

US008008241B2

# (12) United States Patent

# Souter

# (10) Patent No.: US 8,008,241 B2 (45) Date of Patent: Aug. 30, 2011

# (54) AUTOMATIC DISHWASHING DETERGENT COMPOSITION

- (75) Inventor: **Philip Frank Souter**, Morpeth (GB)
- (73) Assignee: The Procter & Gamble Company,

Cincinnati, OH (US)

(\*) Notice: Subject to any disclaimer, the term of this

patent is extended or adjusted under 35

U.S.C. 154(b) by 135 days.

- (21) Appl. No.: 12/397,489
- (22) Filed: Mar. 4, 2009

# (65) Prior Publication Data

US 2009/0233831 A1 Sep. 17, 2009

# (30) Foreign Application Priority Data

Mar. 14, 2008	(EP)	 08152749
Sep. 18, 2008	(EP)	 08164650

(51) **Int. Cl.** 

C11D 3/386 (2006.01) C12N 9/54 (2006.01) C12N 9/28 (2006.01)

- (52) **U.S. Cl.** ...... 510/226; 435/202; 435/221

See application file for complete search history.

# (56) References Cited

# U.S. PATENT DOCUMENTS

4,227,084 A	10/1980	Thomann et al
4,246,612 A	1/1981	Berry et al.
4,760,025 A	7/1988	Estell et al.
4,765,916 A	8/1988	Ogar et al.
4,810,410 A	3/1989	Diakun et al.

4,972,017 A	11/1990	Smith et al.
5,114,611 A	5/1992	Van Kralingen et al.
5,679,630 A	10/1997	Baeck et al.
5,856,164 A	1/1999	Outtrup et al.
6,287,841 B1*	9/2001	Mulleners et al 435/221
6,312,936 B1	11/2001	Poulose et al.
6,380,147 B1	4/2002	Speckmann et al.
6,426,229 B1	7/2002	Yamamoto et al.
6,939,702 B1	9/2005	Vind et al.

#### FOREIGN PATENT DOCUMENTS

DE	102005062984 A1	7/2007
WO	WO 9102792 A1	3/1991
WO	WO 9402618 A1	2/1994
WO	WO 9510591 *	4/1995
WO	WO 0242408 A2	5/2002
WO	WO 2008/010925 A2	1/2008

## OTHER PUBLICATIONS

"Stainzyme: a breakthrough in dishwashing performance" Focus on Surfactants, Elsevier, vol. 2004, No. 12, Dec. 1, 2004, p. 3, XP004699512 ISSN: 1351-4210.

\* cited by examiner

Primary Examiner — Tekchand Saidha (74) Attorney, Agent, or Firm — Melissa G Krasovec; Leonard W Lewis

## (57) ABSTRACT

An automatic dishwashing detergent composition comprising:

- a. at least 0.1 mg of active protease per gram of composition, wherein the protease is a variant of a protease that has at least 70% identity with the amino acid sequence of SEQ ID NO:1. wherein said variant comprise variations in one or more of the following positions: 32, 33, 48-54, 58-62, 94-107, 116, 123-133, 150, 152-156, 158-161, 164, 169, 175-186, 197, 198, 203-216 as compared with the protease in SEQ ID NO:1; and
- at least 0.05 mg of an active low temperature amylase per gram of composition.

# 10 Claims, No Drawings

# AUTOMATIC DISHWASHING DETERGENT COMPOSITION

#### TECHNICAL FIELD

The present invention is in the field of detergents. In particular, it relates to an automatic dishwashing detergent composition comprising a new enzymatic system. The composition provides excellent cleaning and finishing, it is environmentally friendlier than traditional compositions and allows for a more energy efficient automatic dishwashing process.

#### **BACKGROUND**

Automatic dishwashing detergents have improved over time but still some of the consumer needs are unmet in terms of both cleaning and finishing. In recent years there has been an ever increasing trend towards safer and environmentally friendly detergent compositions. This trend imposes addi- 20 tional constrains onto the automatic dishwashing formulator. In terms of energy efficiency and raw material savings, it is desirable to design products which provide good performance even at low temperatures and with a reduction on the amount of chemicals, in particular non-readily biodegradable 25 chemicals.

A frequent problem found in automatic dishwashing is the presence of grit on washed items. Grit is sometimes found in dishware/tableware after the automatic dishwashing process even if the items were free of it before they went into the 30 dishwasher. It seems that grit is formed during the dishwashing process. The mechanism of grit formation is not well understood. It maybe due to the high temperatures and combination of different soils lifted from the soiled items during the dishwashing process. Somehow, the different soils seem to recombine to give rise to grit which deposits onto the surface of the washed items. Once the grit is formed and deposited it is very difficult to remove it. The problem seems to be more acute when detergents in unit dose form are used.

Another frequently found and unsolved problem is the 40 presence of filming and spotting on washed items, this is particularly noticeable in glass and metal items.

In view of the above discussion, an objective of the present invention is to provide a more eco-friendly product that at the same time provides excellent cleaning and finishing benefits. 45

# SUMMARY OF THE INVENTION

According to a first aspect of the present invention, there is provided an automatic dishwashing detergent composition 50 comprising an improved enzymatic system comprising an improved protease (a variant protease as described herein below) in combination with a low temperature amylase (as described herein below). The composition of the invention provides cleaning and finishing benefits across a wide range 55 of temperatures, including low temperatures, improving the energy profile of the dishwashing process. Surprisingly, the composition of the invention allows for a more energy efficient dishwashing processes without compromising in cleaning and finishing.

The variant protease for use in the composition of the invention is a protease wherein the variations are made versus a protease that has at least 70%, preferably at least 90%, more preferably at least 95%, even more preferably at least 99% and especially 100% identity with the amino acid sequence of 65 SEQ ID NO:1. Said variant protease comprises substitutions in one or more of the following positions: 32, 33, 48-54,

2

58-62, 94-107, 116, 123-133, 150, 152-156, 158-161, 164, 169, 175-186, 197, 198, 203-216 as compared with the protease in SEQ ID NO:1. Excellent results in terms of cleaning and finishing (including absence of spotting and filming and excellent shine) have been obtained when the variant protease comprises the following variations: G116V+S126L+P127Q+

Particularly good results are obtained when the variant protease is used in combination with a low temperature amy-10 lase which is either:

- (a) a variant comprising one or more, preferably three or more substitutions in the following positions versus SEQ ID NO: 2:
- 9, 26, 149, 182, 186, 202, 257, 295, 299, 323, 339 and 345; and
- (b) optionally with one or more, preferably all of the substitutions and/or deletions in the following positions: 118, 183, 184, 195, 320 and 458, which if present preferably comprise R118K, D183\*, G184\*, N195F, R320K and/or R458K.

60

15

(c) a variant comprising at least one substitution in the following positions versus SEQ ID NO:4: M202, M208, S255, R172, and/or M261. Preferably said amylase comprises one or more of M202L, M202V, M202S, M202T, M2021, M202Q, M202W, S255N and/or R172Q. Particularly preferred are those comprising the M202L or M202T mutations.

Preferred low temperature amylases include those comprising the following sets of mutations:

- (i) M9L+, M323T;
- (ii) M9L+M202L/T/V/I+M323T;
- (iii) M9L+N195F+M202L/V/I+M323T;
- (iv) M9L+R118K+D183\*+G184\*+R320K+M323T+ R458K;
- (v) M9L+R118K+D183\*+G184\*+M202L/T/V/I; R320K+M323T+R458K;
- M9L+G149A+G182T+G186A+M202L+T257I+ Y295F+N299Y+M323T+A339S+E345R;
- M9L+G149A+G182T+G186A+M202I+T257I+ Y295F+N299Y+M323T+A339S+E345R;
- M9L+R118K+G149A+G182T+D183\*+G184\*+ G186A+M202L+T257I+Y295F+N299Y+R320K+ M323T+A339S+E345R+R458K;
- M9L+R118K+G149A+G182T+D183\*+G184\*+ G186A+N195F+M202L+T257I+Y295F+N299Y+ R320K+M323T+A339S+E345R+R458K; (x) M9L+ R118K+G149A+G182T+D183\*+G184\*+G186A+ M202I+T257I+Y295F+N299Y+R320K+M323T+ A339S+E345R+R458K;
- M9L+R118K+D183\*+D184\*+N195F+M202L+ (xi) R320K+M323T+R458K;
- M9L+R118K+D183\*+D184\*+N195F+M202T+ (xii) R320K+M323T+R458K;
- M9L+R118K+D183\*+D184\*+N195F+M202I+ (xiii) R320K+M323T+R458K;
- M9L+R118K+D183\*+D184\*+N195F+M202V+ R320K+M323T+R458K;
- M9L+R118K+N150H+D183\*+D184\*+N195F+ M202L+V214T+R320K+M323T+R458K; or
- M9L+R118K+D183\*+D184\*+N195F+M202L+ V214T+R320K+M323T+E345N+R458K.

The most preferred amylase is the variant sold under the tradename Stainzyme Plus<sup>TM</sup> (Novozymes A/S, Bagsvaerd, Denmark).

A further preferred set of low temperature amylases are those derived from Bacillus sp. 707, whose sequence is shown 0.0 0,000,2 11 22

as SEQ ID NO: 4, preferably comprising one or more of the following mutations M202, M208, S255, R172, and/or M261. Preferably said amylase comprises one or more of M202L, M202V, M202S, M202T, M202I, M202Q, M202W, S255N and/or R172Q. Particularly preferred are those variants comprising the M202L or M202T mutations.

In preferred embodiments, the composition comprises a high level of amylase, at least 0.2 mg of active amylase per gram of composition, preferably from about 0.2 to about 10, more preferably from about 0.25 to about 6, specially pre- 10 ferred from about 0.3 to about 4 mg of active amylase per gram of composition. It has been found that compositions comprising a high level of amylase help to prevent grit formation during the automatic dishwashing process and provide good cleaning and finishing results. Better results in 15 terms of grit removal can be achieved when the composition comprises a lipase, thus in a preferred embodiment the composition of the invention comprise a lipase, preferably a lipase derived from the Humicola Lanuginosa wild -type that contains the mutations T231R and N233R. Compositions com- 20 prising Lipex® (Novozymes A/S, Bagsvaerd, Denmark) have been found particularly effective in terms of grit prevention.

The cleaning results can be further improved by adding a high level of protease to the composition. Thus, in further preferred embodiments, the composition comprises a high 25 level of protease, in particular at least 1.5 mg of active protease per gram of composition. Preferred levels of protease in the compositions of the invention include from about 1.5 to about 10, more preferably from about 1.8 to about 5 and especially from about 2 to about 4 mg of active protease per grams of composition. The biodegradability of the compositions of the invention is further improved when a high level of enzyme is present. The high level of enzyme allows for a reduction on the level of non-biodegradable ingredients of the product and at the same time improves cleaning and finishing 35 performance.

In especially preferred embodiments, the composition comprises an anti-redeposition agent and/or sulfonated polymer. Excellent finishing results are obtained with compositions comprising an anti-redeposition agent or a sulfonated 40 polymer and in particular compositions comprising a combination thereof. Benefits are seen in terms of reduction/prevention of filming, spotting and improvement on shine. Shine on washed items seem to be an unsolved problem, in particular in stressed cases of highly soiled loads. The compositions of the invention provide shine benefits even under stressed conditions. These benefits, under stressed conditions, are not easily achievable with compositions lacking the enzymatic system of the invention.

In preferred embodiments, the compositions of the invention reduce the particle size of the soil fragments and/or molecular weight as compared to that obtained with traditional detergent compositions. This facilitates the suspension of the soils in the wash liquor. Soil suspension can further be improved by an anti-redeposition agent. The anti-redeposi- 55 tion agent contributes to keep detached soils as individual entities in solution and prevents re-combination that can give rise to grit formation. These agents can also help to detach soils from the soiled surfaces. This in combination with soil suspension contributes to a more effective enzymatic clean- 60 ing and results in better shine and reduced filming and spotting on the washed items. Preferred anti-redeposition agents are non-ionic surfactants, in particular non-ionic surfactants having a phase inversion temperature (PIT) in the range of from about 40 to about 70° C. Compositions comprising 65 non-ionic surfactants having a PIT in this temperature range provide very good cleaning. The anti-redeposition agent may

4

also help the enzymes to get to the soiled substrates. The anti-redeposition agent seems to help with the cleaning during the main wash. Some of the anti-redeposition agent is carried over to the rinse cycle where it helps with sheeting thereby reducing/eliminating filming and spotting. Surfactants, having a PIT in the claimed range, present cleaning properties during the main wash and sheeting properties during the rinse. In other preferred embodiments the anti-redeposition agent is a non-ionic surfactant having a Draves wetting time (as measured using the standard method ISO 8022 under the following conditions; 3-g hook, 5-g cotton skein, 0.1% by weight aqueous solution at a temperature of 25° C.) of less than about 360 seconds, preferably less than 60 seconds.

It is also preferred that the compositions of the invention comprise a metal care agent, in particular a zinc salt.

In preferred embodiments the composition of the invention is in unit dose form. Products in unit dose form include tablets, capsules, sachets, pouches, etc. Preferred for use herein are tablets wrapped with a water-soluble film and water-soluble pouches. The weight of the composition of the invention is from about 10 to about 25 grams, preferably from about 12 to about 24 grams and more preferably from 14 to 22 grams. These weights are extremely convenient for automatic dishwashing product dispenser fit. In the cases of unit dose products having a water-soluble material enveloping the detergent composition, the water-soluble material is not considered as part of the composition.

In preferred embodiments the unit dose form is a water-soluble pouch (i.e., water-soluble film enveloping a detergent composition), preferably a multi-compartment pouch having a plurality of films forming a plurality of compartments. This configuration contributes to the flexibility and optimization of the composition. It allows for the separation and controlled release of different ingredients. Preferably one compartment contains a composition in solid form and another compartment contains a composition in liquid form.

In preferred multi-compartment pouch embodiments two different compartments contain anti-redeposition agent. Preferably the films of these two compartments have different dissolution profiles, allowing the release of the same or different anti-redeposition agents at different times. For example, anti-redeposition agent from one compartment (first compartment) can be delivered early in the washing process to help with soil removal and anti-redeposition agent from another compartment (second compartment) can be delivered at least two minutes, preferably at least five minutes later than the anti-redeposition agent from the first compartment. Ideally, the enzymes should be delivered after the anti-redeposition agent from the first compartment and before the anti-redeposition agent from the second compartment.

Especially preferred for use herein is a multi-compartment pouch comprising two side-by-side compartments superposed onto another compartment wherein at least two different compartments contain two different compositions.

According to the second aspect of the invention, there is provided a method of dishwashing in an automatic dishwashing machine using the detergent composition of the invention and comprising the steps of placing the composition into the product dispenser and releasing it during the main-wash cycle.

In preferred process embodiments, anti-redeposition agents are delivered at two different times of the dishwashing process.

According to the third aspect of the invention, there is provided the use of the detergent composition of the invention for automatic dishwashing at low temperature (i.e., the mainwash temperature is no more than 50, preferably no more than 45 and more preferably no more than 40° C.).

## DETAILED DESCRIPTION OF THE INVENTION

The present invention envisages an automatic dishwashing detergent composition comprising a new enzymatic system. The system comprises an improved protease and a low temperature amylase and optionally a lipase. The compositions of the invention prevent grit formation on washed items, which 10 Amino Acid Identity is one of the problems currently found in automatic dishwashing. The composition provides excellent cleaning and finishing results and it is environmentally friendly in terms of energy and raw material reduction. The present invention also envisages a method of dishwashing using the composition of 15 the invention.

## Enzyme Related Terminology

Nomenclature for Amino Acid Modifications

In describing enzyme variants herein, the following nomenclature is used for ease of reference: Original amino 20 acid(s):position(s):substituted amino acid(s).

According to this nomenclature, for instance the substitution of glutamic acid for glycine in position 195 is shown as G195E. A deletion of glycine in the same position is shown as G195\*, and insertion of an additional amino acid residue such 25 as lysine is shown as G195GK. Where a specific enzyme contains a "deletion" in comparison with other enzyme and an insertion is made in such a position this is indicated as \*36D for insertion of an aspartic acid in position 36. Multiple mutations are separated by pluses, i.e.: S99G+V102N, repre- 30 senting mutations in positions 99 and 102 substituting serine and valine for glycine and asparagine, respectively. Where the amino acid in a position (e.g. 102) may be substituted by another amino acid selected from a group of amino acids, e.g. the group consisting of N and I, this will be indicated by 35

In all cases, the accepted IUPAC single letter or triple letter amino acid abbreviation is employed.

Protease Amino Acid Numbering

The numbering used in this patent is numbering versus the 40 specific protease (PB92) listed as SEQ ID No: 1. An alternative numbering scheme is the so-called BPN' numbering scheme which is commonly used in the art. For convenience the numbering schemes are compared below in Table 1:

TABLE 1

Protease Muta	tion numbering	-
PB92 numbering of this patent (numbering versus SEQ ID NO: 1)	Equivalent BPN' numbering	50
G116V + S126L + P127Q + S128A	G118V + S128L + P129Q + S130A	•
G116V + S126N + P127S +	G118V + S128N + P129S +	
S128A + S160D	S130A + S166D	
G116V + S126L + P127Q + S128A + S160D	G118V + S128L + P129Q + S130A + S166D	55
G116V + S126V + P127E + S128K	G118V + S128V + P129E + S130K	
G116V + S126V + P127M + S160D	G118V + S128V + P129M + S166D	
S128T	S130T	60
G116V + S126F + P127L + S128T	G118V + S128F + P129L + S130T	O.C
G116V + S126L + P127N + S128V	G118V + S128L + P129N + S130V	
G116V + S126F + P127Q	G118V + S128F + P129Q	
G116V + S126V + P127E +	G118V + S128V + P129E +	
S128K + S160D	S130K + S166D	65
G116V + S126R + P127S + S128P	G118V + S128R + P129S + S130P	0.3
S126R + P127Q + S128D	S126R + P129Q + S130D	

6

TABLE 1-continued

	Protease Mutation numbering									
5	PB92 numbering of this patent (numbering versus SEQ ID NO: 1)	Equivalent BPN' numbering								
	S126C + P127R + S128D S126C + P127R + S128G	S128LC + P129R + S130D S128LC + P129R + S130G								

The relatedness between two amino acid sequences is described by the parameter "identity". For purposes of the present invention, the alignment of two amino acid sequences is determined by using the Needle program from the EMBOSS package (http://emboss.org) version 2.8.0. The Needle program implements the global alignment algorithm described in Needleman, S. B. and Wunsch, C. D. (1970) J. Mol. Biol. 48, 443-453. The substitution matrix used is BLO-SUM62, gap opening penalty is 10, and gap extension penalty

The degree of identity between an amino acid sequence of and enzyme used herein ("invention sequence") and a different amino acid sequence ("foreign sequence") is calculated as the number of exact matches in an alignment of the two sequences, divided by the length of the "invention sequence" or the length of the "foreign sequence", whichever is the shortest. The result is expressed in percent identity. An exact match occurs when the "invention sequence" and the "foreign sequence" have identical amino acid residues in the same positions of the overlap. The length of a sequence is the number of amino acid residues in the sequence.

Variant Protease

The variant protease for use herein is a protease with variations versus a protease that has at least 70%, preferably at least 90%, more preferably at least 95%, even more preferably at least 99% and especially 100% identity with the amino acid sequence of SEQ ID NO:1. Said variant protease comprises substitutions in one or more of the following positions: 32, 33, 48-54, 58-62, 94-107, 116, 123-133, 150, 152-156, 158-161, 164, 169, 175-186, 197, 198, 203-216 as compared with the protease in SEQ ID NO:1. Preferably, said protease has substitutions in one or more of the following positions: 60, 94, 97-102, 105, 116, 123-128, 150, 152, 160, 183, 203, 211, 212, 213, 214 and 216. More preferably, the protease 45 comprises mutations in one or more, even more preferably in three or more of the following positions, 116, 126, 127, 128 and 160.

Especially preferred are variants with mutations in each of positions 116, 126, 127 and 128.

Particularly suitable for use in the composition of the invention has been found to be a protease comprising the following specific mutations versus the enzyme of SEQ ID

(i) G116V+S126L+P127Q+S128A (ii) G116V+S126N+P127S+S128A+S160D (iii) G116V+S126L+P127Q+S128A+S160D (iv) G116V+S126V+P127E+S128K (v) G116V+S126V+P127M+S160D (vi) S128T (vii) G116V+S126F+P127L+S128T (viii) G116V+S126L+P127N+S128V (ix) G116V+S126F+P127Q (x) G116V+S126V+P127E+S128K+S160D (xi) G116V+S126R+P127S+S128P (xii) S126R+P127Q+S128D (xiii) S126C+P127R+S128D; or (xiv) S126C+P127R+S128G

Especially preferred for use in the composition of the invention has been found to be a protease comprising the mutations G116V+S126L+P127Q and S128A.

Assay for Protease Activity

Protease activity is measured using Dimethyl Casein <sup>5</sup> (DMC). Release of peptides is initiated via protease action. Protease activity is measured in PU's. 1 PU (protease unit) is the amount of enzyme which hydrolyzes casein such that the initial rate of formation of peptides per minute corresponds to 1 µmole of glycine per minute. 1 KPU is equal to 1000 <sup>10</sup> protease units.

Analysis

A 2,4,6 Trinitrobenzenesulphonic acid (TNBSA) solution and a DMC solution are prepared. All ingredients are from Sigma-Aldrich, Milwaukee, USA, unless otherwise stated. The TNBSA solution is made by dissolving 0.40 mL of TNBSA (Sigma Cat No P-2297) in 50 mL of deionized water. The DMC solution is made by dissolving 5.09 g of Potassium Chloride (Sigma Catalogue No: P-3911) and 1.545 g of Boric 20 Acid (Sigma Catalogue No: B-0399) in 500 mL of deionized water. The solution is stirred for 10 mins to dissolve and then the pH adjusted to 9.0 using 50% NaOH. 2 g of DMC are then added (DMC, British Drug House, Cat No. 79457) and the solution is stirred to dissolve.

 $100~\mu L$  of a dilute enzyme containing sample is added (0.5% sodium sulfite solution with 0.04% calcium chloride; Sigma Catalogue No: S-6672 and Sigma Catalogue No: C-5080, respectively) to 1800  $\mu L$  of DMC solution. The resultant solution is mixed and incubated at 37° C. for 4  $_{30}$  minutes. Then 900  $\mu L$  of TNBSA solution are added to the mixture and incubated for another 5 minutes. The absorbance is read at 415 nm.

Preferably, the variant protease of the invention has an activity of at least 0.3 KNPU per gram of composition, more 35 preferably at least 0.7 KNPU per gram of composition and especially 1 KNPU per gram of composition.

Additional Proteases

In the composition of the invention a mixture of two or more proteases may be used. A mixture of proteases can 40 contribute to an enhanced cleaning across a broader temperature and/or substrate range and provide superior shine benefits, especially when used in conjunction with an anti-redeposition agent and/or a sulfonated polymer.

Suitable proteases for use in combination with the variant 45 protease of the invention include metalloproteases and serine proteases, including neutral or alkaline microbial serine proteases, such as subtilisins (EC 3.4.21.62). Suitable proteases include those of animal, vegetable or microbial origin. Microbial origin is preferred. Chemically or genetically modified 50 mutants are included. The protease may be a serine protease, preferably an alkaline microbial protease or a chymotrypsin or trypsin-like protease. Examples of neutral or alkaline proteases include:

- (a) subtilisins (EC 3.4.21.62), especially those derived 55 from *Bacillus*, such as *Bacillus lentus*, *B. alkalophilus*, *B. subtilis*, *B. amyloliquefaciens*, *Bacillus pumilus* and *Bacillus gibsonii* described in U.S. Pat. Nos. 6,312,936 B1, 5,679,630, 4,760,025, DE102006022216A1 and DE102006022224A1.
- (b) trypsin-like or chymotrypsin-like proteases, such as trypsin (e.g., of porcine or bovine origin), the Fusarium protease described in WO 89/06270 and the chymotrypsin proteases derived from Cellumonas described in WO 05/052161 and WO 05/052146.
- (c) metalloproteases, especially those derived from *Bacillus amyloliquefaciens* decribed in WO 07/044993A2.

8

Preferred commercially available protease enzymes include those sold under the trade names Alcalase®, Savinase®, Primase®, Durazym®, Polarzyme®, Kannase®, Liquanase®, Ovozyme®, Neutrase®, Everlase® and Esperase® by Novozymes A/S (Denmark), those sold under the tradename Maxatase®, Maxacal®, Maxapem®, Properase®, Purafect®, Purafect Prime®, Purafect Ox®, FN3®, FN4® and Purafect OXP® by Genencor International, and those sold under the tradename Opticlean® and Optimase® by Solvay Enzymes.

Especially preferred for use herein in combination with the variant protease of the invention are commercial proteases selected from the group consisting of Properase®, Purafect®, Ovozyme®, Everlase®, Savinase® and FN3®.

Low Temperature Amylases

Preferred amylases for use herein are low temperature amylases. Compositions comprising low temperature amylases allow for a more energy efficient dishwashing processes without compromising in cleaning.

As used herein, "low temperature amylase" is an amylase that demonstrates at least 1.2, preferably at least 1.5 and more preferably at least 2 times the relative activity of the reference amylase at 25° C. As used herein, the "reference amylase" is the amylase of SEQ ID NO:3, commercially available under the tradename of Termamyl<sup>TM</sup> (Novozymes A/S). As used herein, "relative activity" is the fraction derived from dividing the activity of the enzyme at the temperature assayed versus its activity at its optimal temperature measured at a pH of 9.

Preferably low temperature amylases possess one or more of the following properties:

- (a) greater than or equal to 60%, preferably 70%, more preferably 80% and especially 90% of their maximum activity at 50° C.
- (b) greater than or equal to 30%, preferably 40%, more preferably 50%, even more preferably 60% and especially 70% of their maximum activity at 40° C.
- (c) greater than or equal to 20%, preferably 30% more preferably 40% of their maximum activity at 30° C.

Activity may be determined by well-known standard amylase assays described herein below and is assayed between 20 and  $90^{\circ}$  C.

Low temperature amylases for use herein, including chemically or genetically modified mutants (variants), are alkaline amylases possessing at least 90%, preferably 95%, more preferably 98%, even more preferably 99% and especially 100% identity, with those derived from *Bacillus* sp. NCIB 12289, NCIB 12512, NCIB 12513, DSM 9375 (U.S. Pat. No. 7,153,818) DSM 12368, DSMZ no. 12649, KSM AP1378 (WO 97/00324), *Bacillus* sp. 707, KSM K36 or KSM K38 (EP 1,022,334). Preferred low temperature amylases include:

- (a) the variants described in U.S. Pat. No. 5,856,164 and WO99/23211, WO 96/23873, WO00/60060 and WO 06/002643, especially the variants with one or more substitutions in the following positions versus SEQ ID No: 2: 9, 26, 30, 33, 82, 37, 106, 118, 128, 133, 149, 150, 160, 178, 182, 186, 193, 195, 202, 203, 214, 231, 256, 257, 258, 269, 270, 272, 283, 295, 296, 298, 299, 303, 304, 305, 311, 314, 315, 318, 319, 320, 323, 339, 345, 361, 378, 383, 419, 421, 437, 441, 444, 445, 446, 447, 450, 458, 461, 471, 482, 484 that also preferably contain the deletions of D183\* and G184\*.
- (b) variants exhibiting at least 90% identity with the wildtype enzyme from *Bacillus* SP722 (SEQ ID No. 4 in WO06/002643, p. 7-9 of sequence listings), especially

variants with deletions in the 183 and 184 positions and variants described in WO 00/60060, which is incorporated herein by reference.

(c) variants exhibiting at least 95% identity with SEQ ID NO:4, the wild-type enzyme from *Bacillus* sp. 707, 5 especially those comprising mutations in one or more of the following positions M202, M208, S255, R172, and/ or M261.

Preferred commercially available low temperature amylases for use herein are STAINZYME®, STAINZYME 10 PLUS®, STAINZYME ULTRA® and NATALASE® (Novozymes A/S).

In a preferred embodiment the low temperature amylase is a variant with either:

- (a) one or more, preferably three or more substitutions in 15 the following positions versus SEQ ID NO: 2:
  - 9, 26, 149, 182, 186, 202, 257, 295, 299, 323, 339 and 345; and
- (b) optionally with one or more, preferably all of the substitutions and/or deletions in the following positions: 20 118, 183, 184, 195, 320 and 458, which if present preferably comprise R118K, D183\*, G184\*, N195F, R320K and/or R458K.

or:

(c) at least one substitution in the following positions versus SEQ ID NO:4: M202, M208, S255, R172, and/or M261. Preferably said amylase comprises one or more of M202L, M202V, M202S, M202T, M202I, M202Q, M202W, S255N and/or R172Q. Particularly preferred are those comprising the M202L or M202T mutations. 30

Preferred low temperature amylases include those comprising the following sets of mutations:

- (i) M9L+M323T;
- (ii) M9L+M202L/T/V/I+M323T;
- (iii) M9L+N195F+M202L/T/V/I+M323T;
- (iv) M9L+R118K+D183\*+G184\*+R320K+M323T+ R458K;
- $\begin{array}{lll} \text{(v)} & \text{M9L+R118K+D183*+G184*+M202L/T/V/I+R320K+M323T+R458K;} \end{array}$
- (vi) M9L+G149A+G182T+G186A+M202L+T257I+ 40 Y295F+N299Y+M323T+A339S+E345R;
- (vii) M9L+G149A+G182T+G186A+M202I+T257I+ Y295F+N299Y+M323T+A339S+E345R;
- (viii) M9L+R118K+G149A+G182T+D183\*+G184\*+ G186A+M202L+T257I+Y295F+N299Y+R320K+ M323T+A339S+E345R+R458K;
- (ix) M9L+R118K+G149A+G182T+D183\*+G184\*+ G186A+N195F+M202L+T257I+Y295F+N299Y+ R320K+M323T+A339S+E345R+R458K;
- (x) M9L+R118K+G149A+G182T+D183\*+G184\*+ 50 G186A+M202I+T257I+Y295F+N299Y+R320K+ M323T+A339S+E345R+R458K;
- (xi) M9L+R118K+D183\*+D184\*+N195F+M202L+ R320K+M323T+R458K;
- (xii) M9L+R118K+D183\*+D184\*+N195F+M202T+ 55 R320K+M323T+R458K;
- (xiii) M9L+R118K+D183\*+D184\*+N195F+M202I+ R320K+M323T+R458K;
- (xiv) M9L+R118K+D183\*+D184\*+N195F+M202V+ R320K+M323T+R458K;
- (xv) M9L+R118K+N150H+D183\*+D184\*+N195F+ M202L+V214T+R320K+M323T+R458K; or
- (xvi) M9L+R118K+D183\*+D184\*+N195F+M202L+ V214T+R320K+M323T+E345N+R458K.

The most preferred amylase is the variant sold under the 65 tradename Stainzyme Plus<sup>™</sup> (Novozymes A/S, Bagsvaerd, Denmark).

10

Preferred for use herein is a combination of a mixture of two or more low temperature amylases. A mixture of amylases can contribute to an enhanced cleaning across a broader temperature and/or substrate range and provide superior shine benefits, especially when used in conjunction with an anti-redeposition agent and/or a sulfonated polymer. Assay for Alpha-Amylase Activity

Amylase activity is measured using a maltoheptaoside modified with a p-Nitrophenol chromophore (Infinity Amylase Reagent from Thermo Electron, Woburn, Mass., USA, Cat #: TR25421). Release of the chromophore is initiated via amylase action. Amylase activity is measured initially in AMU'ss. 1 AMU (amylase unit) is the amount of enzyme which hydrolyzes PNP-G7 (p-nitrophenyl-alpha,D-maltoheptaoside) carbohydrate substrate such that the initial rate of formation of small carbohydrates (G2-4) per minute corresponds to 1 μmole of 4-Nitrophenol per minute.

The test is run versus a reference enzyme, that of SEQ ID No:3 sold under the tradename Termamyl<sup>TM</sup> (Novozymes A/S). These amylase units (AMUs) are converted into a unit of kNU, using the conversion factor 0.133 mg of Termamyl<sup>TM</sup> corresponds to 1 KNU. Therefore if using the above assay the enzyme sample shows an activity equivalent to that shown by 0.266 mg of Termamyl<sup>TM</sup>, its activity is considered to be 2 KNU.

Analysis

 $200~\mu L$  of dilute enzyme containing sample is added to  $2500~\mu L$  of Infinity amylase reagent. Mix and incubate at  $37^{\circ}$  C. for 4.5 minutes. The absorbance is read at 415 nm.

Preferably, the low temperature amylase in the composition of the invention has an activity of at least 6 KNU, more preferably at least 7.5 KNU per gram of detergent composition.

Additional Enzymes

Additional enzymes suitable for use in the composition of the invention can comprise one or more enzymes selected from the group comprising hemicellulases, cellulases, cellobiose dehydrogenases, peroxidases, proteases, xylanases, lipases, phospholipases, esterases, cutinases, pectinases, mannanases, pectate lyases, keratinases, reductases, oxidases, phenoloxidases, lipoxygenases, ligninases, pullulanases, tannases, pentosanases, malanases,  $\beta$ -glucanases, arabinosidases, hyaluronidase, chondroitinase, laccase, amylases, and mixtures thereof.

In preferred embodiments, such additional enzyme may be selected from the group consisting of lipases, including "first cycle lipases" comprising a substitution of an electrically neutral or negatively charged amino acid with R or K at any of positions 3, 224, 229, 231 and 233 on the wild-type of *Humicola Lanuginosa*, whose sequence is shown as SEQ ID No 1 in pages 5 and 6 of U.S. Pat. No. 6,939,702 B1, preferably a variant comprising T231R and N233R mutations. One such preferred variant is sold under the tradename Lipex® (Novozymes A/S, Bagsvaerd, Denmark).

Enzyme stabilizer components—Suitable enzyme stabilizers include oligosaccharides, polysaccharides and inorganic divalent metal salts, such as alkaline earth metal salts, especially calcium salts. Chlorides and sulphates are preferred with calcium chloride an especially preferred calcium salt according to the invention. Examples of suitable oligosaccharides and polysaccharides, such as dextrins, can be found in WO07/145964A2 which is incorporated herein by reference. In case of aqueous compositions comprising protease, a reversible protease inhibitor, such as a boron compound, inckuding borate and 4-formyl phenyl boronic acid or a tripeptide aldehyde, can be added to further improve stability. Anti-redeposition Agent

Suitable for use herein as anti-redeposition agents are non-ionic surfactants. Traditionally, non-ionic surfactants have been used in automatic dishwashing for surface modification purposes in particular for sheeting to avoid filming and spotting and to improve shine. It has been found that in the 5 compositions of the invention non-ionic surfactants contribute to prevent redeposition of soils.

In preferred embodiments the anti-redeposition agent is a non-ionic surfactant or a non-ionic surfactant system having a phase inversion temperature, as measured at a concentration of 1% in distilled water, between 40 and 70° C., preferably between 45 and 65° C. By a "non-ionic surfactant system" is meant herein a mixture of two or more non-ionic surfactants. Preferred for use herein are non-ionic surfactant systems. They seem to have improved cleaning and finishing properties and better stability in product than single non-ionic surfactants.

Phase inversion temperature is the temperature below which a surfactant, or a mixture thereof, partitions preferentially into the water phase as oil-swollen micelles and above 20 which it partitions preferentially into the oil phase as water swollen inverted micelles. Phase inversion temperature can be determined visually by identifying at which temperature cloudiness occurs.

The phase inversion temperature of a non-ionic surfactant or system can be determined as follows: a solution containing 1% of the corresponding surfactant or mixture by weight of the solution in distilled water is prepared. The solution is stirred gently before phase inversion temperature analysis to ensure that the process occurs in chemical equilibrium. The phase inversion temperature is taken in a thermostable bath by immersing the solutions in 75 mm sealed glass test tube. To ensure the absence of leakage, the test tube is weighed before and after phase inversion temperature measurement. The temperature is gradually increased at a rate of less than 1° C. per minute, until the temperature reaches a few degrees below the pre-estimated phase inversion temperature. Phase inversion temperature is determined visually at the first sign of turbidity.

Suitable nonionic surfactants include: i) ethoxylated nonionic surfactants prepared by the reaction of a monohydroxy 40 alkanol or alkyphenol with 6 to 20 carbon atoms with preferably at least 12 moles particularly preferred at least 16 moles, and still more preferred at least 20 moles of ethylene oxide per mole of alcohol or alkylphenol; ii) alcohol alkoxylated surfactants having a from 6 to 20 carbon atoms and at least one ethoxy and propoxy group. Preferred for use herein are mixtures of surfactants i) and ii).

Another suitable non-ionic surfactants are epoxy-capped poly(oxyalkylated) alcohols represented by the formula:

$$R_1O[CH_2CH(CH_3)O]_x[CH_2CH_2O]_\nu[CH_2CH(OH)R_2] \tag{I}$$

wherein  $R_1$  is a linear or branched, aliphatic hydrocarbon radical having from 4 to 18 carbon atoms;  $R_2$  is a linear or branched aliphatic hydrocarbon radical having from 2 to 26 carbon atoms; x is an integer having an average value of from 0.5 to 1.5, more preferably about 1; and y is an integer having a value of at least 15, more preferably at least 20.

Preferably, the surfactant of formula I, at least about 10 carbon atoms in the terminal epoxide unit [CH<sub>2</sub>CH(OH)R<sub>2</sub>]. Suitable surfactants of formula I, according to the present invention, are Olin Corporation's POLY-TERGENT® SLF-18B nonionic surfactants, as described, for example, in WO 94/22800, published Oct. 13, 1994 by Olin Corporation.

Preferably non-ionic surfactants and/or system to use as anti-redeposition agents herein have a Draves wetting time of less than 360 seconds, preferably less than 200 seconds, more 65 preferably less than 100 seconds and especially less than 60 seconds as measured by the Draves wetting method (standard

12

method ISO 8022 using the following conditions; 3-g hook, 5-g cotton skein, 0.1% by weight aqueous solution at a temperature of 25° C.).

Amine oxides surfactants are also useful in the present invention as anti-redeposition surfactants include linear and branched compounds having the formula:

$$R^{3}(OR^{4})_{x}$$
  $N^{+}(R^{5})_{2}$ 

wherein R<sup>3</sup> is selected from an alkyl, hydroxyalkyl, acylamidopropoyl and alkyl phenyl group, or mixtures thereof, containing from 8 to 26 carbon atoms, preferably 8 to 18 carbon atoms; R<sup>4</sup> is an alkylene or hydroxyalkylene group containing from 2 to 3 carbon atoms, preferably 2 carbon atoms, or mixtures thereof; x is from 0 to 5, preferably from 0 to 3; and each R<sup>5</sup> is an alkyl or hydroxyalkyl group containing from 1 to 3, preferably from 1 to 2 carbon atoms, or a polyethylene oxide group containing from 1 to 3, preferable 1, ethylene oxide groups. The R<sup>5</sup> groups can be attached to each other, e.g., through an oxygen or nitrogen atom, to form a ring structure.

These amine oxide surfactants in particular include  $\rm C_{10}$ - $\rm C_{18}$  alkyl dimethyl amine oxides and  $\rm C_8$ - $\rm C_{18}$  alkoxy ethyl dihydroxyethyl amine oxides. Examples of such materials include dimethyloctylamine oxide, diethyldecylamine oxide, bis-(2-hydroxyethyl)dodecylamine oxide, dimethyldodecylamine oxide, dipropyltetradecylamine oxide, methylethylhexadecylamine oxide, dodecylamidopropyl dimethylamine oxide, cetyl dimethylamine oxide, stearyl dimethylamine oxide, tallow dimethylamine oxide and dimethyl-2-hydroxyoctadecylamine oxide. Preferred are  $\rm C_{10}$ - $\rm C_{18}$  alkyl dimethylamine oxide, and  $\rm C_{10-18}$  acylamido alkyl dimethylamine oxide.

Anti-redeposition agents and in particular non-ionic surfactants may be present in amounts from 0 to 10% by weight, preferably from 0.1% to 10%, and most preferably from 0.25% to 6%.

Sulfonated Polymer

The polymer, if used, is used in any suitable amount from about 0.1% to about 50%, preferably from 1% to about 20%, more preferably from 2% to 10% by weight of the composition. Sulfonated/carboxylated polymers are particularly suitable for the compositions contained in the pouch of the invention

Suitable sulfonated/carboxylated polymers described herein may have a weight average molecular weight of less than or equal to about 100,000 Da, or less than or equal to about 50,000 Da, or from about 3,000 Da to about 50,000, preferably from about 5,000 Da to about 45,000 Da.

As noted herein, the sulfonated/carboxylated polymers may comprise (a) at least one structural unit derived from at least one carboxylic acid monomer having the general formula (I):

wherein R<sup>1</sup> to R<sup>4</sup> are independently hydrogen, methyl, carboxylic acid group or CH<sub>2</sub>COOH and wherein the carboxylic acid groups can be neutralized; (b) optionally, one or more

structural units derived from at least one nonionic monomer having the general formula (II):

$$\begin{array}{c} \mathbb{R}^{5} \\ \downarrow \\ \mathbb{C} \\ \downarrow \\ \mathbb{X} \end{array}$$
 (II)

wherein  $R^5$  is hydrogen,  $C_1$  to  $C_6$  alkyl, or  $C_1$  to  $C_6$  hydroxyalkyl, and X is either aromatic (with  $R^5$  being hydrogen or methyl when X is aromatic) or X is of the general formula (III):

$$\begin{matrix} & & \\ & & \\ C = & O \\ & & \\ Y & & \\ R^6 & & \end{matrix}$$

wherein  $R^6$  is (independently of  $R^5$ ) hydrogen,  $C_1$  to  $C_6$  alkyl, or  $C_1$  to  $C_6$  hydroxyalkyl, and Y is O or N; and at least one structural unit derived from at least one sulfonic acid monomer having the general formula (IV):

$$\begin{array}{c} R^7 \\ \downarrow \\ (A)_t \\ \downarrow \\ (B)_t \\ \downarrow \\ SO_3^- \ M^+ \end{array}$$

wherein  $R^7$  is a group comprising at least one  $sp^2$  bond, A is O, N, P, S or an amido or ester linkage, B is a mono- or polycyclic aromatic group or an aliphatic group, each t is independently 0 or 1, and  $M^+$  is a cation. In one aspect,  $R^7$  is a  $C_2$  to  $C_6$  alkene. In another aspect,  $R^7$  is ethene, butene or propene.

Preferred carboxylic acid monomers include one or more of the following: acrylic acid, maleic acid, itaconic acid, methacrylic acid, or ethoxylate esters of acrylic acids, acrylic and methacrylic acids being more preferred. Preferred sulfonated monomers include one or more of the following: sodium(meth)allyl sulfonate, vinyl sulfonate, sodium phenyl (meth)allyl ether sulfonate, or 2-acrylamido-methyl propane sulfonic acid. Preferred non-ionic monomers include one or more of the following: methyl(meth)acrylate, ethyl(meth) acrylate, t-butyl(meth)acrylate, methyl(meth)acrylamide, ethyl(meth)acrylamide, t-butyl(meth)acrylamide, styrene, or  $\alpha$ -methyl styrene.

Preferably, the polymer comprises the following levels of monomers: from about 40 to about 90%, preferably from about 60 to about 90% by weight of the polymer of one or more carboxylic acid monomer; from about 5 to about 50%, preferably from about 10 to about 40% by weight of the polymer of one or more sulfonic acid monomer; and optionally from about 1% to about 30%, preferably from about 2 to about 20% by weight of the polymer of one or more non-ionic monomer. An especially preferred polymer comprises about 70% to about 80% by weight of the polymer of at least one carboxylic acid monomer and from about 20% to about 30% by weight of the polymer of at least one sulfonic acid monomer.

14

The carboxylic acid is preferably (meth)acrylic acid. The sulfonic acid monomer is preferably one of the following: 2-acrylamido methyl-1-propanesulfonic acid, 2-methacrylamido-2-methyl-1-propanesulfonic acid, 3-methacrylamido-2-hydroxypropanesulfonic acid, allysulfonic acid, methallysulfonic acid, allyloxybenzenesulfonic acid, methallyloxybenzensulfonic acid, 2-hydroxy-3-(2-propenyloxy)propanesulfonic acid, 2-methyl-2-propene-1-sulfonic acid, styrene sulfonic acid, vinylsulfonic acid, 3-sulfopropyl acrylate, 3-sulfopropyl methacrylate, sulfomethylacrylamid, sulfomethylmethacrylamide, and water soluble salts thereof. The unsaturated sulfonic acid monomer is most preferably 2-acrylamido-2-propanesulfonic acid (AMPS).

Preferred commercial available polymers include:
Alcosperse 240, Aquatreat AR 540 and Aquatreat MPS supplied by Alco Chemical; Acumer 3100, Acumer 2000, Acusol 587G and Acusol 588G supplied by Rohm & Haas; Goodrich K-798, K-775 and K-797 supplied by BF Goodrich; and ACP 1042 supplied by ISP technologies Inc. Particularly preferred polymers are Acusol 587G and Acusol 588G supplied by Rohm & Haas.

In the polymers, all or some of the carboxylic or sulfonic acid groups can be present in neutralized form, i.e. the acidic hydrogen atom of the carboxylic and/or sulfonic acid group in some or all acid groups can be replaced with metal ions, preferably alkali metal ions and in particular with sodium ions.

Cleaning Actives

Any traditional cleaning ingredients can be used as part of the compositions of invention. The levels given are weight per cent and refer to the total composition (excluding the water-soluble film in the case of enveloped composition executions). The detergent compositions can be built or unbuilt and comprise one or more detergent active components which may be selected from bleach, bleach activator, bleach catalyst, surfactants, alkalinity sources, enzymes, polymeric dispersants, anti-corrosion agents (e.g. sodium silicate) and care agents. Highly preferred detergent components include a builder compound, an alkalinity source, an anti-redeposition agent, a sulfonated polymer, an enzyme and an additional bleaching agent.

Builder

Builders suitable for use herein include builder which forms water-soluble hardness ion complexes (sequestering builder) such as citrates and polyphosphates e.g. sodium tripolyphosphate and sodium tripolyphosphate hexahydrate, potassium tripolyphosphate and mixed sodium and potassium tripolyphosphate salts and builder which forms hardness precipitates (precipitating builder) such as carbonates e.g. sodium carbonate.

Other suitable builders include amino acid based compound or a succinate based compound. The term "succinate based compound" and "succinic acid based compound" are used interchangeably herein. Examples of suitable amino acid based compounds include MGDA (methyl-glycine-diacetic acid), and salts and derivatives thereof and GLDA (glutamic-N,N-diacetic acid) and salts and derivatives thereof. GLDA (salts and derivatives thereof) is especially preferred according to the invention, with the tetrasodium salt thereof being especially preferred. Other suitable builders are described in U.S. Pat. No. 6,426,229. Particular suitable builders include; for example, aspartic acid-N-monoacetic acid (ASMA), aspartic acid-N,N-diacetic acid (ASDA), aspartic acid-N-monopropionic acid (ASMP), iminodisuccinic acid (IDA), N-(2-sulfomethyl)aspartic acid (SMAS), N-(2-sulfoethyl)aspartic acid (SEAS), N-(2-sulfomethyl) glutamic acid (SMGL), N-(2-sulfoethyl)glutamic acid (SEGL), N-methyliminodiacetic acid (MIDA),  $\alpha$ -alanine-N, N-diacetic acid (α-ALDA), serine-N,N-diacetic acid (SEDA), isoserine-N,N-diacetic acid (ISDA), phenylalanine-

N,N-diacetic acid (PHDA), anthranilic acid-N,N-diacetic acid (ANDA), sulfanilic acid-N,N-diacetic acid (SLDA), taurine-N,N-diacetic acid (TUDA) and sulfomethyl-N,N-diacetic acid (SMDA) and alkali metal salts or ammonium salts thereof.

Preferably the amino acid based compound or succinate based compound is present in the composition in an amount of at least 1 wt %, preferably at least 5 wt %, more preferably at least 10 wt %, and most preferably at least 20 wt %. Preferably these compounds are present in an amount of up to 10 50 wt %, preferably up to 45 wt %, more preferably up to 40 wt %, and most preferably up to 35 wt %. It is preferred that the composition contains 20% wt or less of phosphorous-containing ingredients, more preferably 10% wt or less, most preferably that they are substantially free of such ingredients 15 and even more preferably they are free of such ingredients.

Other builders include homopolymers and copolymers of polycarboxylic acids and their partially or completely neutralized salts, monomeric polycarboxylic acids and hydroxy-carboxylic acids and their salts. Preferred salts of the abovementioned compounds are the ammonium and/or alkali metal salts, i.e. the lithium, sodium, and potassium salts, and particularly preferred salts are the sodium salts.

Suitable polycarboxylic acids are acyclic, alicyclic, heterocyclic and aromatic carboxylic acids, in which case they 25 contain at least two carboxyl groups which are in each case separated from one another by, preferably, no more than two carbon atoms. Polycarboxylates which comprise two carboxyl groups include, for example, water-soluble salts of, malonic acid, (ethyl enedioxy) diacetic acid, maleic acid, diglycolic acid, tartaric acid, tartronic acid and fumaric acid. Polycarboxylates which contain three carboxyl groups include, for example, water-soluble citrate. Correspondingly, a suitable hydroxycarboxylic acid is, for example, citric acid. Another suitable polycarboxylic acid is the homopolymer of acrylic acid. Other suitable builders are disclosed in WO 95/01416, to the contents of which express reference is hereby made.

The builder is typically present at a level of from about 30 to about 80%, preferably from about 40 to about 70% by weight of composition. It is also preferred that the ratio of sequestering builder to precipitating builder is from about 10:1 to about 1:1, preferably from about 8:1 to 2:1. Silicates

Preferred silicates are sodium silicates such as sodium disilicate, sodium metasilicate and crystalline phyllosilicates. 45 Silicates if present are at a level of from about 1 to about 20%, preferably from about 5 to about 15% by weight of composition.

Bleach

Inorganic and organic bleaches are suitable cleaning 50 actives for use herein. Inorganic bleaches include perhydrate salts such as perborate, percarbonate, perphosphate, persulfate and persilicate salts. The inorganic perhydrate salts are normally the alkali metal salts. The inorganic perhydrate salt may be included as the crystalline solid without additional protection. Alternatively, the salt can be coated.

Alkali metal percarbonates, particularly sodium percarbonate are preferred perhydrates for use herein. The percarbonate is most preferably incorporated into the products in a coated form which provides in-product stability. A suitable coating material providing in product stability comprises mixed salt of a water-soluble alkali metal sulphate and carbonate. Such coatings together with coating processes have previously been described in GB-1,466,799. The weight ratio of the mixed salt coating material to percarbonate lies in the range from 1:200 to 1:4, more preferably from 1:99 to 19, and most preferably from 1:49 to 1:19. Preferably, the mixed salt is of sodium sulphate and sodium carbonate which has the

16

general formula Na2S04.n.Na2CO3 wherein n is from 0.1 to 3, preferably n is from 0.3 to 1.0 and most preferably n is from 0.2 to 0.5.

Another suitable coating material providing in product stability, comprises sodium silicate of Si02:Na20 ratio from 1.8:1 to 3.0:1, preferably L8:1 to 2.4:1, and/or sodium metasilicate, preferably applied at a level of from 2% to 10%, (normally from 3% to 5%) Of Si02 by weight of the inorganic perhydrate salt. Magnesium silicate can also be included in the coating. Coatings that contain silicate and borate salts or boric acids or other inorganics are also suitable.

Other coatings which contain waxes, oils, fatty soaps can also be used advantageously within the present invention.

Potassium peroxymonopersulfate is another inorganic perhydrate salt of utility herein.

Typical organic bleaches are organic peroxyacids including diacyl and tetraacylperoxides, especially diperoxydodecanedioc acid, diperoxytetradecanedioc acid, and diperoxyhexadecanedioc acid. Dibenzoyl peroxide is a preferred organic peroxyacid herein. Mono- and diperazelaic acid, mono- and diperbrassylic acid, and Nphthaloylaminoperoxicaproic acid are also suitable herein.

The diacyl peroxide, especially dibenzoyl peroxide, should preferably be present in the form of particles having a weight average diameter of from about 0.1 to about 100 microns, preferably from about 0.5 to about 30 microns, more preferably from about 1 to about 10 microns. Preferably, at least about 25%, more preferably at least about 50%, even more preferably at least about 75%, most preferably at least about 90%, of the particles are smaller than 10 microns, preferably smaller than 6 microns. Diacyl peroxides within the above particle size range have also been found to provide better stain removal especially from plastic dishware, while minimizing undesirable deposition and filming during use in automatic dishwashing machines, than larger diacyl peroxide particles. The preferred diacyl peroxide particle size thus allows the formulator to obtain good stain removal with a low level of diacyl peroxide, which reduces deposition and filming. Conversely, as diacyl peroxide particle size increases, more diacyl peroxide is needed for good stain removal, which increases deposition on surfaces encountered during the dishwashing process.

Further typical organic bleaches include the peroxy acids, particular examples being the alkylperoxy acids and the arylperoxy acids. Preferred representatives are (a) peroxybenzoic acid and its ring-substituted derivatives, such as alkylperoxybenzoic acids, but also peroxy-α-naphthoic acid and magnesium monoperphthalate, (b) the aliphatic or substituted aliphatic peroxy acids, such as peroxylauric acid, peroxystearic acid, ∈-phthalimidoperoxycaproic acid[phthaloiminoperoxyhexanoic acid (PAP)], o-carboxybenzamidoperoxycaproic acid, N-nonenylamidoperadipic acid N-nonenylamidopersuccinates, and (c) aliphatic araliphatic peroxydicarboxylic acids, such as 1,12-diperoxycarboxylic acid, 1,9-diperoxyazelaic acid, diperoxysebacic acid, diperoxybrassylic acid, the diperoxyphthalic acids, 2-decyldiperoxybutane-1,4-dioic acid, N,N-terephthaloyldi (6-aminopercaproic acid).

Bleach Activators

Bleach activators are typically organic peracid precursors that enhance the bleaching action in the course of cleaning at temperatures of 60° C. and below. Bleach activators suitable for use herein include compounds which, under perhydrolysis conditions, give aliphatic peroxoycarboxylic acids having preferably from 1 to 10 carbon atoms, in particular from 2 to 4 carbon atoms, and/or optionally substituted perbenzoic acid. Suitable substances bear O-acyl and/or N-acyl groups of the number of carbon atoms specified and/or optionally substituted benzoyl groups. Preference is given to polyacylated alkylenediamines, in particular tetraacetylethylenediamine

(TAED), acylated triazine derivatives, in particular 1,5-diacetyl-2,4-dioxohexahydro-1,3,5-triazine (DADHT), acylated glycolurils, in particular tetraacetylglycoluril (TAGU), N-acylimides, in particular N-nonanoylsuccinimide (NOSI), acylated phenolsulfonates, in particular n-nonanoyl- or 5 isononanoyloxybenzenesulfonate (n- or iso-NOBS), carboxylic anhydrides, in particular phthalic anhydride, acylated polyhydric alcohols, in particular triacetin, ethylene glycol diacetate and 2,5-diacetoxy-2,5-dihydrofuran and also triethylacetyl citrate (TEAC). Bleach activators if included in 10 the compositions of the invention are in a level of from about 0.1 to about 10%, preferably from about 0.5 to about 2% by weight of the composition.

Bleach Catalyst
Bleach catalysts preferred for use herein include the manganese triazacyclononane and related complexes (U.S. Pat. Nos. 4,246,612, 5,227,084); Co, Cu, Mn and Fe bispyridy-

lamine and related complexes (U.S. Pat. No. 5,114,611); and pentamine acetate cobalt(III) and related complexes (U.S. Pat. No. 4,810,410). A complete description of bleach catalysts suitable for use herein can be found in WO 99/06521, pages 34, line 26 to page 40, line 16. Bleach catalyst if included in the compositions of the invention are in a level of from about 0.1 to about 10%, preferably from about 0.5 to about 2% by weight of the composition.

Metal Care Agents

Metal care agents may prevent or reduce the tarnishing, corrosion or oxidation of metals, including aluminium, stainless steel and non-ferrous metals, such as silver and copper. Suitable examples include one or more of the following:

- (a) benzatriazoles, including benzotriazole or bis-benzotriazole and substituted derivatives thereof. Benzotriazole derivatives are those compounds in which the available substitution sites on the aromatic ring are partially or completely substituted. Suitable substituents include linear or branch-chain C<sub>1</sub>-C<sub>20</sub>-alkyl groups and hydroxyl, thio, phenyl or halogen such as fluorine, chlorine, bromine and iodine.
- (b) metal salts and complexes chosen from the group consisting of zinc, manganese, titanium, zirconium, hafnium, vanadium, cobalt, gallium and cerium salts and/or complexes, the metals being in one of the oxidation states II, III, IV, V or VI. In one aspect, suitable metal salts and/or metal complexes may be chosen from the group consisting of Mn(II) sulphate, Mn(II) citrate, Mn(II) stearate, Mn(II) acetylacetonate, K<sub>2</sub>TiF<sub>6</sub>, K<sub>2</sub>ZrF<sub>6</sub>, CoSO<sub>4</sub>, Co(NO<sub>3</sub>)<sub>2</sub> and Ce(NO<sub>3</sub>)<sub>3</sub>, zinc salts, for 45 example zinc sulphate, hydrozincite or zinc acetate;
- (c) silicates, including sodium or potassium silicate, sodium disilicate, sodium metasilicate, crystalline phyllosilicate and mixtures thereof.

Further suitable organic and inorganic redox-active substances that act as silver/copper corrosion inhibitors are disclosed in WO 94/26860 and WO 94/26859.

Preferably the composition of the invention comprises from 0.1 to 5% by weight of the composition of a metal care agent, preferably the metal care agent is a zinc salt. Unit Dose

Products in unit dose form include tablets, capsules, sachets, pouches, etc. Preferred for use herein are pouches, in particular multi-compartment pouches.

A multi-compartment pouch is formed by a plurality of water-soluble films which form a plurality of compartments. The pouch preferably comprises at least two side-by-side compartments superposed (i.e., placed above) onto another compartment. This disposition contributes to the compactness, robustness and strength of the pouch, additionally, it minimise the amount of water-soluble film required. It only requires three pieces of film to form three compartments. The robustness of the pouch allows also for the use of very thin

films without compromising the physical integrity of the pouch. The pouch is also very easy to use because the compartments do not need to be folded to be used in dispensers of fix geometry. At least two of the compartments of the pouch contain two different compositions. By "different compositions" herein is meant compositions that differ in at least one ingredient.

Preferably, at least one of the compartments contains a solid composition and another compartment a liquid composition, the compositions are preferably in a solid to liquid weight ratio of from about 20:1 to about 1:20, more preferably from about 18:1 to about 2:1 and even more preferably from about 15:1 to about 5:1. The pouch of the invention is very versatile because it can accommodate compositions having a broad spectrum of values of solid:liquid ratio. Particularly preferred have been found to be pouches having a high solid:liquid ratio because many of the detergent ingredients are most suitable for use in solid form, preferably in powder form. The ratio solid:liquid defined herein refers to the relationship between the weight of all the solid compositions and the weight of all the liquid compositions in the pouch.

In other embodiments the solid:liquid weight ratio is from about 2:1 to about 18:1, more preferably from about 5:1 to about 15:1. These weight ratios are suitable in cases in which most of the ingredients of the detergent are in liquid form.

In preferred embodiments the two side-by-side compartments contain liquid compositions, which can be the same but preferably are different and another compartment contains a solid composition, preferably in powder form, more preferably a densified powder. The solid composition contributes to the strength and robustness of the pouch. The liquid compositions contribute to the stability of the pouch, in particular if the solid composition comprises moisture sensitive ingredients (such as bleach). This is more so if the compartments superposed onto the solid-containing compartment cover completely the top surface (i.e. the common solid/liquid surface) of the solid-containing compartment.

For dispenser fit reasons the unit dose form products herein have a square or rectangular base and a height of from about 1 to about 5 cm, more preferably from about 1 to about 4 cm. Preferably the weight of the solid composition is from about 10 to about 22 grams, more preferably from about 15 to about 20 grams and the weight of the liquid compositions is from about 0.5 to about 4 grams, more preferably from about 0.8 to about 3 grams.

The multi-compartment pouch of the invention is very versatile in terms of dissolution profile. In preferred embodiments, at least two of the films which form different compartments have different solubility, under the same conditions, releasing the content of the compositions which they partially or totally envelope at different times. The term "solubility" as used herein is not intended to refer to total solubility of a film but to the point at which the pouch in the wash solution breaks to release its content.

Enzymes, and in particular amylases, can lose stability in product, due to its interaction with bleach and builders (they can destabilize the enzyme by binding to the calcium of the enzymes). In addition, the performance of enzymes in a cleaning solution can be impaired by the alkalinity of the solution, bleach, builders, etc. In preferred embodiments, one of the compositions of the multi-compartment pouch, preferably a solid composition, comprises bleach and another composition, preferably a composition in liquid form, comprises enzymes. It is also preferred that one of the films enclosing the enzyme-comprising composition dissolves prior to the films enclosing the bleach-containing composition during the main-wash cycle of an automatic dishwashing machine, thereby releasing the enzyme-containing composition into the wash liquor prior to the delivery of the bleach-containing composition. This gives the enzymes the possibility to oper-

ate under optimum condition, avoiding interactions with other detergent actives. The pouch provides excellent cleaning. It is preferred that the bleach-containing composition comprises also a builder.

Controlled release of the ingredients of a multi-compart- 5 ment pouch can be achieved by modifying the thickness of the film and/or the solubility of the film material. The solubility of the film material can be delayed by for example crosslinking the film as described in WO 02/102,955 at pages 17 and 18. Other water-soluble films designed for rinse release are described in U.S. Pat. Nos. 4,765,916 and 4,972,017. Waxy coating (see WO 95/29982) of films can help with rinse release. pH controlled release means are described in WO 04/111178, in particular amino-acetylated polysaccharide having selective degree of acetylation.

Other means of obtaining delayed release by multi-compartment pouches with different compartments, where the compartments are made of films having different solubility are taught in WO 02/08380.

Abbreviations Used in the Example

In the example, the abbreviated component identifications 20 have the following meanings:

Carbonate	Anhydrous sodium carbonate
STPP	Sodium tripolyphosphate anhydrous
Silicate	Amorphous Sodium Silicate (SiO <sub>2</sub> :Na <sub>2</sub> O = from 2:1 to 4:1)
Alcosperse 240-D	Sulfonated polymer available from Alco Chemical 95% solids
Percarbonate	Sodium percarbonate of the nominal formula 2Na <sub>2</sub> CO <sub>3</sub> •3H <sub>2</sub> O <sub>2</sub>
TAED	Tetraacetylethylenediamine
Detergency enzyme	available from Novozymes A/S
SLF18	Non-ionic surfactant available from BASF
Neodol 1-9	Non-ionic surfactant available from Shell
DPG	dipropylene glycol

In the following example all levels are quoted in per cent by weight of the composition (either solid or liquid composition).

## **EXAMPLE**

The compositions tabulated below are introduced into a 40 multi-compartment pouch having a first compartment comprising the solid composition (in powder form) and a liquid compartment superposed onto the powder compartment comprising the liquid compositions. The film used is Monosol M8630 film as supplied by Monosol. The weight of the solid composition is 17 grams and the weight of liquid compositions is 2.6 gram.

The pouch also comprises 0.3 mg of active low temperature amylase and 2 mg of active variant protease per gram of composition.

Ingredient	Level (% wt)	
Solid composition		
STPP Carbonate Silicate TAED Zinc carbonate SLF18	35 24 7 0.5 0.5	
Percarbonate Alcosperse 240D Processing aids Liquid composition	15 10 To balance	
DPG SLF18 Neodol 1-9	45 45 3	

20

#### -continued

Ingredient	Level (% wt)
Glycerine	2
Processing aids	To balance

The exemplified pouch is used to wash a soiled load as described herein below in an automatic dishwasher under the conditions described herein below. The washing items present excellent shine.

Substrates/Soils

Corning ware round casserole dish with egg.

- 1 part of butter with 50 cc of egg in microwave 41/2 minutes.
- 2 casserole dishes per run

Stainless steel pot

Painted with 10 grams of cooked and blended Kraft Macaroni and cheese

Baked in over for seven minutes

2 stainless steel pots per run

China Vertex plate

Painted with five grams of cooked and blended Minute

Dry overnight

2 plates per run

Black Ceramic Plates

Painted with 5 grams of a composite soil (TMD) comprising eggs, vegetables, meat, and cereals.

Allowed to dry over night

4 plates per run

TMD soil is made by J&R.

Stainless Steel Spatulas

Painted with five grams of TMD soil

Allowed to dry overnight

4 spatulas per run

35 Test Conditions:

Bank of four machines GE2600

City Water (8 gpg)

Four products

120° F. Inlet Water temperature

Normal cycle/heated dry

Substrates listed above are placed in the dishwasher

50 grams of the TMD soil is added when the main wash cup

The dimensions and values disclosed herein are not to be understood as being strictly limited to the exact numerical values recited. Instead, unless otherwise specified, each such dimension is intended to mean both the recited value and a functionally equivalent range surrounding that value. For example, a dimension disclosed as "40 mm" is intended to mean "about 40 mm".

Every document cited herein, including any cross referenced or related patent or application, is hereby incorporated herein by reference in its entirety unless expressly excluded or otherwise limited. The citation of any document is not an admission that it is prior art with respect to any invention disclosed or claimed herein or that it alone, or in any combination with any other reference or references, teaches, suggests or discloses any such invention. Further, to the extent that any meaning or definition of a term in this document conflicts with any meaning or definition of the same term in a document incorporated by reference, the meaning or definition assigned to that term in this document shall govern.

While particular embodiments of the present invention have been illustrated and described, it would be obvious to those skilled in the art that various other changes and modifications can be made without departing from the spirit and scope of the invention. It is therefore intended to cover in the appended claims all such changes and modifications that are within the scope of this invention.

#### SEQUENCE LISTING

<160> NUMBER OF SEQ ID NOS: 4 <210> SEQ ID NO 1 <211> LENGTH: 269 <212> TYPE: PRT <213 > ORGANISM: Bacillus Alkalophilus PB92 <400> SEQUENCE: 1 Ala Gln Ser Val Pro Trp Gly Ile Ser Arg Val Gln Ala Pro Ala Ala His Asn Arg Gly Leu Thr Gly Ser Gly Val Lys Val Ala Val Leu Asp Thr Gly Ile Ser Thr His Pro Asp Leu Asn Ile Arg Gly Gly Ala Ser 40 Phe Val Pro Gly Glu Pro Ser Thr Gln Asp Gly Asn Gly His Gly Thr 55 His Val Ala Gly Thr Ile Ala Ala Leu Asn Asn Ser Ile Gly Val Leu 70 Gly Val Ala Pro Asn Ala Glu Leu Tyr Ala Val Lys Val Leu Gly Ala Ser Gly Ser Gly Ser Val Ser Ser Ile Ala Gln Gly Leu Glu Trp Ala 105 Gly Asn Asn Gly Met His Val Ala Asn Leu Ser Leu Gly Ser Pro Ser 120 Pro Ser Ala Thr Leu Glu Gln Ala Val Asn Ser Ala Thr Ser Arg Gly 135 Val Leu Val Val Ala Ala Ser Gly Asn Ser Gly Ala Gly Ser Ile Ser Tyr Pro Ala Arg Tyr Ala Asn Ala Met Ala Val Gly Ala Thr Asp Gln 170 Asn Asn Asn Arg Ala Ser Phe Ser Gln Tyr Gly Ala Gly Leu Asp Ile Val Ala Pro Gly Val Asn Val Gln Ser Thr Tyr Pro Gly Ser Thr Tyr Ala Ser Leu Asn Gly Thr Ser Met Ala Thr Pro His Val Ala Gly Ala Ala Ala Leu Val Lys Gln Lys Asn Pro Ser Trp Ser Asn Val Gln Ile 230 235 Arg Asn His Leu Lys Asn Thr Ala Thr Ser Leu Gly Ser Thr Asn Leu 250 245 Tyr Gly Ser Gly Leu Val Asn Ala Glu Ala Ala Thr Arg 260 <210> SEO ID NO 2 <211> LENGTH: 485 <212> TYPE: PRT <213> ORGANISM: Bacillus AA560 <400> SEQUENCE: 2 His His Asn Gly Thr Asn Gly Thr Met Met Gln Tyr Phe Glu Trp Tyr 10 Leu Pro Asn Asp Gly Asn His Trp Asn Arg Leu Arg Ser Asp Ala Ser 25 Asn Leu Lys Asp Lys Gly Ile Ser Ala Val Trp Ile Pro Pro Ala Trp

Lys	Gly 50	Ala	Ser	Gln	Asn	Asp 55	Val	Gly	Tyr	Gly	Ala 60	Tyr	Asp	Leu	Tyr
Asp 65	Leu	Gly	Glu	Phe	Asn 70	Gln	Lys	Gly	Thr	Ile 75	Arg	Thr	Lys	Tyr	Gly 80
Thr	Arg	Asn	Gln	Leu 85	Gln	Ala	Ala	Val	Asn 90	Ala	Leu	Lys	Ser	Asn 95	Gly
Ile	Gln	Val	Tyr 100	Gly	Asp	Val	Val	Met 105	Asn	His	Lys	Gly	Gly 110	Ala	Asp
Ala	Thr	Glu 115	Met	Val	Arg	Ala	Val 120	Glu	Val	Asn	Pro	Asn 125	Asn	Arg	Asn
Gln	Glu 130	Val	Ser	Gly	Glu	Tyr 135	Thr	Ile	Glu	Ala	Trp 140	Thr	Lys	Phe	Asp
Phe 145	Pro	Gly	Arg	Gly	Asn 150	Thr	His	Ser	Asn	Phe 155	Lys	Trp	Arg	Trp	Tyr 160
His	Phe	Asp	Gly	Val 165	Asp	Trp	Asp	Gln	Ser 170	Arg	ГЛа	Leu	Asn	Asn 175	Arg
Ile	Tyr	ГЛа	Phe 180	Arg	Gly	Asp	Gly	Lys 185	Gly	Trp	Asp	Trp	Glu 190	Val	Asp
Thr	Glu	Asn 195	Gly	Asn	Tyr	Asp	Tyr 200	Leu	Met	Tyr	Ala	Asp 205	Ile	Asp	Met
Asp	His 210	Pro	Glu	Val	Val	Asn 215	Glu	Leu	Arg	Asn	Trp 220	Gly	Val	Trp	Tyr
Thr 225	Asn	Thr	Leu	Gly	Leu 230	Asp	Gly	Phe	Arg	Ile 235	Asp	Ala	Val	Lys	His 240
Ile	Lys	Tyr	Ser	Phe 245	Thr	Arg	Asp	Trp	Ile 250	Asn	His	Val	Arg	Ser 255	Ala
	-	-	260		Phe			265			_	-	270	_	
		275			Tyr		280					285			
	290				His	295					300				
305		-	-		Arg 310					315					320
				325	Val				330			_		335	
Glu	Glu	Ala	Leu 340	Glu	Ser	Phe	Val	Glu 345	Glu	Trp	Phe	Lys	Pro 350	Leu	Ala
		355			Thr		360					365			
Gly	Asp 370	Tyr	Tyr	Gly	Ile	Pro 375	Thr	His	Gly	Val	Pro 380	Ala	Met	Lys	Ser
385	Ile	Asp	Pro	Ile	Leu 390	Glu	Ala	Arg	Gln	395	Tyr	Ala	Tyr	Gly	Arg 400
Gln	Asn	Aap	Tyr	Leu 405	Asp	His	His	Asn	Ile 410	Ile	Gly	Trp	Thr	Arg 415	Glu
Gly	Asn	Thr	Ala 420	His	Pro	Asn	Ser	Gly 425	Leu	Ala	Thr	Ile	Met 430	Ser	Asp
Gly	Ala	Gly 435	Gly	Asn	Lys	Trp	Met 440	Phe	Val	Gly	Arg	Asn 445	ГЛЗ	Ala	Gly
Gln	Val 450	Trp	Thr	Asp	Ile	Thr 455	Gly	Asn	Arg	Ala	Gly 460	Thr	Val	Thr	Ile
Asn 465	Ala	Asp	Gly	Trp	Gly 470	Asn	Phe	Ser	Val	Asn 475	Gly	Gly	Ser	Val	Ser 480

-continued

								-	con	tin	ued	
Ile Trp	Val Asn	Lув 485										
<210> SE <211> LE <212> TY <213> OF	ENGTH: 4 PE: PRT		s Lic	heni:	Lform	nis						
<400> SE	EQUENCE :	3										
Ala Asn 1	Leu Asn	Gly Thr 5	Leu	Met	Gln	Tyr 10	Phe	Glu	Trp	Tyr	Met 15	Pro
Asn Asp	Gly Gln 20	His Trp	Arg	Arg	Leu 25	Gln	Asn	Asp	Ser	Ala 30	Tyr	Leu
Ala Glu	His Gly 35	Ile Thr	Ala	Val 40	Trp	Ile	Pro	Pro	Ala 45	Tyr	Lys	Gly
Thr Ser 50	Gln Ala	Asp Val	Gly 55	Tyr	Gly	Ala	Tyr	Asp	Leu	Tyr	Asp	Leu
Gly Glu 65	Phe His	Gln Lys 70	Gly	Thr	Val	Arg	Thr 75	Lys	Tyr	Gly	Thr	Lys 80
Gly Glu	Leu Gln	Ser Ala 85	Ile	ГÀа	Ser	Leu 90	His	Ser	Arg	Asp	Ile 95	Asn
Val Tyr	Gly Asp 100	Val Val	Ile	Asn	His 105	Lys	Gly	Gly	Ala	Asp 110	Ala	Thr
Glu Asp	Val Thr 115	Ala Val	Glu	Val 120	Asp	Pro	Ala	Asp	Arg 125	Asn	Arg	Val
Ile Ser 130	Gly Glu	His Arg	Ile 135	ГÀв	Ala	Trp	Thr	His 140	Phe	His	Phe	Pro
Gly Arg 145	Gly Ser	Thr Tyr 150	Ser	Asp	Phe	Lys	Trp 155	His	Trp	Tyr	His	Phe 160
Asp Gly	Thr Asp	Trp Asp 165	Glu	Ser	Arg	Lys 170	Leu	Asn	Arg	Ile	Tyr 175	Lys
Phe Gln	Gly Lys 180	Ala Trp	Asp	Trp	Glu 185	Val	Ser	Asn	Glu	Asn 190	Gly	Asn
Tyr Asp	Tyr Leu 195	Met Tyr	Ala	Asp 200	Ile	Asp	Tyr	Asp	His 205	Pro	Asp	Val
Ala Ala 210	Glu Ile	Lys Arg	Trp 215	Gly	Thr	Trp	Tyr	Ala 220	Asn	Glu	Leu	Gln
Leu Asp 225	Gly Phe	Arg Leu 230	Asp	Ala	Val	Lys	His 235	Ile	Lys	Phe	Ser	Phe 240
Leu Arg	Asp Trp	Val Asn 245	His	Val	Arg	Glu 250	Lys	Thr	Gly	Lys	Glu 255	Met
Phe Thr	Val Ala 260	Glu Tyr	Trp	Gln	Asn 265	Asp	Leu	Gly	Ala	Leu 270	Glu	Asn
Tyr Leu	Asn Lys 275	Thr Asn		Asn 280	His	Ser	Val	Phe	Asp 285	Val	Pro	Leu
His Tyr 290	Gln Phe	His Ala	Ala 295	Ser	Thr	Gln	Gly	Gly 300	Gly	Tyr	Asp	Met
Arg Lys 305	Leu Leu	Asn Ser 310	Thr	Val	Val	Ser	Lys 315	His	Pro	Leu	Lys	Ala 320
Val Thr	Phe Val	Asp Asn 325	His	Asp	Thr	Gln 330	Pro	Gly	Gln	Ser	Leu 335	Glu
Ser Thr	Val Gln 340	Thr Trp	Phe	Lys	Pro 345	Leu	Ala	Tyr	Ala	Phe 350	Ile	Leu
Thr Arg	Glu Ser 355	Gly Tyr		Gln 360	Val	Phe	Tyr	Gly	Asp 365	Met	Tyr	Gly

#### -continued

Thr Lys Gly Asp Ser Gln Arg Glu Ile Pro Ala Leu Lys His Lys Ile Glu Pro Ile Leu Lys Ala Arg Lys Gln Tyr Ala Tyr Gly Ala Gln His 395 Asp Tyr Phe Asp His His Asp Ile Val Gly Trp Thr Arg Glu Gly Asp Ser Ser Val Ala Asn Ser Gly Leu Ala Ala Leu Ile Thr Asp Gly Pro Gly Gly Ala Lys Arg Met Tyr Val Gly Arg Gln Asn Ala Gly Glu Thr 440 Trp His Asp Ile Thr Gly Asn Arg Ser Glu Pro Val Val Ile Asn Ser 455 Glu Gly Trp Gly Glu Phe His Val Asn Gly Gly Ser Val Ser Ile Tyr 470 Val Gln Arg <210> SEQ ID NO 4 <211> LENGTH: 485 <212> TYPE: PRT <213> ORGANISM: Bacillus sp. 707 <400> SEQUENCE: 4 His His Asn Gly Thr Asn Gly Thr Met Met Gln Tyr Phe Glu Trp Tyr Leu Pro Asn Asp Gly Asn His Trp Asn Arg Leu Asn Ser Asp Ala Ser Asn Leu Lys Ser Lys Gly Ile Thr Ala Val Trp Ile Pro Pro Ala Trp Lys Gly Ala Ser Gln Asn Asp Val Gly Tyr Gly Ala Tyr Asp Leu Tyr Asp Leu Gly Glu Phe Asn Gln Lys Gly Thr Val Arg Thr Lys Tyr Gly Thr Arg Ser Gln Leu Gln Ala Ala Val Thr Ser Leu Lys Asn Asn Gly Ile Gln Val Tyr Gly Asp Val Val Met Asn His Lys Gly Gly Ala Asp 105 Ala Thr Glu Met Val Arg Ala Val Glu Val Asn Pro Asn Asn Arg Asn 120 Gln Glu Val Thr Gly Glu Tyr Thr Ile Glu Ala Trp Thr Arg Phe Asp Phe Pro Gly Arg Gly Asn Thr His Ser Ser Phe Lys Trp Arg Trp Tyr 155 His Phe Asp Gly Val Asp Trp Asp Gln Ser Arg Arg Leu Asn Asn Arg 165 170 Ile Tyr Lys Phe Arg Gly His Gly Lys Ala Trp Asp Trp Glu Val Asp 180 185 Thr Glu Asn Gly Asn Tyr Asp Tyr Leu Met Tyr Ala Asp Ile Asp Met Asp His Pro Glu Val Val Asn Glu Leu Arg Asn Trp Gly Val Trp Tyr 215 Thr Asn Thr Leu Gly Leu Asp Gly Phe Arg Ile Asp Ala Val Lys His 230 235 Ile Lys Tyr Ser Phe Thr Arg Asp Trp Ile Asn His Val Arg Ser Ala 250

_															
Thr	Gly	Lys	Asn 260	Met	Phe	Ala	Val	Ala 265	Glu	Phe	Trp	Lys	Asn 270	Asp	Leu
Gly	Ala	Ile 275	Glu	Asn	Tyr	Leu	Gln 280	Lys	Thr	Asn	Trp	Asn 285	His	Ser	Val
Phe	Asp 290	Val	Pro	Leu	His	Tyr 295	Asn	Leu	Tyr	Asn	Ala 300	Ser	Lys	Ser	Gly
Gly 305	Asn	Tyr	Asp	Met	Arg 310	Asn	Ile	Phe	Asn	Gly 315	Thr	Val	Val	Gln	Arg 320
His	Pro	Ser	His	Ala 325	Val	Thr	Phe	Val	Asp 330	Asn	His	Asp	Ser	Gln 335	Pro
Glu	Glu	Ala	Leu 340	Glu	Ser	Phe	Val	Glu 345	Glu	Trp	Phe	ГÀа	Pro 350	Leu	Ala
Tyr	Ala	Leu 355	Thr	Leu	Thr	Arg	Glu 360	Gln	Gly	Tyr	Pro	Ser 365	Val	Phe	Tyr
Gly	Asp 370	Tyr	Tyr	Gly	Ile	Pro 375	Thr	His	Gly	Val	Pro 380	Ala	Met	Arg	Ser
385 Tys	Ile	Asp	Pro	Ile	Leu 390	Glu	Ala	Arg	Gln	395 Tàa	Tyr	Ala	Tyr	Gly	Lys 400
Gln	Asn	Asp	Tyr	Leu 405	Asp	His	His	Asn	Ile 410	Ile	Gly	Trp	Thr	Arg 415	Glu
Gly	Asn	Thr	Ala 420	His	Pro	Asn	Ser	Gly 425	Leu	Ala	Thr	Ile	Met 430	Ser	Asp
Gly	Ala	Gly 435	Gly	Ser	Lys	Trp	Met 440	Phe	Val	Gly	Arg	Asn 445	Lys	Ala	Gly
Gln	Val 450	Trp	Ser	Asp	Ile	Thr 455	Gly	Asn	Arg	Thr	Gly 460	Thr	Val	Thr	Ile
Asn 465	Ala	Asp	Gly	Trp	Gly 470	Asn	Phe	Ser	Val	Asn 475	Gly	Gly	Ser	Val	Ser 480
Ile	Trp	Val	Asn	Lys 485											

What is claimed is:

1. An automatic dishwashing detergent composition comprising:

- a. at least 0.1 mg of active protease per gram of composition, wherein the protease is a variant of a protease that has at least 70% identity with the amino acid sequence of SEQ ID NO:1. wherein said variant comprise variations in one or more of the following positions: 32, 33, 48-54, 58-62, 94-107, 116, 123-133, 150, 152-156, 158-161, 164, 169, 175-186, 197, 198, 203-216 as compared with the protease in SEQ ID NO:1; and
- b. at least 0.05 mg of an active low temperature amylase per gram of composition, wherein the low temperature amylase is a variant with:
  - a. one or more substitutions in the following positions versus SEQ ID NO: 2: 9, 26, 149, 182, 186, 202, 257, 295, 299, 323, 339 and 345; and, optionally, with one or more of the substitutions and/or deletions in the following positions: 118, 183, 184, 195, 320 and 458, which if present, comprise R118K, D183\*, G184\*, N195F, R320K and/or R458K; or
  - b. at least one substitution in the following positions versus SEQ ID NO:4: M202, M208, S255, R172, and/or M261.
- 2. An automatic dishwashing detergent composition according to claim 1 wherein the protease comprises the variations G116V+S126L+P127Q+S128A.
- 3. An automatic dishwashing detergent composition 65 according to claim 1 comprising at least 1.5 mg of active protease per gram of the composition.

4. An automatic dishwashing detergent composition according to claim 3 further comprising at least about 0.2 mg of low temperature amylase per gram of composition.

30

- 5. An automatic dishwashing detergent composition according to claim 1 further comprising a lipase.
- **6.** An automatic dishwashing detergent composition according to claim 1 further comprising an anti-redeposition agent and/or a sulfonated polymer.
- 7. An automatic dishwashing detergent composition according to claim 1 wherein the composition is in the form of a water-soluble pouch and wherein the weight of the composition is from about 10 to about 25 grams.
- 8. The automatic dishwashing detergent composition according to claim 7, wherein the pouch is a multi-compartment pouch having a plurality of films forming a plurality of compartments.
- 9. The automatic dishwashing detergent composition according to claim 8, wherein the pouch comprises two side-by-side compartments superposed onto another compartment and at least two different compartments, respectively comprising two different compositions.
- 10. A method of dishwashing in an automatic dishwashing machine using a detergent composition according to claim 1 wherein the composition is in unit dose form and the weight of the composition is from about 10 to about 25 grams comprising the steps of placing the pouch into the product dispenser and releasing it during the main-wash cycle.

\* \* \* \* \*