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- (54) METHOD FOR MAKING A PASSIVATING Publication Classification COATING BASED ON ONE OR MORE SILICON COMPOUNDS ON A CERAMIC MATRIX COMPOSITE REINFORCED WITH CARBON FIBRES AND CERAMIC MATRIX COMPOSITE REINFORCED WITH CARBON FIBRES WITH SUCH COATING
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(57) ABSTRACT

A method makes a silicon compound-based passivating coating on a ceramic matrix composite reinforced with carbon fibers. A piece made in a ceramic matrix composite reinforced with carbon fibers is placed in a closed chamber of an oven . A predefined load of solid silicon is placed in the piece. The oven is heated while maintaining inside the chamber predefined medium/low vacuum conditions, to generate silicon vapors inside the chamber . The vapors react coating having composites of the substances with the silicon. The partial pressure of the vacuum, temperature inside the chamber and exposure times of the piece to the silicon vapors to obtain a predefined thickness of the surface coating are chosen. The piece is cooled once the predefined thickness of the passivating coating is reached.

FIG.6

METHOD FOR MAKING A PASSIVATING COATING BASED ON ONE OR MORE SILICON COMPOUNDS ON A CERAMIC MATRIX COMPOSITE REINFORCED WITH CARBON FIBRES AND CERAMIC MATRIX COMPOSITE REINFORCED WITH CARBON FIBRES WITH SUCH COATING

FIELD OF APPLICATION

[0001] The present invention relates to a method for making a passivating coating based on one or more silicon compounds on a ceramic matrix composite (CMC) rein-
forced with carbon fibres and a ceramic matrix composite reinforced with carbon fibres with such coating.

BACKGROUND ART

[0002] Ceramic matrix composites (CMC) are materials designed to combine the properties of polymer composites with the properties of ceramic materials. The properties of CMCs include:

[0003] high mechanical features (polymer composites and ceramic materials);

[0004] high fracture toughness (polymer composites);

[0005] lightness (polymer composites)
[0006] high hardness (ceramic materials)

[0007] high thermal stability (ceramic materials)

[0008] high resistance to oxidation (ceramic materials)

[0009] In the field of non-oxide materials, CMCs can be made with carbon fibres or with ceramic fibres, mainly in silicon carbide (SiC).

[0010] The CMCs made with SiC fibres have a high resistance to oxidation, but have very high costs and generally show medium-low mechanical features. The high costs are mainly due to the cost of the SiC fibres, about 2 orders of magnitude higher than the cost of carbon fibres. Furthermore, SiC fibre-based CMCs are usually densified with longer processes, which typically involve the formation of the SiC matrix from the vapour phase.

of the SiC matrix from the vapour phase.

[0011] Carbon fibre-based CMCs have a lower oxidation resistance than SiC fibre-based CMCs. This aspect limits the maximum operating temperature in an oxidising environment.

[0012] A possible strategy to increase the operating temperature of carbon fibre-based CMCs is to apply protective coatings which reduce or completely prevent the contact of the material with the oxygen of the external environment. [0013] The protective coatings may be applied with chemical vapour deposition (CVD) or physical vapour deposition (PVD) techniques.
[0014] In the case of chemical vapour deposition (CVD), a vapour is generated which reacts

vapours which react with each other at close range with the substrate itself.

[0015] In the case of physical vapour deposition (PVD), the vapour of the substance to be deposited is generated directly. Therefore in PVD, unlike CVD, there is no reaction between substrate and coating.

[0016] The SiC is deposited by CVD methods starting from silicon precursor gases (e.g., silicon tetrachloride, silanes or TEOS) which react with the free carbon of the composite surface. In this case part of the substrate but the coating appears substantially deposited on the surface of the CMC.

[0017] On the other hand, the PVD techniques for making
SiC coatings involve evaporating silicon in an atmosphere
rich in acetylene. Thereby an SiC gas is formed which is
deposited on the substrate in full discontinuity th

tion against the oxidation of the CMC. However, they have a number of disadvantages which limit the use thereof:

- [0019] need to work in ultra-high vacuum conditions; this requires the construction of expensive deposition cham bers of limited size;
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- [0020] high cost of deposition plants;
[0021] need to manage high-danger gases, such as acetylene in the case of the PVD technique or chlorinated gases
in the case of the CVD technique; the chlorinated gases
which decompose during the growth of the coating generate residual gases rich in hydrochloric acid, which is aggressive towards metals and must be managed and disposed of correctly ;
- [0022] the typical deposition rates with the CVD and PVD techniques are around 1 μ m per hour; making highthickness coatings (of the order of tens of μ m) requires
long process times.
[0023] The aforesaid operating limits also entail limita-

tions on the sizes of the pieces to be coated, which tend to be small, of the order of a few tens of centimetres.

[0024] In the reference sector there is therefore a very strong need to make coatings comprising silicon carbide on ceramic matrix composites reinforced with carbon fibres having the possibility of :

- [0025] operating in no-high vacuum conditions, of the order of mbar;
- [0026] operating in the absence of dangerous gases to manage;
- [0027] using less expensive deposition plants with respect to those required to make coatings with the CVD or PVD techniques ;

[0028] reducing process times due to higher deposition rates .

[0029] It should also be noted that the coatings obtained by means of the CVD or PVD techniques are associated with the substrate through a net interface (not graded), which defines a marked area of discontinuity. The presence of this discontinuity area favours the development of cracks and therefore the access of the external environment (oxidant) to the substrate.

(0030) It should also be noted that the CVD deposition technique does not generally allow to obtain coatings free of cracks .

[0031] The cracks are formed due to the different coefficient of thermal expansion CTE between the substrate and coating. SiC has a CTE of $4 \cdot 10 \cdot 6$ K - 1, while a carbon fibre-based CMC has a CTE around $1 \cdot 10 \cdot 6$ K-1. The cracking of the coating therefore forms during cooling, and during the thermal cycles of use since the coating tends to expand and contract more than the substrate.

[0032] It is also known to insert a compatibilisation substrate between the coating and CMC in order to attenuate the mismatch between the coating CTE and CMC by means of a graded structure.

[0033] It is also known to use other types of protection against oxidation and temperature known as TBC (Thermal Barrier Coating) based on oxides, or as UHTCs (Ultra High Temperature Ceramics) based on borides, typically used for metal substrates. TBCs and UHTCs have even more deviated thermo-mechanical properties and have greater compatibility problems with the substrate. In this case, the need to use more than one interlayer between the substrate and coating is well known.

 $[0034]$ A crack-free coating could be obtained with the CVD technique if a very low thickness of the order of a few the coating to deform, tolerating tensile stresses without fracturing. However, a reduced thickness would quickly be consumed in the presence of oxygen due to the establish ment of passive oxidation mechanisms, which involve the oxidation of the SiC with the formation of $SiO₂$, and of active oxidation mechanisms with the formation of SiO.

[0035] Therefore, in the reference sector there is also an interest in making coatings based on one or more silicon interest in making coatings based on one or more silicon
compounds, in particular comprising silicon carbide, on ceramic matrix composites reinforced with carbon fibres which are as free of microcracks as possible and which have
a gradual passage and therefore are more compatibilised with the composite substrate so as to better with that thicknesses of the order of tens of microns .

[0036] Therefore, the object of the present invention is to eliminate, or at least reduce, the aforementioned problems relating to the background art, by providing a method for making a passivating coating based on one or which gives the possibility of making the coating in nonhigh vacuum conditions and in the absence of dangerous gases to manage.

[0037] A further object of the present invention is to provide a method for making a passivating coating based on
one or more silicon compounds on ceramic matrix composites reinforced with carbon fibres which can be implemented with less expensive plants than those required to make coatings with CVD or PVD techniques.

[0038] A further object of the present invention is to provide a method for making a passivating coating based on one or more silicon compounds on ceramic matrix composites reinforced with carbon fibres which allows to reduce the process times being the thickness of coating obtained equal.

[0039] A further object of the present invention is to provide a method for making a passivating coating based on one or more silicon compounds on ceramic matrix composites reinforced with carbon fibres which allows to make coatings which are more compatibilized with the composite substrate so that the substrate itself can better with stand coatings with thicknesses of the order of tens of microns.

[0040] A further object of the present invention is to provide a method for making a passivating coating based on one or more silicon compounds on ceramic matrix composites reinforced with carbon fibres which allows to reduce the formation of microcracks on the final coating.

DESCRIPTION OF THE DRAWINGS

[0041] The technical features of the invention may be clearly found in the contents of the claims presented below and the advantages thereof will become more apparent from the following detailed description, made with reference to the accompanying drawings, which show one or more embodiments thereof, given purely by way of non-limiting examples, in which:

[0042] FIG. 1 shows a simplified flow diagram of the method for making a silicon-compound based passivating coating on a CMC in accordance with the present invention; [0043] FIG. 2 shows a BES SEM image of the surface of a coating comprising SiC obtained on a CMC by the method of the present invention in accordance with example 1;

SiC coating obtained via CVD on the surface of a carbon [0044] FIG. 3 shows an SEM image of the surface of an

fibre-based CMC;
[0045] FIG. 4 shows a BES SEM image of a coating comprising SiC obtained on a CMC coated with a slurry comprising silicon, by the method of the present in accordance with example 2;
[0046] FIG. 5 shows a BES SEM image of a coating

comprising SiC obtained on a CMC coated with a slurry comprising SiC, by the method of the present invention in accordance with example 2;

[0047] FIG. 6 shows a BES SEM image of the surface of a coating obtained in accordance with the invention according to what is provided in example 3;

[0048] FIG. $\overline{7}$ shows a BES SEM image of the surface of PRESENTATION OF THE INVENTION **10046** FIG. *t* shows a BES SEW image of the surface of the coating of FIG. **6** after having been subjected to a PWT test:

[0049] FIG. 8 shows a BES SEM image of a section of the coating of FIG. 7;

[0050] FIG. 9 shows a BES SEM image of the surface of a coating obtained according to the invention on a CMC coated with a slurry containing SiC in accordance with example 4;

[0051] FIG. 10 shows a BES SEM image of the surface of a coating obtained according to the invention on a CMC coated with a slurry containing graphite in accordance with example 4;

[0052] FIG. 11 shows a BES SEM image of the section of a coating obtained according to the invention on a CMC coated with a slurry containing molybdenum in accordance with example 5;

[0053] FIG. 12 shows a BES SEM image of a section of a coating comprising SiC obtained according to the inven tion, relating to the growth interface of the coating characterised by a dendritic structure;
[0054] FIG. 13 shows the graph of the silicon vapour

tension curve as a function of the temperature.

DETAILED DESCRIPTION

[0055] The present invention relates to a method for making a silicon compound-based passivating coating on a ceramic matrix composite reinforced with carbon fibres.

[0056] In accordance with a general form of implementation, the aforesaid method comprises the following operating steps:

a) providing at least one piece made in a ceramic matrix composite (CMC) reinforced with carbon fibres;

b) placing the piece to be coated in a closed chamber of an

c) placing in the closed chamber a predefined load of solid silicon avoiding direct contact between the silicon and the piece to be coated;

d) heating the oven, maintaining inside the closed chamber predefined medium vacuum or low vacuum conditions , so as to generate from said silicon load vapours of silicon inside said closed chamber, which react with one or more substances present on the surface of said piece to be coated leading to the formation of a surface coating comprising
composites of said one or more substances with the silicon.
[0057] As will be discussed below, the expression "sub-
stances present on the surface of the piece to be intended to include both substances which are incorporated into the piece itself and are exposed on the surface of the piece itself, and substances which are not incorporated into the piece itself, but which are deposited on the surface of such piece and are therefore exposed to the vapours of silicon.

[0058] The term "substances" is intended to include both simple elements, such as carbon (C) or metals, and com-

pounds of simple elements, such as silicon carbide (SiC).
[0059] Preferably, the medium vacuum conditions are defined by a partial pressure between 5 mbar and 300 mbar, while the low vacuum conditions are defined by a part

 $[0060]$ In accordance with said general form of implementation, the aforesaid method further comprises the following operating steps:

e) choosing the partial pressure of the vacuum, the maximum temperature inside said closed chamber and the expo sure times of said piece to be coated to the vapours of silicon so as to obtain a predefined thickness of the surface coating on said piece; and
f) cooling said piece once said predefined thickness of the

passivating coating has been reached.
[0061] It has been possible to experimentally verify that the method according to the invention defines a fast and economical technique which allows to grow a passivating coating of silicon compounds (and in particular of silicon carbide SiC, as will be discussed below) on the surface of a carbon fibre-based CMC.

[0062] The cost-effectiveness of this technique derives essentially from the fact that:

- [0063] it does not generate dangerous gases to manage; this excludes the need for expensive and complex man agement of such gases;
- [0064] it operates in non-high vacuum conditions (medium vacuum or low vacuum).

[0065] The method according to the invention can there-
fore be implemented with less expensive deposition plants
than those required to make coatings with the CVD or PVD
techniques.
[0066] Unlike other coating techniques,

to grow on the surface of the piece by simple deposition of materials on the surface of the piece itself. The method according to the invention instead provides for the coating to grow on the surface of the piece also due to the addition of substances which are incorporated into the piece itself and are exposed on the surface. In other words, the coating grows from the material of the piece/substrate itself; this growth is generated by the reactions induced by the vapours of silicon and by the consequent addition of silicon on the surface of the piece to be coated.

- [0067] This leads to a double effect:
[0068] the coating has a close adhesion with the piece itself (which acts as a substrate);
- [0069] the growth of the coating removes from the surface of the piece itself substances which in the absence of the coating would be susceptible to oxidation reactions; there-

fore the growth of the coating not only has the effect of providing protection for the piece , but causes a chemical the layers itself (at least of the surface itself (at least of the surface layers) with passivative phenomena . [0070] The method according to the invention also allows

a 100 um) in a short time compared to other techniques (times to obtain significant coating thicknesses (even of the order of of the order of tens of hours, with respect to times of the order of hundreds of hours). It is also possible to choose a thermal and vacuum profile dedicated to the growth of the coating, as expressly provided in step e) of the method itself and as will be discussed later in the description.

[0071] Generally, on the surface of the piece to be coated made of a ceramic matrix composite (CMC) reinforced with carbon fibres, free carbon is present.
[0072] This free carbon can be constituted entirely or at

least partly by carbon of the carbon reinforcement fibres of the ceramic matrix composite which the piece is made of.

 $[0.073]$ Part of the free carbon present on the surface of the piece can also derive from the matrix which incorporates the carbon fibres. This carbon consists of residual carbon from the pyrolysed carbonaceous preform used to make the piece itself and which did not react with the molten silicon during
the infiltration step which led to the formation of the ceramic
matrix composite.
[0074] Advantageously, as will be resumed in the follow-

ing description, the free carbon present on the surface of the piece can consist at least partly of carbon deposited on the surface of said piece before the aforesaid heating step d). In this case it is the carbon that does not belong to the piece to be coated, but has been added to the piece in particular to favour the kinetics of the reactions between the carbon and vapours of silicon, and therefore to stimulate the growth of

the passivating coating.

19075 | Advantageously, during the heating step d) the vapours of silicon react in particular with the aforesaid free carbon forming silicon carbide as a silicon compound and thus leading to the formation of a passivating coating comprising silicon carbide.
 [0076] The higher the free carbon content on the surface of

the piece , the higher the silicon carbide content in the coating

[0077] Advantageously, as can be seen from the photograph presented in FIG. 12, the passivating coating comprising silicon carbide (light coloured layer) grows on the surface of said piece with a dendritic growth structure at the interface between the piece and the passivating coating . This dendritic structure allows and testifies to a close adhesion between the substrate (piece) and the passivating coating. The dark coloured layer above the passivating coating coating consists of the resin used to incorporate the sample so as to obtain the SEM scan.

[0078] More in detail, the dendritic growth structure is due to the diffusion penetration of the vapours of silicon inside
the first surface layers of the substrate. This leads to the progressive growth of SiC structures which develop at the interface between the substrate and the passivating coating.

[0079] In accordance with a preferred embodiment of the invention, the method comprises a step g) of depositing on all or at least part of the surface of the piece to be coated a layer of powders of one or more materials suitable to react

with the vapours of silicon and are chosen as a function of one or more predefined silicon compounds to be obtained in

the passivating coating.
 [0080] Operatively, this step g) is conducted before the aforesaid heating step d).
 [0081] Advantageously, the powders of said one or more

materials are deposited on the surface of the piece to be coated in the form of a slurry based on water, an organic coated in the form of solvent or a resine in the form of a resin are then solvent or resin are then

evaporated and/or cross-linked before placing the piece to be coated in the closed chamber of the oven.

[0083] Preferably, the aforesaid one or more materials are chosen from the group constituted by graphite, coke, car-
bonaceous material, silicon, crystalline or amorphous silica,

transition metals, silicon carbide and/or other carbides.
[0084] In particular, the presence of graphite powders, coke or carbonaceous particles in the aforesaid layer of powders deposited on the surface of the piece can be provided so as to increase the free carbon present on the surface of the piece and thus favour the formation of silicon carbide during the heating step d) with exposure to the vapours of silicon.

[0085] In particular, the presence of silicon carbide powders in the aforesaid layer of powders deposited on the surface of the piece can be provided so that during the heating step d), with the exposure to the vapours of silicon, the silicon carbide powders become nucleation points for the crystallisation of silicon carbide on the surface of the piece.
[0086] In particular, the presence of silicon powders in the

aforesaid layer of powders deposited on the surface of the piece can be provided so that during the heating step d), with the exposure to the vapours of silicon, the silicon powders liquefy, triggering formation reactions of silicon carbide in the liquid phase, which favour the formation of silicon carbide on the surface of the piece.

[0087] In particular, the presence of crystalline or amorphous silica powders in the aforesaid layer of powders deposited on the surface of the piece can be provided so that during the heating step d), with exposure to the vapours of silicon, the silica powders generate silicon oxide gas which reacts with the free carbon of the piece, forming silicon carbide crystals, the latter becoming nucleation points for the crystallisation of silicon carbide on the surface of the

piece.
[0088] Advantageously, the aforesaid layer of powders can comprise powders of one or more transition metals. The inclusion of transition metal powders in the layer of powders which completely or partially covers the piece to be coated is aimed at the formation of silicides and/or carbo-silicides in the passivating coating as well. Operatively, during the heating step d) with exposure to the vapours of silicon, the aforesaid one or more transition metals can react with the silicon forming silicides and, in the presence of free carbon, forming mixed phases of silicides and carbon-silicides, so that said passivating coating also comprises silicides and/or carbon silicides .

[0089] Generally, the coatings of silicides and carbosilicides, in particular of molybdenum, chromium, titanium, have good thermal stability. In particular, the silicides and carbo-silicides have excellent resistance to oxidation and good mechanical features at high temperatures . The pres ence of silicides and/or carbo-silicides in the passivating coating can therefore be an improvement with respect to silicon carbide SiC alone, which tends to oxidize to form $SiO₂$.

[0090] As already mentioned above, the heating step d) of the oven is carried out maintaining predefined medium vacuum or low vacuum conditions inside the closed cham ber, so as to generate vapours of silicon from said silicon load inside the closed chamber.

[0091] Preferably, the heating step is carried out to generate vapours of silicon from the melting of the silicon load. However, the case in which the vapours of silicon are generated by sublimation of the silicon load is included as well.

[0092] Preferably, the aforesaid heating step d) is conducted in the presence of a partial nitrogen and/or inert gas pressure, so as to prevent pressures lower than the vapour tension of the silicon being reached inside the closed chamber of the oven and thus causing the vaporisation and sublimation of any silicon contained in said piece. The partial nitrogen and/or inert gas pressure is also inserted so as to reduce the wettability of the silicon with respect to the carbon and thereby limit the leakage of any silicon present inside said piece before the formation of said passivating coating.

[0093] Preferably, the heating step d) is carried out at temperatures higher than 1400° C., and in particular higher

than $1,414$ ^o C. (silicon melting temperature) to obtain silicon vapour formation. However, this value depends on the vacuum level or partial pressure of an inert gas or nitrogen introduced into the closed chamber. In this system, nitrogen
cannot be considered an inert gas since at temperatures
above 1400° C. it reacts with the free silicon of the CMC and the silicon of the silicon load to form silicon nitride. Low partial nitrogen pressures (around a few mbar) can in any case be used in the method according to the present invention as they do not involve significant nitriding of the coating and of the substrate (piece to be coated).
[0094] As previously highlighted, silicon melts at 1.414°

with the increasing temperature, as shown in the graph in FIG. 13. Once the maximum heating temperature has been set, it is not advisable to work at pressure values (vacuum) below the vapour tension of the silicon (represented by the curve shown in FIG. 13), if a CMC in which free silicon is present is to be coated. The free silicon inside the CMC would rapidly evaporate, generating voids in the substrate. Therefore, the insertion of partial pressures of nitrogen and inert gas (typically argon) has the purpose of preventing the silicon from undergoing evaporation. C. and the vapour tension thereof increases significantly

[0095] Advantageously, as a function of the vacuum level (pressure inside the closed chamber), the heating step d) can be conducted so as to reach a maximum temperature inside the closed chamber of the oven within a wide temperature range preferably between $1,350^{\circ}$ C. and $1,800^{\circ}$ C., more preferably between $1,400^{\circ}$ C. and $1,800^{\circ}$ C., and even more preferably between $1,450^{\circ}$ C. and $1,600^{\circ}$ C. $[0.096]$ Operatively, the operating parameters which gov-

ern the extent of the thickness of the final coating are :

[0097] the maximum heating temperature of the closed chamber;

[0098] exposure times of the piece to be coated to the vapours.

[0099] More in detail, it has been possible to experimentally verify that, by increasing the maximum heating tem-

perature and the exposure times to the vapours, the final thickness of the passivating coating increases.

[0100] Preferably, the heating step d) comprises the following sub-steps:

dl) heating from room temperature up to a predefined maximum temperature ;

d2) permanence at the maximum temperature; and

d3) cooling from the maximum temperature to room tem

[0101] The vapours of silicon can also form during substeps d1) and $\overline{d}3$ as a function of the temperature and pressure conditions. Therefore, although the formation of

the passivating coating occurs mainly during sub-step d2),
the coating can also grow during sub-steps d1) and d2).
[0102] Operatively, it has also been possible to experi-
mentally verify that the extent of microcracks in

[0103] Regardless of the thermal heating/cooling cycle, the passivating coating to the invention is less subject to microcracks than coatings obtained with traditional techniques due to the fact that it grows starting from materials belonging to the substrate with a dendritic structure and not by simple deposition which could lead to a clear distinction between coating and substrate.

[0104] Advantageously, the solid silicon load can consist of silicon in powder or in pieces.

[0105] In particular, the silicon load can be arranged in the closed chamber inside one or more crucibles.

 $[0106]$ The piece to be coated can be arranged in the closed chamber outside said one or more crucibles to avoid direct contact between liquid silicon and the piece to be coated.

[0107] Alternatively, the piece to be coated can be arranged above at least one of said crucibles with an arranged above at least one of said crucibles with an shurries are shown in Table 3 below.

interposed separator element made of a material substantially impermeable to liquid silicon, preferably BN, Si_3N_4 ,

SiC, B₄C.

ing to the invention are now presented.

Example 1

[0109] A CMC piece obtained by liquid silicon infiltration (LSI) of a long carbon fibre preform was coated by exposure to vapours of silicon in accordance with the method according to the present invention.

[0110] The volumetric ratios of the different components of the densified CMC are shown in Table 1 below.

TABLE 1

	Unit of measurement	LSI - CMC
Density	g / cc	2.05
Open porosity	$\frac{0}{0}$	2.4
SiC content.	Vol %	24
Si content	Vol %	8
C content	Vol $\%$	65

[0111] The CMC piece was placed in a carbon reinforced carbon fibre (CFC) crucible above a boron nitride plate with a hexagonal crystal structure . At the bottom of the crucible 1 kg of silicon powder with a particle size between 0.2 and 0.6 mm was inserted. The crucible was then subjected inside the closed chamber of an oven to a thermal cycle (heating step d) described in Table 2 below.

TABLE 2

Step	т initial Γ ° C.1	final Γ ^o C.1	Heating Rate \lceil [°] C./min]	Duration [min]	Atmosphere
2 3 4	25 1375 1600 1600	1375 1600 1600 25	4 -3	338 225 120 525	vacuum (0.1 mbar) vacuum (0.1 mbar) vacuum (0.1 mbar) vacuum (0.1 mbar)

[0112] At the end of the thermal cycle, the CMC piece had a passivating silicon carbide (SiC) coating, having an average thickness of $80-100 \mu m$. Overall, the thermal cycle (including the heating step and the cooling step) lasted 1,208 min (about 20 hours).

[0113] FIG. 2 shows an SEM image of the SiC coating collected from the backscattered electron signal.
[0114] The passivating coating covered the entire surface of the CMC; it was also found that there was no free carbon

directly exposed to the external environment. A less marked microcrack (seen in FIG. 2) was observed compared to that present in a coating deposited with the CVD technique, an example of which is seen in the image of FIG. 3.

Example 2

[0115] Two CMC samples having the same composition as the CMC of example 1 (Table 1) were coated with two different water-based slurries, containing a vinyl glue (dis-
persion of polyvinyl acetate resins; commercial product
Vinavil used). A first slurry contained silicon powders with
D50 25 µm, while the second slurry containe with $D50$ of 0.6 μ m. The compositions of the two different slurries are shown in Table 3 below.

TABLE 3

Slurry components	$Slurry1$ [wt %]	Slurry2 [wt $%$]
water Vinavil silicon D50 25 µm SiC D50 0.6 µm	66.1 6.6 27.2.	76.6 7.7 15.8
Total	100.0	100.0

[0116] The slurries were applied with a brush and left to dry in air at room temperature for 24 hours.

[0117] The two samples were placed in a crucible above boron nitride supports with a hexagonal crystal structure, as described in example 1. At the bottom of the crucible, 1 kg of powdered silicon with a particle size between 0.2 and 0.6 mm was introduced. The crucible was subjected to the same thermal cycle as in example 1 and described in Table 2.

[0118] It was observed that after thermal treatment (heating step d) the two CMC samples were covered with an SiC coating (passivating coating).
[0119] At the end of the thermal cycle, both CMC samples had a passivating s cycle (including the heating step and the cooling step) lasted 1,208 min (about 20 hours).

[0120] FIGS. 4 and 5 show SEM images collected from the surface of these two samples covered by the passivating coating according to the invention. More in detail, FIG. 4 shows a BES (Backscattered Electrons

[0121] Signals) SEM image of the surface coated with a coating obtained with the silicon slurry (slurry 1). The coating completely covers the substrate but the morphology of the fibre tow texture remains visible. The coating has a microcrack.

[0122] FIG. 5 shows a BES SEM image of the surface
coated with a coating obtained with the SiC slurry (slurry 2).
The coating completely covers the texture of the underlying
CMC. A fairly significant crack is observed, mor

covers the substrate, but the texture of the CMC correlated to the arrangement of the fibre tow remains visible. A very fine crack is also observed. The coating obtained by exposing a surface treated with SiC powder slurry to vapours of silicon also completely covers the texture of the substrate, but has a more pronounced crack.

Example 3

[0124] A large sample (35×25 cm approximately) in CMC consisting of long carbon fibres and Si/SiC matrix which has the same composition as example 1 and reported in Table 1 was coated with an SiC sub micrometric powder slurry. The composition of the slurry is the same based on SiC reported

in example 2 (slurry 2 in Table 3).
[0125] The CMC was then subjected to a thermal cycle described in Table 4 below, which differs from the cycle used in the preceding examples for the insertion of a partial pressure of nitrogen before and during the exposure to the vapours of silicon. Nitrogen was used in particular to inhibit the boiling of the free silicon inside the CMC , as described previously .

TABLE 4

Step	initial [° C.]	final 1° C.1	Heating Rate C/min Г°	Duration [min]	Atmosphere
1	25	1375	4	337.5	vacuum (0.1 mbar)
2	1375	1600		225	Partial pressure
3	1600	1600		120	$N2$ (5-10 mbar) Partial pressure $N2$ (5-10 mbar)
4	1600	1375	-3	75	vacuum (0.1 mbar)
5	1375	25	-3	450	vacuum (0.1 mbar)

[0126] At the end of the thermal cycle, the CMC piece had a passivating silicon carbide (SiC) coating, having an average thickness of 100 μ m. Overall, the thermal cycle (including the heating step and the cooling step) lasted 1,208 min (about 20 hours).

 $[0127]$ The sample was subjected to plasma wind tunnel (PWT) tests. The energy flow of the test was 345 KW/m2 and was applied to the sample surface for 600 seconds. The surface temperature of the CMC stabilised at about 1280º C. after a few tens of seconds.

[0128] After testing, no detachment of the coating was observed and the entire sample showed no detectable weight loss. FIG. 6 shows an SEM image of the surface of the coating before the PWT test, while FIGS. 7 and 8 show two SEM images of the surface and section of the coating

respectively after the PWT test.
[0129] In FIG. 7 (surface after PWT treatment) a microcrack can be detected. The extent of the microcracking of the coating after the PWT test is similar to that observed immediately after the formation of the coating and before the PWT test.

[0130] Prior to the PWT test, no free carbon directly exposed to the outer surface was observed. No detectable deterioration due to oxygen exposure was observed after the

[0131] In some areas seen as lighter coloured areas in FIG.
7 (surface after PWT test), the presence of oxidised silicon was detected.

Example 4

[0132] Two CMC samples, obtained by liquid silicon infiltration (LSI) of a long carbon fibre preform and having the composition reported in Table 5, were coated, the first with the SiC powder slurry of example 2; the second with colloidal graphite applied by spray. The two samples were then exposed to vapours of silicon in the same thermal cycle, performed according to the parameters of Table 4 (example 3).

TABLE 5

	Unit of measurement	LSI - CMC
Density	g/cc	1.94
Open porosity	$\frac{0}{6}$	3
SiC content	Vol %	19
Si content	Vol %	8
C content	Vol %	70

[0133] At the end of the thermal cycle , the CMC piece coated with slurry comprising SiC had a passivating silicon carbide (SiC) coating, having an average thickness of 30-40 μ m. Overall, the thermal cycle (including the heating step

and the cooling step) lasted 1,208 min (about 20 hours).
[0134] At the end of the thermal cycle, the CMC piece
coated with slurry comprising graphite had a passivating
silicon carbide (SiC) coating, having an average thick 50-60 µm. Overall, the thermal cycle (including the heating step and the cooling step) lasted 1,208 min (about 20 hours). [0135] FIG. 9 shows an SEM image of the surface of the coating obtained by covering the CMC with SiC powder slurry, while FIG. 10 shows an SEM image of the surface of the coating obtained by covering the CMC with graphite slurry .

[0136] It was observed that the exposure to vapours of silicon of the CMC sample having the composition reported in Table 5 and coated with SiC powder leads to the formation of a coating with a high content of free silicon. More in detail, in FIG. 9 in dark grey SiC microcrystals are observed embedded in a silicon coating (in light grey).
[0137] The use of a graphitic layer increases the surface

free carbon and favours a greater presence of SiC. In fact, in FIG. 10 an excellent distribution of SiC microcrystals (in dark grey) can be observed, which are smaller than those detected in FIG. 9.

[0138] The CMC prepared with graphitic slurry was sub-
jected to 6 PWT tests lasting 700 seconds each, for a total duration of 70 minutes. When fully operational, a tempera7

ture of 1250° C. was detected on the surface of the sample. Following the entire PWT test, the coating showed no signs of detachment and the CMC showed a minimal weight loss of 0.5%.

Example 5

[0139] A sample of CMC, obtained by liquid silicon infiltration (LSI) of a long carbon fibre preform and having the same composition as the CMC of example 4 and reported
in Table 5, was coated with a water-based slurry having the
composition shown in Table 6. The slurry contains a solid load of molybdenum powder with a grain size with D50 of 1.5 μ m. The slurry was applied with a brush and allowed to air dry at room temperature for 24 hours.

TABLE 6

Slurry components	Slurry3 [wt $%$]	
water polyvinyl alcohol (PVA) molybdenum D50 1.5 um	85.5 2.6 11.9	
Total	100.0	

[0140] The sample was placed in a crucible above boron nitride supports with a hexagonal crystal structure, as described in Example 1. At the bottom of the crucible, 1 kg of powdered silicon with a particle size between 0.2 and 0.6 mm was introduced. The crucible was subjected to the same thermal cycle as in example 3 and described in Table 4.

[0141] It was observed that after thermal treatment (heating step d) the CMC sample was covered with a coating having a mixed composition of silicon carbide (SiC) and molybdenum disilicide ($MoSi₂$) (passivating coat

[0142] In FIG. 11 a bundle of CMC carbon fibres is observed at the bottom in black, a portion of a passivating coating in silicon carbide (SiC) with a thickness of 20-25 μ m in grey, and a layer with a thickness of about 5-10 um of molybdenum disilicide $(MoSi₂)$ in white, adhered above the silicon carbide (SiC) layer.

[0143] At the end of the thermal cycle, the CMC sample had a passivating coating of silicon carbide (SiC) and molybdenum disilicide ($MOSi₂$), having an overall average

thickness of about 20-30 μ m.
[0144] The method according to the invention for making a passivating coating based on one or more silicon compounds, in particular comprising silicon carbide, on ceramic matrix composites reinforced with carbon fibres allows making coatings in non-high vacuum conditions and in the absence of dangerous gases to manage.

[0145] The method according to the invention can further be implemented with less expensive deposition plants than those required to make coatings with the CVD or PVD techniques .

[0146] The method according to the invention allows to reduce the process times with the same coating thickness obtained with respect to depositions by traditional techniques, and in particular CVD and PVD.

[0147] The method according to the invention allows to make passivating coatings based on one or more silicon compounds, and in particular based on SiC, on ceramic matrix composites reinforced with carbon fibres which are more compatibilized with the composite substrate than coat ings of silicon compounds obtained with traditional tech niques. The greater compatibilisation of the coating with the CMC substrate derives in particular from the fact that the coating grows starting from the substrate itself due to the reactions induced by the vapours of silicon, leading to a growth mode of the coating on the substrate characterised by dendritic structures. The latter allow a close adhesion between the substrate and coating. This allows the substrate itself to better withstand coatings with thicknesses of the order of tens of microns.
[0148] Therefore, the main advantages of the method

according to the present invention are the following:
[0149] possibility of working in non-high vacuum condi-

tions, of the order of mbar;

[0150] absence of dangerous gases to manage;

[0151] low cost of equipment if compared with that needed to make a CVD or PVD coating;

[0152] possibility of covering large pieces;

 $[0153]$ high deposition speed: in a few hours it is possible to obtain coatings with a thickness of the order of tens of um.

[0154] obtaining passivating coatings which are more adherent to the substrate and therefore more compatibilized therewith.

[0155] In particular , the method according to the present invention is suitable for being applied to coat pieces in milled and/or ground CMC. In fact, the coating obtainable with the method according to the present invention inte grates the exposed portions of the CMC after milling and/or grinding on the surface, without affecting the remaining

gering on the surface of the surface .
 [0156] The method according to the present invention also lends itself to being applied to coat a CMC component, which has undergone partial oxidation during the use thereof.

[0157] Therefore, the invention, thus conceived, achieves the preset objects.

 $[0158]$ Obviously, in the practical embodiment thereof, it may also take other shapes and configurations different from the one shown above, without thereby departing from the present scope of protection.

[0159] Moreover, all details may be replaced by technically equivalent elements, and any size, shape and material may be used according to the needs.

1. Method for making a silicon compound-based passivating coating on a ceramic matrix composite reinforced with carbon fibres, said method comprising the following steps:

- a) providing at least one piece made in a ceramic matrix composite reinforced with carbon fibres;
- b) placing said piece to be coated in a closed chamber of an oven:
- c) placing in said closed chamber a predefined load of solid silicon avoiding direct contact between the silicon and said piece to be coated;
- d) heating said oven, maintaining inside said chamber predefined medium vacuum conditions, with partial pressure from 5 mbar to 300 mbar, or low vacuum conditions, with partial pressure up to 0.1 mbar, so as to generate from said silicon load vapours of silicon inside said closed chamber, which react with one or more substances present on the surface of said piece leading to formation of a surface coating comprising composites of said one or more substances with the silicon;
- e) choosing the partial pressure of the vacuum, temperature inside said closed chamber and exposure times of said piece to the vapours of silicon so as to obtain a predefined thickness of the surface coating on said piece; and
- f) cooling said piece once said predefined thickness of the passivating coating has been reached.

2. Method according to claim 1, wherein free carbon is present on the surface of said piece and wherein during said heating step d) said vapours of silicon react with said free carbon forming silicon carbide as a silicon compound thus leading to the formation of a passivating coating comprising

silicon carbide.
3. Method according to claim 2, wherein the free carbon
is comprises entirely or at least partly by carbon of the
carbon reinforcement fibres of the ceramic matrix composite

which the piece is made of.
 4. Method according to claim 2, wherein the free carbon comprises at least partly carbon deposited on the surface of said piece before said heating step d).

5. Method according to claim 2 wherein said passivating coating comprising silicon carbide grows on the surface of said piece with a dendritic growth structure at an interface between the piece and the passivating coating.

6. Method according to claim 1, comprising a step g) of depositing on all or at least part of the surface of said piece a layer of powders of one or more materials to react with the vapours of silicon and chosen as a function of one or more passivating coating, said step g) being conducted before said heating step d).

7. Method according to claim 6, wherein said one or more materials are chosen from the group consisting of: graphite, coke, carbonaceous material, silicon, crystalline or amorphous silica, transition metals, silicon carbide and/or other carbides .

8. Method according to claim 6, wherein the powders of said one or more materials are deposited on the surface of said piece in the form of a slurry based on water, on an organic solvent or on a resin and in which the water, said solvent or said resin are made to evaporate and/or cross-link
before placing said piece in said closed chamber.

9. Method according to claim 6, wherein said layer of powders comprises powders of one or more transition metals so that during said heating step with exposure to vapours of silicon said one or more transition metals react with the silicon forming silicides and, in the presence of free carbon,

forming mixed phases of silicides and carbon-silicides, so that said passivating coating also comprises silicides and/or carbon silicides.

10. Method according to claim 1, wherein said heating step d) is conducted in the presence of a partial nitrogen and/or inert gas pressure, so as to prevent pressures lower than the vapour tension of the silicon being reac said closed chamber and causing vaporisation or sublimation of any silicon contained in said piece.

step d) is conducted in the presence of a partial nitrogen 11. Method according to claim 1, wherein said heating
step d) is conducted in the presence of a partial nitrogen
and/or inert gas pressure so as to reduce wettability of the
silicon with respect to the carbon and thereby l

temperature between $1,350^{\circ}$ C. and $1,800^{\circ}$ C.
13. Method according to claim 12, wherein the time

interval of permanence at the maximum temperature and the maximum temperature are chosen as a function of a desired final thickness of the passivating coating to be obtained on the piece to be coated.

14. Method according to claim 1, wherein said load of solid silicon comprises silicon in powder or in pieces arranged in said closed chamber inside one or more crucibles and wherein said piece to be coated is arranged in said closed chamber outside said one or more crucibles to prevent direct contact between liquid silicon and the piece to

15. Method according to claim 1, wherein said load of solid silicon comprises silicon in powder or in pieces arranged in said closed chamber inside one or more cru cibles and wherein said piece to be coated is arranged above

between said crucibles made of a material substantially
impermeable to liquid silicon.
16. Ceramic matrix composite reinforced with carbon
fibres provided with a silicon compound-based passivating
coating, wherein said p dritic growth structure in an interface area with a composite substrate .

17. Ceramic matrix composite reinforced with carbon fibres provided with a silicon compound-based passivating coating, wherein said passivating coating comprises a dendritic growth structure in an interface area with a com substrate, wherein said coating is obtained using the method according to claim 1.

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