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(54) LOW MOLECULAR WEIGHT CATIONC LIPIDS FOR OLIGONUCLEOTIDE DELIVERY

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- (63) Continuation of application No. 14/681,641, filed on Apr. 8, 2015, now Pat. No. 9,458,087, which is a continuation of application No. 13/876,528, filed as application No. PCT/US2011/053556 on Sep. 28, 2011, now Pat. No. 9,029,604.
- (60) Provisional application No. 61/388,201, filed on Sep. 30, 2010.

 (52) **U.S. Cl.** CPC C12N 15/113 (2013.01); A61K 9/5123 (2013.01); C07C 217/08 (2013.01); C12N 15/88 (2013.01); C07C 2101/02 (2013.01); CI2N 2310/14 (2013.01); C12N 2320/32 (2013.01); Y10S 977/783 (2013.01); Y10S 977/916 (2013.01)

US 9,725,720 B2 (10) Patent No.:

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(58) Field of Classification Search None

See application file for complete search history.

(56) References Cited

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(57) ABSTRACT

The instant invention provides for novel cationic lipids that can be used in combination with other lipid components such as cholesterol and PEG-lipids to form lipid nanoparticles with oligonucleotides. It is an object of the instant invention to provide a cationic lipid scaffold that demon strates enhanced efficacy along with lower liver toxicity as a result of lower lipid levels in the liver. The present invention employs low molecular weight cationic lipids with one short lipid chain to enhance the efficiency and tolerabil ity of in vivo delivery of siRNA.

15 Claims, 2 Drawing Sheets

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LOW MOLECULAR WEIGHT CATIONIC LIPIDS FOR OLIGONUCLEOTIDE DELIVERY

This application is a Continuation of U.S. patent appli cation Ser. No. 14/681,641 filed Apr. 8, 2015, now U.S. Pat. No. 9,458,087 issued Oct. 4, 2016, which is a Continuation of U.S. patent application Ser. No. 13/876,528 filed Mar. 28, which is 371 National Phase Entry of International Patent Application No. PCT/US2011/53556 filed Sep. 28, 2011, and which claims benefit under 35 U.S.C. S119(e) of U.S. Provisional Application No. 61/388,201, filed Sep. 30, 2010, the contents of which are incorporated herein by reference in their entirety. 10 15

REFERENCE TO SEQUENCE LISTING SUBMITTED ELECTRONICALLY

The sequence listing of the present application has been submitted electronically via EFS-Web as an ASCII format- 20
ted sequence listing with a file name "MRLMIS00045WOPCTSEQ.txt", creation date of Sep. 29, 2016 and a size of 4,161 bytes. The sequence listing submitted via EFS-Web is part of the specification and is herein incorporated by reference in its entirety.

BACKGROUND OF THE INVENTION

The present invention relates to novel cationic lipids that can be used in combination with other lipid components such as cholesterol and PEG-lipids to form lipid nanoparticles with oligonucleotides, to facilitate the cellular uptake and endosomal escape, and to knockdown target mRNA 30

both in vitro and in vivo.
Cationic lipids and the use of cationic lipids in lipid Cationic lipids and the use of cationic lipids in lipid nanoparticles for the delivery of oligonucleotides, in par 35 ticular siRNA and miRNA, have been previously disclosed. Lipid nanoparticles and use of lipid nanoparticles for the delivery of oligonucleotides, in particular siRNA and miRNA, has been previously disclosed. Oligonucleotides miRNA, has been previously disclosed. Oligonucleotides
(including siRNA and miRNA) and the synthesis of oligo-
nucleotides has been previously disclosed. (See U.S. patent
applications: U.S. 2006/0083780, U.S. 2006/0240554, 2008/0020058, U.S. 2009/0263407 and U.S. 2009/0285881 and PCT patent applications: WO 2009/086558, WO2009/ 127060, WO2009/132131, WO2010/042877, WO2010/ 054384, WO2010/054401, WO2010/054405 and WO2010/ 054406). See also Semple S. C. et al., Rational design of cationic lipids for siRNA delivery, Nature Biotechnology, published online 17 Jan. 2010; doi:10.1038/nbt.1602.
Other cationic lipids are disclosed in U.S. patent appli-40

cations: U.S. 2009/0263407, U.S. 2009/0285881, U.S. 2010/ 0055168, U.S. 2010/0055169, U.S. 2010/0063135, U.S. 2010/0076055, U.S. 2010/0099738 and U.S. 2010/0104629.

Traditional cationic lipids such as CLinDMA and DLinDMA have been employed for siRNA delivery to liver but suffer from non-optimal delivery efficiency along with 55 liver toxicity at higher doses. It is an object of the instant invention to provide a cationic lipid scaffold that demon strates enhanced efficacy along with lower liver toxicity as a result of lower lipid levels in the liver. The present, a result of lower lipid levels in the liver. The present, invention employs low molecular weight cationic lipids with 60 one short lipid chain to enhance the efficiency and tolerabil ity of in vivo delivery of siRNA.

SUMMARY OF THE INVENTION

The instant invention provides for novel cationic lipids that can be used in combination with other lipid components such as cholesterol and PEG-lipids to form lipid nanopar ticles with oligonucleotides. It is an object of the instant invention to provide a cationic lipid scaffold that demon strates enhanced efficacy along with lower liver toxicity as a result of lower lipid levels in the liver. The present invention employs low molecular weight cationic lipids with one short lipid chain to enhance the efficiency and tolerabil ity of in vivo delivery of siRNA.

BRIEF DESCRIPTION OF THE FIGURES

FIG. 1: LNP (Compound 1) efficacy in mice.

FIG. 2. LNP (Compound 1) efficacy in rat (ApoB siRNA). FIG. 3. Cationic lipid (Compound 1) levels in rat liver.

DETAILED DESCRIPTION OF THE INVENTION

The various aspects and embodiments of the invention are directed to the utility of novel cationic lipids useful in lipid nanoparticles to deliver oligonucleotides, in particular, siRNA and miRNA, to any target gene. (See U.S. patent applications: U.S. 2006/0083780, U.S. 2006/0240554, U.S. 2008/0020058, U.S. 2009/0263407 and U.S. 2009/0285881 and PCI patent applications: WO 2009/086558, WO2009/ 127060, WO2009/132131, WO2010/042877, WO2010/ 054384, WO2010/054401, WO2010/054405 and WO2010/ 054406). See also Semple S. C. et al., Rational design of cationic lipids for siRNA delivery, Nature Biotechnology, published online 17 Jan. 2010; doi:10.1038/nbt. 1602.

The cationic lipids of the instant invention are useful components in a lipid nanoparticle for the delivery of oligonucleotides, specifically siRNA and miRNA.

In a first embodiment of this invention, the cationic lipids are illustrated by the Formula A:

45 wherein:

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 $R¹$ and $R²$ are independently selected from H, (C_1-C_6) alkyl, heterocyclyl, and polyamine, wherein said alkyl, heterocyclyl and polyamine are optionally substituted with one to three substituents selected from R', or $R¹$ and $R²$ can be taken together with the nitrogen to which they are attached, to form a monocyclic heterocycle with 4-7 mem bers optionally containing, in addition to the nitrogen, one or two additional heteroatoms selected from N, O and S, said monocyclic heterocycle is optionally substituted with one to three substituents selected from R';

 $R³$ is selected from H and (C_1-C_6) alkyl, said alkyl optionally substituted with one to three substituents selected from R';

R" is independently selected from halogen, R", OR", SR", CN, $CO₂R"$ and $CON(RH)₂$;

R" is independently selected from H and (C_1-C_6) alkyl, wherein said alkyl is optionally substituted with halogen and OH:

n is 0, 1, 2, 3, 4 or 5; and
 L_1 and L_2 are independently selected from C_3 - C_{24} alkyl and C_3-C_{24} alkenyl, said alkyl and alkenyl are optionally substituted with one or more substituents selected from R';

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or any pharmaceutically acceptable salt or stereoisomer thereof.

In a second embodiment, the invention features a com pound having Formula A, wherein:

 $R¹$ and $R²$ are each methyl:

 R^3 is H;

b is 0;

 L_1 is selected from C_3 - C_{24} alkyl and C_3 - C_{24} alkenyl; and

 L_2 is selected from C_3-C_9 alkyl and C_3-C_9 alkenyl;

or any pharmaceutically acceptable salt or Stereoisomer 10 thereof.

In a third embodiment, the invention features a compound having Formula A, wherein:

 $R¹$ and $R²$ are each methyl;

 R^3 is H;

n is 0;

 L_1 is selected from C_3-C_9 alkyl and C_3-C_9 alkenyl; and

 L_2 is selected from C_3 -C₂₄ alkyl and C₃-C₂₄ alkenyl;

or any pharmaceutically acceptable salt or stereoisomer thereof.

In a fourth embodiment, the invention features a com pound having Formula A, wherein:

 $R¹$ and $R²$ are each methyl;

 R^3 is H;

n is 1:

 L_1 is selected from C_3 - C_{24} alkyl and C_3 - C_{24} alkenyl; and

 L_2 is selected from C_3 - C_9 alkyl and C_3 - C_9 alkenyl; or any pharmaceutically acceptable salt or stereoisomer

thereof.

In a fifth embodiment, the invention features a compound having Formula A, wherein: 30

 $R¹$ and $R²$ are each methyl;

 R^3 is H;

- L_1 is selected from C_3 - C_{24} alkyl and C_3 - C_{24} alkenyl; and 35
- L_2 is selected from C_3 - C_9 alkyl and C_3 - C_9 alkenyl;
- or any pharmaceutically acceptable salt, or stereoisomer thereof.

Specific cationic lipids are:

- (2S)- N,N-dimethyl-1-(9Z,12Z)-octadeca-9,12-dien-1- yloxylundecan-2-amine (Compound 1);
- (2S)-1-(9Z,12Z)-octadeca-9,12-dien-1-yloxylundecan-2 amine (Compound 2);
(2S)-1-[(9Z,12Z)-octadeca-9,12-dien-1-yloxy]dodecan-2-
- amine (Compound 3); (2R)-1-(9Z,12Z)-octadeca-9,12-dien-1-yloxydodecan-2-
- amine (Compound 4); (2S)-1-(9Z,12Z)-octadeca-9,12-dien-1-yloxyldecan-2-
- amine (Compound 5);
(2S)-1-[(9Z,12Z)-octadeca-9,12-dien-1-yloxy]nonan-2-
- amine (Compound 6);
- (2S)- N,N-dimethyl-1-(9Z,12Z)-octadeca-9,12-dien-1 yloxyltridecan-2-amine (Compound 7);
- (2S)- N,N-dimethyl-1-(9Z,12Z)-octadeca-9,12-dien-1 yloxylnonan-2-amine (Compound 8):
- (2R)- N,N-dimethyl-1-(9Z,12Z)-octadeca-9,12-dien-1 yloxydodecan-2-amine (Compound 9);
- (2S)- N,N-dimethyl-1-(9Z,12Z)-octadeca-9,12-dien-1- yloxydodecan-2-amine (Compound 10);
- yloxy]decan-2-amine (Compound 11); and
- $(2S, 12Z, 15Z)$ $-M, N$ -dimethyl-1-(octyloxy)henicosa-12,15-dien-2-amine (Compound 12);
- dien-2-amine (Compound 12); (2R,12Z,15Z)-1-(decyloxy)-N,N-dimethylhenicosa-12, 15
- $(2R, 12Z, 15Z)$ -1-(hexyloxy)-N,N-dimethylhenicosa-12,15dien-2-amine (Compound 14);
- (2R,12Z,15Z)-1-(hexadecyloxy)-N,N-dimethylhenicosa-12, 15-dien-2-amine (Compound 15);
- $(2R, 12Z, 15Z)$ $-M$, N -dimethyl-1-(undecyloxy)henicosa-12,
- 15-dien-2-amine (Compound 16);
N,N-dimethyl-2-{[(9Z,12Z)-octadeca-9,12-dien-1-yloxy]
methyl}undecan-1-amine (Compound 17);
N,N-dimethyl-3-{[(9Z,12Z)-octadeca-9,12-dien-1-yloxy]
methyl}dodecan-1-amine (Compound 18); and
-
- (2S) N,N-dimethyl-1-({8-[(1R,2R)-2-{[(1S,2S)-2-pentyl-
cyclopropyl]methyl}cyclopropyl]octyl}oxy)tridecan-2amine (Compound 19);
- or any pharmaceutically acceptable salt or stereoisomer thereof.
- 15 useful in the preparation of lipid nanoparticles. In another embodiment, the cationic lipids disclosed are
	- In another embodiment, the cationic lipids disclosed are useful components in a lipid nanoparticle for the delivery of oligonucleotides.

In another embodiment, the cationic lipids disclosed are useful components in a lipid nanoparticle for the delivery of siRNA and miRNA.

In another embodiment, the cationic lipids disclosed are useful components in a lipid nanoparticle for the delivery of siRNA.

The cationic lipids of the present invention may have asymmetric centers, chiral axes, and chiral planes (as described in: EX. Eliel and S. H. Wilen, Stereochemistry of Carbon. Compounds, John Wiley & Sons, New York, 1994, pages 1119-1190), and occur as racemates, racemic mix isomers and mixtures thereof, including optical isomers, being included in the present invention, in addition, the cationic lipids disclosed herein may exist as tautomers and both tautomeric forms are intended to be encompassed by the scope of the invention, even though only one tautomeric structure is depicted.

45 carbon or on different carbons, so long as a stable structure It is understood that substituents and substitution patterns on the cationic lipids of the instant invention can be selected by one of ordinary skill in the art to provide cationic lipids that are chemically stable and that can be readily synthesized by techniques known in the art, as well as those methods set forth below, from readily available starting materials. If a substituent is itself substituted with more than one group, it is understood that these multiple groups may be on the same results.

50 that are chemically stable and that can be readily synthesized It is understood that one or more Si atoms can be incorporated into the cationic lipids of the instant invention by one of ordinary skill in the art to provide cationic lipids by techniques known in the art, from readily available starting materials.

55 60 compounds of Formula A. For example, different isotopic 65 In the compounds of Formula A, the atoms may exhibit their natural isotopic abundances, or one or more of the atoms may be artificially enriched in a particular isotope having the same atomic number, but an atomic mass or mass number different, from the atomic mass or mass number predominantly found in nature. The present invention is meant to include all suitable isotopic variations of the forms of hydrogen (H) include protium (^1H) and deuterium (^{2}H) . Protium is the predominant hydrogen isotope found in nature. Enriching for deuterium may afford certain thera peutic advantages. Such as increasing in vivo half-life or reducing dosage requirements, or may provide a compound, useful as a standard for characterization of biological samples. Isotopically-enriched compounds within Formula

n is 2:

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A can be prepared without undue experimentation by con ventional techniques well known to those skilled in the art or by processes analogous to those described in the Scheme and Examples herein using appropriate isotopically-en riched reagents and/or intermediates.

As used herein, "alkyl" means a straight chain, cyclic or branched saturated aliphatic hydrocarbon having the speci fied number of carbon atoms.

As used herein, "alkenyl' means a straight chain, cyclic or branched, unsaturated aliphatic hydrocarbon having the 10 specified number of carbon atoms including but not limited to diene, triene and tetraene unsaturated aliphatic hydrocar bons.

Examples of a cyclic "alkyl" or "alkenyl are:

As used herein, "heterocyclyl" or "heterocycle" means a
4- to 10-membered aromatic or nonaromatic heterocycle containing from 1 to 4 heteroatoms selected from the group consisting of O, N and S, and includes bicyclic groups. "Heterocyclyl" therefore includes, the following: benzoiim-35 dazolyl, benzofuranyl, benzofurazanyl, benzopyrazolyl, carbolinyl, cinnolinyl, furanyl, imidazolyl, indolinyl, indolyl, indolazinyl, indazolyl, isobenzofuranyl, isoindolyl, iso quinolyl, isothiazolyl, isoxazolyl, naphthpyridinyl, oxadiaz- 40 olyl, oxazolyl, oxazoline, isoxazoline, oxetanyl, pyranyl. pyrazinyl, pyrazolyl pyridazinyl, pyridopyridinyl, pyridazi nyl, pyridyl, pyrimidyl, pyrrolyl, quinazolinyl, quinolyl, quinoxalinyl, tetrahydropyranyl, tetrazolyl, tetraZolopyridyl, thiadiazolyl, thiazolyl, thienyl, triazolyl azetidinyl, 1,4 dioxanyl, hexahydroazepinyl, piperazinyl, piperidinyl, pyr rolidinyl, morpholinyl, thiomorpholinyl, dihydrobenzoimi dazolyl, dihydrobenzofuranyl, dihydrobenzothiophenyl, dihydrobenzoxazolyl, dihydrofuranyl, dihydroiimdazolyl, dihydroindolyl, dihydroisooxazolyl, dihydroisothiazolyl, 50 dihydrooxadiazolyl, dihydrooxazolyl, dihydropyrazinyl, dihydropyrazolyl, dihydropyridinyl, dihydropyrimidinyl, dihydropyrrolyl, dihydroquinolinyl, dihydroietrazolyl, dihy drothiadiazolyl, dihydrothiaxolyl, dihydrothienyl, dihydro triazolyl, dihydroazetidinyl, methylenedioxybenzoyl, tetra- 55 hydrofuranyl, and tetrahydrothienyl, and N-oxides thereof all of which are optionally substituted with one to three substituents selected from R".

As used herein, "polyamine" means compounds having two or more amino groups. Examples include putrescine, cadaverine, spermidine, and spermine. 60

As used herein, "halogen" means Br, Cl, F and I.
In an embodiment of Formula A, $R¹$ and $R²$ are independently selected from H and (C_1-C_6) alkyl, wherein said alkyl is optionally substituted with one to three substituents 65 selected from R', or R^1 and R^2 can be taken together with the nitrogen to which they are attached to form a monocyclic

heterocycle with 4-7 members optionally containing, in addition to the nitrogen, one or two additional heteroatoms selected from N, O and S, said monocyclic heterocycle is optionally substituted with one to three substituents selected from R'.
In an embodiment of Formula A. $R¹$ and $R²$ are indepen-

dently selected from H, methyl, ethyl and propyl, wherein said methyl, ethyl and propyl are optionally substituted with one to three substituents selected from R', or $R¹$ and $R²$ can be taken together with the nitrogen to which they are attached to form a monocyclic heterocycle with 4-7 mem bers optionally containing, in addition to the nitrogen, one or two additional heteroatoms selected from N, O and S, said monocyclic heterocycle is optionally substituted with one to three substituents selected from R'.
In an embodiment of Formula A, R^1 and R^2 are indepen-

dently selected from H, methyl, ethyl and propyl.

In an embodiment of Formula A, R^1 and R^2 are each methyl.

In an embodiment of Formula A, $R³$ is selected from H and methyl.

In an embodiment of Formula A, $R³$ is H.

In an embodiment of Formula A, R' is R".

In an embodiment of Formula A, R" is independently selected from H. methyl, ethyl and propyl, wherein said methyl, ethyl and propyl are optionally substituted with one

or more halogen and OH.
In an embodiment of Formula A, R" is independently selected from H, methyl, ethyl and propyl.

In an embodiment of Formula A, n is 0, 1 or 2.

In an embodiment of Formula A, n is 0 or 1.

In an embodiment of Formula A, n is 0.

In an embodiment of Formula A, L_1 is selected from C_3-C_{24} alkyl and C_3-C_{24} alkenyl, which are optionally substituted with halogen and OH.

In an embodiment of Formula A, L_1 is selected from C_3-C_{24} alkyl and C_3-C_{24} alkenyl.

In an embodiment of Formula A, L_1 is selected from C_3 - C_{24} alkenyl.

In an embodiment of Formula A, L_1 is selected from C_{12} -C₂₄ alkenyl.

In an embodiment of Formula A, L_1 is C_{18} alkenyl. In an embodiment of Formula A, L_1 is:

In an embodiment of Formula A, L_1 is C_8 alkyl.

In an embodiment of Formula A, L_2 is selected from C_3-C_{24} alkyl and C_3-C_{24} alkenyl, which are optionally substituted with halogen and OH.

In an embodiment of Formula A, $L₂$ is selected from C_3-C_{24} alkyl and C_3-C_{24} alkenyl.

In an embodiment of Formula A, L_2 is selected from C_3-C_{24} alkenyl.

In an embodiment of Formula A, L_2 is selected from C_{12} -C₂₄ alkenyl.

In an embodiment of Formula A, L_2 is C_{19} alkenyl. In an embodiment of Formula A, L_2 is:

In an embodiment of Formula A, L_2 is selected from C_3 - C_9 alkyl and C_3 - C_9 alkenyl, which are optionally substituted with halogen and OH.

In an embodiment of Formula A, L_2 is selected from C_5 -C₉ alkyl and C_5 -C₉ alkenyl, which are optionally substi-5 tuted with halogen and OH.

In an embodiment of Formula A, L_2 is selected from C_7 - C_9 alkyl and C_7 - C_9 alkenyl, which are optionally substituted with halogen and OH.

In an embodiment of Formula A , L_2 is selected from 10 C_3 - C_9 alkyl and C_3 - C_9 alkenyl.

In an embodiment of Formula A, L_2 is selected from C_5 - C_9 alkyl and C_5 - C_9 alkenyl.

In an embodiment of Formula A, L_2 is selected from C_7 - C_9 alkyl and C_7 - C_9 alkenyl.

In an embodiment of Formula A, L_2 is C_3-C_9 alkyl.

In an embodiment of Formula A, L_2 is C_5-C_9 alkyl.

In an embodiment of Formula A, L_2 is C_7 -C₉ alkyl.

In an embodiment of Formula A, L_2 is C_9 alkyl.
In an embodiment of Formula A, "heterocyclyl" is pyro- 20

lidine, piperidine, morpholine, imidazole or piperazine.

In an embodiment of Formula A, "monocyclic heterocy-clyl" is pyrolidine, piperidine, morpholine, imidazole or piperazine.

In an embodiment of Formula A. "polyamine' is 25 putrescine, cadaverine, spermidine or spermine.

In an embodiment, "alkyl" is a straight chain saturated aliphatic hydrocarbon having the specified number of carbon atOmS.

In an embodiment, "alkenyl" is a straight chain unsatu- 30 rated aliphatic hydrocarbon having the specified number of carbon atoms.

Included in the instant invention is the free form of cationic lipids of Formula A, as well as the pharmaceutically acceptable salts and stereoisomers thereof. Some of the 35 isolated specific cationic lipids exemplified herein, are the protonated salts of amine cationic lipids. The terra "free form" refers to the amine cationic lipids in non-salt form. The encompassed pharmaceutically acceptable salts not only include the isolated saits exemplified for the specific 40 cationic lipids described herein, but also all the typical pharmaceutically acceptable salts of the free form of cat ionic lipids of Formula A. The free form of the specific salt cationic lipids described may be isolated using techniques known in the art. For example, the free form may be 45 regenerated by treating the salt with a suitable dilute aque ous base solution Such as dilute aqueous NaOH, potassium carbonate, ammonia and Sodium bicarbonate. The free forms may differ from their respective salt forms somewhat, in certain physical properties, such as solubility in polar sol- 50 vents, but the acid and base salts are otherwise pharmaceu tically equivalent to their respective free forms for purposes of the invention.

The pharmaceutically acceptable salts of the instant cat ionic lipids can be synthesized from the cationic lipids of 55 this invention which contain a basic or acidic moiety by conventional chemical methods. Generally, the salts of the basic cationic lipids are prepared either by ion exchange chromatography or by reacting the free base with stoichioinorganic or organic acid in a suitable solvent or various combinations of solvents. Similarly, the salts of the acidic compounds are formed by reactions with the appropriate inorganic or organic base. metric amounts or with an excess of the desired salt-forming 60

Thus, pharmaceutically acceptable salts of the cationic 65 lipids of this invention include the conventional non-toxic salts of the cationic lipids of this invention as formed by

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reacting a basic instant cationic lipids with an inorganic or organic acid. For example, conventional non-toxic salts chloric, hydrobromic, sulfuric, sulfamic, phosphoric, nitric and the like, as well as salts prepared from organic acids such as acetic, propionic, succinic, glycolic, stearic, lactic, malic, tartaric, citric, ascorbic, pamoic, maleic, hydroxymaleic, phenylacetic, glutamic, benzoic, salicylic, sulfanilic, 2-acetoxy-benzoic, fumaric, toluenesulfonic, methanesulfo nic, ethane disulfonic, oxalic, isethionic, trifluoroacetic (TFA) and the like.

When the cationic lipids of the present invention are acidic, suitable "pharmaceutically acceptable salts" refers to salts prepared form pharmaceutically acceptable non-toxic bases including inorganic bases and organic bases. Salts derived from inorganic bases include aluminum, ammonium, calcium, copper, ferric, ferrous, lithium, magnesium, manganic salts, manganous, potassium, sodium, zinc and the like. Particularly preferred axe the ammonium, calcium, magnesium, potassium and sodium salts. Salts derived from pharmaceutically acceptable organic non-toxic bases include salts of primary, secondary and tertiary amines, substituted amines including naturally occurring substituted amines, cyclic amines and basic ion exchange resins, such as arginine, betaine caffeine, choline, N,N¹-dibenzylethylene-
diamine, diethyiamin, 2-diethylaminoethanol, 2-dimethylaminoethanol, ethanolamine, ethylenediamine, N-ethylmorpholine, N-ethylpiperidine, glucamine, glucosamine, histidine, hydrabamine, isopropylamine, lysine, methylglu-camine, morpholine, piperazine, piperidine, polyamine res ins, procaine, purines, theobromine, triethylamine, trimeth ylamine tripropylamine, tromethamine and the like.

The preparation of the pharmaceutically acceptable salts described above and other typical pharmaceutically accept able salts is more fully described by Berg et al., "Pharma ceutical Salts," *J. Pharm. Sci.*, 1977:66:1-19.
It will also be noted that the cationic lipids of the present

invention are potentially internal salts or zwitterions, since under physiological conditions a deprotonated acidic moiety in the compound, such as a carboxyl group, may be anionic, and this electronic charge might then be balanced off Internally against the cationic charge of a protonated or alkylated basic moiety, such as a quaternary nitrogen atom.

EXAMPLES

Examples provided are intended to assist in a further understanding of the invention. Particular materials employed, species and conditions are intended to be further illustrative of the invention and not limitative of the reason able scope thereof. The reagents utilized in synthesizing the cationic lipids are either commercially available or are

readily prepared by one of ordinary skill in the art.
Synthesis of the novel cationic lipids is a linear process starting from epichlorohydrin (i) (General Scheme 1). Epoxide opening, ring closure with lipid alkoxide delivers epoxy ether intermediate ii. Original addition to the epoxide pro vides secondary alcohol intermediate iii. Mitsinobu inver sion with azide followed by reduction yields primary amine intermediates V. Reductive animation provides the tertiary amine derivatives vi.

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An alternative synthesis of the novel cationic lipids start ing from epichlorohydrin (i) is depicted in General Scheme 2. Epoxide opening, ring closure with lipid Grignard deliv ide provides secondary alcohol intermediate iii. Mitsinobu inversion with azide followed by reduction yields primary $_{30}$ amine intermediates V. Reductive animation provides the tertiary amine derivatives vi. 25

GENERAL SCHEME 2 Ω Ω \overline{C} $\overline{L_1 \cup R}$ $\overline{L_1 \cup R}$ $\overline{L_1 \cup R}$ THE L_2 SnCl₄ i vii. $\frac{\text{DTBAD, TPP}}{\text{DPPA}}$ H_0 H_1 DPPA L_2 iii. N_3 \overline{O} \overline{C} $\overline{CPP,H_2O}$ L_2 iv 11 RCHO, NaBH(OAc) ₃ L_2 \mathbf{v} R¹ $e^{2\lambda N}$ \sim $e^{\lambda L_1}$ L2 vi

Synthesis of doubly homologated cationic lipids xiii begins with ketone vii. Peterson olefination generates the unsaturated amide Xi. Conjugate reduction with L-Selectride gives amide xii. Amide reduction with lithium aluminum hydride gives cationic lipids xiii.

Synthesis of the homologated cationic lipids X (General Scheme 3) begins with oxidation of intermediate iii to ketone vii using Dess-Martin Periodinane. Conversion of the ketone to the nitrile viii is accomplished with TOSMIC. gives primary amine ix. Reductive animation provides cationic lipids X. Reduction of the nitrite with lithium aluminum hydride 65 (2S) N,N-dimethyl-1-(9Z,12Z)-octadeca-9,12 dien-1-yloxylundecan-2-amine (Compound 1)

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Triphenyl phosphine (14.4g, 55 mmol) was dissolved in THF and cooled to 0°C. under nitrogen. Di-tertbutyl azodi

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A 250 ml, rb flask was charged with magnetic stirbar, tetrabutyl ammonium bromide (TBAB, 2.72 g, 8.4 mmol), linoleyl alcohol (225 g, 884 mmol), and sodium hydroxide (50.7 g, 1.2 mol), then cooled in an ice bath. The (S)epichlorohydrin (156 g, 1.69 mol) was added slowly over 2 hours and then warmed to ambient temperature and stirred overnight, 259 mL of hexane was added and allowed to stir for 15 mins, then mixture was filtered and organic layer was concentrated in vacuo. The product was purified using 0-10% ethyl acetate/hexane gradient on 330 g silica column to give $(2R)$ -2- $\{[(9Z,12Z)$ -octadeca-9,12-dien-1-yloxyl] methyl}oxirane. ¹H NMR (CDCl₃, 300 mHz) δ 0.90-0.86 $(m, 3H), 1.29$ (s, 16H), 1.55-1.64 (m, 2H), 2.00-2.07 (m, ²⁵) 4H), 2.58-2.61 (m, 1H), 2.74-2.80 (m, 3H), 3.12-3.15 (m, 1H)3.34-3.52 (m, 3H), 3.67-3.72 (dd, J=12 Hz, 1H) 5.30 5.35 (m, 4H); HRMS (m+1) calc'd 323.2872. found 323.2951.

carboxylate (13.7 g, 59.5 mmol) was added slowly and the reaction was stirred for 30 mins. Then the alcohol $(20 \text{ g}, 45.8)$ mmol) was added dropwise and allowed to stir for 10 mins, then diphenyl phosphorylazide (15.1 g, 55 mmol) was added and allowed to stir overnight, warming to ambient tempera ture. The reaction was evaporated to dryness in vacuo and directly loaded onto a silica gel column and eluted with 0-10% ethyl acetate/hexane gradient to provide (2S)-2 azidoundecyl (9Z,12Z)-octadeca-9,12-dien-1-yl ether which was carried directly into the next reaction without charac terization.

The epoxide $(15 \text{ g}, 46.5 \text{ mmol})$ was dissolved in THF and $_{45}$ cooled to 0° C. under stream of Nitrogen. Octyl Grignard (25.6 mL 2M solution, 51.2 mmol) was added dropwise and then heated in microwave at 120° C. for one hour. The precipitate was filtered off and the solvent evaporated in column and eluted with 0-10% gradient (hexane-ethyl acetate) to give (2R)-1-[(9Z,12Z)-octadeca-9,12-dien-1yloxylundecan-2-ol. LC/MS (m+1)=437.6. vacuo. The crude oil was directly loaded onto a silica gel $_{50}$

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Triphenyl phosphine (4.54g, 17.3 mmol) and the azide (8 g, 17.3 mmol) were dissolved in THF. The reaction mixture was split into 3 µw tubes and Irradiated at 120° C. for 1 hour each. Considerable pressure built in each tube so care should be noted. LC indicated 100% conversion to phosphoimine intermediate. To each, tube was added ~3 mL of water and the reaction irradiated for 10 min at 120° C. The reaction mixtures were combined and concentrated to remove organic solvent. Hexane was added to precipitate phosphine oxides which were filtered through sintered glass funnel. The solvent was then removed in vacuo. The crude product was purified using HPLC with 30 min run and 60-100% water/acetonitrile gradient. The combined HPLC fractions were neutralized with sodium bicarbonate evaporated in vacuo. The pure product was partitioned between water/ hexanes. The organic layer was dried over sodium sulfate, filtered and evaporated in vacuo to afford $(2S)$ -1-[$(9Z, 12Z)$ octadeca-9,12-dien-1-yloxylundecan-2-amine (2), 1 H NMR (CDC1,300 mHz) & 0.88-0.87(m, 6H), 1.25-129 (s.32H), 1.54-1.54 (m, 2H), 2.03-2.05 (m, 4H), 2.23 (s. 2H), 2.75 2.76 (m, 2H), 2.96 (m, 1H), 3.13–3.18 (m. 1H), 3.38-3.45 (m, 3H), 5.31-5.38 (m, 4H); LC/MS (m+1)=436.7.

by triacetoxy borohydride (5.1 g, 24.1 mmol). The reaction was stirred at ambient temperature for 15 mins. LC/MS indicated 100% conversion to product. Added 1M NaOH until basic and extracted with hexane and washed with water. Retained organic layer and removed solvent in vacuo. Purified using 60-100% water/acetonitrile 30 min gradient on C8 HPLC. Combined fractions and added sodium bicar bonate and evaporated organics in vacuo. The product, was partitioned between water/hexanes and the organics were dried over sodium sulfate, filtered and evaporated in vacuo to deliver $(2S)$ —N,N-dimethyl-1- $[(9Z, 12Z)$ -octadeca-9,12dien-1-yloxylundecan-2-amine (1) . ¹H NMR (CDCl₃, 300) mHz) δ 0.88-0.87 (m, 6H), 1.285 (s, 33H), 1.55 (m, 2H), 1.80 (m, 1H), 2.00-2.05 (s, 4H), 2.29-2.31 (2H), 2.50 (m, 1H), 2.76-1.77 (m, 2H), 3.36-3.51 (m, 6H), 5.34-5.36 (m, 4H); LC/MS (m+1)=464.9.

The primary amine (3.5g, 8 mmol) was dissolved in THF and formaldehyde (3.26 g. 40.2 mmol) was added, followed

Compounds 3-11 are novel cationic lipids and were prepared according to General Scheme 1 above.

intermediate chloro-alcohol was purified via flash chroma tography (silica, 0-35% ethyl acetate/hexanes). The alcohol

A round bottomed flask was charged with magnetic stir bar, copper cyanide (1.45 g, 16.2 mmol), epichlorohydrin 60 (15 g, 162 mmol) and purged with nitrogen. THF was added, the solution cooled to -78° C. and linoleyl Grignard (68.8 g, 195 mmol) was added slowly. After addition of Grignard the reaction was allowed to warm to ambient temperature. The reaction was quenched with Saturated ammonium chloride 65 solution and extracted with ether. The organics were dried over sodium sulfate, filtered and evaporated in vacuo. The

was dissolved in THF and allowed to stir with solid NaOH pellets at ambient temperature for 16 hours, then filtered off NaOH and washed organic layer with water. The organics were dried over sodium, sulfate, filtered and evaporated in vacuo to provide $(2S)-2-[(10Z,13Z)-nonadeca-10,13-dien-10]$ 1-yl]oxirane. ¹H NMR (CDC1₃, 300 mHz) δ 0.87-0.90 (m, 3H), 1.27-1.52 (m, 22H), 2.01-2.19 (m, 4H), 2.40-2.46 (m, 1H), 2.71-2.76 (m, 3H), 2.89-2.91 (m. 1H), 5.30-5.36 (m, 4H); LC/MS (m+H+acetonitrile)=349.5.

and cooled to 0°C. To this solution was added tin chloride 15 flash chromatography (0-20% ethyl acetate/hexanes) to give (1.63 mmol, 1.63 mL of a 1M solution). The epoxide (5 g, (2R 127 157) 1 (octyloxy)henicose 12.15 die 16.3 mmol) was added to the reaction mixture dropwise and $(2R,12Z,15Z)$ -1-(octyloxy)henicosa-12,15-dien-2-ol.
the reaction was aged for 1 hour at 0° C. The reaction was LC/MS (m+H)=437.6. the reaction was aged for 1 hour at 0° C. The reaction was

The alcohol (2.55 g, 19.6 mmol) was dissolved in DCM evaporated in vacuo, dissolved in hexanes and purified by

The alcohol was carried on to final Compound 12 as 30 described for Compound 1. $\rm{^1H}$ NMR (CDCl₃, 300 mHz) δ 0.85 – 0.091 (m, 0 H), 1.272 (s, 34 H), 1.40 (m, 1 H), 1.57 (m, 1H), 1.65 (s, 4H), 2.01-2.08 (3H), 2.30 (m, 6H), 2.52 (m, 1H), 2.75-2.79 (m, 2H), 3.29-3.4 (m, 2H), 3.46-3.51 (dd. Compounds 13-16 are novel cationic lipids and were prepared according to General Scheme 2 above.

19 N,N-dimethyl-2-(9Z,12Z)-octadeca-9,12-dien-1 yloxy]methyl}undecan-1-amine (Compound 17)

To a solution of alcohol iii (15g, 34.3 mmol) in dichlo romethane (50 mL) was added Dess-Martin Periodinane (14.6 g., 343 mmol) and the reaction was stirred at ambient temperature for 16 hours. The solids were filtered and the filtrate partitioned between water/DCM. The organics were ²⁰ dried over sodium sulfate, filtered and evaporated, in vacuo. Purification by flash chromatography (silica, 0-15% ethyl acetate/hexanes) gave ketone vii. LC/MS (M+H)=435.6.

To a solution of ketone vii (10g, 23.0 mmol) in DME (40 mL) was added TOSMIC (5.8 g. 29.9 mmol) and the solution was cooled to 0° C. To the cooled solution was $_{40}$ added potassium tert-butoxide (46 mmol. 46 mL of a 1M solution in tBuOH) dropwise. After 30 minutes the reaction was partitioned between hexanes and water. The organics were dried over sodium sulfate, filtered and evaporated in vacuo. Purification by flash chromatography (silica, 0-10% ethyl acetate/hexanes) gave nitrile viii. LC/MS (M+H) $=446.6.$ 45

To a solution of nitrile viii (4.6 g. 10.4 mmol) in THF (25 mL) was added lithium aluminum hydride (0.8 g. 20.7 mmol) at ambient temperature. The reaction was quenched with sodium sulfate decahydrate solution and the solids were filtered. The filtrate was dried over sodium sulfate, filtered 65 and evaporated in vacuo to give crude amine ix which was carried directly into next reaction, LC/MS (M+H)=450.6. 60

A solution of amine ix $(4.7 \text{ g}, 10.3 \text{ mmol})$ and formalde-
rde $(2.5 \text{ g}, 31.1 \text{ mmol})$ in THF (25 mL) was treated with 15 hyde (2.5 g. 31.1 mmol) in THF (25 mL) was treated with sodium, triacetoxyborohydride (6.6%, 31.1 mmol) at ambi ent temperature. After aging for 15 minutes, the reaction was quenched with 1M sodium hydroxide and partitioned between water and hexanes. The organics were dried over sodium sulfate, filtered and evaporated in vacuo. Purification ²⁰ by preparative reverse phase chromatography (C8 column,

acetonitrile/water gradient) gave compound 17. LC/MS $(M+H)=479.6.$ ¹H NMR (CDCl₃, 400 mHz) δ 5.36 (m, 4H), 3.38 (m, 3H), 3.26 (m, 1H), 2.75 (t, J=6.4 Hz, 2H), 2.22 (m, 1H), 2.19 (s, 6H), 2.04 (m, 5H), 1.71 (m, 1H), 1.54 (m, 2H), 1.28 (m, 32H), 0.83 (m, 6H).

N,N-dimethyl-3-(9Z,12Z)-octadeca-9,12-dien-1 yloxy]methyl}dodecan-1-amine (Compound 18)

45 A solution of silyl amide $(12.4 \text{ g}, 78 \text{ mmol})$ in THF (50 m) mL) was cooled to -78° C. and treated with nBuLi (62.4 mmol, 25 mL of a 2.5M solution) and aged for 10 minutes.
To this solution was transferred ketone vii (12 g, 27.6 mmol) in a small portion of dry THF. The reaction was aged 15 minutes then warmed to ambient temperature, quenched with sodium bicarbonate solution and partitioned between water and hexanes. The organics were dried over sodium sulfate, filtered and evaporated in vacuo to give amide xi LC/MS (M+H)=505.6.

Amide xi (7 g, 13.9 mmol) was treated with L-Selectride (55.6 mmol. 55.6 mL of a 1M solution) in a microwave vial. The reaction was sealed and irradiated in a microwave reaction set at 70° C. for 16 hours. The reaction was then diluted with dichloromethane and quenched by careful addi tion of sodium perborate solid until effervescence stopped. The solids were filtered and the filtrate evaporated in vacuo to give xii. LC/MS $(M+H)=507.6$.

Compound Xiv was carried on to final compound 19 as outlined for compound 1 above, LC/MS (M+H)=520.8.

Compound 20 is DLinkC2DMA as described in Nature Biotechnology, 2010, 28, 172-176, WO 2010/042877 A1, WO 2010/048536 A2, WO 2010/088537 A2, and WO 2009/ 5 127O6O A1.

higher pH with a mixing ratio in the range of 1:1 to 1:3 vol:vol but targeting 1:2 vol:vol. This buffered solution is at a temperature in the range of 15-40°C., targeting 30-40°C. The mixed LNPs are held from 30 minutes to 2 hrs prior to an anion exchange filtration step. The temperature during incubating is in the range of 15-40°C., targeting 30-40°C.

Compound 21 is MC3 as described in WO 2010/054401, and WO 2010/144740 A1.

After incubating the solution is filtered through a 0.8 um filter containing an anion exchange separation step. This

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LNP Compositions

The following lipid nanoparticle compositions (LNPs) of the instant invention are useful for the delivery of oligo- $30₁₀$ nucleotides, specifically siRNA and miRNA:

Cationic Lipid/Cholesterol/PEG-DMG 56.6/38/5.4:

Cationic Lipid/Cholesterol/PEG-DMG 60/38/2:

Cationic Lipid/Cholesterol/PEG-DMG 67.3/29/3.7:

Cationic lipid/Cholesterol/PEG-DMG 49.3/47/3.7:

Cationic Lipid/Cholesterol/PEG-DMG 50.3/44.3/5.4;

Cationic Lipid/Cholesterol/PEG-C-DMA/DSPC 40/48/2/ 10:

Cationic Lipid/Cholesterol/PEG-DMG/DSPC 40/48/2/10: and

Cationic Lipid/Cholesterol/PEG-DMG/DSPC 58/30/2/10. LNP Process Description:

The Lipid Nano-Particles (LNP) are prepared by an impinging jet process. The particles are formed by mixing lipids dissolved in alcohol with siRNA dissolved in a citrate 45 buffer. The mixing ratio of lipids to siRNA are targeted at 45-55% lipid and 65-45% siRNA. The lipid solution contains a novel cationic lipid of the instant invention, a helper lipid (cholesterol), PEG (e.g. PEG-C-DMA, PEG-DMG) lipid, and DSPC at a concentration of 5-15 mg/ml with a 50 target of 9-1.2 mg/mL in an alcohol, (for example ethanol). The ratio of the lipids has a mole percent range of 25-98 for the cationic lipid with a target of 35-65, the helper lipid has a mole percent range from 0-75 with a target of 30-50, the PEG lipid has a mole percent range from 1-1.5 with a target 55 of 1-6, and the DSPC has a mole percent range of 0-15 with a target of 0-12. The siRNA solution contains one or more siRNA sequences at a concentration range from 0.3 to 1.0 mg/mL with a target of 0.3-0.9 mg/mL in a sodium citrate buffered salt solution with pH in the range of 3.5-5. The two 60 liquids are heated to a temperature in the range of $15-40^{\circ}$ C., targeting $30-40^{\circ}$ C., and then mixed in an impinging jet mixer instantly forming the LNP. The teelD has a range from 0.25 to 1.0 mm and a total flow rate from 10-600 mL/min. The combination of flow rate and tubing ID has effect of 65 controlling the particle size of the LNPs between 30 and 200 nm. The solution is then mixed with a buffered solution at a

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in the mage of 1:1 to 1:3

LCMS (M+H)=2008.

May have prime with a mixing antis in the range of 1:1 to 1:3

CMS (M+H)=2008.

May be a large and EV of the base of the strength of the strength of the streng process uses tubing IDS ranging from 1 mm ID to 5 mm ID and a flow rate from 10 to 2000 mL/min. The LNPs are concentrated and diafiltered via an ultrafiltration process where the alcohol is removed and the citrate buffer is exchanged for the final buffer solution such as phosphate buffered saline. The ultrafiltration process uses a tangential flow filtration format (TFF). This process uses a membrane nominal molecular weight cutoff range from 30-500 KD. The membrane format can be hollow fiber or flat sheet cassette. The TFF processes with the proper molecular weight cutoff retains the LNP in the retentate and the filtrate or permeate contains the alcohol; citrate buffer; final buffer wastes. The TFF process Is a multiple step process with an initial concentration to a siRNA concentration of 1-3 mg/mL. Following concentration, the LNPS solution is dia filtered against the final buffer for 10-20 volumes to remove the alcohol and perform buffer exchange. The material is then concentrated an additional 1-3 fold. The final steps of the LNP process are to sterile filter the concentrated LNP solution and vial the product.

Analytical Procedure:

1) siRNA Concentration
The siRNA duplex concentrations are determined by Strong Anion-Exchange High-Performance Liquid Chromatography (SAX-HPLC) using Waters 2695 Alliance system (Water Corporation, Milford Mass.) with a 2996 PDA detec tor. The LNPs, otherwise referred to as RNAi Delivery Vehicles (RDVs), are treated with 0.5% Triton X-100 to free total siRNA and analyzed by SAX separation using a Dionex BioLC DNAPac PA 200 (4x250 mm) column with UV detection at 254 nm. Mobile phase is composed of A: 25 mM NaClO₄, 10 mM Tris, 20% EtOH, pH 7.0 and B: 250 mM NaClO₄, 10 mM Tris, 20% EtOH, pH 7.0 with liner gradient from 0-15 min and flow rate of 1 ml/min. The siRNA amount is determined by comparing to the siRNA standard curve. 2) Encapsulation Rate

Fluorescence reagent SYBR Gold is employed for RNA quantitation to monitor the encapsulation rate of RDVs. RDVs with or without Triton X-100 are used to determine the free siRNA and total siRNA amount. Tire assay is performed using a SpectraMax M5e microplate spectropho

tometer from Molecular Devices (Sunnyvale, Calif.).
Samples are excited at 485 nm and fluorescence emission was measured at 530 nm. The siRNA amount is determined by comparing to the siRNA standard curve.

Encapsulation rate=(1=free siRNA/total siRNA)x 100%

3) Particle Size and Polydispersity

RDVs containing 1 ug siRNA are diluted to a final volume of 3 ml with 1×PBS. The particle size and polydispersity of $_{10}$ the samples is measured by a dynamic light scattering method using ZetaPALS instrument (Brookhaven Instru ments Corporation, Holtsville, N.Y.). The scattered intensity is measured with He—Ne laser at 25° C. with a scattering angle of 90° . 15

4) Zeta Potential Analysis

RDVs containing 1 ug siRNA are diluted to a final volume of 2 ml with 1 mM Tris buffer (pH 7.4). Electrophoretic mobility of samples is determined using ZetaPALS instru ment (Brookhaven Instruments Corporation, Holtsville, N.Y.) with electrode and He—Ne laser as a light source. The Smoluchowski limit is assumed in the calculation of Zeta potentials.

5) Lipid Analysis

 μ individual lipid concentrations are determined by Reverse μ_{25} Phase High-Performance liquid Chromatography (RP HPLC) using Waters 2695 Alliance system (Water Corpo ration, Milford Mass.) with a Corona charged aerosol detector (CAD) (ESA Biosciences, Inc, Chelmsford, Mass.). tor (CAD) (ESA Biosciences, Inc, Chelmsford, Mass.). Individual lipids in RDVs are analyzed, using an Agilent 30 Zorbax SB-CIS (50x4.6 mm, 1.8 um particle size) column with CAD at 60° C. The mobile phase is composed of A: 0.1% TFA in H_2O and B: 0.1% TFA in IPA. The gradient changes from 60% mobile phase A and 40% mobile phase B from time 0 to 40% mobile phase A and 60% mobile phase 35 B at 1.00 min; 40% mobile phase A and 60% mobile phase B from 1.00 to 5.00 min; 40% mobile phase A and 60% mobile phase B from 5.00 minto 25% mobile phase A and 75% mobile phase B at 10.00 min; 25% mobile phase A and 75% mobile phase B from 10.00 min to 5% mobile phase A $_{40}$ and 95% mobile phase B at 15.00 min; and 5% mobile phase Aand 95% mobile phase B from 15.00 to 60% mobile phase A and 40% mobile phase B at 20.00 min with flow rate of 1 ml/min. The individual lipid concentration is determined by comparing to the standard curve with all the lipid $_{45}$ components in the RDVs with a quadratic curve fit. The molar percentage of each lipid is calculated based on its

Utilizing the above described LNP process, specific LNPs with the following ratios were identified:
Nominal Composition:

Cationic Lipid/Cholesterol/PEG-DMG 60/38/2

Cationic Lipid/Cholesterol/PEG-DMG/DSPC 58/30/2/10

 $AGU - 2'$ OCH,

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Nominal Composition Cationic Lipid/Cholesterol/PEG-DMG 60/38/2 Cationic Lipid/Cholesterol/PEG-DMG/DSPC 40/48/2/10 Cationic Lipid/Cholesterol/PEG-DMG/DSPC 58/30/2/10

ApoB siRNA

- $-iB-CUUUAACAAUUCCUGAAAUTST-iB-3'$ (SEQ ID NO.: 3)
- 3'-UsUGAAAUUGUUAAGGACUSUSUSA-5' (SEQ ID NO.: 4)

AUGC - Ribose

iB - Inverted deoxy abasic

UC - 2 " Fluoro

- AGT 2' Deoxy
- AGU 2' OCH

USA - phophorothioate linkage

Example 1

Mouse In Vivo Evaluation of Efficacy

50 55 60 LNPS utilizing Compounds 1-12, in the nominal compo sitions described immediately above, were evaluated for in vivo efficacy. The siRNA targets the mRNA transcript for the firefly (Photinus pyralis) luciferase gene (Accession $#M15077$). The primary sequence and chemical modification pattern of the luciferase siRNA is displayed above. The in vivo luciferase model employs a transgenic mouse in which the firefly luciferase coding sequence is present in all cells. ROSA26-LoxP-Stop-LoxP-Luc (LSL-Luc) transgenic mice licensed from the Dana Farber Cancer Institute are Induced to express the Luciferase gene by first removing the LSL sequence with a recombinant Ad-Cre virus (Vector Biolabs). Due to the organotropic nature of the virus, expres sion is limited to the liver when delivered via tail vein
injection. Luciferase expression levels in liver are quantitated by measuring light output, using an IVIS imager (Xenogen) following administration of the luciferin sub strate (Caliper Life Sciences). Pre-dose luminescence levels are measured prior to administration of the RDVs. Luciferin in PBS (15 mg/mL) is intraperitoneally (IP) injected in a volume of 150 uL. After a four minute incubation period mice are anesthetized with isoflurane and placed in the IVIS imager. The RDVs (containing siRNA) in PBS vehicle were tail vein injected n a volume of 0.2 mL. Final dose levels ranged from 0.1 to 0.5 mg/kg siRNA. PBS vehicle alone was dosed as a control. Mice were imaged 48 hours post dose using the method described above. Changes in luciferin light output directly correlate with luciferase mRNA levels and represent an indirect measure of luciferase siRNA activity. In vivo efficacy results are expressed as % inhibition of luminescence relative to pre-dose luminescence levels. Sys temic administration of the luciferase siRNA RDVs decreased luciferase expression in a dose dependant manner. Greater efficacy was observed in mice dosed with Com pound 1 containing RDVs than with the RDV containing the octyl-CLinDMA (OCD) cationic lipid (FIG. 1). OCD is known and described in WO2010/021865.

Example 2

65 Rat In Vivo Evaluation of Efficacy and Toxicity LNPS utilizing compounds in the nominal compositions described above, were evaluated for in vivo efficacy and increases in alanine amino transferase and aspartate amino transferase in Sprague-Dawley (Crl:CD(SD) female rats (Charles River Labs). The siRNA targets the mRNA tran script for the ApoB gene (Accession #NM 019287). The primary sequence and chemical modification pattern of the 5 ApoB siRNA is displayed above. The RDVs (containing siRNA) in PBS vehicle were tail vein injected in a volume of 1 to 1.5 ml. Infusion rate is approximately 3 ml/min. Five rats were used in each dosing group. After LNP adminis tration, rats are placed in cages with normal diet and water 10 present. Six hours post dose, food is removed from the cages. Animal necropsy is performed 24 hours after LNP dosing. Rats are anesthetized under isoflurane for 5 minutes, then maintained under anesthesia, by placing them in nose cones continuing the delivery of isoflurane until ex-sangui- 15 nation is completed. Blood is collected from the vena cava using a 23 gauge butterfly Venipuncture set and aliquoted to serum separator vacutainers for serum chemistry analysis. Punches of the excised caudate liver lobe are taken and placed in RNALater (Ambion) for mRNA analysis. Pre served liver tissue was homogenized and total RNA isolated using a Qiagen bead mill and the Qiagen miRNA-Easy RNA isolation kit following the manufacturer's instructions. Liver ApoB mRNA levels were determined by quantitative RT PCR. Message was amplified from purified RNA utilizing a 25 rat ApoB commercial probe set (Applied Biosystems Cat #RN01499054_ml). The PCR reaction was performed on an ABI 7500 instrument with a 96-well Fast Block. The ApoB mRNA level is normalized to the housekeeping PPIB (NM 011149) mRNA. PPIB mRNA levels were determined by 30 RT-PCR using a commercial probe set (Applied Biosytems Cat. No. Mm00478295_ml). Results are expressed as a ratio of ApoB mRNA/PPIB mRNA. All mRNA data is expressed relative to the PBS control dose. Serum ALT and AST analysis were performed on the Siemens Advia 1800 Clini-35 cal Chemistry Analyzer utilizing the Siemens alanine ami notransferase (Cat#03039631) and aspartate aminotransferase (Cat#03039631) reagents. Similar efficacy was observed in rats dosed with Compound 1 containing RDV DLinKC2DMA (Compound 20) or MC3 (Compound 21, FIG. 2). Additionally, 3 out of 4 rats treated with 3 mg/kg DLinkC2DMA (Compound 20) failed to survive 48 hours and 2 out of 4 rats treated with 3 mg/kg MC3 (Compound 21) failed to survive 48 hours, 1 out of 4 rats treated with 10 45 mg/kg Compound 1 Survived at 48 hours post dose. than with the RDV containing the cationic lipid 40

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Example 3

Determination of Cationic Lipid Levels in Rat Liver

Liver tissue was weighed into 20-ml vials and homog enized in 9 v/w of water using a GenoGrinder 2000 (OPS Diagnostics, 1600 strokes/min, 5 min). A 50 uL aliquot, of each tissue homogenate was mixed with 300 uL of extrac tion/protein precipitating solvent (50/50 acetonitrile/metha nol containing 500 nM internal standard) and the plate was centrifuged to sediment precipitated protein. A Volume of 200 uL of each supernatant was then transferred to separate wells of a 96-well plate and 10 ul samples were directly analyzed by LC/MS-MS.

Standards were prepared by spiking known amounts of a methanol stock solution of compound into untreated rat liver homogenate (9 vol water/weight liver). Aliquots (50 uL) each standard/liver homogenate was mixed with 300 uL of extraction/protein precipitating solvent (50/50 acetonitrile/ methanol containing 500 nM internal standard) and the plate was centrifuged to sediment precipitated protein. A volume of 200 uL of each supernatant was transferred to separate wells of a 96-well plate and 10 μ l of each standard was directly analyzed by LC/MS-MS.

Absolute quantification versus standards prepared and extracted from liver homogenate was performed using an Aria LX-2 HPLC system (Thermo Scientific) coupled to an API 4000 triple quadrupole mass spectrometer (Applied Biosystems). For each run, a total of 10 uL sample was injected onto a RDS Hypersil C8 HPLC column (Thermo, 50×2 mm, 3 μ m) at ambient temperature.

Mobile Phase A:

95% $H₂O/5%$ methanol/10 mM ammonium formate/0.1% formic acid Mobile Phase B: 40% methanol/60% n-propa nol/10 mMammonium formate/0.1% formic acid. The flow rate was 0.5 mL/min and gradient elution profile was as follows: hold at 80% A for 0.25 min, linear ramp to 100% B over 1.6 min, hold at 100% B for 2.5 min, then return and hold at 80% A for 1.75 min. Total runtime was 5.8 min. API 4000 source parameters were CAD: 4, CUR: 15, GS1: 65, GS2:35, IS: 4000, TEM: 550, CXP: 15, DP: 60, EP: 10. In rats dosed with Compound 1 containing RDV liver levels were lower than with the RDV containing the cationic lipid DLinKC2DMA (Compound 20) or MC3 (Compound 21, FIG. 3).

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What is claimed is: 1. A cationic lipid of Formula A: wherein:

- R^1 and R^2 are independently selected from H, (C_1-C_6) alkyl, heterocyclyl, and polyamine, wherein said alkyl, heterocyclyl and polyamine are optionally substituted with one to three substituents selected from R', or R^1 55 and $R²$ can be taken together with the nitrogen to which they are attached to form a monocyclic heterocycle with 4-7 members optionally containing, in addition to the nitrogen, one or two additional heteroatoms selected from N, O and S, said monocyclic heterocycle 60 is optionally substituted with one to three substituents selected from R';
- $R³$ is selected from H and (C_1-C_6) alkyl, said alkyl optionally substituted with one to three substituents selected from R':
- R" is independently selected from halogen, R", OR". SR". CN, $CO₂R"$ and $CON(R")₂$;

R" is independently selected from H and (C_1-C_6) alkyl, wherein said alkyl is optionally substituted with halo gen and OH:

n is 0, 1, 2, 3, 4 or 5; and one of L_1 and L_2 is

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65 and the other is C_3-C_{24} alkyl and C_3-C_{24} alkenyl, said alkyl and alkenyl are optionally substituted with one or more substituents selected from R';

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or any pharmaceutically acceptable salt or stereoisomer thereof.

2. The cationic lipid of claim 1, wherein R_1 and R_2 are independently selected from the group consisting of H, methyl, ethyl and propyl.

3. The cationic lipid of claim 2, wherein R_1 and R_2 each are methyl.

4. The cationic lipid of claim 3, wherein R_3 is H or methyl.

5. The cationic lipid of claim 4, wherein R_3 is H.

6. The cationic lipid of claim 1, wherein n is 0, 1 or 2. 10

7. The cationic lipid of claim 6, wherein n is 0 or 1.

8. The cationic lipid of claim 7, wherein n is 0.

9. Alipid nanoparticle comprising a cationic lipid of claim 1.

10. The lipid nanoparticle of claim 9, wherein the lipid 15 nanoparticle further comprises an oligonucleotide.

11. The lipid nanoparticle of claim 10, wherein the oligonucleotide is an siRNA or miRNA.

12. The lipid nanoparticle of claim 11, wherein the oligonucleotide is an siRNA. 2O

13. The lipid nanoparticle of claim 9, wherein the lipid nanoparticle further comprises cholesterol and PEG-DMG.

14. The lipid nanoparticle of claim 9, wherein the lipid nanoparticle further comprises cholesterol, PEG-DMG and $DSPC.$ 25

15. The lipid nanoparticle of claim 9, wherein the lipid nanoparticle further comprises cholesterol, PEG-C-DMA and DSPC.

> $\frac{1}{2}$ \mathbf{r} \rightarrow