield 47%).

MS[M+H]+=744

3) Synthesis of Intermediate I111

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Under nitrogen atmosphere, 15 g of Intermediate I110, 23.7 g of boron triiodide, and 5.4 g of triphenylborane were stirred using 150 ml of dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 5.9 g of Intermediate I111 (yield 39%).

I111

MS[M+H]+=752

## 4) Synthesis of Compound 108

After 2 g of Intermediate I111, 0.74 g of 4a,9a-dimethyl-6-phenyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 0.77 g of sodium-tert-butoxide, and 0.04 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.7 g of Compound 108 (yield 64%).

MS[M+H]+=993

Synthesis Example 109. Synthesis of Compound 109

## 1) Synthesis of Intermediate I112

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 $Cl$ 
 $Cl$ 
 $Cl$ 
 $I112$ 

After 40 g of 4a,9a-dimethyl-6-(trimethylsilyl)-2,3,4,4a, 9,9a-hexahydro-1H-carbazole, 33.0 g of 1-bromo-3,5-di-

chlorobenzene, 42.2 g of sodium-tert-butoxide, and 2.24 g of bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate: hexane column to obtain 39 g of Intermediate I112 (yield 64%).

## MS[M+H]+=419

## 2) Synthesis of Intermediate I113

After 30 g of Intermediate I112, 23.0 g of di([1,1'-biphenyl]-3-yl), 20.7 g of sodium-tert-butoxide, and 1.10 g of bis(tri-tert-butylphosphine)palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 4 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 24 g of Intermediate I113 (yield 48%).

I113

# MS[M+H]+=704

## 3) Synthesis of Intermediate I114

Under nitrogen atmosphere, 15 g of Intermediate I113, 25.0 g of boron triiodide, and 5.7 g of triphenylborane were stirred using 150 ml of dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 6.0 g of Intermediate I114 (yield 40%).

## MS[M+H]+=712

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## 4) Synthesis of Compound 109

After 2 g of Intermediate I114, 0.77 g of 4a,9a-dimethyl-6-(trimethylsilyl)-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 0.81 g of sodium-tert-butoxide, and 0.04 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.8 g of Compound 109 (yield 68%).

MS[M+H]+=949

Synthesis Example 110. Synthesis of Compound 110

#### 1) Synthesis of Intermediate I115

After 40 g of 8-(tert-butyl)-6a,11a-dimethyl-6,6a,11,11a-tetrahydro-5H-benzo[a]carbazole, 29.6 g of 1-bromo-3,5-dichlorobenzene, 37.8 g of sodium-tert-butoxide, and 2.0 g 65 of bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting

mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 38 g of Intermediate I115 (yield 64%).

MS[M+H]+=451

2) Synthesis of Intermediate I116

After 30 g of Intermediate I115, 23.8 g of 5-(tert-butyl)-N-(3-(tert-butyl)phenyl)-[1,1'-biphenyl]-2-amine, 19.2 g of sodium-tert-butoxide, and 1.02 g of bis(tri-tert-butylphosphine)palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 4 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 25 g of Intermediate I116 (yield 49%).

Under nitrogen atmosphere, 15 g of Intermediate I116, 35 22.8 g of boron triiodide, and 5.2 g of triphenylborane were stirred using 150 ml of dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to 40 obtain 6.1 g of Intermediate I117 (yield 40%).

MS[M+H]+=780

4) Synthesis of Compound 110

818

After 2 g of Intermediate I117, 0.70 g of 6a,11a-dimethyl-6,6a,11,11a-tetrahydro-5H-benzo[a]carbazole, 0.81 g of sodium-tert-butoxide, and 0.04 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.9 g of Compound 110 (yield 68%).

MS[M+H]+=993

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Synthesis Example 111. Synthesis of Compound 111

1) Synthesis of Intermediate I118

After 40 g of 6a,11a-dimethyl-6,6a,11,11a-tetrahydro-5H-benzo[a]carbazole, 36.2 g of 1-bromo-3,5-dichlorobenzene, 46.2 g of sodium-tert-butoxide, and 2.5 g of bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the

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resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 41 g of Intermediate I118 (yield 65%).

MS[M+H]+=395

2) Synthesis of Intermediate I119

After 30 g of Intermediate I118, 12.9 g of diphenylamine, 21.9 g of sodium-tert-butoxide, and 1.17 g of bis(tri-tert-butylphosphine)palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 4 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 22 g of Intermediate I119 (yield 55%).

MS[M+H]+=528

3) Synthesis of Intermediate I120

$$\bigcap_{N} \bigcap_{N} \bigcap_{N$$

820

Under nitrogen atmosphere, 15 g of Intermediate I119, 33.4 g of boron triiodide, and 7.6 g of triphenylborane were stirred using 150 ml of dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 5.9 g of Intermediate I120 (yield 39%).

MS[M+H]+=535

4) Synthesis of Compound 111

After 2 g of Intermediate I120, 0.75 g of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 1.1 g of sodium-tert-butoxide, and 0.06 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.6 g of Compound 111 (yield 61%).

Synthesis Example 112. Synthesis of Compound 112

## 1) Synthesis of Intermediate I21

After 30 g of Intermediate I109, 12.0 g of diphenylamine, 20.5 g of sodium-tert-butoxide, and 1.09 g of bis(tri-tert-butylphosphine)palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 4 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 21 g of Intermediate 65 l121 (yield 53%).

## 2) Synthesis of Intermediate I122

Under nitrogen atmosphere, 15 g of Intermediate I121, 31.7 g of boron triiodide, and 7.2 g of triphenylborane were stirred using 150 ml of dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 6.3 g of Intermediate I122 (yield 41%).

I122

## MS[M+H]+=563

## 3) Synthesis of Compound 112

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After 2 g of Intermediate I121, 1.39 g of 6-(tert-butyl)-9a-methyl-4a-phenyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 1.0 g of sodium-tert-butoxide, and 0.06 g of bis(tritert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate: hexane column, and then recrystallized to obtain 1.6 g of Compound 112 (yield 53%).

MS[M+H]+=846

Synthesis Example 113. Synthesis of Compound

# 1) Synthesis of Intermediate I123

-continued

Cl

HO

N

I123

After 40 g of 6-chloro-4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 31.7 g of 3-bromo-5-methylphenol, 48.9 g of sodium-tert-butoxide, and 2.6 g of bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred at 70° C. for 8 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 38 g of Intermediate I123 (yield 66%).

MS[M+H]+=341.88

2) Synthesis of Intermediate I124

NfO NfO NfO

After 30 g of Intermediate I123, 23.6 ml of 1,1,2,2,3,3, 4,4,4-nonafluorobutane-1-sulfonyl fluoride, and 36.4 g of potassium carbonate were put into 400 ml of methyl chloride, the resulting mixture was stirred at room temperature for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 45 g of Intermediate I124 (yield 82%).

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## 2) Synthesis of Intermediate I125

After 30 g of Intermediate I124, 20.9 g of bis(3-(tertbutyl) phenyl) amine, 47.0 g of cesium carbonate, 0.83 g of  $_{60}\,$ bis(dibenzylidineacetone)palladium(0), and 1.38 g of Xphos were put into 600 ml of xylene, the resulting mixture was refluxed and stirred for 8 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 19 g 65 of Intermediate I125 (yield 65%).

Under nitrogen atmosphere, 15 g of Intermediate I125, 29.1 g of boron triiodide, and 6.6 g of triphenylborane were stirred using 150 ml of dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-40 purified with ethyl acetate:hexane, and then recrystallized to obtain 5.8 g of Intermediate I126 (yield 38%).

## MS[M+H]+=614

## 4) Synthesis of Compound 113

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After 2 g of Intermediate I126, 0.55 g of diphenylamine, <sup>20</sup> 0.9 g of sodium-tert-butoxide, and 0.05 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.6 g of Compound 113 (yield 66%).

MS[M+H]+=746

Synthesis Example 114. Synthesis of Compound 114

# 1) Synthesis of Intermediate I127

After 40 g of 6-(tert-butyl)-4a,9a-dimethyl-2,3,4,4a,9,9a-65 hexahydro-1H-carbazole, 31.9 g of 1-bromo-3-chloro-5-methylbenzene, 44.8 g of sodium-tert-butoxide, and 2.4 g of

bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred at 70° C. for 8 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 41 g of Intermediate I127 (yield 69%).

MS[M+H]+=382

2) Synthesis of Intermediate I128

After 30 g of Intermediate I127, 20.3 g of sodium-tert-butoxide, and 1.07 g of bis(tri-tert-butylphosphine)palladium(0) were put into 500 ml of toluene under nitrogen atmosphere, a solution in which 23.6 g of 5-(tert-butyl)-N-(3-chlorophenyl)-[1,1'-biphenyl]-2-amine was dissolved in toluene was added dropwise thereto when the resulting mixture began to boil, and stirred for 2 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 28 g of Intermediate I128 (yield 58%).

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Under nitrogen atmosphere, 15 g of Intermediate I128, 25.9 g of boron triiodide, and 5.9 g of triphenylborane were stirred using 150 ml of dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 5.4 g of Intermediate I129 (yield 36%).

MS[M+H]+=690

4) Synthesis of Compound 114

830

After 2 g of Intermediate I128, 0.82 g of bis(-4-(tert-butyl)phenyl)amine, 0.9 g of sodium-tert-butoxide, and 0.05 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.9 g of Compound 114 (yield 70%). MS[M+H]+=935

Synthesis Example 115. Synthesis of Compound 115

1) Synthesis of Intermediate I130

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After 40 g of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 26.9 g of 1,3-dibromo-5-chlorobenzene, 57.3 g of sodium-tert-butoxide, and 1.0 g of bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 4 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 33 g of Intermediate 1130 (yield 65%).

MS[M+H]+=512

2) Synthesis of Intermediate I131

Under nitrogen atmosphere, 15 g of Intermediate I130, 34.4 g of boron triiodide, and 7.8 g of triphenylborane were stirred using 150 ml of dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-purified with ethyl acetate:hexane, and then recrystallized to obtain 6.9 g of Intermediate I131 (yield 45%).

MS[M+H]+=519

3) Synthesis of Compound 115

-continued

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N
N
N
115

After 2 g of Intermediate I131, 1.08 g of bis(4-(tert-butyl) phenyl)amine, 1.11 g of sodium-tert-butoxide, and 0.06 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 2.1 g of Compound 115 (yield 71%).

MS[M+H]+=764

Synthesis Example 116. Synthesis of Compound 116

1) Synthesis of Compound 116

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After 2 g of Intermediate I131, 0.78 g of 4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 1.11 g of sodium-tert-butoxide, and 0.06 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.8 g of Compound 116 (yield 68%).

MS[M+H]+=684

Synthesis Example 117. Synthesis of Compound

1) Synthesis of Intermediate I132

After 40 g of 6-(tert-butyl)-4a,9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 35.1 g of 1-bromo-3,5-dichlorobenzene, 44.8 g of sodium-tert-butoxide, and 0.79 g of bis(tri-tert-butylphosphine)palladium(0) were put into 800 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 6 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 38 g of Intermediate I132 (yield 61%).

f Intermediate I13 MS[M+H]+=403 2) Synthesis of Intermediate I133

40 After 30 g of Intermediate I132, 17.1 g of 4a,5,7,9a-tetramethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 21.5 g of sodium-tert-butoxide, and 1.14 g of bis(tri-tert-butylphos-phine)palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 4 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 21 g of Intermediate I133 (yield 47%).

I133

MS[M+H]+=596

3) Synthesis of Intermediate I134

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Under nitrogen atmosphere, 15 g of Intermediate I133, 29.6 g of boron triiodide, and 6.7 g of triphenylborane were stirred using 150 ml of dichlorobenzene at 160° C. for 6  $^{20}$ hours. The reaction was completed, and the resulting product was extracted at room temperature, and then columnpurified with ethyl acetate:hexane, and then recrystallized to obtain 6.1 g of Intermediate I134 (yield 40%).

MS[M+H]+=604

4) Synthesis of Compound 117

-continued

117

After 2 g of Intermediate I134, 0.92 g of 4a,9a-dimethyl-6-phenyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 0.95 g of sodium-tert-butoxide, and 0.05 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.6 g of Compound 117 (yield 57%).

MS[M+H]+=845

Synthesis Example 118. Synthesis of Compound 118

1) Synthesis of Compound 118

After 2 g of Intermediate I131, 1.28 g of 10H-spiro [acridine-9,9'-fluorene], 0.93 g of sodium-tert-butoxide, and 0.04 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.9 g of Compound 118 (yield 61%).

MS[M+H]+=814

Synthesis Example 119. Synthesis of Compound 119

## 1) Synthesis of Intermediate I135

838

After 30 g of Intermediate I132, 19.2 g of 5-(tert-butyl)-4a-9a-dimethyl-2,3,4,4a,9,9a-hexahydro-1H-carbazole, 17.9 g of sodium-tert-butoxide, and 1.14 g of bis(tri-tert-butylphosphine)palladium(0) were put into 600 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 4 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column to obtain 23 g of Intermediate I135 (yield 49%).

MS[M+H]+=624

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2) Synthesis of Intermediate I136

Under nitrogen atmosphere, 15 g of Intermediate I135, 28.3 g of boron triiodide, and 6.3 g of triphenylborane were stirred using 150 ml of dichlorobenzene at 160° C. for 6 hours. The reaction was completed, and the resulting product was extracted at room temperature, and then column-

purified with ethyl acetate:hexane, and then recrystallized to obtain 5.6 g of Intermediate I136 (yield 37%).

MS[M+H]+=632

3) Synthesis of Compound 119

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After 2 g of Intermediate I136, 1.10 g of 5'H-spiro [dibenzo[b,d]silole-5,10'-dibenzo[b,e][1,4]azasiline], 0.91 g of sodium-tert-butoxide, and 0.05 g of bis(tri-tert-butylphosphine)palladium(0) were put into 30 ml of toluene under nitrogen atmosphere, the resulting mixture was stirred for 3 hours. After the completion of the reaction, the resulting product was extracted, and then purified with an ethyl acetate:hexane column, and then recrystallized to obtain 1.8 g of Compound 119 (yield 60%).

840

Experimental Example

Example 1

A glass substrate thinly coated with indium tin oxide (ITO) to have a thickness of 1.500 Å was put into distilled water in which a detergent was dissolved, and ultrasonically washed. In this case, a product manufactured by Fischer Co. was used as the detergent, and distilled water, which had been filtered twice with a filter manufactured by Millipore Co., was used as the distilled water. After the ITO was washed for 30 minutes, ultrasonic washing was conducted twice repeatedly using distilled water for 10 minutes. After the washing using distilled water was completed, ultrasonic washing was conducted by using isopropyl alcohol, acetone, and methanol solvents, and the resulting product was dried and then transported to a plasma washing machine. Furthermore, the substrate was washed by using oxygen plasma for 5 minutes, and then was transported to a vacuum deposition machine.

The following Formula [HAT] was thermally vacuumdeposited to have a thickness of 50 Å on a transparent ITO electrode, which was prepared as described above, thereby forming a hole injection layer. The following Formula [NPB] was vacuum-deposited to have a thickness of 1,100 Å on the hole injection layer, thereby forming a hole transport layer. The following Formula [HT-A] was vacuumdeposited to have a thickness of 200 Å on the hole transport layer, thereby forming an electron blocking layer. Next, 2 wt % of Compound 1 as a blue light emitting dopant based on the total weight of a light emitting layer and 9-(naphthalen-1-vl)-10-(4-(naphthalen-2-vl)phenyl)anthracene [BH] as a host were vacuum-deposited to have a thickness of 300 Å on 35 the electron blocking layer, thereby forming the light emitting layer. [TPBI] and the following Formula [LiQ] were vacuum-deposited at a weight ratio of 1:1 on the light emitting layer, thereby forming a first electron transport layer having a thickness of 200 Å. [LiF] was vacuumdeposited on the first electron transport layer, thereby forming a second electron transport layer having a thickness of 100 Å. Aluminum was deposited to a thickness of 1,000 Å on the second electron transport layer, thereby forming a negative electrode. In the aforementioned procedure, the deposition rate of the organic material was maintained at 0.4 to 0.9 Å/sec, the deposition rates of lithium fluoride of the second electron transport layer and aluminum of the negative electrode were maintained at 0.3 Å/sec and 2 Å/sec, respectively, and the degree of vacuum during the deposition was maintained at 1×10-7 to 5×10-8 torr, thereby manufacturing an organic light emitting device.

Examples 2 to 119 and Comparative Examples 1 and 2

Organic light emitting devices were manufactured in the same manner as in Example 1, except that the compounds in the following Table 1 were used instead of Compound 1 in Example 1.

The efficiencies, service lives, and color coordinates (based on 1931 CIE color coordinate) of the organic light emitting devices, manufactured in Examples 1 to 119 and Comparative Examples 1 and 2, at a current density of 10 mA/cm² were measured, and the results thereof are shown in the following Table 1.

TABLE 1

Efficiency   CIE   CIE   life   (cd/A)   (x)   (y)   T95 (hr)			Color co	Service	
Example 1 (Compound 1) 7.23 0.150 0.066 112 Example 2 (Compound 2) 6.75 0.141 0.061 107 Example 3 (Compound 4) 7.22 0.149 0.065 111 Example 4 (Compound 4) 7.22 0.149 0.065 111 Example 5 (Compound 6) 7.30 0.150 0.074 118 Example 6 (Compound 6) 7.30 0.150 0.074 118 Example 7 (Compound 7) 7.23 0.149 0.066 110 Example 8 (Compound 8) 6.72 0.141 0.066 100 Example 8 (Compound 9) 7.43 0.150 0.070 125 Example 10 (Compound 10) 6.77 0.141 0.066 100 Example 10 (Compound 11) 6.77 0.141 0.063 109 Example 12 (Compound 12) 6.75 0.141 0.064 110 Example 13 (Compound 14) 7.21 0.150 0.095 116 Example 16 (Compound 15) 7.23 0.150 0.095 116 Example 16 (Compound 16) 7.23 0.150 0.095 116 Example 17 (Compound 17) 7.22 0.150 0.095 116 Example 18 (Compound 18) 6.77 0.141 0.063 109 Example 19 (Compound 19) 7.22 0.150 0.095 116 Example 10 (Compound 15) 7.23 0.150 0.095 116 Example 16 (Compound 16) 7.23 0.150 0.095 116 Example 17 (Compound 17) 7.22 0.150 0.095 116 Example 18 (Compound 19) 6.78 0.141 0.066 107 Example 20 (Compound 20) 7.21 0.150 0.095 116 Example 21 (Compound 21) 7.22 0.150 0.095 116 Example 22 (Compound 20) 7.21 0.150 0.095 116 Example 23 (Compound 20) 7.21 0.150 0.095 116 Example 24 (Compound 21) 7.22 0.150 0.097 116 Example 25 (Compound 27) 7.21 0.150 0.095 116 Example 26 (Compound 27) 7.21 0.150 0.095 116 Example 27 (Compound 28) 6.77 0.141 0.066 107 Example 28 (Compound 29) 7.21 0.150 0.095 116 Example 29 (Compound 23) 6.77 0.141 0.066 107 Example 26 (Compound 24) 7.21 0.150 0.095 116 Example 27 (Compound 25) 7.21 0.150 0.095 116 Example 28 (Compound 27) 6.77 0.141 0.066 107 Example 27 (Compound 28) 7.21 0.150 0.095 116 Example 28 (Compound 29) 7.21 0.150 0.095 116 Example 30 (Compound 30) 6.80 0.141 0.066 107 Example 30 (Compound 31) 6.80 0.141 0.066 107 Example 31 (Compound 31) 6.80 0.141 0.066 107 Example 31 (Compound 31) 6.80 0.141 0.068 108 Example 31 (Compound 31) 6.80 0.141 0.068 108 Example 31 (Compound 31) 6.80 0.141 0.069 107 Example 40 (Compound 40) 6.89 0.141 0.069 107 Example 40 (Compound 41) 6.89 0.141 0.06					
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Example 47 (Compound 47)         7.21         0.150         0.092         116           Example 48 (Compound 48)         7.22         0.150         0.095         116           Example 49 (Compound 49)         6.78         0.141         0.069         108           Example 50 (Compound 50)         6.88         0.141         0.067         107           Example 51 (Compound 51)         6.75         0.141         0.063         107           Example 52 (Compound 52)         6.77         0.141         0.063         107           Example 53 (Compound 53)         7.21         0.150         0.095         116           Example 54 (Compound 54)         7.21         0.150         0.097         116           Example 56 (Compound 55)         6.78         0.141         0.069         108           Example 56 (Compound 57)         6.75         0.141         0.068         109           Example 58 (Compound 57)         6.75         0.141         0.063         107           Example 59 (Compound 58)         6.75         0.141         0.063         107           Example 59 (Compound 59)         7.20         0.150         0.097         119           Example 60 (Compound 60)         7.23         0.150 <td></td> <td></td> <td></td> <td></td> <td></td>					
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Example 50 (Compound 50) 6.88 0.141 0.071 109 Example 51 (Compound 51) 6.75 0.141 0.067 107 Example 52 (Compound 52) 6.77 0.141 0.063 107 Example 53 (Compound 53) 7.21 0.150 0.095 116 Example 54 (Compound 55) 6.78 0.141 0.069 108 Example 55 (Compound 56) 6.70 0.141 0.068 109 Example 56 (Compound 57) 6.75 0.141 0.068 109 Example 58 (Compound 57) 6.75 0.141 0.069 107 Example 58 (Compound 59) 7.20 0.150 0.097 119 Example 59 (Compound 59) 7.20 0.150 0.097 119 Example 60 (Compound 60) 7.23 0.150 0.097 119 Example 61 (Compound 61) 6.77 0.141 0.071 109 Example 62 (Compound 62) 6.88 0.141 0.081 110 Example 63 (Compound 63) 6.73 0.141 0.066 109 Example 64 (Compound 64) 6.78 0.141 0.062 109 Example 65 (Compound 65) 6.73 0.141 0.070 109					
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Example 52 (Compound 52)       6.77       0.141       0.063       107         Example 53 (Compound 53)       7.21       0.150       0.095       116         Example 54 (Compound 54)       7.21       0.150       0.097       116         Example 55 (Compound 55)       6.78       0.141       0.069       108         Example 56 (Compound 56)       6.70       0.141       0.068       109         Example 57 (Compound 57)       6.75       0.141       0.063       107         Example 58 (Compound 58)       6.75       0.141       0.069       107         Example 59 (Compound 59)       7.20       0.150       0.097       119         Example 60 (Compound 60)       7.23       0.150       0.099       117         Example 61 (Compound 61)       6.77       0.141       0.071       109         Example 62 (Compound 62)       6.88       0.141       0.081       110         Example 63 (Compound 63)       6.73       0.141       0.062       109         Example 65 (Compound 65)       6.78       0.141       0.062       109         Example 65 (Compound 65)       6.73       0.141       0.070       109					
Example 53 (Compound 53)       7.21       0.150       0.095       116         Example 54 (Compound 54)       7.21       0.150       0.097       116         Example 55 (Compound 55)       6.78       0.141       0.069       108         Example 55 (Compound 56)       6.70       0.141       0.068       109         Example 57 (Compound 57)       6.75       0.141       0.063       107         Example 58 (Compound 58)       6.75       0.141       0.069       107         Example 59 (Compound 59)       7.20       0.150       0.097       119         Example 60 (Compound 60)       7.23       0.150       0.099       117         Example 61 (Compound 61)       6.77       0.141       0.071       109         Example 63 (Compound 62)       6.88       0.141       0.081       110         Example 64 (Compound 64)       6.78       0.141       0.062       109         Example 65 (Compound 65)       6.73       0.141       0.070       109					
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Example 56 (Compound 56)     6.70     0.141     0.068     109       Example 57 (Compound 57)     6.75     0.141     0.063     107       Example 58 (Compound 58)     6.75     0.141     0.069     107       Example 59 (Compound 59)     7.20     0.150     0.097     119       Example 60 (Compound 60)     7.23     0.150     0.099     117       Example 61 (Compound 61)     6.77     0.141     0.071     109       Example 62 (Compound 62)     6.88     0.141     0.081     110       Example 63 (Compound 63)     6.73     0.141     0.062     109       Example 64 (Compound 65)     6.73     0.141     0.070     109	Example 54 (Compound 54)		0.150	0.097	
Example 57 (Compound 57)     6.75     0.141     0.063     107       Example 58 (Compound 58)     6.75     0.141     0.069     107       Example 59 (Compound 59)     7.20     0.150     0.097     119       Example 60 (Compound 60)     7.23     0.150     0.099     117       Example 61 (Compound 61)     6.77     0.141     0.071     109       Example 62 (Compound 62)     6.88     0.141     0.081     110       Example 63 (Compound 63)     6.73     0.141     0.066     109       Example 64 (Compound 64)     6.78     0.141     0.062     109       Example 65 (Compound 65)     6.73     0.141     0.070     109					
Example 58 (Compound 58)     6.75     0.141     0.069     107       Example 59 (Compound 59)     7.20     0.150     0.097     119       Example 60 (Compound 60)     7.23     0.150     0.099     117       Example 61 (Compound 61)     6.77     0.141     0.071     109       Example 62 (Compound 62)     6.88     0.141     0.081     110       Example 63 (Compound 63)     6.73     0.141     0.066     109       Example 64 (Compound 64)     6.78     0.141     0.062     109       Example 65 (Compound 65)     6.73     0.141     0.070     109					
Example 59 (Compound 59)     7.20     0.150     0.097     119       Example 60 (Compound 60)     7.23     0.150     0.099     117       Example 61 (Compound 61)     6.77     0.141     0.071     109       Example 62 (Compound 62)     6.88     0.141     0.081     110       Example 63 (Compound 63)     6.73     0.141     0.066     109       Example 64 (Compound 64)     6.78     0.141     0.062     109       Example 65 (Compound 65)     6.73     0.141     0.070     109					
Example 60 (Compound 60)     7.23     0.150     0.099     117       Example 61 (Compound 61)     6.77     0.141     0.071     109       Example 62 (Compound 62)     6.88     0.141     0.081     110       Example 63 (Compound 63)     6.73     0.141     0.066     109       Example 64 (Compound 64)     6.78     0.141     0.062     109       Example 65 (Compound 65)     6.73     0.141     0.070     109					
Example 61 (Compound 61)     6.77     0.141     0.071     109       Example 62 (Compound 62)     6.88     0.141     0.081     110       Example 63 (Compound 63)     6.73     0.141     0.066     109       Example 64 (Compound 64)     6.78     0.141     0.062     109       Example 65 (Compound 65)     6.73     0.141     0.070     109					
Example 63 (Compound 63) 6.73 0.141 0.066 109 Example 64 (Compound 64) 6.78 0.141 0.062 109 Example 65 (Compound 65) 6.73 0.141 0.070 109	Example 61 (Compound 61)				
Example 64 (Compound 64) 6.78 0.141 0.062 109 Example 65 (Compound 65) 6.73 0.141 0.070 109					
Example 65 (Compound 65) 6.73 0.141 0.070 109					
	Example 65 (Compound 65)	7.20	0.141	0.070	115

**844**TABLE 1-continued

			Color coordinate		Service
5	Material	Efficiency (cd/A)	CIE (x)	CIE (y)	life T95 (hr)
	Example 67 (Compound 67)	7.20	0.150	0.097	115
	Example 68 (Compound 68)	6.78	0.141	0.069	108
	Example 69 (Compound 69)	6.78	0.141	0.071	108
10	Example 70 (Compound 70)	6.68	0.140	0.054	110
	Example 71 (Compound 71)	6.72	0.140	0.051	110
	Example 72 (Compound 72) Example 73 (Compound 73)	6.72 6.71	$0.140 \\ 0.140$	0.051 0.051	110 110
	Example 74 (Compound 74)	6.75	0.140	0.051	111
	Example 75 (Compound 75)	6.70	0.140	0.051	110
	Example 76 (Compound 76)	6.69	0.140	0.054	110
15	Example 77 (Compound 77)	6.70	0.140	0.054	110
13	Example 78 (Compound 78)	6.70	0.140	0.054	109
	Example 79 (Compound 79)	6.75	0.140	0.050	111
	Example 80 (Compound 80)	7.43	0.150	0.070	125
	Example 81 (Compound 81)	7.43	0.150	0.070	125
	Example 82 (Compound 82)	7.43	0.150	0.070	125
20	Example 83 (Compound 83)	7.43	0.150	0.070	124
	Example 84 (Compound 84) Example 85 (Compound 85)	7.44 7.44	0.150 0.150	0.071 $0.072$	125 125
	Example 86 (Compound 86)	7.44	0.150	0.072	125
	Example 87 (Compound 87)	7.44	0.150	0.071	125
	Example 88 (Compound 88)	7.44	0.150	0.072	125
	Example 89 (Compound 89)	7.45	0.150	0.076	125
25	Example 90 (Compound 90)	7.47	0.150	0.077	127
	Example 91 (Compound 91)	7.48	0.150	0.077	126
	Example 92 (Compound 92)	7.48	0.150	0.077	126
30	Example 93 (Compound 93)	7.48	0.150	0.077	125
	Example 94 (Compound 94)	7.49	0.150	0.079	125
	Example 95 (Compound 95)	7.44	0.150	0.071	124
	Example 96 (Compound 96)	6.75	0.140	0.054	111
	Example 97 (Compound 97) Example 98 (Compound 98)	6.75 6.75	$0.140 \\ 0.140$	0.053 0.053	110 110
	Example 98 (Compound 99)	6.75	0.140	0.053	109
	Example 100 (Compound 100)	6.79	0.140	0.051	111
	Example 101 (Compound 101)	7.42	0.150	0.072	124
35	Example 102 (Compound 102)	7.42	0.150	0.074	124
33	Example 103 (Compound 103)	7.43	0.150	0.072	125
	Example 104 (Compound 104)	7.41	0.150	0.073	125
	Example 105 (Compound 105)	7.11	0.140	0.053	127
	Example 106 (Compound 106)	7.32	0.140	0.055	125
	Example 107 (Compound 107)	7.30	0.150	0.056	125
40	Example 108 (Compound 108)	7.39 7.41	0.150 0.150	0.060 0.058	130 120
	Example 109 (Compound 109) Example 110 (Compound 110)	7.41	0.150	0.058	126
	Example 111 (Compound 111)	7.15	0.140	0.057	127
45	Example 112 (Compound 112)	7.22	0.150	0.060	129
	Example 113 (Compound 113)	7.38	0.150	0.066	121
	Example 114 (Compound 114)	7.43	0.140	0.051	121
	Example 115 (Compound 115)	7.40	0.140	0.057	133
	Example 116 (Compound 116)	7.20	0.140	0.055	139
	Example 117 (Compound 117)	7.55	0.150	0.051	140
50	Example 118 (Compound 118)	7.57	0.150	0.050	131
	Example 119 (Compound 119)	7.55	0.150	0.051	131
	Comparative Example 1	6.11	0.174	0.190	95
50	(Compound BD-1)	2.89	0.120	0.022	15
	Comparative Example 2 (Compound BD-2)	2.69	0.120	0.032	13
	(Compound DD-2)				

From Table 1, it can be confirmed that Examples 1 to 119, in which the compound of the present application including a non-aromatic pentagonal ring including N in the molecule is used, have better efficiency and service life characteristics than Comparative Example 1 in which the compound (BD-1) in which a benzene ring is fused to a pentagonal ring including N to form an aromatic ring is used and Comparative Example 2 in which the compound (BD-2) in which all benzene rings around a boron atom are bonded to each other to form a fused ring is used. Further, in general, when the color coordinate value is reduced, the service life characteristic deteriorates, but it can be confirmed that the compound of the present invention implements a dark blue color

due to the low color coordinate value, and as a result, the color purity is excellent and the service life characteristic is also improved.

The invention claimed is:

1. A compound of Formula 1:

Formula 1 10
$$A2 \xrightarrow{R_1)_{n1}} B \xrightarrow{R_2} Ar_1 Ar_2$$

$$A1 \xrightarrow{R_3)_{n3}} Ar_4$$

$$Ar_3$$

wherein in Formula 1:

Ar<sub>1</sub> and Ar<sub>4</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted alkynyl group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted cycloalkyl group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted aryl group, or a substituted or unsubstituted aryl group, or adjacent groups are bonded to each other to form a substituted or unsubstituted aliphatic hydrocarbon ring;

Ar<sub>2</sub> and Ar<sub>3</sub> are bonded to each other to form a substituted or unsubstituted aliphatic hydrocarbon ring having 3 to 60 carbon atoms or are bonded to each other to form a substituted or unsubstituted monocyclic to tricyclic aliphatic hydrocarbon ring having 3 to 60 carbon 40 atoms, in which an aliphatic hydrocarbon ring or an aromatic hydrocarbon ring is fused or unfused;

A1 and A2 are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted boron group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted or unsubstituted or unsubstituted alkynyl group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted or unsubstituted or unsubstituted aryl group, or a substituted or unsubstituted aryl group, or are bonded to each other to form a substituted or unsubstituted or unsubstituted ring;

R<sub>1</sub> to R<sub>3</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted cycloalkyl group, a substituted or unsubstituted or unsubstituted or unsubstituted or unsubstituted heterocyclic group; and

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n1 to n3 are each an integer from 0 to 3, and when n1 to n3 are each 2 or more, substituents in a plurality of parentheses are the same as or different from each other

**2**. The compound of claim **1**, wherein Formula 1 is any one of the following Formula 1-1 or 1-2:

Formula 1-1
$$Ar_{5} \xrightarrow{Ar_{6}} N \xrightarrow{R_{1}} Ar_{1}$$

$$Ar_{7} \xrightarrow{Ar_{8}} N \xrightarrow{Ar_{4}} Ar_{3}$$

$$(R_{3})_{n3}$$

Formula 1-2

A12
$$\begin{array}{c} (R_1)_{n1} \\ B \\ N \\ Ar_1 \\ Ar_2 \\ (R_3)_{n3} \end{array}$$
Formula

wherein in Formulae 1-1 and 1-2:

the definitions of  $R_1$  to  $R_3$ ,  $Ar_1$  to  $Ar_4$ , and n1 to n3 are the same as those defined in Formula 1;

A11 and A12 are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted boron group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted aryloxy group, as a substituted or unsubstituted aryloxy group; and

Ar<sub>5</sub> to Ar<sub>8</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted aryl group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted aryl group, or adjacent groups are bonded to each other to form a substituted or unsubstituted ring.

**3**. The compound of claim **2**, wherein Formula 1-1 is any one of the following Formulae 3 to 7:

Formula 3

Formula 4 15

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$$Ar_{5}$$
 $Ar_{7}$ 
 $Ar_{8}$ 
 $R_{23}$ 
 $R_{23}$ 
 $R_{21}$ 
 $R_{21}$ 

Formula 5  $Ar_{5} \xrightarrow{Ar_{6}} N \xrightarrow{R_{23}} R_{23} \xrightarrow{R_{22}} (R_{25})_{p3}$   $R_{23} \xrightarrow{R_{23}} R_{23}$ 

Formula 6

$$Ar_{5}$$

$$Ar_{7}$$

$$Ar_{8}$$

$$R_{23}$$

$$R_{23}$$

$$R_{23}$$

$$R_{24}$$

$$R_{25}$$

$$R_{26}$$

Formula 7  $(R_1)_{n1} \qquad \qquad (R_2)_{n2}$ 

$$Ar_{5}$$
 $Ar_{6}$ 
 $Ar_{7}$ 
 $Ar_{8}$ 
 $R_{23}$ 
 $R$ 

wherein in Formulae 3 to 7:

the definitions of  $R_1$  to  $R_3$ ,  $Ar_5$  to  $Ar_8$ , and n1 to n3 are the same as those defined in Formula 1-1;

R<sub>21</sub> to R<sub>27</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted avloxy group, a substituted or unsubstituted avl group, or a substituted or unsubstituted proup;

p1 is an integer from 0 to 8;

p2 to p4 are each an integer from 0 to 14;

p5 is an integer from 0 to 20; and

when p1 to p5 are each 2 or more, the substituents in a plurality of parentheses are the same as or different from each other.

**4**. The compound of claim **1**, wherein Formula 1 is any one of the following Formula 2-1 or 2-2:

Formula 2-1

$$R_{31}$$
 $R_{31}$ 
 $R_{31}$ 
 $R_{32}$ 
 $R_{31}$ 
 $R_{31}$ 
 $R_{32}$ 
 $R_{33}$ 

Formula 2-2

wherein in Formulae 2-1 and 2-2:

the definitions of A1, A2, R<sub>1</sub> to R<sub>3</sub>, and n1 to n3 are the same as those defined in Formula 1;

R<sub>31</sub> to R<sub>35</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted or unsubstituted or unsubstituted alkenyl group, a substituted or unsubstituted alkenyl group, a substituted alkoxy group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted group, or a substituted or unsubstituted group, a substituted or unsubstituted group, or a substituted or unsubstituted group;

adjacent R<sub>35</sub>s are optionally bonded to each other to form a substituted or unsubstituted aliphatic hydrocarbon ring;

r33 is an integer from 0 to 8;

r34 and r35 are each an integer from 0 to 4; and

when r33 to r35 are each 2 or more, substituents in a plurality of parentheses are the same as or different 5 from each other.

5. The compound of claim 2, wherein Formula 1-1 is any one of the following Formulae 8 to 10:

Formula 8

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$$(R_1)_{n1}$$
 $R_4$ 
 $R_5$ 
 $R_6$ 
 $R_7$ 
 $R_7$ 
 $R_8$ 
 $R_8$ 
 $R_9$ 
 $R_9$ 

Formula 9

$$(R_1)_{n1}$$

$$(R_2)_{n2}$$

$$Ar_1$$

$$Ar_2$$

$$(R_3)_{n3}$$

Formula 10

$$R_{8}$$
 $Y_{1}$ 
 $Y_{2}$ 
 $R_{9}$ 
 $X_{1}$ 
 $X_{1}$ 
 $X_{2}$ 
 $X_{1}$ 
 $X_{2}$ 
 $X_{3}$ 
 $X_{40}$ 
 $X_{1}$ 
 $X_{1}$ 
 $X_{2}$ 
 $X_{3}$ 
 $X_{1}$ 
 $X_{2}$ 
 $X_{3}$ 
 $X_{40}$ 

wherein in Formulae 8 to 10:

the definitions of R<sub>1</sub> to R<sub>3</sub>, n1 to n3, and Ar<sub>1</sub> to Ar<sub>4</sub> are the same as those defined in Formula 1-1;

R<sub>4</sub> to R<sub>9</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen  $\,^{50}$ group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted cycloalkyl group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heterocyclic group;

Y<sub>1</sub> and Y<sub>2</sub> are bonded to each other to form a substituted or unsubstituted aliphatic hydrocarbon ring; and

Cy1 are a substituted or unsubstituted aromatic hydrocarbon ring.

6. The compound of claim 1, wherein Formula 1 is any one of the following Formulae 101 to 108:

850

Formula 102

Formula 101

Formula 103

$$Ar_{103}$$
 $N$ 
 $(R_{12})_{n12}$ 
 $Ar_{1}$ 
 $Ar_{2}$ 
 $Ar_{3}$ 
 $(R_{3})_{n3}$ 

Formula 104

$$A_{105}$$
 $A_{106}$ 
 $A_{106}$ 

Formula 105

$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $Ar_1$ 
 $Ar_2$ 
 $R_{18})_{n16}$ 
 $R_{16}$ 
 $(R_{17})_{n15}$ 

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$$(R_{18})_{n16}$$
 $R_{16}$ 
 $(R_{19})_{n17}$ 

wherein in Formulae 101 to 108:

the definitions of A1, A2, R<sub>1</sub> to R<sub>3</sub>, n1 to n3, and Ar<sub>1</sub> to Ar<sub>4</sub> are the same as those defined in Formula 1;

 $Q_1$  is  $C(R_{48})(R_{49})$  or  $Si(R_{48})(R_{49})$ ;

Q<sub>2</sub> is C or Si; R<sub>3</sub>', R<sub>11</sub> to R<sub>20</sub>, and R<sub>41</sub> to R<sub>49</sub> are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubsti- 60 tuted alkyl group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted cycloalkyl group, a substituted or unsubstituted 65 aryl group, or a substituted or unsubstituted heterocyclic group, and n3' are an integer from 0 to 3;

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n11 to n14 and n41 are each an integer form 0 to 2; n15 is an integer from 0 to 8;

n16 to n18 and n42 to n47 are each an integer from 0 to

when n3', n15 to n18, and n42 to n47 are each 2 or more, the substituents in a plurality of parentheses are the same as or different from each other;

when n11 to n14 and n41 are each 2, the substituents in a plurality of parentheses are the same as or different from each other; and

 $Ar_{101}$  to  $Ar_{106}$  are the same as or different from each other, and are each independently hydrogen, deuterium, a substituted or unsubstituted alkyl group, a substituted or unsubstituted aryl group, or a substituted or unsubstituted heteroaryl group.

7. A compound of any one of the following Formulae 12 to 18, 20 to 22, and 24 to 39:

Formula 107

Formula 108

Formula 106

Formula 12

$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $(R_{109})_{m1}$ 
 $(R_3)_{n3}$ 

Formula 13

$$(R_{1})_{n1}$$

$$(R_{2})_{n2}$$

$$(R_{111})_{m3}$$

$$(R_{3})_{n3}$$

$$(R_{112})_{m4}$$

Formula 14

$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $(R_{113})_{m5}$ 
 $(R_{114})_{m6}$ 

Formula 15

$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $(R_{116})_{m7}$ 
 $(R_{3})_{n3}$ 

Formula 16

-continued

-continued

$$(R_{117})_{m9} \\ (R_{1})_{n1} \\ (R_{2})_{n2} \\ (R_{3})_{n3}$$

$$(R_{1})_{n1}$$

$$(R_{2})_{n2}$$

$$(R_{136})_{m20}$$

$$(R_{3})_{n3}$$

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Formula 24

Formula 22

Formula 17 
$$(R_{119})_{m11} \qquad \qquad 20$$
 
$$R_{120})_{m12} \qquad 25$$
 
$$(R_{3})_{n3}$$

$$(R_1)_{n1}$$
 $R_2)_{n2}$ 
 $(R_{143})_{m23}$ 
 $(R_3)_{n3}$ 

Formula 18 30

$$(R_{121})_{m13}$$
 $R_{122}$ 
 $(R_{1})_{n1}$ 
 $(R_{2})_{n2}$ 
 $(R_{126})_{m14}$ 
 $(R_{126})_{m14}$ 
 $(R_{123})_{n3}$ 
 $(R_{123})_{n3}$ 
 $(R_{124})_{m13}$ 
 $(R_{126})_{m14}$ 
 $(R_{126})_{m14}$ 
 $(R_{126})_{m14}$ 
 $(R_{126})_{m14}$ 
 $(R_{126})_{m14}$ 
 $(R_{127})_{m13}$ 
 $(R_{128})_{m14}$ 
 $(R_{128})_{m14}$ 
 $(R_{128})_{m14}$ 
 $(R_{128})_{m14}$ 

$$(R_1)_{n1}$$
 $R_2)_{n2}$ 
 $(R_{145})_{m25}$ 
 $(R_3)_{n3}$ 
 $(R_{146})_{m26}$ 

Formula 20

Formula 26

$$(R_{132})_{m16}$$
 $(R_{1})_{n1}$ 
 $(R_{2})_{n2}$ 
 $(R_{133})_{m17}$ 
 $(R_{3})_{n3}$ 

$$(R_{148})_{m28}$$

$$(R_{147})_{m27}$$

$$(R_{3})_{n3}$$

Formula 21

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$$(R_{1})_{n1}$$
 $(R_{2})_{n2}$ 
 $(R_{135})_{m19}$ 
 $(R_{134})_{m18}$ 

Formula 27]
$$(R_1)_{n1} \qquad (R_2)_{n2} \qquad (R_{150})_{m30}$$

$$(R_{149})_{m29} \qquad (R_{3})_{n3}$$

15

-continued

Formula 28 
$$\mathbb{R}_{153}$$
 
$$\mathbb{R}_{153}$$
 
$$\mathbb{R}_{153}$$
 
$$\mathbb{R}_{153}$$
 
$$\mathbb{R}_{153}$$

$$(R_1)_{n1}$$
 $(R_2)_{n2}$ 
 $(R_{156})_{m32}$ 
 $(R_3)_{n3}$ 
 $(R_3)_{n3}$ 

Formula 30

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$$R_{161}$$
 $R_{161}$ 
 $R_{161}$ 
 $R_{161}$ 
 $R_{161}$ 
 $R_{161}$ 
 $R_{162}$ 
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 $R_{163}$ 
 $R_{163}$ 

Formula 32
$$R_{165}$$

$$R_{167}$$

$$R_{167}$$

$$R_{167}$$

$$R_{168}$$

$$R_{168}$$

$$R_{168}$$

$$R_{168}$$

$$R_{168}$$

$$R_{168}$$

$$R_{168}$$

Formula 33

$$(R_{171})_{m43} = (R_{170})_{m42}$$

$$(R_{171})_{m43} = (R_{170})_{m42}$$

Formula 34

$$(R_{173})_{m45}$$
 $(R_{173})_{m45}$ 
 $(R_{174})_{m46}$ 
 $(R_{175})_{m47}$ 
 $(R_{176})_{m48}$ 

Formula 35

$$(R_{177})_{m49}$$
 $(R_{3})_{n4}$ 
 $(R_{178})_{m50}$ 
 $(R_{179})_{m51}$ 

-continued

Formula 36  $(R_1)_{n1}$  $(R_2)_{n2}$ 5  $(R_{181})_{m53}$ 10  $(R_{182})_{m54}$ 15  $(R_{184})_{m56}$ 

-continued

Formula 39

$$(R_{193})_{m65}$$
 $(R_{193})_{m67}$ 
 $(R_{194})_{m66}$ 
 $(R_{196})_{m68}$ 
 $(R_{197})_{m69}$ 
 $(R_{198})_{m70}$ 

 $Q_1$  is  $C(R_{199})(R_{200})$  or  $Si(R_{199})(R_{200})$ ;  $Q_2$  is C or Si;

 $R_1$  to  $R_3$  and  $R_{109}$  to  $R_{126}$ ,  $R_{132}$  to  $R_{137}$ , and  $R_{143}$  to  $R_{200}$ are the same as or different from each other, and are each independently hydrogen, deuterium, a halogen group, a nitrile group, a substituted or unsubstituted silyl group, a substituted or unsubstituted boron group, a substituted or unsubstituted alkyl group, a substituted or unsubstituted alkenyl group, a substituted or unsubstituted alkynyl group, a substituted or unsubstituted alkoxy group, a substituted or unsubstituted aryloxy group, a substituted or unsubstituted cycloalkyl group, a substituted or unsubstituted aryl group, a substituted or unsubstituted amine group, or a substituted or unsubstituted heterocyclic group;

wherein in Formulae 12 to 18, 20 to 22, and 24 to 39:

n4 is an integer from 0 to 2, and when n4 is 2, the substituents in two parentheses are the same as or different from each other; and

n1 to n3, m1 to m14, m16 to m21, and m23 to m70 are each an integer from 0 to 3, and

when n1 to n3, m1 to m14, m16 to m21, and m23 to m70 are each 2 or more, the substituents in two or more parentheses are the same as or different from each other.

8. The compound of claim 1 that is any one of the following compounds:

Formula 37

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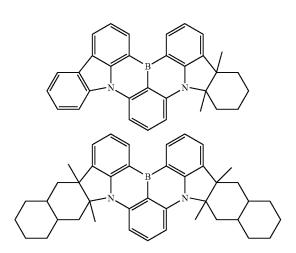
35

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$$(R_{185})_{m57}$$
 $(R_{3})_{m4}$ 
 $(R_{187})_{m59}$ 
 $(R_{188})_{m60}$ 
 $(R_{188})_{m60}$ 

50 [Formula 38]  $(R_1)_{n1}$ 55  $(R_2)_{n2}$ 60  $(R_{192})_{m64}$ 



$$\bigcup_{D} \bigcup_{N} \bigcup_{N} \bigcup_{D} \bigcup_{N} \bigcup_{N$$

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$$\begin{array}{c|c} & & & & \\ & & & \\ D & & & \\ D & & & \\ \end{array}$$

$$\bigcup_{D} \bigcup_{N} \bigcup_{N$$

$$\bigcup_{N} \bigcup_{N} \bigcup_{N$$

906 -continued

65

-continued

5

N

N

10

15

25 30 N N N 40

50 55 N N 60 -continued

B

N

N

N

Si

-continued

-continued

N N N 

-continued

$$S_{S_{1}}$$
 $S_{N}$ 
 $S_{N}$ 

**9**. A compound that is any one of the following compounds:

N N N 

-continued

B N N

N N 

-continued

-continued

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. An organic light emitting device comprising: a first electrode;

a second electrode provided to face the first electrode; and an organic material layer having one or more layers provided between the first electrode and the second electrode,

wherein one or more layers of the organic material layer comprise the compound of claim 1.

- 11. The organic light emitting device of claim 10, wherein the organic material layer comprises a hole injection layer or a hole transport layer, and the hole injection layer or the hole transport layer comprises the compound.
- 12. The organic light emitting device of claim 10, wherein 5 the organic material layer comprises an electron transport layer or an electron injection layer, and the electron transport layer or the electron injection layer comprises the compound.
- 13. The organic light emitting device of claim 10, wherein the organic material layer comprises a light emitting layer, and the light emitting layer comprises the compound.
- 14. The organic light emitting device of claim 10, wherein the organic material layer comprises a light emitting layer, and the light emitting layer comprises the compound as a dopant of the light emitting layer.
- 15. The compound of claim 1, wherein Ar<sub>2</sub> and Ar<sub>3</sub> are bonded to each other to form a substituted or unsubstituted cyclohexane, a substituted or unsubstituted tetradecahydrophenanthrene, a substituted or unsubstituted tetrahydronaphthalene, or a substituted or unsubstituted decahydronaphthalene.
  - **16**. An organic light emitting device comprising: a first electrode;
  - a second electrode provided to face the first electrode; and an organic material layer having one or more layers provided between the first electrode and the second electrode.
  - wherein one or more layers of the organic material layer comprise the compound of claim 7.
- 17. The organic light emitting device of claim 16, wherein:
  - a) the organic material layer comprises a hole injection layer or a hole transport layer, and the hole injection layer or the hole transport layer comprises the compound; or

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- b) the organic material layer comprises an electron transport layer or an electron injection layer, and the electron transport layer or the electron injection layer comprises the compound; or
- c) the organic material layer comprises a light emitting layer, and the light emitting layer comprises the compound; or
- d) the organic material layer comprises a light emitting layer, and the light emitting layer comprises the compound as a dopant of the light emitting layer.
- **18**. An organic light emitting device comprising: a first electrode;
- a second electrode provided to face the first electrode; and an organic material layer having one or more layers provided between the first electrode and the second electrode.
- wherein one or more layers of the organic material layer comprise the compound of claim 9.
- ${\bf 19}.$  The organic light emitting device of claim  ${\bf 18},$   $^{20}$  wherein:
  - a) the organic material layer comprises a hole injection layer or a hole transport layer, and the hole injection layer or the hole transport layer comprises the compound; or
  - b) the organic material layer comprises an electron transport layer or an electron injection layer, and the electron transport layer or the electron injection layer comprises the compound; or
  - c) the organic material layer comprises a light emitting layer, and the light emitting layer comprises the compound; or
  - d) the organic material layer comprises a light emitting layer, and the light emitting layer comprises the compound as a dopant of the light emitting layer.

\* \* \* \* \*